[I-1]

Surface Analysis of Sulfur-Treated GaAs

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Sulfur treatments have recently been reported which electrically passivate GaAs surfaces. From the variety of experimental results reported on sulfur-treated GaAs surfaces, it is clear that the treatments can improve the surface electronic properties. This paper presents the results of an investigation of the chemistry of GaAs surfaces after various sulfur treatments including ammonium sulfide and sodium sulfide liquid solutions and hydrogen sulfide gas. Auger electron spectroscopy(AES) and x-ray photoelectron spectroscopy(XPS) have been used to determine the surface chemical composition and bonding. Different wet chemical etches and substrate heating temperatures were used in order to vary the chemical compostion of the GaAs surface from Ga-rich to As-rich. The As/Ga ratio ranged from 0.48 to 5.98.

Ammonium sulfide and sodium sulfide solutions etch the GaAs leaving near-stoichiometric or slightly As-rich surfaces free of elemental As or oxides. Similar surfaces result from etching with NH₄OH and exposing to H₂S. Samples etched with HCl and H₂SO₄: H₂O₂: H₂O and exposed to H₂S are As-rich due to the presence of elemental As. The amount of the elemental As decreases rapidly with temperature and is undetectable for temperatures above 200°C in a H₂S ambient.

Sulfur bonds to both Ga and As surface atoms, whichever are available as a result of substrate processing. Only one bonding state is observed for either Ga-S or As-S bonding independent of the sulfur treatment process. In addition, sulfur bonds to both elemental As and substrate As with the same binding energy and similar concentration. For most samples, the sulfur coverage is less than a monolayer although samples heated at 350°C in a H₂S ambient show more than a monolayer sulfur coverage.

References

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