

(EPON 828 grade) with epoxy equivalent value of 184. The curing agent and a reactive additive were MDA and SN, respectively. The loading of Al_2O_3 treated with silane coupling agent was from 0 to 10 phr.

2-2 Specimen Preparation

The mixture of DGEBA, MDA, SN and Al_2O_3 was degassed and mechanically stirred at 80°C. The reacting mixture at gel state was cast into the pre-heated mold coated with releasing agent at 80°C. The system was cured at 80°C for 1.5 hr, firstly, and at 150°C for 1.0 hr, secondly[3]. The thermally cured specimen at higher temperature than T_g was annealed to the room temperature. It was post cured at 150°C at the convection oven for 4 days.

2-3. Thermal Analysis

T_g was measured from the DSC curves of cured and post-cured alumina filled DGEBA/MDA/SN system for 4 days at 150°C. Initial decomposition temperature (T_d) and most rapidly decomposing temperature (T_p) were detected by using TGA at the heating rates of 5, 10, 15, 20 °C/min. The stream rate of ambient nitrogen gas was 80 ml/min[4].

3. Results and Discussion

3-1. DSC Analysis

Fig. 1 shows the DSC thermograms of DGEBA/MDA/SN- Al_2O_3 composite system at the heating rate of 10 °C/min. The glass transition

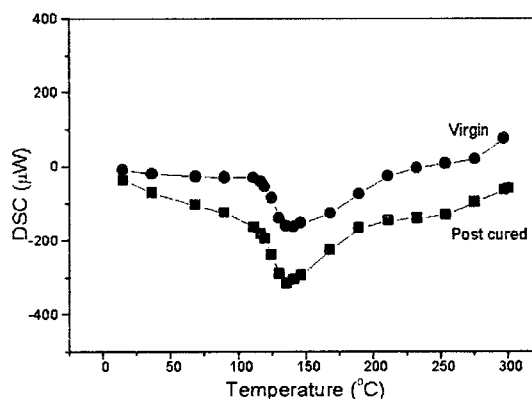


Fig. 1. Effects of post cure on DSC curves of DGEBA/MDA/SN- Al_2O_3 system.

temperatures (T_g) of virgin and post-cured system for 4 days at 150°C were 121 and 124°C, respectively. Under the post curing temperature of 150°C, T_g was not changed apparently. But at higher temperature the value will be increased as observed at the literature[5].

3-2. TG Analysis

The TG curves of the composite at the heating rates of 5, 10, 15 and 20 °C/min are shown in Fig. 2. On the TG thermograms, the weight of the sample is almost stable below 325°C depending on the heating rates and above the temperature, the weight loss rate increases rapidly during the short temperature range due to the removal of low molecular weight components produced by the pyrolysis of the polymeric main chain at high temperature. The thermograms shift toward higher temperature as

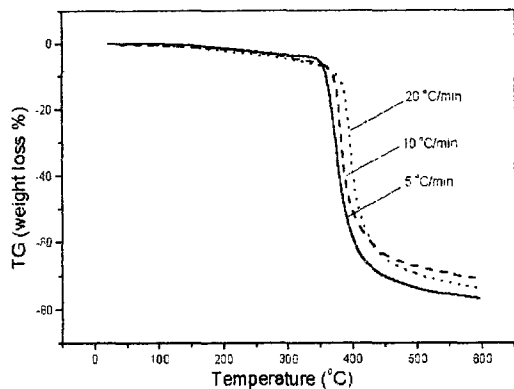


Fig. 2. Effects of heating rates on TG thermograms of DGEBA/MDA/SN system filled with Al_2O_3 .

the heating rate increased. The initial decomposing temperature (T_d) increased with the increment of heating rate. These values at 5, 10, 15 and 20 $^{\circ}C/min$ were 359.7, 372.4, 378.1 and 382.5, respectively.

3-3. DTG Analysis

Fig. 3 shows the DTG thermograms of DGEBA/MDA/SN- Al_2O_3 at various heating rates. As the heating rate increased the thermograms shift toward right and the peak temperature where the polymeric insulator was decomposed most rapidly increased. On the DTG curve, it is clear that the sample degraded in one stage and the peak temperature of the curve increased with the increment of heating rate. These values at 5, 10, 15 and 20 $^{\circ}C/min$ were 365.1, 377.2, 389.4, and 392.1, respectively. On the DTG curves, one stage weight loss curve was observed.

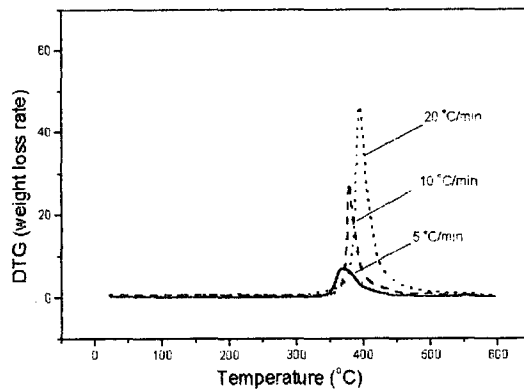


Fig. 3. Effects of heating rates on DTG thermograms of DGEBA/MDA/SN system filled with Al_2O_3 .

3-3. Post-curing Time on TG and DTG

Fig. 4. shows the effects of post curing condition on thermal decomposing temperature. As the heating rates increased the initial decomposing temperature and most decomposing temperature (T_p) increased.

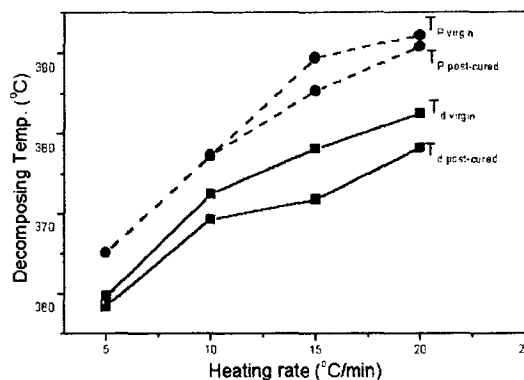


Fig. 4. Effects of post curing period on thermal properties of DGEBA/MDA/SN system filled with Al_2O_3 .

To increase the thermal stability of Al₂O₃ filled epoxy resin system for the application of high power electric equipment under high thermal stress, we are concentrating on the optimum curing condition, mixing ratio and surface treatment of the filler[6]. The detail things will be presented next.

References

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