

# Cathodoluminescent properties of $Y_2O_2S:Eu$ phosphors synthesized by citric acid-gel method

Yoichiro Nakanishi

Research Institute of Electronics, Shizuoka University

3-5-1 Johoku, Hamamatsu 432-8011, Japan

Phone : +81-53-478-1346, E-mail : nakanishi@rie.shizuoka.ac.jp

## Abstract

$Y_2O_2S:Eu$  phosphors with fine particle have been synthesized by citric acid-gel method. In this method,  $Na_2S_2O_3$  was added to  $Y_2O_3:Eu$  precursor, then the mixture was fired at  $1000^\circ C$  for 3h in  $S_2$  atmosphere. When the  $Y_2O_3:Eu$  precursor was pre-annealed at  $400\sim 700^\circ C$  before the firing of the mixture, the fine particles with a diameter of around  $1\ \mu m$  were obtained after the firing. The phosphor pre-annealed at  $400^\circ C$  showed a luminance and CIE color coordinates of  $2350\ cd/m^2$  and  $(0.61, 0.37)$ , respectively, at under excitation of 3 kV and  $60\ \mu A/cm^2$ .

## 1. Introduction

Field emission displays (FEDs) are one of the future flat-panel display technologies [1,2]. At the low voltage excitation (lower than 5 kV), the phosphors which are used in CRTs can not show high luminance and efficiency in FEDs because of their high resistivity [3] and deterioration of the phosphor surface. Therefore, phosphors must have low resistivity to suppress the charging-up, moreover, they must have stable surface for high-density electron beam irradiation in order to suppress deterioration.

$Y_2O_3:Eu$  and  $Y_2O_2S:Eu$  are available as red emitting phosphors. A main emission of  $Y_2O_3:Eu$  phosphor is due to the transition of  $^5D_0 \rightarrow ^7F_2$  with a peak at about 611 nm, whose color purity is not good. Moreover, the band gap of  $Y_2O_3$  is about 6 eV, which takes place charging-up on the surface of the  $Y_2O_3:Eu$  phosphor. Whereas the peak wavelength of

$Y_2O_2S:Eu$  is about 630 nm due to  $^5D_0 \rightarrow ^7F_3$  transition, whose color coordinates is very close that of red point in NTSC. Moreover, the band gap of  $Y_2O_2S$  is about 4 eV which is smaller than  $Y_2O_3$ , although the energy gap is too large for low voltage FEDs.

Therefore, in this investigation, the synthesis of  $Y_2O_2S:Eu$  phosphors by citric acid-gel method was tried and the morphology, structural and luminescent properties the synthesized phosphors will be reported.

## 2. Experimental

Figure 1 shows a flow chart of the synthesis of

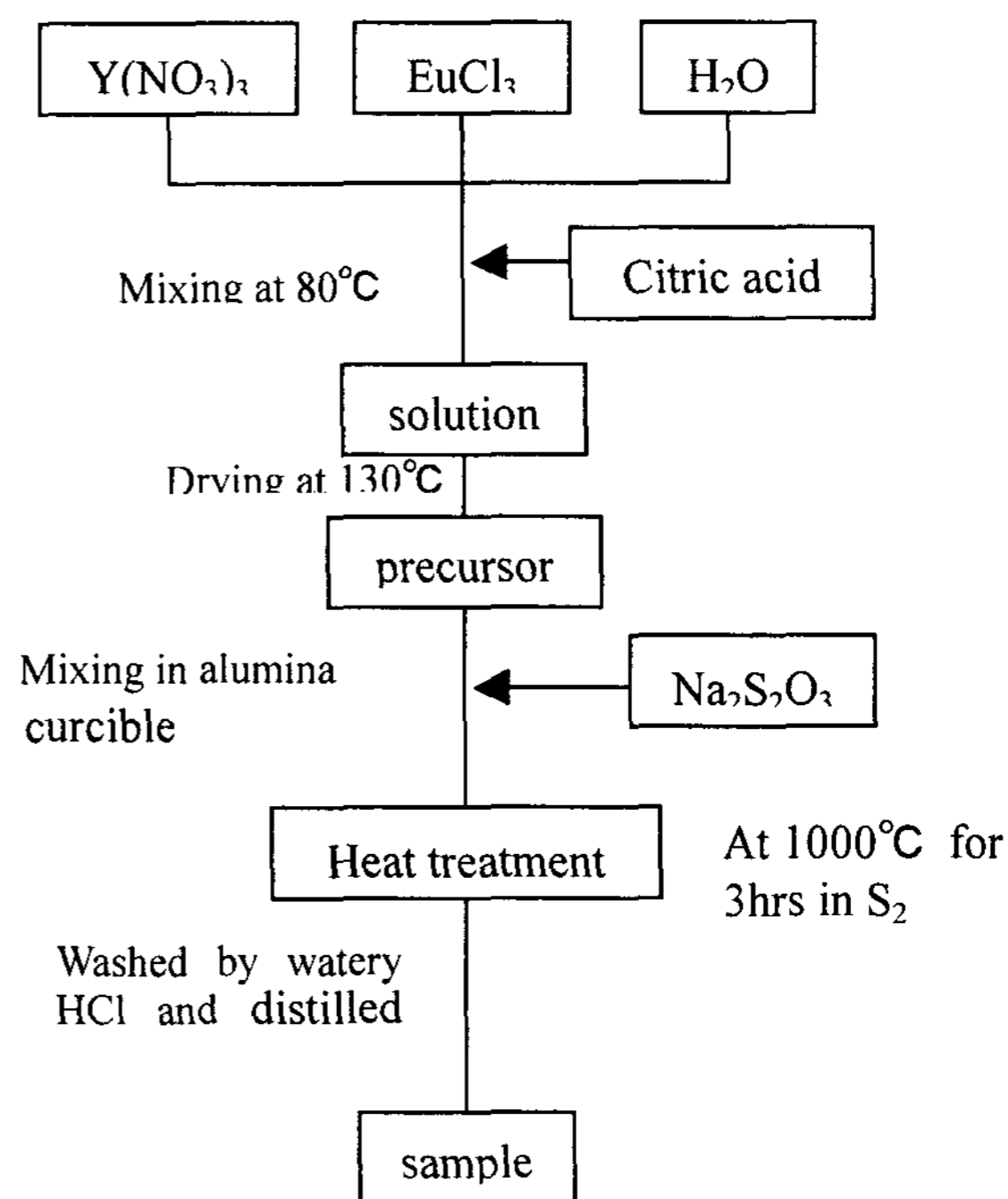


Fig.1 Flow chart of synthesis of phosphors by citric acid-gel method.

the  $Y_6O_2S:Eu$  phosphor by the citric acid-gel method. The citric acid is added to control reaction rate, as a result, the size of the synthesized phosphor can be controlled. A pre-annealing of the precursor was also tried before the addition of  $Na_2S_2O_3$  to the precursor. The pre-annealing was carried out at 400 to 700°C for 1 h in air.  $Na_2S_2O_3$  was mixed with the precursor with or without the pre-annealing, then the mixture was fired at 1000°C for 3h in  $S_2$  atmosphere. After the firing, the samples were washed in dilute HCl and distilled water in turn.

Structural properties and morphology of the phosphors were characterized by X-ray diffraction (XRD) measurement and SEM observation, respectively. The luminescent properties of the phosphors was characterized under excitation of electron beam of energy lower than 3 kV and 60  $\mu A/cm^2$ .

### 3. Results and discussion

#### 3.1 Synthesis of $Y_2O_2S:Eu$ phosphors

Figure 2 shows XRD curves of (a) precursor, (b) 700°C-annealed precursor, (c) phosphor fired the mixture without pre-annealing, (d) phosphor fired the mixture with pre-annealing at 400°C and (e) 700°C. The sample annealed the precursor at 1000°C for 1h in air shows  $Y_2O_3$  structure. It is seen from (c) to (e) that the samples fired the mixtures of the precursor and  $Na_2S_2O_3$  in  $S_2$  shows the formation of the  $Y_2O_2S$  without  $Y_2O_3$  phase regardless of the pre-annealing.

Figure 3 shows CL spectra of the  $Y_2O_3:Eu$  and  $Y_2O_2S:Eu$  phosphors corresponding to (b) and (c), respectively, in Fig.2, where the phosphors were excited with 3 kV and 60  $\mu A/cm^2$ . It is seen that each spectrum corresponds to emission spectrum of well known  $Y_2O_3:Eu$  and  $Y_2O_2S:Eu$  phosphors, respectively.

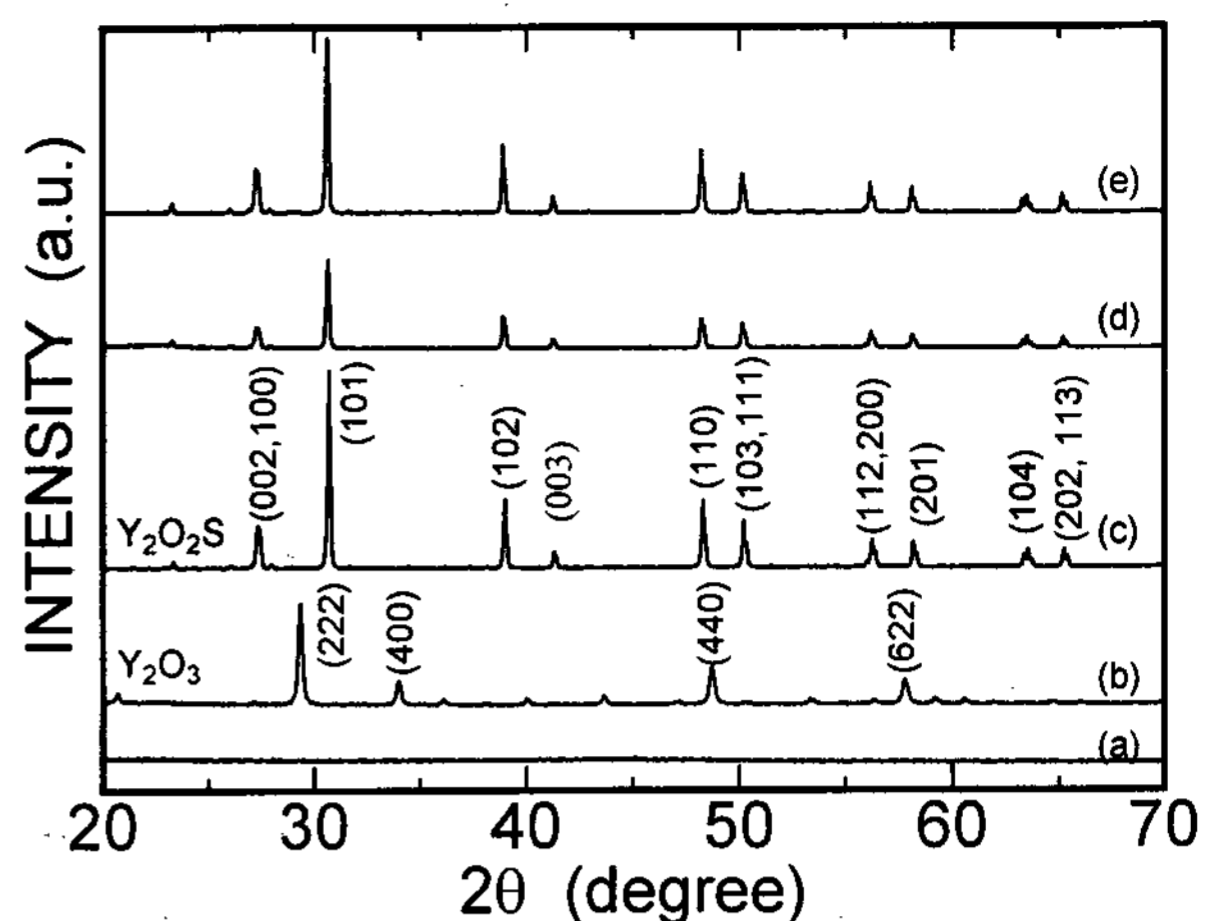


Fig.2 XRD curves of (a) precursor, (b) annealed at 1000°C without  $Na_2S_2O_3$ , (c) non pre-annealed, (d) pre-annealed at 400°C and (e) pre-annealed at 700°C.

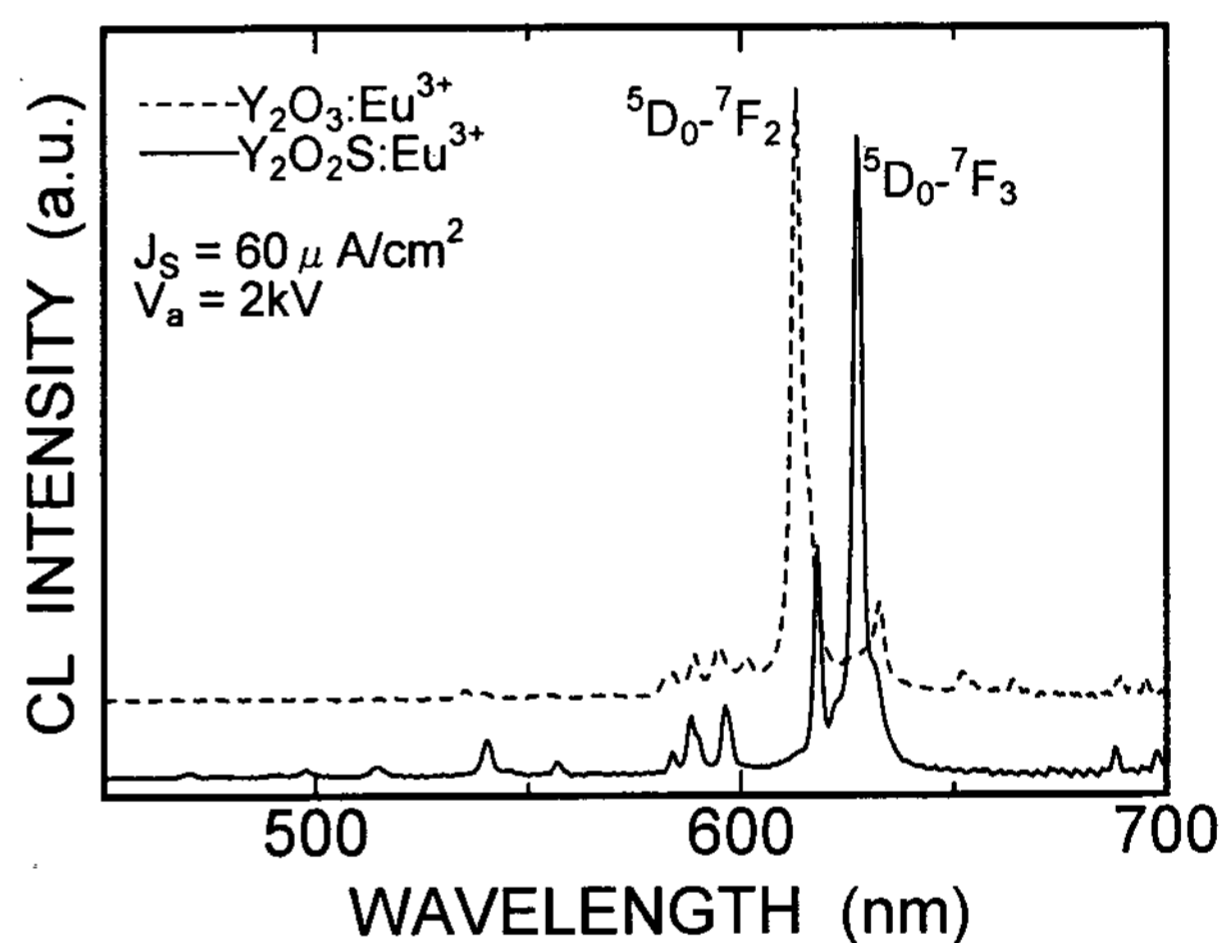


Fig.3 CL spectra of  $Y_2O_3:Eu$  and  $Y_2O_2S:Eu$  phosphors

These results show that  $Y_2O_2S:Eu$  phosphor could be successfully synthesized by the citric acid-gel method.

#### 3.2 Morphology of $Y_2O_2S:Eu$ phosphor

Figure 4 shows SEM photographs of  $Y_2O_2S:Eu$  phosphors (a) without pre-annealing, (b) with pre-annealing at 400°C and (c) with pre-annealing at 700°C. It is seen from (a) that the phosphors with grain size around or smaller than 5  $\mu m$  when no pre-annealing. On the other hand, it should be noticed that by the pre-annealing, the grain sizes of

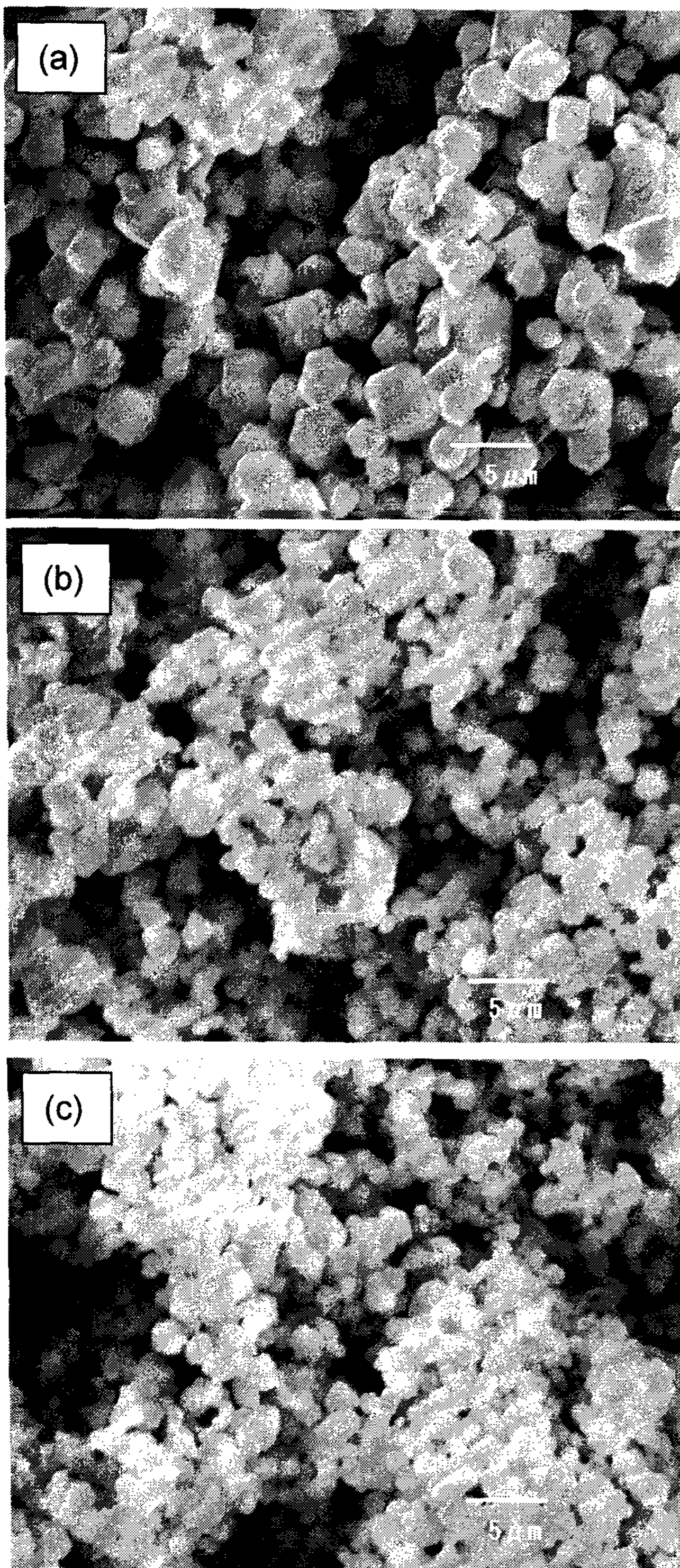


Fig.4 SEM photographs of  $Y_2O_2S:Eu$  phosphors (a) non pre-annealed, (b) pre-annealed at (b)  $400^\circ C$  and (c)  $700^\circ C$ .

the phosphor particles could be made small to around  $1 \mu m$  or less and became uniform.

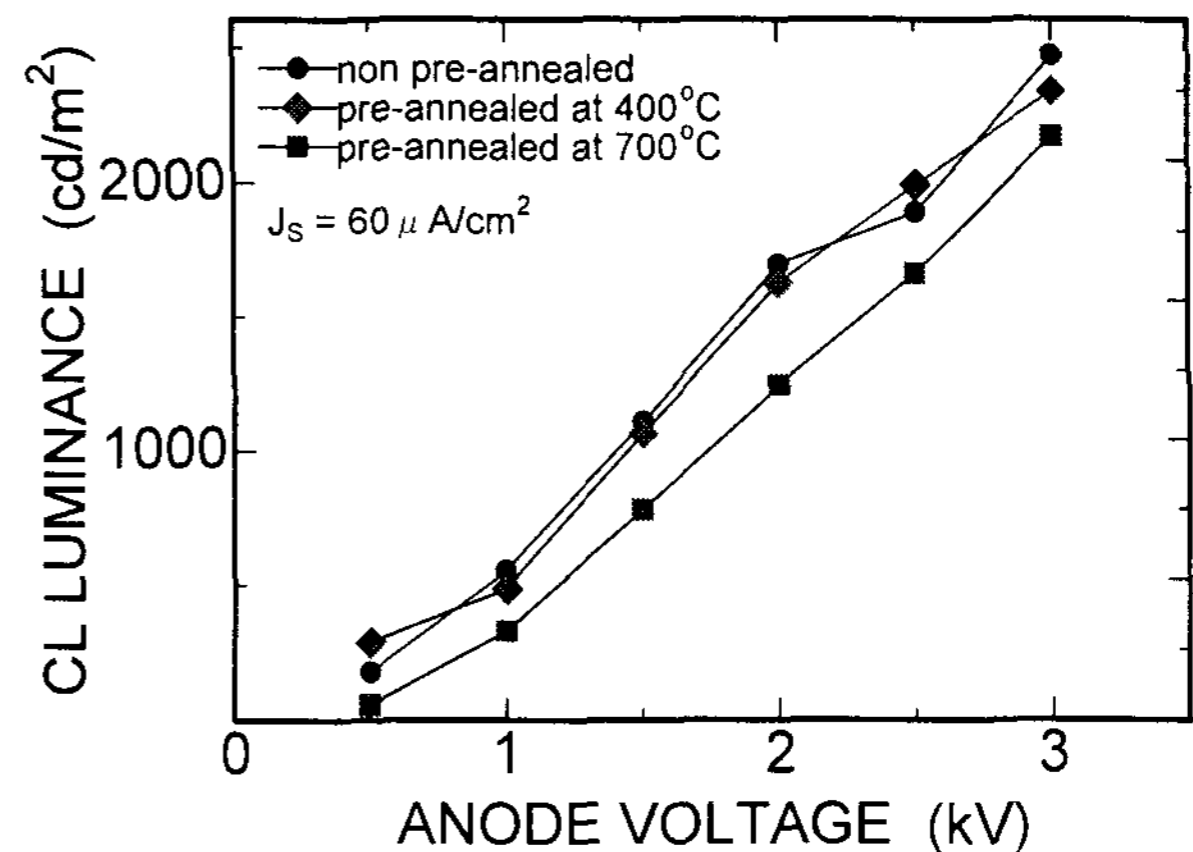


Fig.5 L-V characteristics of  $Y_2O_2S:Eu$  phosphors shown in Fig.4.

### 3.3 CL properties of $Y_2O_2S:Eu$ phosphors

Figure 5 shows CL luminance vs. anode voltage characteristics of  $Y_2O_2S:Eu$  phosphors shown in Fig.4. It is seen that almost same characteristics are obtained for non pre-annealing and pre-annealing at  $400^\circ C$ , and the maximum luminance is about  $2500 \text{ cd/m}^2$  under excitation at  $3 \text{ kV}$ ,  $60 \mu A/cm^2$ . Whereas the luminance of the phosphor pre-annealed at  $700^\circ C$  is a little lower over whole anode voltage.

The reason for this result might be due to the fine particle, however, the detail is not understood.

Figure 6 shows luminous efficiency vs. anode voltage characteristics of the phosphors shown in Fig.5. It is seen that the phosphors without the pre-annealing and with the pre-annealing at  $400^\circ C$  show luminous efficiency of about  $4 \text{ lm/W}$

The luminance and luminous efficiency obtained in this investigation is expected as the phosphor material for FEDs.

## 4. Summary

Red emitting  $Y_2O_2S:Eu$  phosphors were synthesized by citric acid-gel method. It was confirmed from the measurements of XRD and CL spectra that nearly perfect  $Y_2O_2S:Eu$  phosphors could be synthesized by the method. Moreover, it

was found that the fine particles with diameter of about 1  $\mu\text{m}$  or less could be synthesized. The phosphors without the pre-annealing and with pre-annealing at 400°C showed a luminance and luminous efficiency of about 2500  $\text{cd}/\text{m}^2$  and 4  $\text{lm}/\text{W}$ , respectively, under excitation at 3 kV and  $60\mu\text{A}/\text{cm}^2$ . The phosphors with these properties are promising for low voltage FEDs

**5. References**

- [1] R.O.Peterson, Extd. Abs. of 1<sup>st</sup> Int. Conf. on Sci. and Technol. of Display Phosphors p.11 (San Diego, CA, 1995).
- [2] S.Ito, H.Toki, F.Kataoka, K.Tamura and Y.Sato, Extd. Abs. of 3<sup>rd</sup> Int. Conf. on Sci. and Technol. of Display Phosphors p.275 (Huntington Beach, CA, 1997).
- [3] H.Kominami, T.Nakamura, Y.Nakanishi and Y.Hatanaka, Jpn. J. Appl. Phys. 35, 1600 (1996).

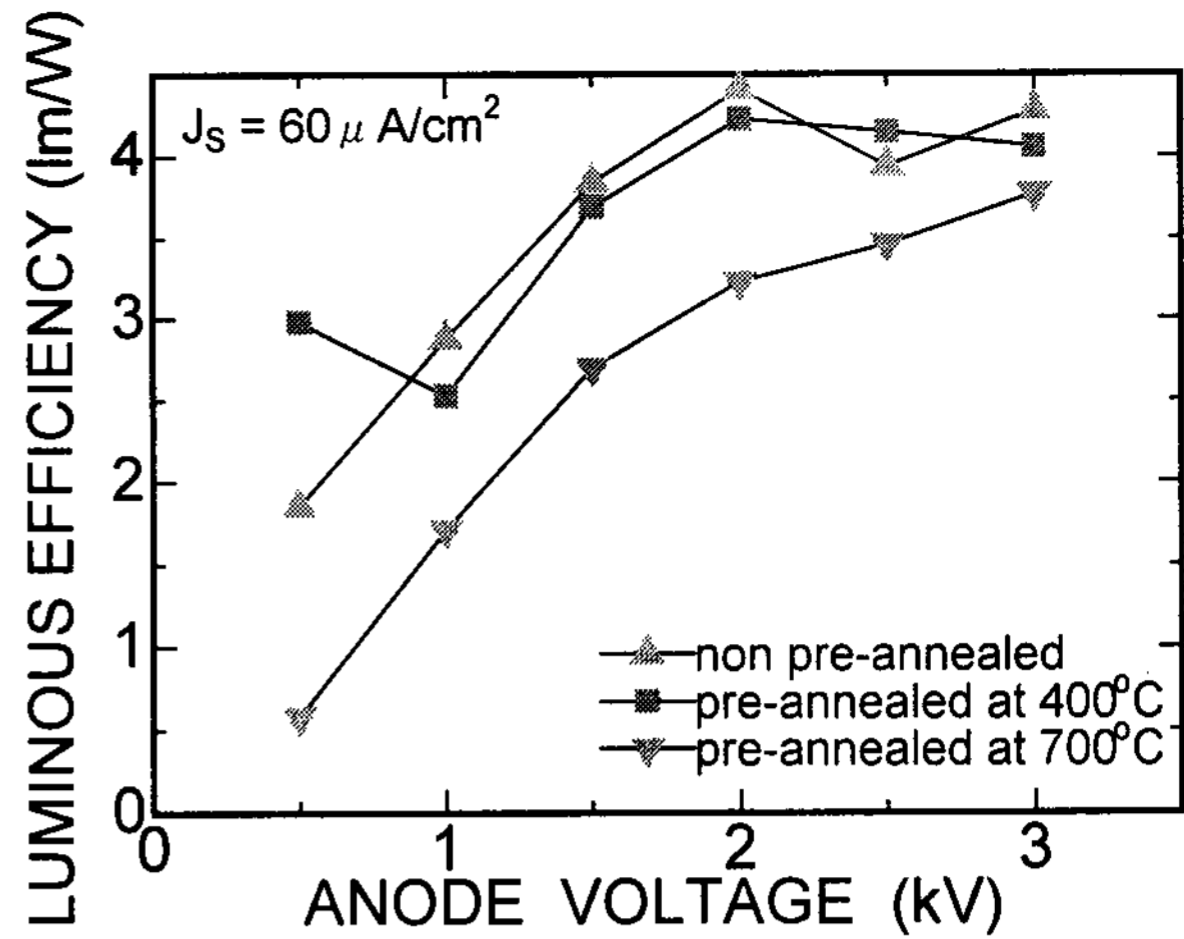


Fig.6  $\eta$ -V characteristics of  $\text{Y}_2\text{O}_2\text{S}:\text{Eu}$  phosphors shown in Fig.4.