## Development of Carbon Dioxide capture process using strongly basic Ionic liquid.

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For the purpose of developing advanced new absorbent for carbon dioxide, solubility of carbon dioxide using strongly basic ionic liquid, such as [bmim]phenolate, is studied with various temperatures and pressures ranges. [bmim]phenolate is easily synthesized using [bmim]Cl (C<sub>6</sub>H<sub>15</sub>ClN<sub>2</sub>) and Na-phenoxide ( $C_6H_5NaO\cdot 3H_2O$ ) with simple method and shows greater performance at capturing CO<sub>2</sub>. CO<sub>2</sub> solubility of [bmim]phenolate is measured with VLE (Vapor Liquid Equilibrium) apparatus, and its purity is assessed by <sup>1</sup>H-NMR spectra and <sup>1</sup>H-NMR spectra. Solubility data of carbon dioxide of some conventional ILs are also measured for comparism, and other properties of [bmim]phenolate are reported.

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