

Effects of Fe Ions Substitution on the Structure and Magnetic Properties of $\text{Ba}_{0.95}\text{Bi}_{0.05}\text{TiO}_3$

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1. Introduction

BaTiO_3 (BTO) is known as a classical ferroelectric material with Curie temperature (T_C) at 120°C . It belongs to tetragonal perovskite structure at room temperature and has a wide range of application such as capacitor, positive temperature coefficient of resistance thermostat, and piezo electric sensors [1]. It has been reported that the substitution of Bi^{3+} ion on Ba^{2+} site resulted an increase in dielectric constant and the T_C [1,2]. On comparison the other hand, transition metal ions doped BTO are reported to have a magnetic ordering [3]. However, the ferroelectricity of those materials is suppressed [4]. Based on those reports, we synthesized $\text{Ba}_{0.95}\text{Bi}_{0.05}\text{Ti}_{1-x}\text{Fe}_x\text{O}_3$ ($0 \leq x \leq 0.1$) in order to study the effects of Fe ions on the structure and magnetic properties of Bi doped BTO.

2. Experimental

Polycrystalline samples of $\text{Ba}_{0.95}\text{Bi}_{0.05}\text{Ti}_{1-x}\text{Fe}_x\text{O}_3$ ($0 \leq x \leq 0.1$) were prepared via solid-state reaction. The structure of samples was confirmed by powder x-ray diffraction (XRD) measurements using $\text{CuK}\alpha$ radiation. Isothermal magnetization measurements at room temperature were measured using VSM (Lakeshore 7300).

3. Results and Discussion

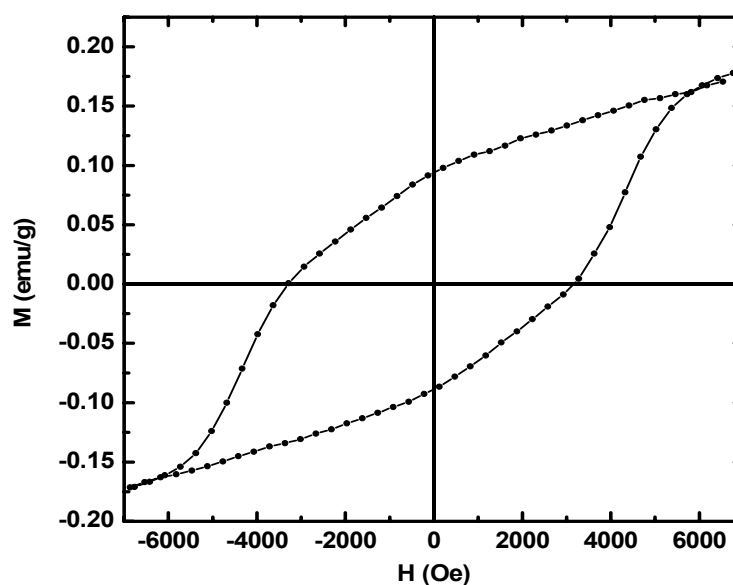


Fig. 1. The $M(H)$ hysteresis loops of $\text{Ba}_{0.95}\text{Bi}_{0.05}\text{Ti}_{0.9}\text{Fe}_{0.1}\text{O}_3$ at room temperature.

XRD data indicate that $\text{Ba}_{0.95}\text{Bi}_{0.05}\text{TiO}_3$ has a tetragonal structure, but for $x \geq 5\%$ of Fe ions substitution, the tetragonal structure is gradually merged in hexagonal structure. The large hysteresis of MH curve at room temperature for $\text{Ba}_{0.95}\text{Bi}_{0.05}\text{Ti}_{0.9}\text{Fe}_{0.1}\text{O}_3$ is shown in Fig. 1. The magnetic moments increase when the doping amount increases. It is believed that the magnetic properties come from the superexchange between pentahedral Fe^{3+} ions and also between pentahedral and octahedral Fe^{3+} ions[5].

4. References

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