# Phytochemical Study on Randia siamensis

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**Abstract** [] From the roots of *Randia siamensis*, n-mannitol, a mixture of  $\beta$ -sitosterol and campesterol, oleanolic acid acetate, oleanolic acid-3- $\alpha$ -L-arabinoside and mesembryanthemoidigenic acid as a sapogenin were isolated and characterized.

**Keywords** [] Randia siamensis, Rubiaceae, p-Mannitol, Sterol mixture, Oleanolic acid acetate, Oleanolic acid-3-α-L-arabinoside, Mesembryanthemoidigenic acid.

Randia siamensis (Rubiaceae) is an erect shrub distributed in Thailand, the roots of which have been used in a folkloric medicine for inducing abortion. In the course of search for the active substances, five compounds I-V were isolated and characterized.

On concentration of the MeOH extract of the roots, compound I, mp  $166-7^{\circ}$ ,  $[\alpha]_{D}^{23}=+10.16^{\circ}$   $(c=2.07, H_{2}O)$ , was crystallized, which was identified as p-mannitol by direct comparison with an authentic sample (mmp, co-TLC). Column chromatography of the CHCl<sub>3</sub> soluble fraction of the MeOH extract on silica gel using a solvent (CHCl<sub>3</sub> $\rightarrow$ EtOAc, gradient) to give compounds II, III & IV.

Compound II, mp  $138-40^{\circ}$ , was identified as a mixture of  $\beta$ -sitosterol (64.5%) and campesterol (35.5%) (MS, GLC).

Compound III, mp 248-50°,  $[\alpha]_D^{23} = +103.07^\circ$  (c=0.28, MeOH) showed positive Liebermann-Burchard test and strong absorption bands at

3200(OH), 1740, 1260 (acetate) and 1695 cm<sup>-1</sup> (acid) in its IR. Its MS showed a molecular ion peak at m/z 498 (0.8%) and other peaks at m/z 452 [M<sup>+</sup>-(COOH+H), 1.47], 438 (M<sup>+</sup>-CH<sub>3</sub>COOH, 4.39), 248 (RDA, 100) and 203 (RDA-COOH, 75.4). The <sup>1</sup>H NMR (60 MHz, CDCl<sub>3</sub>, TMS) exhibited signals at δ0.88–1.13 (7×Me), 2.03 (3H, s, OAc), 4.60 (IH, m, H-3), and 5.30 (IH, m, H-12). These spectral data were in agreement with those for the structure of oleanolic acid acetate. It was confirmed by deacetylation of III to give oleanolic acid, mp 309-10° (mmp, co-TLC).

Compound IV, mp  $242-4^{\circ}$ ,  $[\alpha]_D^{23} = +47.5^{\circ}$  (c=0.04, MeOH), gave positive reaction in Liebermann-Burchard and Molisch tests and showed strong absorption bands at 3400 (OH), 1700 (acid) and 1000-1100 cm<sup>-1</sup> (glycoside). Acid hydrolysis of IV gave oleanolic acid (mmp, co-TLC) and L-arabinose (co-TLC, GLC). Methanolysis of permethylated product of IV, mp  $170-2^{\circ}$  (M+ 644) gave methyl oleanolate, mp  $201-2^{\circ}$  (mmp, co-TLC) and methyl-2, 3, 4-tri-O-methyl-L-arabinopyranoside (GLC).

The  $\alpha$ -orientation of the glycosidic linkage was suggested not only from the J value (J=6Hz) of the anomeric proton signal of its peracetylated product, but also from the molecular rotation difference (+12.1°) between IV and oleanolic acid ([M]D of methyl- $\alpha$ -L-arabinopyranoside=+28.37°, that of  $\beta$ -form=

 $+402.62^{\circ})^{1)}$ . Therefore the structure of IV was elucidated as  $3-O-\alpha-L$ -arabinopyranosyl oleanolic acid, which is a rare natural compound. The presence of IV in the plants has previously been only reported in Fatsia japonica<sup>2)</sup> and Patrinia scabiosaefolia<sup>3)</sup>.

Acid hydrolysis of the BuOH soluble fraction of the MeOH extract, column chromatography on silica gel and elution with  $C_6H_6$ – $Et_2O$  (4:1) gave compound V, mp 335-8°,  $[\alpha]_D^{23}+47.2^\circ$  (c=0.18, MeOH) which showed positive Liebermann-Burchard test and strong absorption bands at 3100 (OH) and 1700 cm<sup>-1</sup> (acid) in its IR.

Methylation of V with  $CH_2N_2$  and subsequent acetylation with  $Ac_2O$  and pyridine gave a methylester, mp  $209-10^{\circ}$  and a methylester diacatate, mp  $234-7^{\circ}$ , respectively.

Oxidation of the methylester diacetate with SeO<sub>2</sub> yielded a heteroannular diene compound, which showed triple UV maxima at 241, 251, and 261 nm (log  $\varepsilon$ , 4.18, 4.23 and 4.15), typical of 11:12, 13:18 diene of the oleanane series<sup>41</sup>. The MS of V showed a molecular ion peak at m/z 472 (1.57%) and other peaks at m/z 454 (M<sup>+</sup>-H<sub>2</sub>O, 1.76), 441 (M<sup>+</sup>-CH<sub>2</sub>OH, 2.17), 264 (RDA, 72.4), 233 (RDA-CH<sub>2</sub>OH, 100), 201 [RDA-(COOH+H<sub>2</sub>O), 35.39], indicating that V has an  $\beta$ -amyrin skeleton with one hydroxyl group and one carboxyl group at rings D/E and one hydroxyl group at rings A/B<sup>51</sup>.

The appearance of the base peak at m/z 233 which corresponded the loss of a CH<sub>2</sub>OH group from RDA fragment with D/E ring and saponification rate (33%) of V methylester by 10% KOH in EtOH for 8 hr suggested that a CH<sub>2</sub>OH group preferred to be located at C-17<sup>5</sup>, 6).

However, the formation of a monobromo  $\gamma$ -lactone by treatment of V with Br<sub>2</sub> HOAc and the appearance in NMR of V mentylester diacetate of the highest angular methyl signal in

upfield region from 0.77 ppm<sup>7)</sup> strongly supported the precence of a COOH group at C-17 in V.

Treatment of V methylester with LiAlH<sub>4</sub> gave triol, mp 250-7° which was identified as a 28, 29-dihydroxy- $\beta$ -amyrin by direct comparison with an authentic sample (mmp, co-TLC). From the above results, the structure of V was elucidated as  $3\beta$ , 29-dihydroxy-olean-12-en-28-oic acid. A direct comparison (mmp, co-TLC and MS) with an authentic sample of mesembryanthemoidigenic acid<sup>8,9</sup>, kindly supplied by Dr. Daloze of Universite Libre de Bruxelles, confirmed the identity of these two terpenoids.

#### EXPERIMENTAL METHODS

#### Isolation

The roots of Randia siamensis were extracted with hot MeOH. Concentrating and cooling the hot MeOH extract gave a precipitate which was recrystallized from MeOH to give compound I (D-mannitol) as colorless needles, mp  $166-7^{\circ}$ ,  $[\alpha]_{\rm D}^{23} = +10.16^{\circ}$  (c=2.07, H<sub>2</sub>O) (mmp, co-TLC).; acetate, mp  $118-120^{\circ}$  (mmp, co-TLC, NMR).

The filterate after separation of I was partitioned between equal volumes of n-hexane and  $H_2O$ . The aqueous layer was extracted with  $CHCl_3$  and subsequently with BuOH.

The CHCl<sub>3</sub> extract was evaporated and chromatographed on a Si gel and eluted with CHCl<sub>3</sub> - EtOAc (gradient) to give compounds II, III and IV.

Compound II (a mixture of  $\beta$ -sitosterol and campesterol)

Colorless needles from MeOH, mp 138-40°, LB; positive (pink-blue), IR  $\nu_{\rm max}^{\rm KBr}$  cm<sup>-1</sup>; 3400 (OH), 790-840 (trisubstituted double bond); MS

m/z(rel. int.);414 (M<sub>1</sub><sup>+</sup>,100), 400 (M<sub>2</sub><sup>+</sup>, 37.8), 329 (M<sub>1</sub>-C<sub>5</sub>H<sub>9</sub>O, 100), 315 (M<sub>2</sub>-C<sub>5</sub>H<sub>9</sub>O, 27.1), 303 (M<sub>1</sub>-C<sub>7</sub>H<sub>11</sub>O, 47,6), 289 (M<sub>2</sub>-C<sub>7</sub>H<sub>11</sub>O, 20.4), 275 (M<sub>1</sub>-C<sub>9</sub>H<sub>15</sub>O, 22.7), 273 (M-side chain, 79.6), 261 (M<sub>2</sub>-C<sub>9</sub>H<sub>15</sub>O, 11.1), 255 (M-side chain-H<sub>2</sub>O, 76.9), which was found to consist of β-sitosterol (64.5%, T<sub>R</sub> 3.2 min) and campesterol (35.5%, T<sub>R</sub> 2.8 min) by GLC (column, 3% OV-I, 60-80 mesh, 4mm×1.5m; column temp., 270°; injector temp., 300°; N<sub>2</sub>, 45 ml/min).

Acetate, mp 118-20°, NMR (CDCl<sub>3</sub>, 60 MHz, TMS):  $\delta 0.67$ -1.0 (6×Me), 2.02 (s, 3H, OAc), 4.60 (1H, br s, H-3), 5.40 (m, lH, olefin). Compound III (oleanolic acid acetate)

Colorless needles from MeOH, mp 248  $50^{\circ}$   $[\alpha]_D^{23} = +103.07^{\circ} (c=0.28, MeOH)$ .

Compound IV (oleanolic acid-3- $\alpha$ -t-arabinoside) Colorless needles from MeOH, mp 242-1°,  $[\alpha]_D^{23} = +47.5^{\circ} (c=0.04, MeOH)$ .

# Deacetylation of III

A sample (10mg) of III was heated in 10% NH<sub>4</sub>OH in MeOH (5ml) for 5 hr. After concentration, it was crystallized from MeOH to yield colorless needles, mp 309-10°, identified as oleanolic acid (mmp, co-TLC),

### Acid Hydrolysis of IV

A solution of IV (8mg) in 5% H<sub>2</sub>SO<sub>4</sub> in EtOH (5ml) was refluxed for 5 hr. and concentrated under reduced pressure to remove EtOH. After addition of H<sub>2</sub>O, the resulting ppt. was filtered and crystallized from McOH to afford colorless needles, mp 309·10°, identified as oleanolic acid (mmp, co-TLC).

The filtrate was neutralized with BaCO<sub>3</sub>, filtered and concentrated under reduced pressure.

The residue was found to be L-arabinose by TLC(cellulose plate, pyridine-EtOAc-HOAc-H<sub>2</sub>O = 36:36:7:21, Rf 0.38) and GLC of the TMS derivative (column OV-I (3%), 60-80 mesh,

 $1.5 \times 4$ mm; column temp.,  $160^{\circ}$ ; injector temp.,  $180^{\circ}$ ;  $N_2$  45ml/min;  $T_R$  3.6).

Permethylation of IV Followed by Methanolysis Permethylation of IV (30mg) according to the method described by Brimacombe, et al.  $^{10}$  followed by purification by column chromatography with  $C_6H_6$ -Et<sub>2</sub>O-MeOH (8:2:0.5) and crystallization from MeOH yielded prisms, mp

170-2°, MS m/z(%): 644 (M+, 0.2), 453 (M-PM Ara. 61.6), 262 (RDA, 47.4), 203 (RDA-

COOCH<sub>3</sub>, 100), 175 (PM Ara. 33.4).

The permethylether (10mg) was refluxed in 5% HCl-MeOH (3ml) for 5hr. The reaction mixture was poured onto ice and filtered and the ppt was crystallized from MeOH to give methyloleanolate, mp 201-2° (mmp, co-TLC). The filtrate was extracted with CHCl<sub>3</sub>. The CHCl<sub>3</sub> solution was evaporated. The residue was found to be methyl-2, 3-4-tri-O-methyl-L-arabinopyranoside by GLC (column, 5% NPGS 4mm×1.5m; colum temp., 180°; injector temp., 200°; N<sub>2</sub>, 45ml/min; T<sub>R</sub> 1.8).

#### Acetylation of IV

A sample (30mg) of IV was acetylated with acetic anhydride and pyridine in the usual way. The reaction product was crystallized form MeOH to give colorless needles, mp 148-50°, NMR (CDCl<sub>3</sub>, 60 MHz, TMS):  $\delta$ 0.84-1.15 (7×Me), 2.05(6H, s, 2×OAc), 2.08(3H, s, OAc) 4.21 (1H, d, J=6, anomeric H).

# Compound V (mesembryanthemoidigenic acid)

Hydrolysis of the BuOH extract gave a mixture of sapogenins, which was chromatographed on a Si gel column and eluted with  $C_6H_6$ -Et<sub>2</sub>O (1:1) to give V as colorless needles from MeOH, mp 335-8°,  $[\alpha]_D^{23}+47.2^\circ$  (c=0.18, MeOH).

#### Methylation of V

A sample (30mg) of V was esterified when dissolved in MeOH and treated with CH<sub>2</sub>N<sub>2</sub>.

The product was crystallized from MeOH as colorless needles, mp 209-10°, IR  $\nu_{max}^{KBr}cm^{-1}$ :1725 (ester).

#### Acetylation of V-methylester

A sample (30mg) of V-methylester was acetylated as above and crystallized from MeOH to give colorless needles, mp 234-7°, IR  $\nu_{\rm max}^{\rm KBr}$  cm<sup>-1</sup>: 1725 (ester), 1230 (acetate); NMR (CDCl<sub>3</sub>, 80 MHz, TMS):  $\delta$ 0.72-1.10 (6×Me), 2.0 (3H, s, OAc), 2.03 (3H, s, OAc), 3.60 (3H, s, OMe), 4.15 (1H, t, H-3), 5.25 (1H, m, H-12).

## SeO<sub>2</sub> Oxidation of V-methylester Acetate

A solution of V-methylester (5mg) and freshly prepared SeO<sub>2</sub> (5mg)in HOAc (1ml) was heated for 2hr. The solution was filtered, diluted with H<sub>2</sub>O and extracted with Et<sub>2</sub>O. The residue was chromatographed to give a diene as colorless material. UV  $\lambda_{\text{max}}^{\text{EtOH}}$  nm(log $\varepsilon$ ): 241 (4.18), 251(4.23), and 261(4.15).

# Saponification of V-methylester

A solution of V-methylester (10mg) in 10% KOH in EtOH was refluxed for 8hr and concentrated under reduced pressure to remove EtOH. After addition of H<sub>2</sub>O, extracted with Et<sub>2</sub>O (ester, 67%). The aqueous mother liquor neutralized with 5% HCl and extracted with Et<sub>2</sub>O (acid, 33%).

#### LiAlH<sub>4</sub> Reduction of V

A sample of V (10mg) was dissolved in dry THF (50ml) and LiAlH<sub>4</sub> (50mg) was added slowly with stirring and then refluxed for 3hr. The reaction mixture was filtered and concentrated to give 28, 29 dihydroxy- $\beta$ -amyrin, mp 250- $7^{\circ}$  (mmp, co-TLC).

## Monobromo- $\gamma$ -Lactone of V

To a solution of V (5mg) and NaOAc (2mg) in HOAc (1ml) was added dropwise a solution of bromine in HOAc (3%, 1ml). The reaction mixture was kept at room temp. for 3hr. and then poured into  $H_2O$  (5ml) containing  $Na_2S_2O_3$ 

(500mg) to discharge excess bromine. The ppt was filtered, washed with thoroughly with  $H_2O$ , dried and crystallized from MeOH to give V-bromolactone, IR  $\nu_{\rm max}^{\rm KBr}$  cm<sup>-1</sup>: 1800 ( $\gamma$ -lactone).

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