

Activated Nitriles in Heterocyclic Synthesis: Syntheses of Thiazole, Pyrazole and 4H-1,4-Benzothiazine Derivatives

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Abstract—4-Arylazo-3-phenyl-5-aminopyrazoles (**5a,b**) and substituted hydroxythiazoles **8a,b** were synthesized from the reaction of **4a,b** with hydrazine hydrate and mercaptoacetic acid respectively. Compounds **5a,b** and **8a,b** were also obtained from coupling of **2a,b** with **6** and **7**, respectively. 4H-1,4-Benzothiazine **11** was prepared from **1** and **10**. The reaction of the diazonium salts **2a-c** with diethyl 3-amino-2-cyanopent-2-en-1,5-dicarboxylate **12** was also reported.

Keywords—Sulfonamides, thiazole, aminopyrazole, thiazine.

The first azo dyes containing the sulphonamide or sulphonamido moieties were synthesized in 1909 and were found to exhibit colour fastness¹. This class of dyestuffs was subjected to many biomedical studies²⁻⁷ and many compounds showed high antibacterial activities³⁻⁵ and therapeutic significance^{6,7}.

This work was conducted as a part of our programme directed towards developing simple procedures for the synthesis of azoles^{8,9}, fused azoles¹⁰ and azines¹¹ of promising biological activity^{12,13}.

In the first part of the work new thiazole and pyrazole derivatives containing sulphonamido moiety were prepared. Benzoylacetone nitrile **1** was allowed to react with aryl diazonium chlorides containing p-sulphonamido substituent **2a-c** to afford the arylhydrazones **4a-c**. The IR spectra of **4a-c** revealed the presence of conjugated cyano function which was in a good agreement with the hydrazone structure **4** and not the anticipated tautomer **3**. The pyrazoles **5a,b** were obtained by the action of hydrazine hydrate on **4a,b**. The IR spectra of **5a,b** showed the absence of the cyano absorption band and the presence of the amino function. Structure **5a,b** were further established in an independent synthesis via coupling of **2a,b** with 3-phenyl-5-aminopyrazole¹⁴ **6**.

The aryl diazonium salt **2a** was coupled with 3-phenacyl-4-hydroxythiazole¹⁵ **7** to give a product of molecular formula C₁₉H₁₆N₄O₅S₂ for which the structure possibilities **8** and **9** had to be investigated. Structure **9** was readily ruled out based on the formation

of **8a,b** from **4a,b** and mercaptoacetic acid. The same behaviour was also observed on reacting **2b** with **7**.

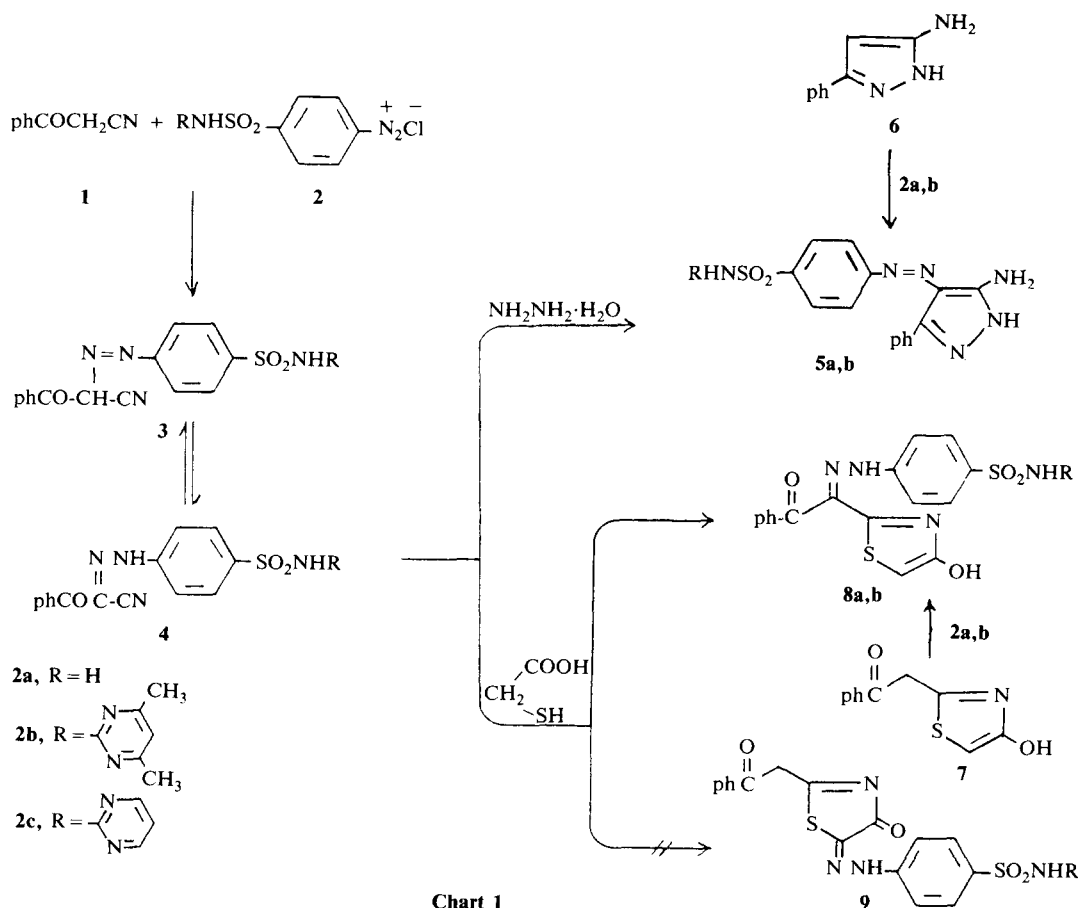
In this part of the work we report on the preparation of a new benzothiazine derivative. 4-Methyl-2-mercaptoaniline **10** was allowed to react with **1** to afford 4H-1,4-benzothiazine **11**. Formation of **11** from **1** and **10** is assumed to proceed via elimination of water and H₂. This type of reactions is in parallel to the recently reported reaction of forming benzothiazines from o-mercaptoanilines and 1,3-diketones¹⁶.

We recently reported¹⁷ on the formation of pyrazolo-[1,5-c] [1,2,4] triazine when the diazonium salt of 3-antipyrinyl-5-aminopyrazole was coupled with diethyl 3-amino-2-cyanopent-2-en-1,5-dicarboxylate **12**. However, in the present work, when the diazonium salts **2a-c** were coupled with **12** the open structures **13a-c** were obtained. ¹H-NMR spectra revealed the presence of two ester groups which agrees with the hydrazone structures **13a-c**. Trials to cyclize **13** into **14** by the action of acetic anhydride or acetic acid failed.

EXPERIMENTAL METHODS

All melting points are uncorrected. Recorded yields correspond to the pure products. IR (KBr) spectra were carried out using a Pye Unicam SP-1100 Spectrophotometer. ¹H-NMR spectra were measured on a Varian EM-360 Spectrometer (60 MHz), using tetramethylsilane as an internal standard and chemical shifts are expressed as δ values. The microanalysis was done at the Microanalytical Units at Cairo and

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Coupling of benzoylacetonitrile 1 with the aryl diazonium salts 2a-c

Formation of the arylhydrazones 4a-c: A solution of benzoylacetonitrile (1) (0.01 mol) in ethanol: pyridine (1:1, 30 ml) was treated with a saturated sodium acetate solution (10 ml), then with 2a-c. The mixture was left in the refrigerator for 24 h and the products so formed were collected by filtration, crystallized and identified as 4a-c.

4a: mp. 233-235°C; from ethanol-DMF; yield 2.4g (75%); IR (cm⁻¹): 3370, 3280 (NH), 2220 (conjugated C=N), 1680 (C=O), 1640 (C=N), 1600 (N=N). (Found: C, 54.83; H, 3.63; N, 17.03. C₁₅H₁₂N₄O₃S (328.35) requires C, 54.87; H, 3.69; N, 17.06%).

4b: mp. 228-230; from ethanol-DMF, yield 3.0g (70%); IR (cm⁻¹): 3300-3100 (NH), 2220 (conjugated C=N), 1685 (C=O), 1640 (C=N), 1605 (N=N). (Found: C, 58.07; H, 4.28; N, 19.36. C₂₁H₁₈N₆O₃S

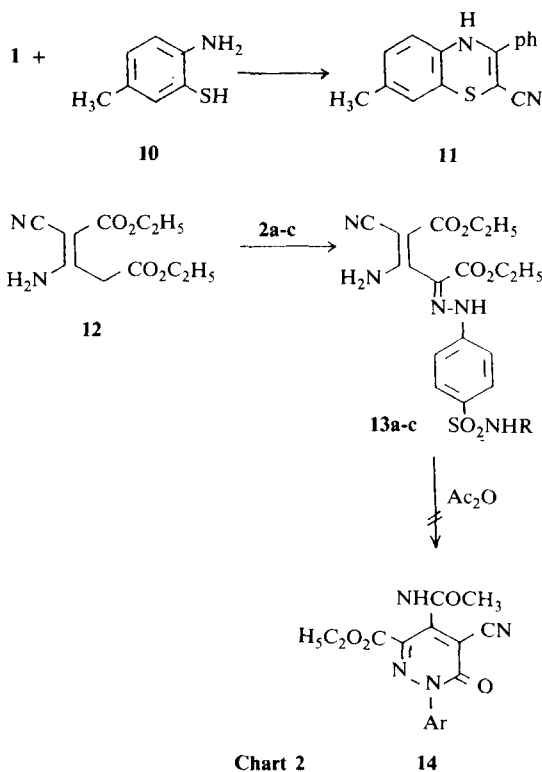
(434.46) requires C, 58.05; H, 4.18; N, 19.35%).

4c: mp. 260-262°C; from ethanol-DMF; yield 3.25g (80%); IR (cm⁻¹): 3240, 3140 (NH); 2220 (conjugated C=N), 1680 (C=O), 1620 (C=N), 1590 (N=N). (Found: C, 56.19; H, 3.41; N, 20.62. C₁₉H₁₄N₆O₃S (406.42) requires C, 56.15; H, 3.47; N, 20.68%).

Formation of 3-phenyl-4-aryldiazo-5-aminopyrazole (5a,b)

(a) From 4a,b and hydrazine hydrate: Hydrazine hydrate (0.05 mole) (98%) was added to each of compounds 4a,b (0.02 mole). The reaction mixture was heated on a water bath for 15 minutes. The product deposited after cooling was triturated with cold ethanol, then crystallized and identified as 5a,b.

(b) Form 3-phenyl-5-aminopyrazole (6) and 2a,b: A cold solution of 3-phenyl-5-aminopyrazole (6) (0.01 mol in ethanol: H₂O (1:1, 50 ml) treated with a saturated sodium acetate solution (20 ml), then with 2a,b. The reaction mixture was kept at room temper-



ature overnight. The solid product was collected by filtration, crystallized and identified as **5a,b**. Compounds **5a,b** formed by this method were identical (mp. 's, mixed mp. 's and IR) with those prepared by method (a).

5a: mp. >300°C, from ethanol-DMF, yield 2.0g (60%); IR (cm⁻¹): 3400-3200 (NH, NH₂), 1630 (C=N), 1610 (N=N). (Found: C, 52.59; H, 4.07; N, 24.49. C₁₅H₁₄N₆O₂S (342.38) requires C, 52.62; H, 4.12; N, 24.55%).

5b: mp. 185-187°C; from ethanol-DMF, yield 2.9g (65%); IR (cm⁻¹): 3400-3100 (NH, NH₂), 1630 (C=N), 1600 (N=N). (Found: C, 56.22; H, 4.47; N, 25.01. C₂₁H₂₀N₈O₂S (448.51) requires C, 56.23; H, 4.49; N, 24.98%).

Synthesis of **8a,b**

(a) From 3-phenacyl-4-hydroxythiazol (7) and 2a,b: A cold solution of **7** (0.01 mol) in ethanol:H₂O mixture (1:1, 100 ml) was treated with saturated sodium acetate solution (10 ml), then with either **2a** or **2b**. The reaction mixture was left overnight in the refrigerator and the solid product was collected by filtration, crystallized from the proper solvent and identified as **8a,b**.

(b) From 4a,b and mercaptoacetic acid: **4a,b** (0.01 mol) and mercaptoacetic acid (0.01 mol) in dry pyridine were refluxed for 6 h. The solvent was removed *i. vac.* and the remaining product was triturated for several times with pet. ether (40-60%). The solid product, so formed, was identified (mp. 's and mixed mp. 's) as **8a,b**.

8a: mp. 205-207°C; from ethanol, yield 2.7g (68%); IR (cm⁻¹): 3400, 3290, 3160 (OH, NH, NH₂), 1685 (exocyclic C=O), 1620 (C=N), 1590 (N=N). (Found: C, 5.68; H, 3.57; N, 13.96. C₁₇H₁₄N₄O₄S₂ (402.43) requires C, 50.73; H, 3.51; N, 13.92%).

8b: mp. 145-147°C, from ethanol, yield 3.2g (63%); IR (cm⁻¹): 3300-3200, 3100 (OH, NH), 1690 (C=O, exocyclic), 1620 (C=N), 1600 (N=N). (Found: C, 54.40; H, 4.01; N, 16.50. C₂₃H₂₀N₆O₄S₂ (508.58) requires C, 54.31; H, 3.96; N, 16.53%).

Synthesis of 4H-1,4-benzothiazine (11)

Benzoylacetone nitrile (**1**) (0.01 mol), 4-methyl-2-mercaptoaniline (**10**) (0.01 mol) and dimethylsulphoxide (8 ml) were stirred together and heated at 140-145°C for 40 minutes. The product crystallized on cooling was filtered off and crystallized from ethanol to give **11**, mp. 187-189°C; yield 2.3g (87%); IR (cm⁻¹): 3310, 3080 (NH), 2200 (conjugated C=N); ¹H-NMR (ppm): 2.5 (s, 3H, CH₃); 6.6-7.45 (m, 8H, aromatic protons); 9.6 (s, 1H, NH). (Found C, 72.68; H, 4.48; N, 10.56. C₁₆H₁₂N₂S (264.35) requires C, 72.69; H, 4.58; N, 10.60%).

Synthesis of **13a-c**

To a stirred, cold solution of the active methylene compound **12** (0.01 mol) in ethanol: pyridine (5:1, 6 ml), the aryl diazonium salt **2a-c** (0.01 mol) was added in portionwise over a period of 30 minutes at <5°C. The reaction mixture was left overnight in the refrigerator. The product was filtered, crystallized from the proper solvent and identified as **13a-c**.

13a: mp. 210-212°C; from ethanol; yield 3.5g (85%); IR (cm⁻¹): 3400, 3320, 3100 (NH, NH₂), 2220 (conjugated C≡N); 1730, 1700 (two C=O, ester); 1630 (C=N), 1595 (N=N). (Found: C, 46.90; H, 4.72; N, 17.10. C₁₆H₁₉N₅O₆S (409.41) requires C, 46.94; H, 4.68; N, 17.11%).

13b: mp. 135-137°C; from ethanol; yield 3.9g (76%); IR (cm⁻¹): 3500-3100 (NH, NH₂), 2210 (conjugated C≡N); 1720, 1710 (two C=O, ester), 1630 (C=N); 1600 (N=N). (Found: C, 51.30; H, 4.81; N, 18.97. C₂₂H₂₅N₇O₆S (515.54) requires C, 51.25; H, 4.89; N, 19.02%).

13c: mp. 143-145°C; from ethanol, yield 3.9g (80%); IR (cm⁻¹): 3500-3240 (NH, NH₂), 2220 (conjugated

(C≡N), 1700, 1695 (two C=O, ester); 1630 (C=N), 1590 (N=N). (Found: C, 47.25; H, 4.31; N, 20.13. C₂₀H₂N₇O₆S (487.49) requires C, 47.27; H, 4.34; N, 20.11%). ¹H-NMR (ppm): 1.1-1.45 (s, 6H, two CH₃), 4.0-4.4 (octet, 8H, two CH₂), 6.85-8.5 (m, 7H, aromatic protons); 8.65-9.2 (brs, 2H, NH₂); 10.90-12.50 (s, 2H, two NH).

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