Reactions with Coumarinohydrazidoyl Halides: Synthesis of Some New Coumarinoyl Pyrazole and 1,2,4-Traizole Derivativies via Dipolar Cycloaddition Reactions

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Several new coumarinoyl pyrazole and coumarionyl-1,2,4-triazole derivatives were synthesised via dipolar cycloaddition reactions with 3-coumarinohydraziodoyl halide. Structure were based on elemental analysis and spectral data.

Key words: Coumarin, Hydrazidoyl halides, Pyrazoles, 1,2,4-triazoles, Cycloaddition reactions

INTRODUCTION

Interest in the synthesis and chemistry of hydrazidoyl halides has relived during the last few years. In this respect, we have recently published a series of publications (Ismail, 1991a,b; Ismail et al., 1991) dealing with the subject. As a continuations to the effort done in this field it was thus decided to extend our investigations to cover the area of cycloaddition reaction with nitrilimines derived from coumarinoyl hydrazidoyl halides. The newly synthesised heterocyclic derivatives contain both the coumarin ring and the pyrazole or 1,2,4-triazole ring which make them highly promising for both biological activity studies (Mathur et al., 1965; Cingolani et al., 1969; Essawy et al., 1980; Fliege et al., 1984) as well as for further chemical transformations leading to other heterocycles.

MATERIALS AND METHODS

All melting points are uncorrected. IR (KBr) were recorded on a pye Unicome Sp. 1100 spectrometer. $^1\text{H-NMR}$ spectra were recorded in DMSO-d₆ on a Varian EM 390 90 MHz spectrometer using TMS as an internal standard and chemical shifts are expressed as δ ppm units. The microanalyses were performed at the Microanalytical Center of Cairo University using Perkin-Elmer 2400 CHN Analyzer.

Compound 1 was prepared according to literature procedure (Ismail et al., 1991), which equivalent

amount of triethylamine to give the corresponding nitrilimine **2**, which was taken as the staring material. Compound **2** reacted with Schiff base, ethyl acrylate, acrylonitrile, N-phenyl maleimide and some α,β unsaturated nitriles, according to the following procedure.

General Procedure

A solution of 1 (0.01 mole) in dry benzene was treated with (0.01 mole) of triethylamine dropwise with stirring. After complete addition (30 min.) each reagent (0.01 mole) was added and the reaction mixture was when refluxed for 3 h. The solid products which separated after cooling were filtered off and washed with petroleum ether, then crystallized from the proper solvents to give the new compounds listed in (Table I).

RESULTS AND DISCUSSION

Thus it has been found that α -(3-coumarinyl)- β -bromoglyoxal- β -phenyl hydrazone (1) (Ismail et al., 1991) reacted with the equivalent amount of triethylamine to generate the corresponding nitrilimine 2 which was taken as the starting material for the present study.

Compound **2** reacted also with ethyl acrylate to afford a reaction product which was formulated as 3-(3'-coumarinoyl)-5-ethoxy carbonyl-pyrazole **5**. Structure of **5** was estabilished on the basis of correct elemental analysis and spectral data. IR spectrum of **5** showed bands at 1690, 1670, 1650 of (three CO) groups in addition to a band at 1620 of (C=N) group. The $^1\text{H-NMR}$ spectrum revealed the presence of the $\text{Co}_2\text{C}_2\text{H}_5$ group (t, 3H, 1.3 and q, 2H, 4.3) in addition to a multiplet at 6.8-7.4 corresponding to pyrazole H-

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Comp.	Colour (solvent)	M.P. (°C)	Yield (%)	Mol. Formula	% Analysis		Calcd./Found	
					C	Н	N	S
4	Brown	260	60	C ₃₀ H ₂₁ N ₃ O ₃	76.4	4.6	8.9	
	(benzene)				76.2	4.4	8.7	
5	Orange	145	60	$C_{22}H_{17}N_2O_5$	67.8	4.4	7.2	
	(CCI₄)				67.6	4.2	7.0	
6	Brown	220	60	$C_{20}H_{11}N_3O_3$	70.4	3.2	12.3	
	(ethanol)				70.2	3.2	12.1	
8	Brown	185	60	$C_{27}H_{15}N_3O_5$	70.3	3.3	9.1	
	(ethanol)				70.1	3.1	9.0	
11a	Orange	210	50	$C_{26}H_{15}N_3O_3$	74.8	3.6	10.1	
	(tolouene)				74.6	3.5	10.0	
11b	Orange	150	50	$C_{28}H_{20}N_2O_5$	72.4	4.3	6.0	
	(toluene)				72.2	4.1	6.0	
11c	Orange	180	50	$C_{26}H_{17}N_3O_4$	71.7	3.9	9.7	
	(toluene)				71.5	3.7	9.5	
11d	Orange	160	50	$C_{26}H_{17}N_3O_3S$	69.2	3.8	9.3	7.1
	(toluene)				69.0	3.7	9.1	6.9

Table I. Characterization data of the newly synthesized derivatives

Scheme 1

4, coumarinoyl and aromatic protons (11H). The disappearance of any signal in the region of δ 4.5-6.5 ppm proves that the compound suffered dehydrogenation under the applied reaction conditions.

Similar to its behaviour towards ethyl acrylate, compound **2** reacted with acrylonitrile to yield the dipolar cycloaddition reaction **6**. IR of **6** showed the presence of bands at 1680, 1660 of (two CO) and a band at 2220 of (CN). ¹H-NMR spectrum revealed the presence of signals of pyrazole H-4 (s, 6.6) in addition to

the multiplet (δ 6.8-7.5 ppm) corresponding to cumarinoyl and aromatic protons (10H). These results proved that the reaction product suffered also dehydrogenation under the applied reaction conditions. The possibility that both **5**, **6** could be the corresponding 4-substituted derivatives was readly ruled out based on the data previously reported by several investigators (Tewari et al., 1983a,b; Ezmirly et al., 1988; Shawali et al., 1990).

Furthermore, 2 cycloadded to N-phenyl-maleimide

7 in dry benzene and the reaction product was formulated as 1.5-diphenyl-1-3-coumarinoyl pyrrolino(3,4-c) pyrazole-2,6-dione **8**. Compound **8** gave the absorption bands of two CO groups in its IR spectrum in addition to the two widely separated bands at 1780 and 1710 cm $^{-1}$ characteristic for the (-CO-NR-CO) grouping. Its 1 H-NMR did not reveal except the multiplet at 6.8-7.6 8 ppm corresponding to coumarinoyl and aromatic protons. This result proved the dehydrogenation of the reaction product under the applied reaction conditions.

The course of cycloaddition of some α.β-unsaturated nitriles 9 was also investigated. Thus, it has been found that 9a-d reacted with 2 to yield cycloaddition reaction products with a molar ratio of 1:1. These reaction products could then be formulated as the 4-phenyl-5-substitued pyrazole derivatives 11a-d respectively. The IR spectrum of each 11a-d did not show any nitrile absorption. The disappearance of any signals for pyrazols protons in the ¹H-NMR spectra of these products indicated that these products were dehydrocyanated under the applied reaction conditions as compounds 11a-d were assumed to be formed via initial formation of the addition products 10a-d respectively which lost HCN to yield the final isolable 11a-d respectively. Again the regioselectivity problem was solved and structures were established to be the 1.4-diphenyl pyrazole derivatives rather than the isomeric 1.5-diphenyl pyrazole based on previous reports in this respect (Tewari et al., 1983a,b; Ezmirly et al., 1988; Shawali et al., 1990).

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