

Petrography and mineral chemistry of the granitic rocks in the Po Eun-Sogrisan Area, Korea

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ABSTRACT: The granitic rocks in the Po Eun - Sogrisan area are composed of the Jurassic Po Eun granodiorite and the Cretaceous Sogrisan granites. The latter can be divided into three rock types: coarse-grained biotite granite, porphyritic biotite granite and granite porphyry. Petrographical observations, especially focusing on the quartz-feldspar intergrowth texture, suggest that the Sogrisan granites has emplaced at shallower level and crystallized more rapidly than the Po Eun granodiorite. The F, Cl contents and the Fe/(Fe+Mg) ratios of biotite and muscovite in the Sogrisan granites are higher than those in the Po Eun granodiorite. The anorthite contents of plagioclase in the Po Eun granodiorite are higher than in the Sogrisan granites. Ilmenite in the Sogrisan granites is more enriched in Mn and depleted in Fe than that in the Po Eun granodiorite. The whole-rock magnetic susceptibility values (in 10^{-6} emu/g unit) are higher in the Sogrisan granites (33~144) than the Po Eun granodiorite (9~12), indicating that the former generally belongs to magnetite-series granitoid and the latter to ilmenite-series one. The Sogrisan granites has solidified under more oxidizing environment than the Po Eun granodiorite, judging from the whole-rock magnetic susceptibility measurements as well as the chemical compositions of biotite and ilmenite.

Key words: Po Eun granodiorite, Sogrisan granites, petrography, mineral chemistry, whole-rock magnetic susceptibility

INTRODUCTION

In the central Ogcheon belt (the Po Eun-Sogrisan area), Jurassic and Cretaceous granitic rocks are closely associated. Since 1970's many studies on these granitic rocks have been carried out (for example: Kim, 1971; Lee, 1971; Lee and Park, 1981; Shibata *et al.*, 1983; Kim and Shin, 1990; Yun and Kim, 1990; Jin *et al.*, 1992). However, petrographical and mineralogical differences between the Jurassic and Cretaceous granites in the belt are hardly known. It is necessary to determine these differences prior to investigating the genetical relationship between the two granitic masses of different age.

The purposes of this study are to describe the occurrence, petrography and mineral chemistry of the Jurassic Po Eun granodiorite and the Cre-

taceous Sogrisan granitic rocks, and to examine their emplacement conditions and crystallization processes.

OUTLINE OF GEOLOGY

In the study area early Paleozoic metasedimentary rocks (so called "Ogcheon Supergroup") and Mesozoic granitic rocks are mainly cropped out (Fig. 1). Mesozoic granitic rocks can be divided into Jurassic granites (166 Ma) and Cretaceous ones (74~82 Ma) judging from their K-Ar biotite ages (Kim, 1971).

Paleozoic strata is mainly composed of phyllite, schist, amphibolite and limestone. They had suffered regional metamorphism of greenschist facies (Kim, 1971), and had been strongly deformed in middle to late Paleozoic (Cluzel *et al.*, 1990)

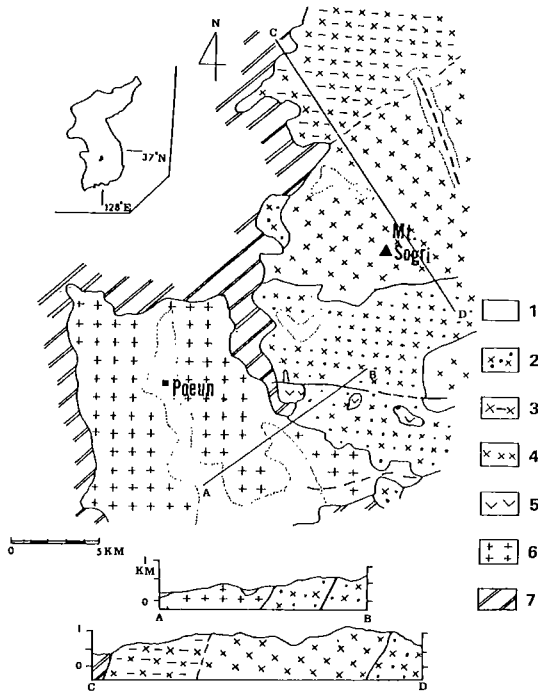


Fig. 1. Geologic map of the Poemun - Mt. Sogri area. 1; Alluvium, 2; Cretaceous granite porphyry (KGPH), 3; Cretaceous porphyritic biotite granite (KPGR), 4; Cretaceous coarse-grained biotite granite (KBGR), 5; Volcanic rocks, 6; Jurassic Poemun biotite granodiorite (JPGD), 7; Paleozoic Ogcheon Supergroup

or early Mesozoic time (Otoh *et al.*, 1990). The Jurassic Poemun granitic rocks (hereafter JPGD) occur as batholith in the southwestern part of the study area. They show granodioritic compositions in mode (Table 1, Fig. 2). Pyroxene- to amphibole-hornfels in the metasediments develop around the JPGD (Lee and Park, 1981). In the western margin of the JPGD mass, the granitic rocks were slightly deformed. Muscovite-bearing, fine-grained felsic dikes several tens of centimeters in width intruded the JPGD with random directions.

Cretaceous granitic rocks (so called "Sogrisan Granites", hereafter KSGR) are texturally divided into coarse-grained biotite granite (KBGR), porphyritic biotite granite (KPGR) and granite porphyry (KGPH). The KBGR and the KPGR show granitic compositions in mode (Table 1, Fig. 2). The KPGR gradually changes to the KBGR in the

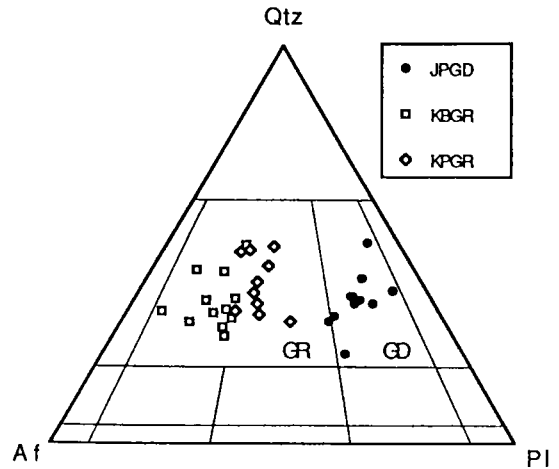


Fig. 2. Triangular diagram of modal compositions for the granitic rocks based on IUGS classification. Abbreviations: GR, Granite; GD, Granodiorite; Qtz, quartz; Af, alkali feldspar; Pl, plagioclase; JPGD, KBGR and KPGR are the same as those in Fig. 1.

Table 1. Average modal compositions of the granitic rocks

| | JPGD | KBGR | KPGR |
|------|------|------|------|
| | 12* | 12* | 15* |
| Qtz | 31.2 | 34.7 | 37.8 |
| Af | 14.9 | 45.6 | 35.8 |
| Pl | 42.6 | 17.1 | 22.3 |
| Bt | 9.6 | 1.6 | 2.9 |
| Mus | 0.5 | 0.1 | 0.2 |
| Zr | tr | 0.1 | 0.1 |
| Ap | 0.2 | 0.1 | 0.1 |
| Sph | 0.3 | - | - |
| Opq | 0.1 | 0.3 | 0.4 |
| All | tr | - | tr |
| Chl | 0.2 | 0.1 | 0.1 |
| C.I. | 10.9 | 2.3 | 3.8 |

Abbreviations: Qtz=quartz; Af=alkali feldspar; Pl=plagioclase; Bt=biotite; Mus=muscovite; Zr=zircon; Ap=apatite; Sph=sphene; Opq=opaque minerals; All=allanite; Chl=chlorite; C.I.=color index; *=number of analyses.

northern part of Mt. Sogri (Sogrisan). The KBGR and some of the KPGR have miarolitic cavities or pegmatitic pockets, ranging from 1 cm to 30 cm in length. Euhedral quartz, alkali feldspar and rarely calcite have grown in the cavities. The amount of cavities increases toward the southern part of the KBGR. A number of pegmatitic and

aplitic dikes intruded into the KBGR and the KPGR. The KGPH, intruding the JPGD and the KBGR, is divided into three types by their phenocryst assemblages and color: pale gray granite porphyry, pinkish quartz feldspar porphyry and dark gray felsite. They show random distributions. The strata of volcanic tuff of andesitic to dacitic compositions are observed within the KGPH mass. The tuffaceous rocks are either lithic or crystalline. Purple lithic tuff contains sub-angular andesitic and dacitic fragments (0.5 cm to 10 cm in length). Black crystalline tuff has quartz, alkali feldspar and opaque minerals, showing flow structure. It is likely that the volcanic tuff has erupted before the intrusion of the KGPH since they contain no granitic rock fragments.

DESCRIPTION OF THE GRANITIC ROCKS

Jurassic Poenu Granodiorite (JPGD)

The JPGD is medium-grained equigranular granodiorite, and is mainly composed of plagioclase, quartz, alkali feldspar and biotite. Spene, apatite, zircon, allanite and ilmenite are also observed as accessory minerals. Secondary minerals are muscovite and chlorite. Plagioclase is euhedral to subhedral and shows strong compositional zonation of andesine to oligoclase. It has sericitized core and sometimes exhibits oscillatory zoning. Biotite, apatite, opaque minerals and early crystallized plagioclase are observed within plagioclase grains as inclusions. Subhedral quartz forms an aggregate of several grains, showing slightly undulatory extinction. It contains biotite and rarely spene. Alkali feldspar is mainly microcline and contains inclusions of plagioclase, quartz, biotite, spene and early microcline. In the deformed rocks, myrmekites occur along the marginal parts of microcline. Dark brown anhedral biotite fills the interstice, and shows linear arrangement in the deformed rocks. Large euhedral spene contains a number of anhedral ilmenite

grains like Jurassic Andong granodiorite (Lee and Lee, 1991). Muscovite-bearing, fine-grained felsic dike (JMFD) is composed of quartz, microcline and muscovite with subordinate amounts of plagioclase and biotite. Judging from the whole-rock chemistry, the JMFD is considered to be *in situ* fractionation product of the JPGD (Jwa *et al.*, in prep.).

Cretaceous Coarse-grained Biotite Granite (KBGR)

Pinkish KBGR is mainly composed of quartz, alkali feldspar, plagioclase and biotite. Accessory minerals are apatite, zircon, allanite, magnetite and ilmenite. Chlorite and muscovite are secondary minerals. Subhedral to anhedral quartz exhibits undulatory extinction. Some of large quartz grains contain biotite and rarely secondary muscovite along their fractures. Embayed quartz grains are usually intergrown with alkali feldspar. Alkali feldspar occurs as anhedral perthitic orthoclase filling the interstices. Large grains show the Carlsbad twinning and contain plagioclase, biotite and anhedral quartz as inclusions. Subhedral plagioclase is oligoclase in composition and forms an aggregate of several grains. Greenish brown biotite contains apatite and zircon, and is partly altered to chlorite and muscovite. Accessory allanite and magnetite accompanied by ilmenite are associated with biotite.

Cretaceous Porphyritic Biotite Granite (KPGR)

The KPGR is leucocratic fine- to medium-grained porphyritic biotite granite and consists of quartz, alkali feldspar, plagioclase and biotite. Apatite, zircon, allanite, magnetite and ilmenite are accessory minerals. Phenocrysts are euhedral quartz (up to 10 mm) and perthitic orthoclase (up to 15 mm). Quartz phenocrysts show slightly undulatory extinction, and rarely contains secondary muscovite. Perthitic orthoclase phenocrysts usually contain plagioclase, biotite and anhedral quartz.

Table 2. Chemical compositions and structural formulae of biotites

| sample | KPGR | | | | KBGR | | | | JPGD | | | | | |
|--------------------------------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|
| | K0101 | K0104 | K1406 | K1408 | K1504 | K1606 | J0209A | J1715 | J1716 | J1718 | J1719A | J1720 | J0713 | J0715 |
| SiO ₂ | 36.03 | 36.18 | 35.64 | 35.72 | 35.10 | 34.94 | 35.18 | 35.96 | 36.96 | 36.12 | 36.46 | 35.86 | 35.27 | 35.46 |
| Al ₂ O ₃ | 14.37 | 12.72 | 14.01 | 14.52 | 13.02 | 12.78 | 18.44 | 15.99 | 15.68 | 15.26 | 15.34 | 15.92 | 17.33 | 17.05 |
| TiO ₂ | 3.41 | 3.30 | 3.37 | 3.12 | 3.37 | 2.42 | 3.02 | 3.28 | 2.42 | 3.05 | 3.35 | 3.36 | 3.21 | 3.46 |
| FeO* | 27.83 | 28.43 | 28.82 | 27.59 | 31.63 | 33.41 | 23.27 | 24.25 | 23.74 | 24.12 | 23.10 | 24.35 | 23.61 | 24.01 |
| MnO | 1.91 | 1.00 | 1.97 | 1.83 | 0.73 | 0.86 | 0.47 | 0.42 | 0.36 | 0.36 | 0.35 | 0.35 | 0.62 | 0.56 |
| MgO | 3.08 | 5.40 | 3.30 | 3.50 | 3.74 | 3.04 | 6.40 | 6.75 | 7.86 | 7.91 | 8.00 | 7.11 | 6.55 | 6.64 |
| CaO | 0.00 | 0.07 | 0.00 | 0.00 | 0.04 | 0.05 | 0.00 | 0.01 | 0.00 | 0.02 | 0.00 | 0.01 | 0.00 | 0.00 |
| Na ₂ O | 0.09 | 0.09 | 0.15 | 0.12 | 0.19 | 0.16 | 0.08 | 0.06 | 0.04 | 0.05 | 0.05 | 0.06 | 0.05 | 0.07 |
| K ₂ O | 9.23 | 9.06 | 9.02 | 9.14 | 8.86 | 8.31 | 9.52 | 9.46 | 9.57 | 9.52 | 9.60 | 9.68 | 9.74 | 9.69 |
| Cl | 0.15 | 0.19 | 0.16 | 0.14 | 0.32 | 0.27 | 0.02 | 0.01 | 0.02 | 0.01 | 0.02 | 0.02 | 0.04 | 0.04 |
| =O | -0.03 | -0.04 | -0.04 | -0.03 | -0.07 | -0.06 | -0.01 | -0.00 | -0.01 | -0.00 | 0.00 | 0.00 | -0.01 | -0.01 |
| F | 1.96 | 1.03 | 2.06 | 1.68 | 1.52 | 0.90 | 0.24 | 0.65 | 0.69 | 0.41 | 0.39 | 0.54 | 0.27 | 0.34 |
| =O | -0.83 | -0.43 | -0.87 | -0.52 | -0.64 | -0.38 | -0.10 | -0.27 | -0.29 | -0.17 | -0.17 | -0.22 | -0.11 | -0.15 |
| Total | 97.20 | 97.00 | 97.59 | 96.81 | 97.81 | 96.70 | 96.53 | 96.29 | 96.94 | 96.66 | 96.49 | 97.04 | 96.57 | 97.16 |
| Recalculated on 22 oxygens | | | | | | | | | | | | | | |
| Si | 5.7402 | 5.7489 | 5.6895 | 5.7090 | 5.6405 | 5.6979 | 5.4045 | 5.5710 | 5.6629 | 5.5830 | 5.6135 | 5.5382 | 5.4498 | 5.4472 |
| Al IV | 2.2598 | 2.2511 | 2.3105 | 2.2910 | 2.3595 | 2.3021 | 2.5955 | 2.4290 | 2.3371 | 2.4170 | 2.3865 | 2.4618 | 2.5502 | 2.5528 |
| VI | 0.4390 | 0.1319 | 0.3293 | 0.4438 | 0.1054 | 0.1548 | 0.7431 | 0.4907 | 0.5020 | 0.3635 | 0.3970 | 0.4354 | 0.6052 | 0.5459 |
| Ti | 0.4090 | 0.3948 | 0.4046 | 0.3751 | 0.4065 | 0.2964 | 0.3485 | 0.3825 | 0.2801 | 0.3549 | 0.3871 | 0.3903 | 0.3733 | 0.3991 |
| Fe | 3.7075 | 3.7785 | 3.8480 | 3.6880 | 4.2507 | 4.5564 | 2.9902 | 3.1423 | 3.0505 | 3.1191 | 2.9749 | 3.1452 | 3.0507 | 3.0847 |
| Mn | 0.2578 | 0.1342 | 0.2661 | 0.2474 | 0.0994 | 0.1193 | 0.0617 | 0.0555 | 0.0469 | 0.0465 | 0.0450 | 0.0463 | 0.0806 | 0.0722 |
| Mg | 0.7322 | 1.2787 | 0.7858 | 0.8347 | 0.8969 | 0.7397 | 1.4655 | 1.5589 | 1.7995 | 1.8232 | 1.8365 | 1.6360 | 1.5091 | 1.5208 |
| Ca | 0.0000 | 0.0117 | 0.0006 | 0.0000 | 0.0070 | 0.0092 | 0.0002 | 0.0017 | 0.0006 | 0.0028 | 0.0019 | 0.0019 | 0.0000 | 0.0003 |
| Na | 0.0267 | 0.0275 | 0.0448 | 0.0335 | 0.0602 | 0.0496 | 0.0247 | 0.0168 | 0.0119 | 0.0155 | 0.0158 | 0.0176 | 0.0157 | 0.0208 |
| K | 1.8762 | 1.8370 | 1.8362 | 1.8637 | 1.8168 | 1.7286 | 1.8664 | 1.8698 | 1.8749 | 1.8774 | 1.8847 | 1.9062 | 1.9190 | 1.8986 |
| Cl | 0.0392 | 0.0524 | 0.0438 | 0.0370 | 0.0869 | 0.0761 | 0.0056 | 0.0020 | 0.0059 | 0.0059 | 0.0049 | 0.0043 | 0.0091 | 0.0090 |
| F | 0.9867 | 0.5168 | 1.0417 | 0.8480 | 0.7729 | 0.4658 | 0.1175 | 0.3191 | 0.3357 | 0.2013 | 0.1924 | 0.2650 | 0.1300 | 0.1724 |
| OH | 2.9741 | 3.4308 | 2.5203 | 3.1150 | 3.1402 | 3.4581 | 3.8769 | 3.6789 | 3.6584 | 3.7928 | 3.8027 | 3.7307 | 3.8609 | 3.8186 |
| Fe/Fe+Mg | 0.8351 | 0.7472 | 0.8304 | 0.8154 | 0.8258 | 0.8603 | 0.6711 | 0.6684 | 0.6290 | 0.6311 | 0.6183 | 0.6578 | 0.6690 | 0.6698 |
| F/(F+OH) | 0.2491 | 0.1309 | 0.2140 | 0.2140 | 0.1975 | 0.1187 | 0.0294 | 0.0798 | 0.0840 | 0.0504 | 0.0482 | 0.0663 | 0.0326 | 0.0432 |

*; total Fe as FeO

Smaller grains in the groundmass are composed of anhedral quartz, orthoclase, plagioclase and biotite. Micrographic intergrowths of quartz and alkali feldspar are frequently developed in the groundmass. Subhedral plagioclase is oligoclase and has sericitized cores. Dark brown biotite with serrate grain boundaries (up to 5 mm) contains apatite, zircon and opaque minerals. Magnetite is intimately associated with ilmenite.

Cretaceous Granite Porphyry (KGPH)

Phenocrysts of pale gray granite porphyry (30

-40% in volume percentage) are orthoclase, plagioclase, quartz and biotite in order of abundance. Euhedral orthoclase (up to 7 mm, about 40% in phenocrysts) shows the Carlsbad twinning, and contains plagioclase and anhedral quartz. Subhedral plagioclase is strongly sericitized. Anhedral quartz (up to 3 mm, about 30%) shows embayed grain boundaries. Anhedral, red to dark brown biotite oxidized at subsolidus condition is associated with opaque minerals. Groundmass is composed of anhedral quartz, orthoclase, plagioclase and biotite. Pale green anhedral hornblende is

rarely observed.

Pinkish quartz feldspar porphyry has many small cavities (up to 2 mm). Subhedral quartz phenocryst has sharp and/or locally embayed grain boundaries. Large euhedral orthoclase (up to 9 mm) displays the Carlsbad twinning, and contains occasionally anhedral quartz and plagioclase. Anhedral biotite, plagioclase and opaque minerals occur as minor constituents. Groundmass is cryptocrystalline and/or microcrystalline and shows micrographic and spherulitic textures.

Dark gray felsite is mainly composed of small anhedral quartz and spherulite. Secondary calcite, chloritized biotite and opaque minerals are rarely observed.

MINERALOGY

The chemical compositions of biotite, plagioclase, alkali feldspar, ilmenite and muscovite in the granitic rocks were analyzed by an electron microprobe analyzer (JXA-733 Mark II, JEOL) of the Geological Institute, University of Tokyo, following the procedure described by Nakamura and Kushiro (1970). Acceleration voltage and beam current conditions except for the analysis of ilmenite were 15 kV and 1.2×10^{-8} A, respectively. Ilmenite was analyzed at 20 kV of acceleration voltage.

Biotite

Subhedral to anhedral biotite usually contains apatite, zircon and opaque minerals. It partly alters to chlorite and muscovite. Some grains in the KBGR suffered secondary oxidation caused by the breakdown to Fe-Ti oxides, quartz and alkali feldspar. In the KPGR, biotite sometimes occurs as large grains (5 to 8 mm).

Table 2 lists the chemical compositions of biotite. The Mg/(Mg+Fe) ratios of biotite in the JPGD are generally higher than 0.3, and those in the KSGR (the KPGR and the KBGR) is lower than 0.3. Total Al contents of biotite in the JPGD

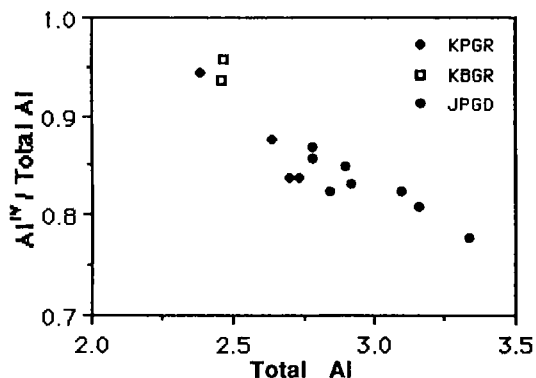


Fig. 3. Al^{IV}/Total Al plotted against Total Al of biotites.

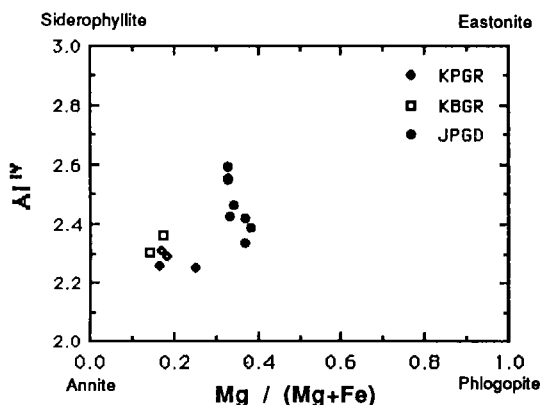


Fig. 4. Al^{IV} plotted against Mg/(Mg+Fe) of biotites.

range from 2.7 to 3.4, whereas those in the KSGR is lower than 2.7. As shown in Fig. 3, the ratios of tetrahedral Al (Al^{IV}) to total Al contents decrease with the increase of total Al contents, indicating the increase of the ratios of octahedral Al (Al^{VI}). Biotite of the JPGD shows more variation in Al^{IV} contents than that of KSGR (Fig. 4). Concerning the substitutions of elements between octahedral and tetrahedral sites, biotite of the JPGD seems to have been substituted as follows: $Ti^{VI} + 2Al^{IV} = Mg^{VI} + 2Si^{IV}$. But elemental substitution in biotite of the KSGR is not clear (Fig. 5).

The fluorine contents of the KSGR range from 0.4 to 1.0, whereas those of the JPGD are lower than 0.3. Biotite of the JPGD is almost chlorine-free. The relative ratios of F, Cl and OH in hydroxyl sites are shown in Fig. 6. Hydroxyl ion

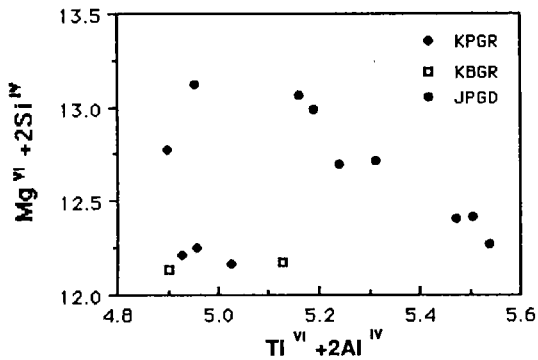


Fig. 5. $(Mg^{VI} + 2Si^{IV})$ plotted against $(Ti^{VI} + 2Al^{IV})$ of biotites.

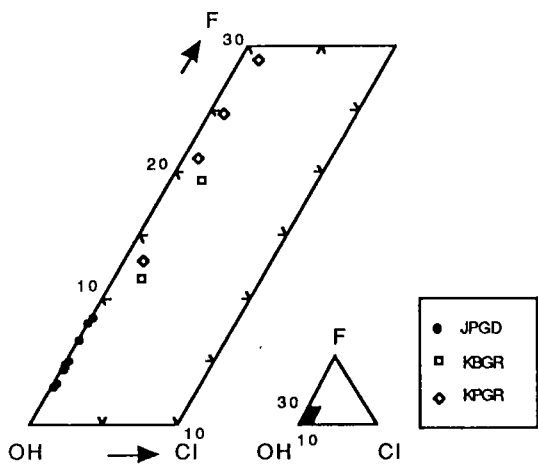


Fig. 6. Triangular diagram of F-OH-Cl contents in hydroxyl sites of biotites.

(OH) in biotite of the KSGR is more replaced by F and Cl than that of the JPGD. Fluorine enrichment in biotite of the KSGR corroborates the experimental result that fluorine is concentrated in granitic melts relative to aqueous fluids (Webster and Holloway, 1990).

Plagioclase

Plagioclase of the JPGD is strongly zoned, and has sericitized cores. Some grains exhibit oscillatory zonings. Albite twinned plagioclase of the KSGR is partly sericitized, and forms an aggregate of several grains. Chemical compositions and structural formulae of plagioclase are listed in

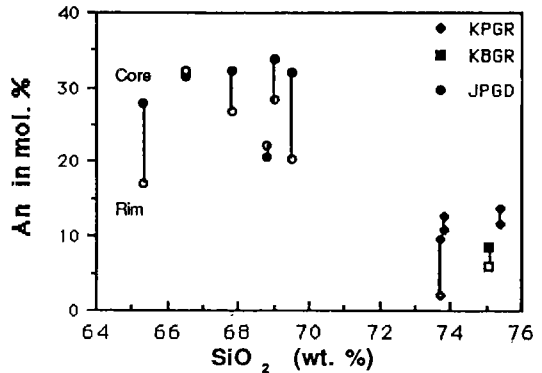


Fig. 7. The An contents (mol.%) of plagioclases plotted against SiO_2 (wt.%) of the host granitic rocks.

Table 3. The anorthite (An) contents of plagioclase in the JPGD are greater than those in the KSGR. Most of them are normally zoned, but some grains reversely zoned. The differences of An contents between core and rim in the JPGD are generally larger than in the KSGR (Fig. 7). Therefore, the granitic magma of the JPGD seems to have had longer duration in plagioclase primary field defined by Tuttle and Bowen (1958) than that of the KSGR, regardless of pressure difference between two granitic masses.

Alkali Feldspar

Alkali feldspar is microcline in the JPGD and perthitic orthoclase in the KSGR. Microcline in the JPGD has crystallized later than any other constituents including plagioclase, biotite and quartz. Perthitic orthoclase in the KPGR usually occurs as large phenocryst. String and patch type albite exsolutions are predominant. Anhedronal alkali feldspar in the KSGR is intergrown with quartz. It is thus expected that the liquidus of alkali feldspar in the KSGR was higher than that of plagioclase, while this state was reverse in the JPGD. The chemical compositions of alkali feldspar are shown in Table 4. The geothermometer of coexisting plagioclase and alkali feldspar determined by Powell and Powell (1977) yield that the temperature of KSGR was 350°C to 490°C, and

Table 3. Chemical compositions of plagioclases

| Rock | KSGR | | | | | | | | | | | | JPGD | | | | | | | | | | | | JMFD | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
|-------------------|-------|--------|-------|-------|-------|-------|-------|-------|--------|--------|-------|--------|--------|-------|-------|-------|-------|-------|-------|--------|-------|-------|--------|--------|--------|--------|---------------------------|-------|-------|-------|-------|-------|-------|--------|-------|-------|--------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|------|
| | K1406 | | | K1408 | | | K0101 | | | K1606 | | | J0209A | | | J1715 | | | J1716 | | | J1718 | | | J1719A | | | J1720 | | | J0713 | | | J0209B | | | J0716B | | | | | | | | | | | | | | | | |
| | core | rim | core | rim | core | rim | core | rim | core | rim | core | rim | core | rim | core | rim | core | rim | core | rim | core | rim | core | rim | core | rim | core | rim | core | rim | core | rim | core | rim | core | rim | core | rim | | | | | | | | | | | | | | | |
| SiO ₂ | 65.90 | 64.67 | 65.28 | 65.74 | 67.02 | 68.75 | 66.66 | 67.27 | 60.65 | 62.01 | 59.60 | 61.57 | 61.69 | 64.67 | 60.42 | 60.25 | 58.69 | 61.53 | 62.12 | 63.33 | 59.18 | 63.31 | 67.42 | 67.66 | 68.41 | 69.28 | 20.78 | 21.34 | 21.29 | 21.18 | 20.33 | 19.33 | 20.17 | 19.71 | 24.70 | 23.82 | 25.03 | 24.21 | 23.95 | 22.02 | 24.33 | 24.77 | 25.67 | 23.99 | 23.66 | 23.12 | 24.69 | 22.56 | 20.40 | 20.17 | 19.61 | 19.41 | |
| TiO ₂ | 0.01 | 0 | 0.05 | 0.04 | 0.04 | 0.02 | 0.01 | 0.00 | 0.01 | 0.04 | 0.02 | 0.00 | 0.02 | 0.01 | 0.01 | 0.02 | 0.03 | 0.00 | 0.00 | 0.00 | 0.02 | 0.03 | 0.02 | 0.02 | 0.01 | 0.00 | 0.00 | 0.14 | 0.14 | 0.11 | 0.06 | 0.13 | 0.04 | 0.13 | 0.09 | 0.03 | 0.06 | 0.01 | 0.05 | 0.04 | 0.12 | 0.00 | 0.02 | 0.11 | 0.08 | 0.11 | 0.05 | 1.34 | 0.02 | 0.01 | 0.01 | 0.04 | 0.02 |
| MnO | 0.03 | 0.03 | 0.02 | 0.01 | 0.01 | 0.00 | 0.02 | 0.04 | 0.03 | 0.00 | 0.00 | 0.00 | 0.01 | 0.00 | 0.00 | 0.00 | 0.05 | 0.02 | 0.01 | 0.02 | 0.05 | 0.02 | 0.00 | 0.03 | 0.03 | 0.02 | 0.02 | 0.01 | 0 | 0.01 | 0.01 | 0.00 | 0.01 | 0.00 | 0.01 | 0.00 | 0.01 | 0.01 | 0.02 | 0.01 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.01 | 0.00 | 0.01 | 0.01 | 0.01 | 0.01 | 0.00 | 0.00 |
| MgO | 0.01 | 0 | 0.02 | 0 | 0.01 | 0.01 | 0.00 | 0.01 | 0.00 | 0.01 | 0.01 | 0.01 | 0.02 | 0.01 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.01 | 0.01 | 0.01 | 0.00 | 2.29 | 2.63 | 2.86 | 2.44 | 1.93 | 0.41 | 1.74 | 1.24 | 6.74 | 5.60 | 7.12 | 5.91 | 5.73 | 3.52 | 6.44 | 6.66 | 7.75 | 5.80 | 5.56 | 4.63 | 6.46 | 4.23 | 1.69 | 1.32 | 0.35 | 0.46 | |
| Na ₂ O | 10.16 | 9.77 | 9.65 | 9.80 | 9.62 | 10.93 | 9.56 | 9.91 | 7.69 | 8.31 | 7.52 | 8.22 | 8.13 | 9.38 | 7.49 | 7.64 | 7.25 | 8.20 | 8.22 | 8.79 | 7.23 | 9.06 | 10.35 | 10.81 | 11.29 | 11.28 | 0.46 | 0.4 | 0.52 | 0.35 | 0.75 | 0.16 | 1.10 | 1.23 | 0.15 | 0.23 | 0.22 | 0.12 | 0.13 | 0.19 | 0.34 | 0.16 | 0.13 | 0.19 | 0.17 | 0.20 | 0.50 | 0.20 | 0.10 | 0.12 | 0.14 | 0.11 | |
| Total | 99.78 | 100.07 | 99.70 | 99.66 | 99.84 | 99.66 | 99.37 | 99.48 | 100.01 | 100.11 | 99.53 | 100.09 | 99.71 | 99.93 | 99.03 | 99.52 | 99.68 | 99.81 | 99.86 | 100.16 | 99.56 | 99.41 | 100.00 | 100.14 | 99.88 | 100.49 | Recalculated on 8 oxygens | | | | | | | | | | | | | | | | | | | | | | | | | | |
| Ca | 0.108 | 0.124 | 0.136 | 0.115 | 0.091 | 0.019 | 0.083 | 0.059 | 0.321 | 0.266 | 0.342 | 0.282 | 0.273 | 0.166 | 0.310 | 0.319 | 0.372 | 0.276 | 0.187 | 0.219 | 0.312 | 0.202 | 0.079 | 0.062 | 0.016 | 0.201 | 0.869 | 0.832 | 0.827 | 0.838 | 0.819 | 0.927 | 0.820 | 0.848 | 0.663 | 0.714 | 0.653 | 0.706 | 0.701 | 0.802 | 0.652 | 0.661 | 0.630 | 0.707 | 0.708 | 0.753 | 0.631 | 0.780 | 0.878 | 0.916 | 0.957 | 0.941 | |
| Na | 0.026 | 0.022 | 0.029 | 0.020 | 0.042 | 0.009 | 0.062 | 0.070 | 0.009 | 0.013 | 0.012 | 0.007 | 0.007 | 0.010 | 0.020 | 0.009 | 0.007 | 0.011 | 0.009 | 0.011 | 0.029 | 0.011 | 0.006 | 0.007 | 0.008 | 0.006 | 10.8 | 12.6 | 13.7 | 11.8 | 9.6 | 2.0 | 8.6 | 6.0 | 32.3 | 26.8 | 33.9 | 28.3 | 27.8 | 17.0 | 31.6 | 32.2 | 36.9 | 27.8 | 20.7 | 22.3 | 32.1 | 20.3 | 8.2 | 6.3 | 1.7 | 2.2 | |
| Ab | 86.6 | 84.8 | 83.3 | 86.1 | 86.0 | 97.1 | 85.0 | 86.9 | 66.8 | 71.9 | 64.9 | 71.0 | 71.5 | 81.9 | 66.4 | 66.9 | 62.4 | 71.1 | 78.3 | 76.6 | 64.9 | 78.6 | 91.2 | 93.0 | 97.5 | 97.2 | 2.6 | 2.6 | 3.0 | 2.1 | 4.4 | 0.9 | 6.4 | 7.1 | 0.9 | 1.3 | 1.2 | 0.7 | 0.7 | 1.1 | 2.0 | 0.9 | 0.7 | 1.1 | 1.0 | 1.1 | 3.0 | 1.1 | 0.6 | 0.7 | 0.8 | 0.6 | |
| Or | 2.6 | 2.6 | 3.0 | 2.1 | 4.4 | 0.9 | 6.4 | 7.1 | 0.9 | 1.3 | 1.2 | 0.7 | 0.7 | 1.1 | 2.0 | 0.9 | 0.7 | 1.1 | 1.0 | 1.1 | 3.0 | 1.1 | 0.6 | 0.7 | 0.8 | 0.6 | *; total Fe as FeO | | | | | | | | | | | | | | | | | | | | | | | | | | |

Table 4. Chemical compositions of alkali feldspars
a) Perthitic orthoclases in the Cretaceous Sogrisan granites

| Rock | KPGR | | | | | | | | KBGR | | | |
|--------------------------------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|
| | K0101 | | K0104 | | K1406 | | K1408 | | K1606 | | K1613 | |
| | af | pl | af | pl | af | pl | af | pl | af | pl | af | pl |
| SiO ₂ | 65.20 | 69.04 | 66.27 | 68.23 | 64.92 | 68.21 | 66.23 | 67.85 | 64.62 | 69.13 | 65.18 | 68.25 |
| Al ₂ O ₃ | 18.01 | 18.77 | 17.97 | 19.67 | 18.10 | 19.40 | 18.38 | 19.90 | 17.72 | 19.00 | 18.11 | 19.34 |
| TiO ₂ | 0.01 | 0.00 | 0.05 | 0.03 | 0.01 | 0.00 | 0.04 | 0.00 | 0.02 | 0.00 | 0.06 | 0.02 |
| FeO* | 0.06 | 0.04 | 0.03 | 0.03 | 0.08 | 0.05 | 0.05 | 0.04 | 0.00 | 0.04 | 0.03 | 0.00 |
| MnO | 0.03 | 0.00 | 0.02 | 0.04 | 0.00 | 0.06 | 0.03 | 0.03 | 0.00 | 0.00 | 0.08 | 0.02 |
| CaO | 0.04 | 0.02 | 0.04 | 0.89 | 0.00 | 0.62 | 0.01 | 1.09 | 0.00 | 0.01 | 0.00 | 0.56 |
| MgO | 0.01 | 0.01 | 0.00 | 0.01 | 0.01 | 0.00 | 0.01 | 0.00 | 0.00 | 0.01 | 0.00 | 0.01 |
| Na ₂ O | 0.68 | 7.88 | 1.72 | 10.44 | 0.61 | 11.24 | 2.39 | 10.60 | 0.20 | 11.17 | 0.25 | 10.59 |
| K ₂ O | 15.42 | 4.91 | 13.72 | 0.29 | 15.52 | 0.30 | 12.8 | 0.24 | 16.43 | 0.10 | 16.28 | 0.60 |
| Total | 99.46 | 100.67 | 99.82 | 99.63 | 99.25 | 99.88 | 100.03 | 99.75 | 99.19 | 99.55 | 99.99 | 99.39 |
| Recalculated on 8 oxygens | | | | | | | | | | | | |
| Ca | 0.0021 | 0.0010 | 0.0020 | 0.0417 | 0.0000 | 0.0293 | 0.0003 | 0.0509 | 0.0000 | 0.0045 | 0.0000 | 0.0263 |
| Na | 0.0610 | 0.6710 | 0.1501 | 0.8867 | 0.0545 | 0.9549 | 0.2114 | 0.9007 | 0.0185 | 0.9471 | 0.0224 | 0.9027 |
| K | 0.9102 | 0.2750 | 0.8030 | 0.0162 | 0.9182 | 0.0165 | 0.7486 | 0.0133 | 0.9760 | 0.0053 | 0.9595 | 0.0335 |
| An | 0.2 | 0.1 | 0.2 | 4.4 | 0.0 | 2.9 | 0.0 | 5.3 | 0.0 | 0.4 | 0.0 | 2.7 |
| Ab | 6.3 | 70.9 | 15.7 | 93.9 | 5.6 | 95.4 | 22.0 | 93.3 | 1.9 | 99.0 | 2.3 | 93.8 |
| Or | 93.5 | 29.0 | 84.1 | 1.7 | 94.4 | 1.7 | 78.0 | 1.4 | 98.1 | 0.6 | 97.7 | 3.5 |

*, total Fe as FeO

b) Microclines in Jurassic Poemun granodiorite

| Rock | JPGD | | | | | | JMFD | |
|--------------------------------|--------|--------|--------|--------|--------|--------|--------|--------|
| | J0209A | J1715 | J1716 | J1719A | J1720 | J0713 | J0209B | J0716B |
| SiO ₂ | 65.04 | 65.49 | 65.51 | 65.40 | 65.69 | 65.06 | 65.72 | 65.90 |
| Al ₂ O ₃ | 18.38 | 18.12 | 18.12 | 18.11 | 18.30 | 18.02 | 18.14 | 17.91 |
| TiO ₂ | 0.05 | 0.00 | 0.02 | 0.06 | 0.03 | 0.04 | 0.01 | 0.01 |
| FeO* | 0.02 | 0.02 | 0.06 | 0.03 | 0.00 | 0.06 | 0.03 | 0.01 |
| MnO | 0.02 | 0.02 | 0.00 | 0.01 | 0.03 | 0.01 | 0.02 | 0.01 |
| CaO | 0.00 | 0.00 | 0.01 | 0.02 | 0.06 | 0.01 | 0.00 | 0.03 |
| Na ₂ O | 0.79 | 0.86 | 0.83 | 1.05 | 1.58 | 0.96 | 0.48 | 0.57 |
| K ₂ O | 15.33 | 15.23 | 15.17 | 15.1 | 14.34 | 15.15 | 15.72 | 15.85 |
| Total | 99.63 | 99.73 | 99.73 | 99.78 | 100.03 | 99.31 | 100.34 | 100.35 |
| Recalculated on 8 oxygens | | | | | | | | |
| Ca | 0.0000 | 0.0002 | 0.0005 | 0.0007 | 0.0030 | 0.0006 | 0.0000 | 0.0015 |
| Na | 0.0708 | 0.0769 | 0.0744 | 0.0936 | 0.1399 | 0.0858 | 0.0427 | 0.0503 |
| K | 0.0931 | 0.8952 | 0.8910 | 0.8871 | 0.8391 | 0.8968 | 0.9322 | 0.9278 |
| An | 0.0 | 0.0 | 0.1 | 0.1 | 0.3 | 0.1 | 0.0 | 0.2 |
| Ab | 7.3 | 7.9 | 7.7 | 9.5 | 14.2 | 8.7 | 4.4 | 5.1 |
| Or | 92.7 | 92.1 | 92.2 | 90.4 | 85.5 | 91.2 | 95.6 | 94.7 |

that of the JPGD was about 250°C at 1 to 3 kbar. These results, however, cannot give any useful informations on the magmatic crystallization

because their temperatures are considerably lower than those expected for solidus of granitic magma.

Table 5. Chemical compositions of ilmenites

| Rock | JPGD | | | | | KPGR | | | |
|--------------------------------|--------|--------|--------|--------|--------|--------|--------|--------|--------|
| | J0209A | J1715 | J1716 | J1718 | J1720 | K0101 | K0104 | K1406 | K1408 |
| SiO ₂ | 0.05 | 0.02 | 0.04 | 0.04 | 0.02 | 0.08 | 0.02 | 0.03 | 0.06 |
| Al ₂ O ₃ | 0.04 | 0.01 | 0.01 | 0.03 | 0.01 | 0.01 | 0.01 | 0.04 | 0.03 |
| TiO ₂ | 52.14 | 52.37 | 52.13 | 52.35 | 51.48 | 50.10 | 50.09 | 49.77 | 52.97 |
| FeO* | 37.79 | 41.97 | 44.44 | 42.41 | 45.13 | 29.78 | 38.58 | 32.84 | 35.56 |
| MnO | 9.16 | 5.18 | 2.79 | 4.20 | 3.13 | 16.85 | 9.59 | 11.71 | 7.14 |
| MnO | 0.04 | 0.02 | 0.05 | 0.01 | 0.01 | 0.01 | 0.00 | 0.01 | 0.01 |
| CaO | 0.02 | 0.01 | 0.22 | 0.16 | 0.02 | 0.02 | 0.05 | 0.00 | 0.01 |
| Na ₂ O | 0.00 | 0.00 | 0.00 | 0.01 | 0.00 | 0.00 | 0.01 | 0.00 | 0.01 |
| Na ₂ O | 0.05 | 0.01 | 0.01 | 0.01 | 0.01 | 0.01 | 0.01 | 0.03 | 0.00 |
| V ₂ O ₃ | 0.28 | 0.14 | 0.21 | 0.18 | 0.25 | 0.21 | 0.19 | 0.21 | 0.25 |
| Cr ₂ O ₃ | 0.01 | 0.01 | 0.03 | 0.00 | 0.01 | 0.00 | 0.03 | 0.00 | 0.02 |
| Total | 99.27 | 99.59 | 99.69 | 99.22 | 99.81 | 96.86 | 98.36 | 94.43 | 95.79 |
| Recalculated on 3 oxygens | | | | | | | | | |
| Si | 0.0013 | 0.0006 | 0.0009 | 0.0010 | 0.0005 | 0.0020 | 0.0005 | 0.0007 | 0.0016 |
| Ti | 0.9930 | 0.9926 | 0.9877 | 0.9781 | 0.9602 | 0.9744 | 0.9621 | 0.9975 | 1.0489 |
| Fe | 0.8004 | 0.8846 | 0.9364 | 0.8811 | 0.9360 | 0.6441 | 0.8240 | 0.7319 | 0.7830 |
| Mg | 0.0014 | 0.0060 | 0.0017 | 0.0049 | 0.0059 | 0.0005 | 0.0000 | 0.0002 | 0.0004 |
| Ca | 0.0005 | 0.0003 | 0.0060 | 0.0421 | 0.0007 | 0.0050 | 0.0013 | 0.0000 | 0.0002 |
| Mn | 0.1966 | 0.1107 | 0.0596 | 0.0884 | 0.0657 | 0.3692 | 0.2074 | 0.2639 | 0.1592 |
| Al | 0.0011 | 0.0004 | 0.0030 | 0.0009 | 0.0002 | 0.0004 | 0.0003 | 0.0012 | 0.0009 |
| V | 0.0056 | 0.0029 | 0.0043 | 0.0036 | 0.0300 | 0.0044 | 0.0039 | 0.0045 | 0.0053 |
| Cr | 0.0002 | 0.0020 | 0.0005 | 0.0000 | 0.0009 | 0.0001 | 0.0005 | 0.0000 | 0.0004 |

*; total Fe as FeO

Ilmenite

Anhedral ilmenite in the JPGD occurs in euhedral sphene as inclusion, and that in the KSGR accompanies magnetite. The chemical compositions of ilmenite are shown in Table 5. Total Fe contents of ilmenite in the KPGR are lower, and Mn contents higher than those in the JPGD (Fig. 8).

Muscovite

Muscovite is mainly observed in core of plagioclase in the JPGD, and anhedral muscovite (about 2.4 vol.%) in felsic dike (JMFD) shows interstitial feature. Anhedral muscovite in KSGR is observed in large quartz grains. Table 6 displays the chemical compositions and structural formulae of muscovite. The F/(F+OH) ratios of

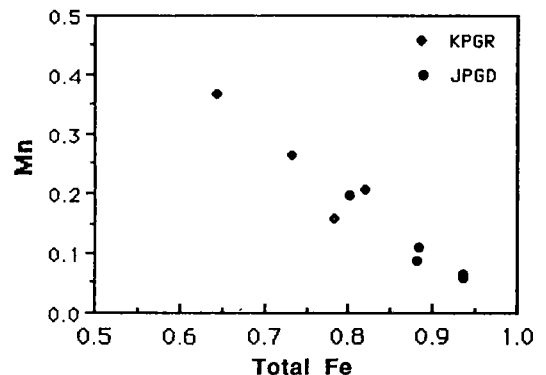


Fig. 8. Mn contents plotted against total Fe of ilmenite.

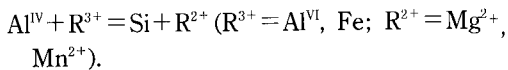
muscovite in the KSGR are higher and the Mg/(Mg + Fe) ratios are lower than those in the JPGD, which are similar to biotite compositions. It seems that muscovites have experienced the celadonite substitution, that is, octahedral Al is replaced by

Table 6. Chemical compositions and structural formulae of muscovites

| Rock | KSGR | | | JPGD | | | | | JMFD | |
|--------------------------------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|
| | K1406 | K1504 | K1606 | J0209A | J1716 | J1718 | J0713 | J0715 | J0209B | J0716B |
| SiO ₂ | 47.88 | 47.78 | 47.57 | 49.40 | 49.00 | 49.46 | 49.44 | 49.83 | 47.32 | 48.55 |
| Al ₂ O ₃ | 34.05 | 32.43 | 36.64 | 32.00 | 37.03 | 32.33 | 32.59 | 31.54 | 34.45 | 30.81 |
| TiO ₂ | 0.06 | 0.25 | 0.01 | 0.45 | 0.00 | 0.10 | 0.02 | 0.35 | 0.39 | 0.47 |
| FeO* | 4.81 | 5.57 | 0.83 | 2.94 | 0.09 | 2.62 | 3.05 | 3.79 | 2.81 | 5.36 |
| MnO | 0.20 | 0.20 | 0.14 | 0.05 | 0.00 | 0.00 | 0.06 | 0.02 | 0.03 | 0.02 |
| MgO | 0.41 | 0.41 | 0.01 | 1.30 | 0.08 | 1.74 | 0.95 | 1.54 | 0.47 | 1.01 |
| CaO | 0.00 | 0.00 | 0.01 | 0.01 | 0.03 | 0.02 | 0.00 | 0.00 | 0.01 | 0.01 |
| Na ₂ O | 0.19 | 0.45 | 0.60 | 0.21 | 0.12 | 0.14 | 0.21 | 0.22 | 0.40 | 0.18 |
| K ₂ O | 7.83 | 8.57 | 9.62 | 9.70 | 9.56 | 10.20 | 10.06 | 9.18 | 9.66 | 9.71 |
| Cl | 0.00 | 0.01 | 0.00 | | 0.01 | 0.00 | 0.00 | 0.01 | | 0.01 |
| =O | 0.00 | 0.00 | 0.00 | | 0.00 | 0.00 | 0.00 | 0.00 | | 0.00 |
| F | 0.96 | 1.52 | 1.19 | | 0.08 | 0.09 | 0.14 | 0.13 | | 0.33 |
| =O | -0.40 | -0.64 | -0.50 | | -0.03 | -0.04 | -0.06 | -0.06 | | -0.14 |
| Total | 95.99 | 96.55 | 96.12 | 96.06 | 95.97 | 96.66 | 96.46 | 96.55 | 95.54 | 96.32 |
| Recalculated on 22 oxygens | | | | | | | | | | |
| Si | 6.2821 | 6.3202 | 6.2373 | 6.4695 | 6.3390 | 6.4582 | 6.4704 | 6.4993 | 6.2470 | 6.4229 |
| Al IV | 1.7179 | 1.6798 | 1.7627 | 1.5305 | 1.6610 | 1.5418 | 1.5296 | 1.5007 | 1.7530 | 1.5771 |
| VI | 3.5481 | 3.3773 | 3.8997 | 3.4296 | 3.9847 | 3.4315 | 3.4965 | 3.3486 | 3.6078 | 3.2270 |
| Ti | 0.0054 | 0.0248 | 0.0007 | 0.0439 | 0.0003 | 0.0097 | 0.0019 | 0.0345 | 0.0387 | 0.0467 |
| Fe | 0.4752 | 0.5554 | 0.0816 | 0.2899 | 0.0082 | 0.2579 | 0.3003 | 0.3725 | 0.2791 | 0.5332 |
| Mn | 0.0226 | 0.0219 | 0.0159 | 0.0055 | 0.0000 | 0.0000 | 0.0063 | 0.0025 | 0.0030 | 0.0026 |
| Mg | 0.0798 | 0.0815 | 0.0011 | 0.2529 | 0.0150 | 0.3391 | 0.1853 | 0.2984 | 0.0925 | 0.1997 |
| Ca | 0.0003 | 0.0000 | 0.0007 | 0.0020 | 0.0036 | 0.0023 | 0.0006 | 0.0040 | 0.0014 | 0.0018 |
| Na | 0.0940 | 0.1146 | 0.1512 | 0.0528 | 0.0311 | 0.0348 | 0.0529 | 0.0549 | 0.1027 | 0.0452 |
| K | 1.3113 | 1.4464 | 1.6092 | 1.6210 | 1.5778 | 1.6992 | 1.6797 | 1.5269 | 1.6275 | 1.6390 |
| Cl | 0.0008 | 0.0014 | 0.0000 | | 0.0013 | 0.0000 | 0.0004 | 0.0017 | | 0.0019 |
| F | 0.3985 | 0.6370 | 0.4947 | | 0.0330 | 0.0359 | 0.0575 | 0.0055 | | 0.1385 |
| OH | 3.6007 | 3.3616 | 3.5053 | | 3.9657 | 3.9641 | 3.9421 | 3.9928 | | 3.8596 |

*; total Fe as FeO

Mg and Fe, and tetrahedral Al by Si (Fig. 9). The substitutional correlation among these elements can be deduced as follows (from Konings *et al.*, 1988) :



From the TiO₂-FeO* (total Fe)-MgO triangular diagram (Fig. 10), muscovites of the granitic rocks in the study area generally have low TiO₂ contents, but those of KSGR have relatively high FeO* contents. Compared with the chemical compositions of muscovites in leucocratic granite of Millevaches massif studied by Monier *et al.* (1984), muscovites in the study area have been

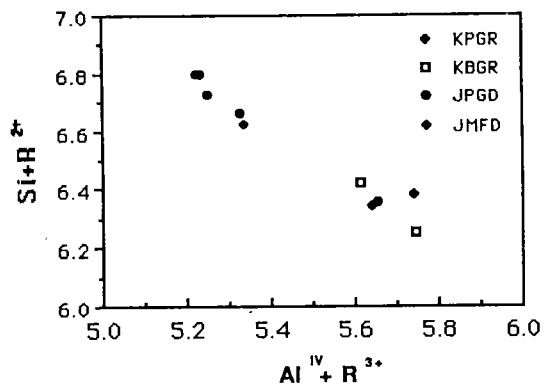


Fig. 9. (Si+R²⁺) plotted against (Al^{IV}+R³⁺) of muscovites.

formed by secondary alteration at subsolidus state.

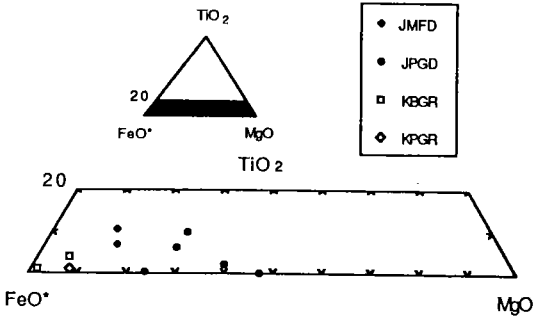


Fig. 10. Triangular diagram of TiO₂-FeO* (total Fe)-MgO contents of muscovites.

DISCUSSION

The characteristic quartz-feldspar intergrowth texture in the granitic rocks in the study area appears in two modes: myrmekitic in the JPGD and micrographic in the KSGR. In the Mesozoic granitic rocks in Korea, the myrmekitic texture appears in the Jurassic granites and the micrographic texture in the Cretaceous ones (for example, Iiyama and Fonteilles, 1981). The petrographical observations for the granitic rocks in the study area corroborates the previous reports. The different quartz-feldspar intergrowth in the Mesozoic granites in Korea indicates the different emplacement condition in two granitic bodies of different age. Generally the Jurassic granitic pluton emplaced deeper part of the crust than the Cretaceous one (Cho and Kwon, 1994). Thus the different intergrowth texture is likely to be controlled by the depth, more precisely by the decompressing and cooling path at the level the magma has emplaced.

In the Cretaceous granitic rocks, the micrographic texture represents the simultaneous crystallization of quartz and alkali feldspar under the undercooling with isothermal decompression (Lee, 1991). The exsolution or degassing of vapour phases, evidenced from the existence of miarolitic cavity or pegmatitic pocket, increases the degree of undercooling. In addition, the spherulitic texture of quartz-feldspar intergrowth sometimes occurs in the KSGR. This texture is generally

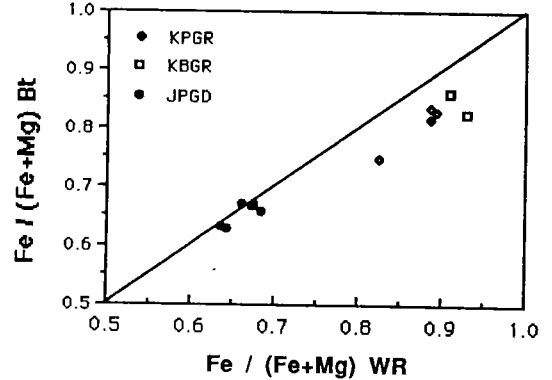


Fig. 11. Fe/(Mg+Fe) of biotites plotted against Fe/(Mg+Fe) of the host granitic rocks.

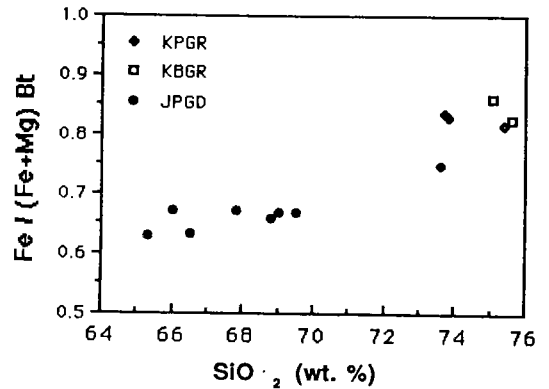


Fig. 12. Fe/(Mg+Fe) of biotites plotted against SiO₂ (wt.%) of the host granitic rocks.

formed by the rapid crystallization from the viscous granitic magma. These petrographic results suggest that the KSGR has rapidly solidified at the shallower level. The myrmekitic texture can be formed in three ways : (1) from the volume for volume replacement of the marginal parts of alkali feldspar in contact with plagioclase (Shelly, 1964), (2) from the micropressure quenching during the separation of an aqueous phase as crystallization progresses (Hibbard, 1979), and (3) from the substitution for alkali feldspar induced by deformation at subsolidus state (Simpson and Wintsch, 1989). The replacement with co-precipitation of quartz cannot wholly be explained by exsolution from the host alkali feldspar, as it requires a net gain of sodium and calcium with a loss of pota-

ssium from the system. The latter two ways need local tectonic events such as deformation for the formation of the intergrowth. In the JPGD, the myrmekitic texture is easily observed along the marginal part of microcline in the deformed samples. Though we have no evidence whether the role of the deformation is direct or indirect, it seems like that the deformation has more or less affected the formation of the intergrowth. Nevertheless the origin of the texture is still uncertain and more studies are needed. Alkali feldspar is microcline in the JPGD and perthitic orthoclase in the KSGR. Since microcline is more stable phase at the lower temperature field than the perthitic orthoclase, the occurrence of microcline in the JPGD may indicate that the JPGD has more slowly solidified at the deeper level than the KSGR.

Compared with whole-rock chemistry (Jwa *et al.*, in prep.), the Fe/(Fe+Mg) ratios in biotite of the KSGR are lower than those of whole-rock, and those of the JPGD are nearly the same as those of whole-rock (Fig. 11). This discrepancy of Fe/(Fe+Mg) ratios between mineral and whole-rock in the KSGR would have been caused by fractional crystallization of mafic minerals including magnetite (Tsusue *et al.*, 1984). Thus it seems likely that the KSGR has been crystallized under more oxidizing conditions than the JPGD. Furthermore, the Fe/(Fe+Mg) ratios of biotite in the JPGD slightly increase with the increase of whole-rock SiO₂ contents, whereas those in the KSGR are dispersed (Fig. 12). This is due to the crystallization of biotite in the JPGD under buffered condition of oxygen fugacity (Wones and Eugster, 1965; Czamanske *et al.*, 1981).

The compositional differences of ilmenite are caused by the condition of crystallization, especially by the oxygen fugacity. Ilmenite in the KPGR has lower total Fe contents and higher Mn contents than that in the JPGD. Czamanske and Mihalik (1972) reported that ilmenite crystallized under more oxidizing condition is enriched in Mn and depleted in Fe. Therefore, ilmenite of the

KPGR accompanied by magnetite had crystallized under more oxidizing condition than that of the JPGD.

The whole-rock magnetic susceptibility of the granitic rocks was measured by Kappameter KT-5. The magnetic susceptibility of the KSGR ranges from 33 to 144 in 10⁻⁶ emu/g, which is higher than that of the JPGD (9~12). In the KSGR, the values of the KGPH, KPGR and KBGR are 33 to 80, 108 to 114 and 90 to 144 in 10⁻⁶ emu/g, respectively. Ishihara (1977) suggested the boundary value of the whole-rock magnetic susceptibility between magnetite-series and ilmenite-series granitoids is about 100 in 10⁻⁶ emu/g. However, the reported boundary values in some granitic masses are generally lower than the suggested one. For example, the boundary value for the Inje-Hongcheon granitic mass is 35 (Jwa, 1990) and that for the Tokuwa batholith is 45 in 10⁻⁶ emu/g (Shimizu, 1986). The higher magnetic susceptibility values represent that the granitic rocks belongs to the magnetite-series granitoids and emplaced in more oxidizing environment. Considering the boundary value of 35~45 in 10⁻⁶ emu/g, the KSGR generally belongs to the magnetite-series granitoid, whereas the JPGD to ilmenite-series. Thus the KSGR magma was solidified under the more oxidation environment than the JPGD magma. The whole-rock magnetic susceptibility data corroborate the result from the chemistry of biotite and ilmenite.

CONCLUSIONS

The studies on petrology and mineralogy of the granitic rocks in Poenu-Sogrisan area lead to the following conclusions.

1) The granitic rocks in study area are composed of Jurassic Poenu granodiorite and Cretaceous Sogrisan granitic rocks. The latter can be divided into coarse-grained biotite granite, porphyritic biotite granite and granite porphyry.

2) From the petrographical observations, it can be said that the Sogrisan granitic rocks, compared with the Poenu granodiorite, had been formed

from shallower-level emplaced and more rapidly crystallized granitic magma.

3) The F, Cl contents and the Fe/(Fe+Mg) ratios of biotite and muscovite in the Sogrisan granitic rocks are higher than those in the Poemun granodiorite. The An contents of plagioclase in the Poemun granodiorite are higher than in the Sogrisan granitic rocks. Alkali feldspar occurs as microcline in the Poemun granodiorite and as perthitic orthoclase in the Sogrisan granitic rocks. Ilmenite in the Sogrisan granitic rocks is more enriched in Mn and depleted in Fe than that in the Poemun granodiorite.

4) The Sogrisan granitic rocks had been solidified under more oxidation environment than the Poemun granodiorite, judging from the chemical compositions of minerals and the whole-rock magnetic susceptibility measurements.

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보은-속리산 지역의 화강암류에 대한 암석기재 및 광물화학

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요약 : 보은-속리산 지역의 화강암류는 주라기 보은 화강섬록암과 백악기 속리산 화강암류로 구성된다. 속리산 화강암류는 다시 세 가지의 암상, 즉 조립질 흑운모 화강암, 반상 흑운모 화강암 및 화강반암으로 세분된다. 암석에 대한 기재적인 관찰로부터, 속리산 화강암류는 보은 화강섬록암에 비해 천소에 관입하고 보다 빠르게 고결된 화강암질 마그마로부터 생성되었음을 알 수 있다. 광물화학에서는 흑운모와 백운모의 F, Cl 함량과 Fe/(FeMg) 비가 보은 화강섬록암에 비해 속리산 화강암류에서 높게 나타나며, 사장석의 아노르사이트 함량은 보은 화강섬록암에서 높게 나타난다. 티탄철석의 경우 속리산 화강암류가 보은 화강섬록암보다 Mn이 풍부하고 Fe가 결핍된 경향을 보인다. 전암 대자율의 측정 결과는 속리산 화강암류가 $33 \sim 144 \times 10^{-6}$ emu/g로 보은 화강섬록암의 $9 \sim 12 \times 10^{-6}$ emu/g보다 높게 나타났으며, 이로부터 속리산 화강암류가 주로 자철석계열의 화강암류에, 보은 화강섬록암이 티탄철석계열의 화강암류에 속함을 알 수 있다. 전암 대자율의 결과와 흑운모 및 티탄철석의 화학조성 등은 속리산 화강암류가 보은 화강섬록암에 비해 보다 산화적인 환경에서 고결되었음을 지시한다.

핵심어 : 보은 화강섬록암, 속리산 화강암류, 암석기재, 광물화학, 전암 대자율