

# Further Study on the Constituents of *Allium tuberosum* leaves

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In the course of continuous work on the leaves of *Allium tuberosum* (Liliaceae), compounds 1 and 2 were isolated from the ethylacetate and butanol-soluble fraction, and identified as N-*p*-coumaroyl tyramine and bis(*p*-hydroxyphenyl) ether, respectively, on the basis of spectral data and physicochemical results.

**Key words :** *Allium tuberosum*, bis(*p*-hydroxyphenyl) ether, N-*p*-coumaroyl tyramine

## INTRODUCTION

*Allium tuberosum* Rottler (Liliaceae) is a perennial herb which is cultivated widely and the leaves are used for food. According to the dictionary of Chinese drugs (Shanghai Science & Technological Publisher, 1985), they have been used for treatment of abdominal pain, diarrhea, hematemesis, snakebite and asthma. In the previous papers (Choi *et al.*, 1988, 1992), the isolation of amino acids, adenosine and  $\beta$ -carboline alkaloid was reported. In the course of continuous work on this plant part, additional two compounds were isolated. This paper deals with the isolation and characterization of the compounds.

## MATERIALS AND METHODS

The mps were taken on an Electrothermal digital melting point apparatus and are uncorrected. The IR spectrum was determined in KBr tablet on a Shimadzu IR-400 spectrophotometer. The  $^1\text{H-NMR}$  and  $^{13}\text{C-NMR}$  spectra were recorded with a Bruker-AM 300 spectrometer; Chemical shifts are given on a  $\delta$  (ppm) scale with tetramethylsilane. The EIMS spectra were taken with a Hewlett-Packard 5985B GC/MS spectrometer operating at 70eV.

### Isolation

This was carried out as described previously (Choi *et al.*, 1992).

The EtOAc soluble fraction (10 g) was subjected to silica gel column chromatography (solvent: EtOAc) to yield compound 1 (150 mg). The BuOH soluble frac-

tion (60 g) was subjected to silica gel column chromatography (solvent:  $\text{CHCl}_3$ -MeOH (gradient)) to yield compound 2 (120 mg).

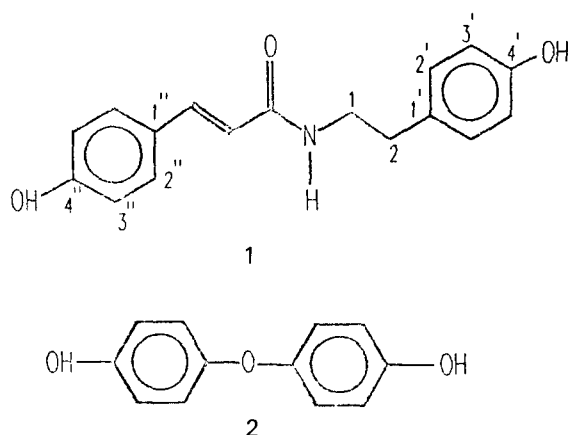
**Compound 1 :** Yield:  $1.5 \times 10^{-1}\%$  (dry weight), Amorphous powder from MeOH,  $\text{FeCl}_3$ ; +, mp; 235~236°C, MS ( $m/z$ , %); 283 ( $M^+$ , 2.0), 164 ( $\text{C}_9\text{O}_3\text{H}_8$ , 65.7), 147 ( $\text{C}_9\text{O}_2\text{H}_7$ , 100), 120 ( $\text{C}_8\text{OH}_6$ , 51.9),  $^1\text{H-NMR}$  (DMSO- $d_6$ , 300 MHz)  $\delta$ ; tyramine moiety 2.60 (2H, t,  $J=7.40$  Hz, H-1), 3.31 (2H, t,  $J=7.40$  Hz, H-2), 6.78 (2H, d,  $J=8.60$  Hz, H-3' & 5'), 7.38 (2H, d,  $J=8.60$  Hz, H-2' & 6'), coumaroyl moiety 6.36 (1H, d,  $J=15.70$  Hz, H- $\alpha$ ), 6.66 (2H, dd,  $J=8.40$  & 2.0 Hz, H-3'' & 5''), 7.00 (2H, dd,  $J=8.40$  & 2.0 Hz, H-2'' & 6''), 7.29 (1H, d,  $J=15.70$  Hz, H- $\beta$ ),  $^{13}\text{C-NMR}$  (DMSO- $d_6$ , 75.5 MHz)  $\delta$ ; tyramine moiety 34.39 (C-2), 40.62 (C-1), 115.86 (C-3' & 5'), 129.38 (C-2' & C-6'), 129.48 (C-1'), 155.56 (C-4'), coumaroyl moiety 115.04 (C-3'' & C-5''), 118.73 (C- $\alpha$ ), 125.89 (C-1''), 129.08 (C-2'' & 6''), 138.49 (C- $\beta$ ), 158.70 (C-4''), 165.23 (C=O).

**Compound 2 :** Yield:  $1.2 \times 10^{-1}\%$  (dry weight), Amorphous prism from aqueous-MeOH, mp; 340.5°C (decomp.), IR (KBr,  $\text{cm}^{-1}$ ); 3019 (OH), 2464, 1992, 1680 (aromatic), 1451, 1461, 1391, 1226, 1098, 997, 846, 760, 584, 563, 542, 431, MS ( $m/z$ , %); 202 ( $M^+$ , 5.5), 112 (96.0), 69 (100),  $^1\text{H-NMR}$  (DMSO- $d_6$ , 300 MHz)  $\delta$ ; 10.97 (1H, br. s, OH), 10.78 (1H, br. s, OH), 7.37 (2H, d,  $J=7.60$  Hz), 5.44 (2H, d,  $J=7.60$  Hz),  $^{13}\text{C-NMR}$  (DMSO- $d_6$ , 75.5 MHz)  $\delta$ ; 164.24, 151.43, 142.04 ( $\times 2$ ), 100.16 ( $\times 2$ ).

## RESULTS AND DISCUSSION

An EtOAc-soluble fraction of the leaves of *A. tuberosum* was repeatedly chromatographed over silica gel to afford compound 1 as amorphous powder, mp 235~236°C. The mass spectrum of compound 1 showed a molecular ion peak at  $m/z$  283 cor-

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responding to  $C_{17}H_{17}NO_3$  and the  $^{13}C$ -NMR spectrum showed 13 unique types of carbon; one  $\alpha,\beta$ -unsaturated carbonyl carbon, eight aromatic carbons, two methylene carbons and two vinylic carbons. Since the formula of compound **1** shows 17 carbon atoms, four of eight aromatic carbons must be an element of symmetry. The  $^1H$ -NMR spectrum of **1** showed two ortho-coupled doublets each of two protons with a J value of 8.6 Hz at  $\delta$  7.38 and 6.78 ppm, indicating the presence of a 1,4-di-substituted benzene ring and two trans-coupled vinylic protons at  $\delta$  6.36 and 7.29 with a J value of 15 Hz. The signals at  $\delta$  7.00 and 6.66 were assignable to the two ortho-coupled protons of another aromatic ring, respectively, which were long range coupled to the vinylic proton as a double doublet with J values of 8.4 and 2.0 Hz. Two triplets at 3.31 and 2.60 could be assignable to two methylene protons. These data together with the appearance of intense peaks at  $m/z$  147 ( $C_9O_2H_7$ , 100) and 120 ( $C_8OH_8$ , 51.9) suggested the compound **1** to be N-*p*-coumaroyl tyramine. It was further confirmed by comparison of  $^{13}C$ -NMR spectral data with those reported in the literature (Zhao *et al*, 1992). The presence of this compound in the plants has previously been reported in the tuber of *Allium bakeri* (Okuyama *et al*, 1986), *Asimina triloba* (Zhao *et al*, 1992) and *Solanum melongena* (Yoshihara *et al*, 1978). This is the first report of its occurrence from this plant. It is of significance that *A. tuberosum* contains this amide because it shows anti-platelet aggregation effect (Okuyama *et al*, 1986) and cytotoxic activities against MCF-7 (human breast carcinoma), A-549 (human lung cancer) and HT-29 (human colon cancer) cell lines (Zhao *et al*, 1992).

The BuOH-soluble fraction of the leaves of *A. tu-*

*berosum* was also repeatedly chromatographed over silica gel to afford compound **2**. Compound **2**,  $C_{12}H_{10}O_3$  (double bond equivalent=8), which gave mp 340.5°C, showed a molecular ion peak at  $m/z$  202 in the mass spectrum. The IR spectrum of **2** displayed absorption bands at 3019 and 1680  $cm^{-1}$ , indicating the presence of hydroxyl group and aromatic ring in the molecule. The  $^1H$ -NMR spectrum of **2** showed two ortho-coupled doublets each of two protons with a J value of 7.60 Hz at  $\delta$  7.37 and 5.44 ppm, indicating the presence of a 1,4-di-substituted benzene ring and two singlets at  $\delta$  10.98 and 9.52 assignable to two phenolic protons, respectively. The  $^{13}C$ -NMR spectrum of **2** showed 4 unique types of carbon, whereas the formula of this compound shows 12 carbon atoms. This is due to an element of symmetry, which has already been shown from the  $^1H$ -NMR. Therefore, the structure was indicated as a symmetrical bis (*p*-hydroxyphenyl) ether from an analysis of the mass and  $^{13}C$ -NMR. This is the first report of its occurrence in nature to our best knowledge.

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