

Journal of Korean Institute of surface Engineering
Vol. 29, No. 6, Dec., 1996

PREFERRED ORIENTATION AND MICROSTRUCTURE OF MOD DERIVED SrBi_{2x}Ta₂O₉ THIN FILMS WITH BI CONTENT x

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ABSTRACT

SrBi_{2x}Ta₂O₉ ferroelectric thin films were prepared on platinized silicon substrates using MOD proces, and crystallization behavior of the films was investigated with variation of the annealing temperature and Bi content x. Crystalline phase of bismuth layered perovskite structure was formed even by baking the films at 800°C for 5 minutes in air, and was not changed by annealing at temperatures raning from 700°C to 900°C for 1 hour in oxygen ambient. When SrBi_{2x}Ta₂O₉ thin films (0.8 ≤ x ≤ 1.6) were annealed at 800°C, Preferred orientation of the films along c-axis was observed with x ≥ 1.2. With increasing Bi content x, surface morphology of the films was changed from equiaxed grains to elongated grains.

INTRODUCTION

Ferroelectric thin films for use in non-volatile random access memories have drawn much attention in recent years, since they exhibit an electrically switchable remanant polarization^[1-3]. Lead zirconate titanate (PZT) thin films have been extensively investigated to apply for ferroelectric memory devices^[1, 4, 5]. However, reliability of Pt/PZT/Pt capacitors, such as polarization fatigue and imprint, limits the realization of the commercial ferroelectric memories^[6, 7].

To suppress the polarization fatigue in PZT based capacitors, many investigations have been performed to replace Pt electrode with conducting oxide electrodes such as RuO₂ and IrO₂^[8-10]. In this configuration, oxide electrodes help to control the oxygen vacancy concentration at the electrode/ferroelectric

interface, which is known to be responsible for the fatigue phenomena^[8-10]. Oxide electrodes have been proven suitable for controlling polarization fatigue. However, their electrical conductivities are not as high as those of Pt electrode, resulting in undesirable characteristics of the device such as high RC time constant and high leakage currents^[11]. In addition, non-ferroelectric secondary phases may be formed by reactions between PZT film and oxide electrode^[12].

To overcome the polarization fatigue in ferroelectric capacitors without such detrimental side-effects caused by oxide electrodes, SrBi₂Ta₂O₉ ferroelectric thin films of bismuth layered perovskite structure has been developed recently^[1, 2, 13, 14]. SrBi₂Ta₂O₉ may be written as (Bi₂O₂)²⁺(SrTa₂O₇)²⁻ consisted of two perovskite-like (SrTa₂O₇)²⁻ layers, alternating with a layer of (Bi₂O₂)²⁺, along the c-axis^[7, 13].

SrBi₂Ta₂O₉ thin film capacitors with Pt electrode can be fatigue-free due to the fact that the oxygen vacancy concentration at the electrode/ferroelectric interface is reduced by entrapping of oxygen vacancies in (Bi₂O₂)²⁺ layers^[7].

Electrical properties of SrBi₂Ta₂O₉ thin films are largely dependent on the stoichiometry and preferred orientation of the films^[7, 15]. As bismuth volatility during annealing process of sol-gel or MCD derived SrBi₂Ta₂O₉ thin films usually occurs, addition of excess bismuth oxide precursor into a coating solution is required to prevent the bismuth deficiency of the films. In this study, SrBi_{2x}Ta₂O₉ thin films (0.8 ≤ x ≤ 1.6) are fabricated by MOD process on the platinized silicon substrates, and preferred orientation and microstructural behavior of the films have been investigated with Bi content x.

EXPERIMENTAL PROCEDURE

SrBi_{2x}Ta₂O₉ (SBT) thin films were prepared on Pt(100nm)/Ti(5nm)/SiO₂(100nm)/Si substrates by MOD method. Sr-2-ethylhexanate [Sr(CH₃(CH₂)₃CH(C₂H₅)COO)₂], Bi-2-ethylhexanate and Ta-2-ethylhexanate were used as starting materials, and the concentration of each precursor solutions was 0.5 M. In order to prepare spin-coating solution, 2-ethylhexanate metal precursors were mixed with Sr:Bi:Ta mole ratio of 1:2x:2 (0.8 ≤ x ≤ 1.6), and stirred for several hours using magnetic stirrer. Mixed precursor solutions were then diluted with n-butyl acetate [CH₃CO₂(CH₂)₃CH₃] to make the concentration of stock solution 0.25 M. To modify hydrolysis and condensation rates of sol, 1 M

NH₄OH or pure H₂O was added into the stock solution. Without depending on pH of the hydrolysis solution, no gelation or precipitation occurred for H₂O/SBT mole ratio up to unity. However, white precipitates were formed when H₂O/SBT more than unity. Formation of precipitates were accelerated water (H₂O/SBT mole ratio=1) and used to fabricate SBT films.

SBT films were deposited on platinized silicon (100) substrates by spin-casting at 3000 rpm for 40 seconds. The coated gel films were dried and baked in air at 250°C for 10 minutes and at 800°C for 5 minutes, respectively. This procedure was repeated several times to obtain the desired film thickness of 200nm. Finally, films were annealed at 700°C ~ 900°C for 1 hour in oxygen ambient.

Thermal analysis (TG/DTA) of gel powder prepared by heating the coating solutions at 180°C for 20 hours was carried out up to 900°C at a heating rate of 5°C/min. Crystalline phase of MOD derived films were characterized by low-angle X-ray diffractometry with incident angle of 5°. Microstructure of the film surface was observed using scanning electron microscopy (SEM) and atomic force microscopy (AFM). Composition of the films was determined by energy dispersive spectroscopy (EDS).

RESULTS AND DISCUSSION

TG/DTA curves of the SBT gel powder are illustrated in Fig. 1. Weight loss of about 30% occurred at temperature between 250°C and 400°C due to the burn-out of organics, which could be confirmed by the exothermic peak at this temperature range on the DTA

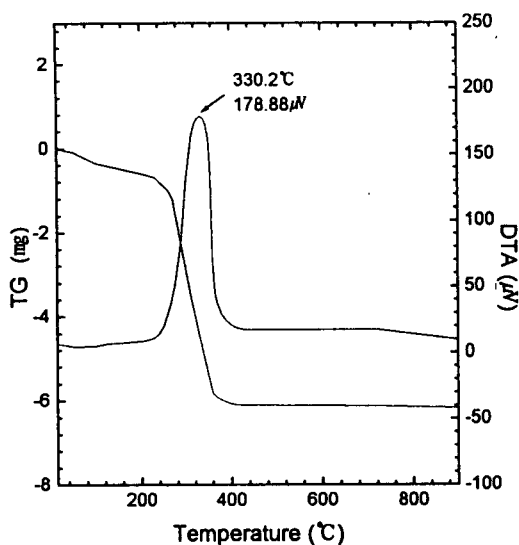


Fig. 1 TG/DTA curves of $\text{SrBi}_2\text{Ta}_2\text{O}_9$ gel powder

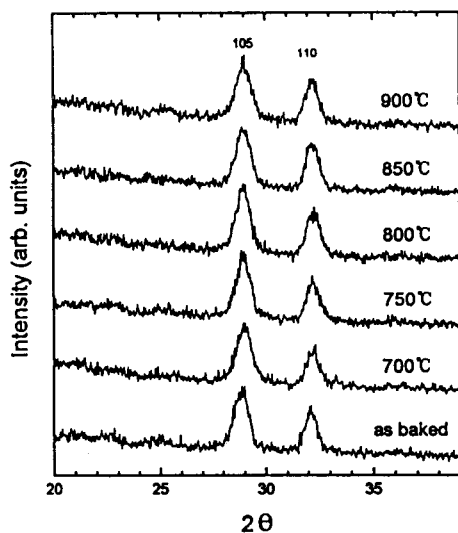


Fig. 2 XRD patterns of $\text{SrBi}_2\text{Ta}_2\text{O}_9$ thin films annealed at various temperatures.

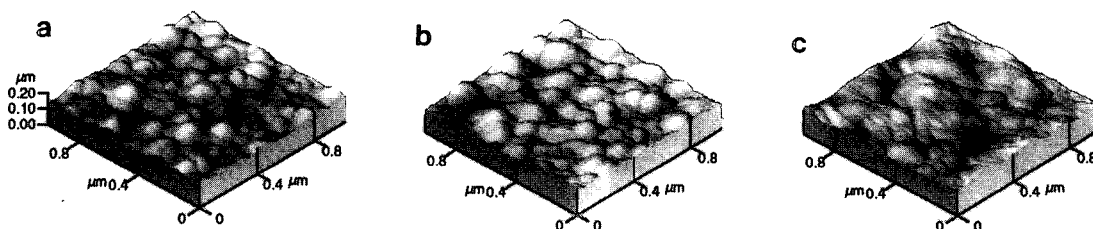


Fig. 3 AFM images of $\text{SrBi}_2\text{Ta}_2\text{O}_9$ thin films annealed at (a) 700°C, (b) 800°C, and (c) 900°C

curve. Based on this result, drying temperature for the spin-coated films was determined to be 250°C, and films with crack-free surface could be obtained after multiple coatings.

XRD patterns of SBT($x=1$ in $\text{SrBi}_{2-x}\text{Ta}_2\text{O}_9$) thin films, annealed at various temperatures, are shown in Fig. 2. MOD derived SBT films could be crystallized to the bismuth layered perovskite structure even by baking at 800°C for 5 minutes in air. Crystalline phase of the SBT films was not changed by annealing at temperatures ranging from 700°C to 900°C for 1 hour in oxygen ambient. All SBT films, just baked at 800°C or baked and annealed at temperatures of 700°C~900°C in oxygen atmosphere, were observed to be non-oriented.

AFM images and root mean square (RMS) surface roughness of SBT films are shown in Fig. 3 and Fig. 4, respectively. The scan area was $1\mu\text{m} \times 1\mu\text{m}$. SBT films were composed of equiaxed the ins without depending on the annealing temperature, and the grain size of the films increased from 100nm to 200nm with increasing annealing temperature from 700°C to 900°C. When annealed at temperatures up to 800°C, the surface of SBT films was quite smooth and flat with RMS surface roughness of about 13nm and was changed little with annealing temperature. However, surface roughness of the films increased significantly by

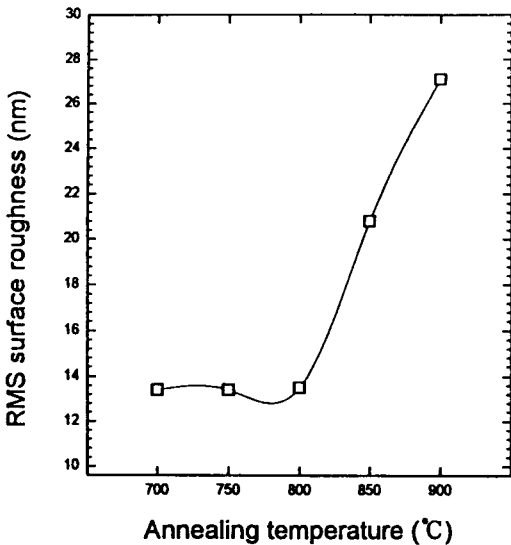


Fig. 4 Surface roughness of $\text{SrBi}_2\text{Ta}_2\text{O}_9$ thin films vs. annealing temperature.

annealing at temperatures above 800°C , and reached to $\sim 28\text{nm}$ for the film annealed at 900°C . This increment of the surface roughness could be attributed to the grain growth at higher annealing temperature.

Bismuth oxide in SBT has relatively high vapor pressure compared with those of strontium and tantalum oxides. Thus, bismuth volatility occurs during annealing process of the as-deposited SBT thin films at elevated temperatures, resulting in non-stoichiometry of the film composition. As lead deficiency of sol-gel derived PZT films deteriorates ferroelectric properties of PZT films^[1,4,5], remanent polarization of SBT thin films decreases with bismuth deficiency of the films. Addition of excess bismuth oxide precursor into coating solutions is helpful to realize the stoichiometric composition of SBT films^[7]. If bismuth content in SBT thin films is enriched to a high level, however, formation of non-ferroelectric second phases and decrease of electric

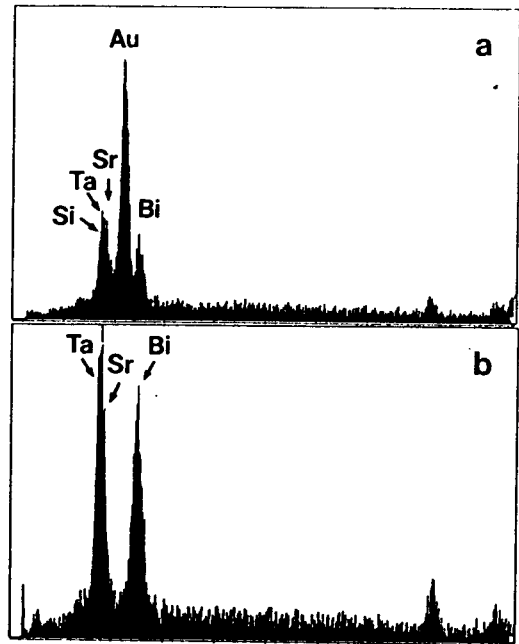


Fig. 5 EDS spectrum of (a) $\text{SrBi}_2\text{Ta}_2\text{O}_9$ thin films and (b) powders

cal resistivity have been reported^[7]. Thus, bismuth content in coating solutions should be controlled to have proper stoichiometry of the annealed SBT films.

SBT thin films were prepared by spin-casting of $\text{SrBi}_x\text{Ta}_2\text{O}_9$ solutions ($0.8 \leq x \leq 1.6$) and annealing at 800°C for 1 hour in oxygen atmosphere. To characterize the composition of $\text{SrBi}_x\text{Ta}_2\text{O}_9$ films, EDS analysis was conducted for films fabricated on platinized silicon substrates. As shown in Fig. 5(a), however, $K_{\alpha 1}$ peak of silicon substrate (1.740 eV) is located close to Ta $M_{\alpha 1}$ (1.710 eV) and $M_{\beta 1}$ (1.766 eV) peaks^[17]. Also $K_{\beta 1}$ peak energy of Si (1.832 eV) coincides with those of Sr $L_{\alpha 1}$ (1.806 eV) and $\beta 1L$ (1.872 eV), resulting in the inability of the quantitative composition analysis^[17]. To eliminate Si substrate peaks in EDS spectrum, $\text{SrBi}_x\text{Ta}_2\text{O}_9$ powders were

fabricated by drying each coating solution at 250°C and annealing at 800°C for 1 hour in oxygen ambient. As shown in Fig. 5(b), EDS analysis was performed on $\text{SrBi}_{2x}\text{Ta}_2\text{O}_9$ powders with carbon coating. Since Ta $M_{\alpha 1}$ and Sr $L_{\alpha 1}$ peaks are detected together at narrow energy range, it would be still difficult to analyze the content of each element accurately even with deconvolution of the peaks^[18]. Compared to strontium and tantalum oxides, bismuth oxide is volatilized more easily during annealing of MOD derived SBT films, and compositional variation of the films is caused mainly by bismuth-oxide volatility. Thus, the composition of $\text{SrBi}_{2x}\text{Ta}_2\text{O}_9$ powders was determined as Bi/(Sr+Ta) mole ratio, and is illustrated in Fig. 6 as a function of Bi content x in coating solutions. Bi/(Sr+Ta) mole ratio increased almost linearly from 0.24 to 0.48 with increase of Bi content x from 0.8 to 1.6. Overall mole ratio of Bi/(Sr+Ta) in Fig. 6 is lower than the expected value in the annealed thin films, which may be due to

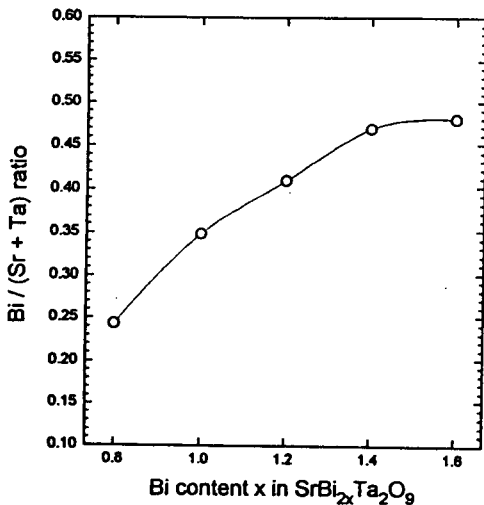


Fig. 6 Bi/(Sr+Ta) mole ratio of SrBi_2O_9 powders vs. Bi content x .

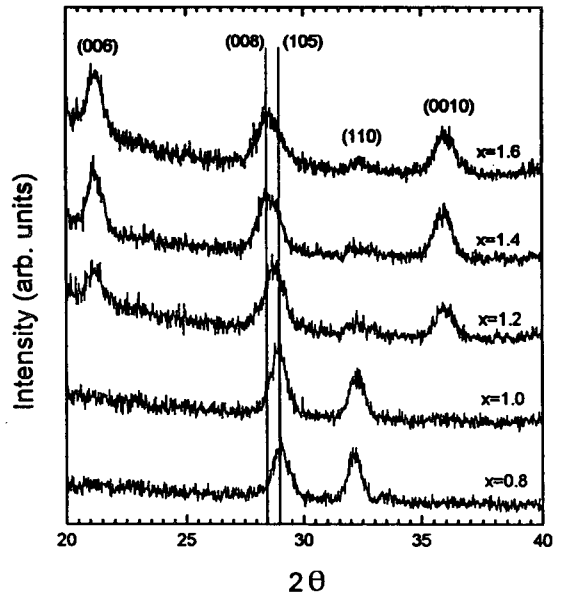


Fig. 7 XRD patterns of $\text{SrBi}_{2x}\text{Ta}_2\text{O}_9$ thin films vs. Bi content x .

the heavy vaporization of bismuth oxide in $\text{SrBi}_{2x}\text{Ta}_2\text{O}_9$ powders to larger surface area compared with thin films.

XRD patterns of $\text{SrBi}_{2x}\text{Ta}_2\text{O}_9$ films are illustrated in Fig. 7. For all composition x ranging from 0.8 to 1.6, SBT phase of the bismuth layered perovskite structure was formed by annealing at 800°C for 1 hour. For SBT thin films with $x \leq 1.0$, only (105) and (110) diffraction peaks were observed. In contrast, (006), (008) and (0010) diffraction peaks newly appeared at $x=1.2$ and their intensities increased with increasing bismuth content x , indicating films were preferentially oriented along the c -axis.

Fig. 8 shows SEM micrographs of $\text{SrBi}_{2x}\text{Ta}_2\text{O}_9$ thin films with various bismuth content x . Contrary to equiaxed grains of non-oriented $\text{SrBi}_{2x}\text{Ta}_2\text{O}_9$ films ($x=0.8$ and 1.0), elongated grains have been observed on the films with c -axis orientation ($x \geq 1.2$), as reported by

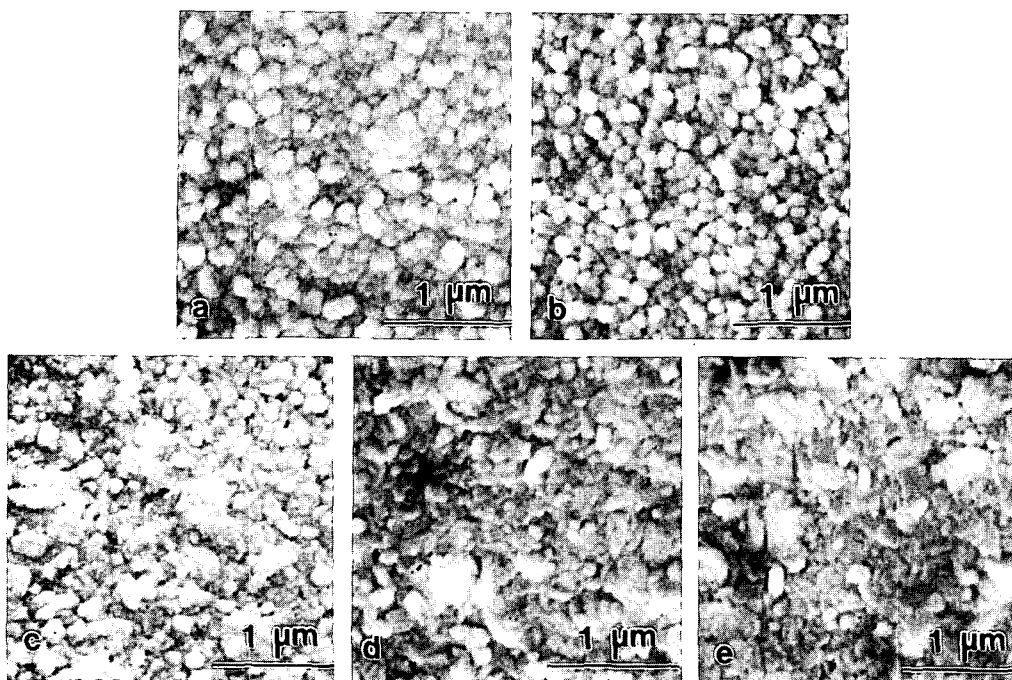


Fig. 8 SEM micrographs of $\text{SrBi}_x\text{Ta}_2\text{O}_9$ thin films with Bi content x of (a) 0.8, (b) 1.0, (c) 1.2, (d) 1.4, and (e) 1.6.

others^[16]. The amount of the elongated grains increased with bismuth content x . From these results, it could be suggested that these elongated grains are oriented along the c -axis which is perpendicular to the film surface. Surface roughness of $\text{SrBi}_x\text{Ta}_2\text{O}_9$ thin films was 13nm~16nm, and was not changed with bismuth content.

Polarization of the compounds with bismuth layered perovskite structure is largely dependent on the crystal orientation. $\text{SrBi}_2\text{Ta}_2\text{O}_9$ may be written as $(\text{Bi}_2\text{O}_2)^{2+}(\text{SrTa}_2\text{O}_7)^{2-}$ consisted of two perovskite-like $(\text{SrTa}_2\text{O}_7)^{2-}$ layers, alternating with a non-ferroelectric layer of $(\text{Bi}_2\text{O}_2)^{2+}$, along the c -axis^[7,13]. As ferroelectric properties of $\text{SrBi}_2\text{Ta}_2\text{O}_9$ depend on O-Ta-O chains in the perovskite layer perpendicular the c -axis, remanant polarization decreases with c -axis preferred orienta-

tion^[7,13,15]. However, it has been reported that the coercive field and electrical conductivity of layered perovskites are lower in the c -axis^[14]. Since strong $\langle 001 \rangle$ orientation was obtained for MOD derived $\text{SrBi}_x\text{Ta}_2\text{O}_9$ films with x ranging from 1.2 to 1.6, electrical properties for frim application is expected to be optimum in this range.

CONCLUSIONS

$\text{SrBi}_x\text{Ta}_2\text{O}_9$ thin films ($0.8 \leq x \leq 1.6$), prepared on platinized silicon substrates using MOD process, were crystallized without formation of second phases by annealing at 800 °C for 1 hour in oxygen ambient. Preferred orientation in $\text{SrBi}_x\text{Ta}_2\text{O}_9$ thin films was largely affected by bismuth content x in the coating solutions, i.e., stoichiometry of the an

nealed films. Thin films with $x \leq 1.0$ did not exhibit the preferred orientation. However, c -axis orientation could be observed for films with $x \geq 1.2$, and was enhanced with increasing bismuth content x . Contrary to the non-oriented films, $\text{SrBi}_{2x}\text{Ta}_2\text{O}_9$ thin films with c -axis orientation were composed of elongated grains of which amount increased with bismuth content x . Thus, it could be suggested that these elongated grains were oriented along the c -axis perpendicular to the film surface.

ACKNOWLEDGMENT

This work was supported by the Electronics and Telecommunications Research Institute of Korea.

REFERENCES

1. J. F. Scott and C. A. Araujo, *Science*, **246** (1989) 1400
2. C. A. Araujo, J. D. Cuchiaro, L. D. McMillan, M. C. Scott and J. F. Scott, *Nature*, **374** (1995) 627
3. S. K. Dey and R. Zuleeg, *Ferroelectrics*, **108** (1990) 37
4. C. V. R. Vasant Kumar, R. Pascual and M. Sayer, *J. Appl. Phys.*, **71** (1992) 864
5. M. Shimizu and T. Shiosaki, *Mat. Res. Soc. Symp. Proc.*, **361** (1994) 295
6. T. Mihara, H. Watanabe and C. A. Araujo, *Jpn J. Appl. Phys.*, **33** (1994) 3996
7. H. Watanabe, T. Mihara, H. Yoshimori and C. A. Araujo, *Jpn J. Appl. Phys.*, **34** (1995) 5240
8. R. Ramesh, J. Lee, T. Sands, V. G. Keramidas and O. Auciello, *Appl. Phys. Lett.*, **64** (1994) 2517
9. K. R. Bellur, H. N. Al-Shareef, S. H. Rou, K. D. Gifford, O. Auciello and A. I. Kingon, *Proc. 8th ISAF* (1992)
10. C. K. Kwok, D. P. Vijay, S. B. Desu, N. R. Parkih and E. A. Hill, *Proc. 4th Int'l Symp. Integrated Ferroelectrics* (1992) 412
11. R. Dat, J. K. Lee, O. Auciello and A. I. Kingon, *Appl. Phys. Lett.*, **67** (1995) 572
12. O. Auciello, H. N. Al-Shareef, K. D. Gifford, D. J. Lichtenwalner, R. Dat, K. R. Bellur, A. I. Kingon and R. Ramesh, *Mat. Res. Soc. Symp. Proc.*, **341** (1994) 341
13. H. Tabata, H. Tanaka and T. Kawai, *Jpn. J. Appl. Phys.*, **34** (1995) 5146
14. S. Y. Chen, X. F. Chen and I. W. Chen, *Mat. Res. Soc. Symp. Proc.*, **361** (1994) 15
15. D. P. Vijay, S. B. Desu, M. Nagata, X. Zhang and T. C. Chen, *Mat. Res. Soc. Symp. Proc.*, **361** (1995) 3
16. T. Atsuki, N. Soyama, T. Yonezawa and K. Ogi, *Jpn. J. Appl. Phys.*, **34** (1995) 5096
17. *International Tables for X-ray Crystallography*, Vol. IV, Birmingham, England, Kynoch Press (1974) pp. 6-43
18. "Energy-Dispersive X-ray Microanalysis," Kevex Corporation, Foster City (1983) 33