# Synthesis and Antitumor Evalution of *cis*-(1,2-Diaminoethane) dichloroplatinum (II) Complexes Linked to 5- and 6-Methyleneuracil and -uridine Analogues

### Jack C. Kim, Min-Hwa Lee and Soon-Kyu Choi\*

Department of Chemistry, College of Natural Science, Pusan National University and \*Dong-A University, Pusan 609-735, Korea

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The search for platinum (II)-based compounds with improved therapeutic properties was prompted to design and synthesize a new family of water-soluble, third generation cis-diaminedichloroplatinum (II) complexes linked to uracil and uridine. Six heretofore unreported uracil and uridine-platinum (II) complexes are; [N-(uracil-5-yl-methyl)ethane-1,2-di-amine]dichloroplatinum (II) (3a), [N-(uracil-6-yl-methyl)ethane-1,2-diamine] dichloroplatinum (II) (3b), {[N-(2', 3'.5'-tri-O-acetyl)uridine-5-yl-methyl] ethane-1,2-diamine\dichloroplatinum (II) (6a), \{[N-(2',3', 5'-tri-O-acetyl) uridine-6-yl-methyl]ethane-1,2-diamine}dichloroplatinum (II) (6b),[N-(uridine-5yl-methyl)ethane-1,2-diamine]dichloroplatinum (II) (7a), [N-(uridine-6-yl-methyl)ethane-1,2-diamineldichloroplatinum (II) (7b). These analogues were prepared from the key starting materials, 5-chloromethyluracil (1a) and 6-chloromethyluracil (1b) which were reacted with ethylenediamine to afford the respective 5-[(2-aminoethyl)amino] methyluracil (2a) and 6-[(2-aminoethyl)amino]methyluracil (2b). The cis-platin complexes 3a and 3b were obtained through the reaction of the respective 2a and 2b with potassium tetrachloroplatinate (II). The heterocyclic nucleic acid bases 1a and 1b were efficiently introduced on the  $\beta$ -D-ribose ring via a Vorbruggen-type nucleoside coupling procedure with hexamethyldisilazane, trimethylchlorosilane and stannic chloride under anhydrous acetonitrile to yield the stereospecific β-anomeric 5chloromethyl-2',3',5'-tri-O-acetyluridine (4a) and 6-chloromethyl-2',3',5'-tri-O-acetyluridine (4b), respectively. The nucleosides 4a and 4b were coupled with ethylenediamine to provide the respective 5-[(amino-ethyl)amino]methyl-2',3',5'-tri-O-acetyluridine (5a) and 6-[(aminoethyl)amino] methyl-2',3',5'-tri-O-acetyluridine (5b). The diamino-uridines 5a and 5b were reacted with potassium tetrachloroplatinate (II) to give the novel nucleoside complexes, 6a and 6b, respectively which were deacetylated into the free nucleosides, 7a and 7b by the treatment with CH<sub>3</sub>ONa. The cytotoxic activities were evaluated against three cell lines (FM-3A, P-388 and J-82) and none of the synthesized compounds showed any significant activity.

**Key words**: *cis*-Diamminedichloroplatinum (II), [N-(uracil-5-yl-methyl) ethane-1,2-diamine]dichloroplatinum (II), [N-(uracil-6-yl-methyl)ethane-1,2-diamine] dichloroplatinum (II), {[N-(2',3',5'-tri-O-acetyl)uridine-5-yl-methyl] ethane-1,2-diamine}dichloroplatinum (II), {[N-(2',3',5'-tri-O-acetyl)uridine-6-yl-methyl]ethane-1,2-diamine}dichloroplatinum (II), [N-(uridine-5-yl-methyl) ethane-1,2-diamine}dichloroplatinum (II), antitumor activities, human bladder carcinoma cell (J-82), Mouse lymphoid neoplasma cell (P-388), Mouse mammary carcinoma cell (FM-3A)

#### INTRODUCTION

Since the successful introduction of cisplatin into oncology practice over 20 years ago, many analogs have been synthesized in attempts to either overcome the particular toxic side effects of the parent drug, or to broaden its clinical spectrum of antitumor activity (Korolkovas, 1988; Nicolini, 1988). But their widespread

clinical use is limited by inherent resistance (limited activity against many common human cancers), by intrinsic or acquired drug resistance (reduced efficacy upon repeated treatment) and by their relative toxic side effects (Andrews, 1992). The carboplatin *cis*-diamine-(1,1-cyclobutane dicarboxylato)platinum (II) is the only clinically successful second generation plartinem (II) complexes (Harland, *et al.*, 1984). It does not exhibit significant nephrotoxicity and emesis, and its a latively lower toxicities as compared to those of cisp'atin have been related to the greater pharmaco-

kinetic stability of its 1,1-cyclobutane-carboxylate ligand in solution (Bitha *et al.*, 1989). Nevertheless, it still has two other draw-backs. Just like cisplatin, it only effects a narrow range of tumors and causes the development of resistance in the tumor cell. In considering a third generation platinum complexes for clinical development, a key focus of drug design will be the ability of an agent to improve a broader spectrum of activity, increased clinical efficacy, less severe side-effects, lack of cross-resistance to cisplatin and enhanced water-solubility for systematic use (Burchenal *et al.*, 1979).

The need for platinum (II) complexes with improved characteristics stimulate our drug design of several biologically important nucleic acid base (uracil and uridine)-based ligands that may have greater watersolubility and less general systematic toxicity than cisplatin. We have synthesized six heretofore unreported platinum (II) complexes of the amino analogues of uracil and uridine; uracil analogues, [N-(uracil-5-yl-methvl)ethane-1,2-diamineldichloroplatinum (II) (3a) and [N-(uracil-6-yl-methyl)ethane-1,2-diamine]dichloroplatinum (II) (3b) and uridine nucleoside analogues, {[N-(2',3', 5'-tri-O-acetyl)uridine-5-yl-methyl]ethane-1,2-diamine} dichloroplatinum (II) (6a), {[N-(2',3',5'-tri-O-acetyl)uridine-6-yl-methyl]ethane-1,2-diamine} dichloroplatinum (II) (**6b**), [N-(uridine-5-yl-methyl)ethane-1,2-diamine]dichloro- platinum (II) (7a) and [N-(uridine-6-yl-methyl)ethane-1,2-diamine]dichloro- platinum (II) (7b). Antitumor activities of those synthesized compounds were evaluated against the following three cell lines;

- (a) mouse lymphoid neoplasma cell (P-388)
- (b) mouse mammary carcinoma cell (FM-3A)
- (c) human bladder carcinoma (J-82)

#### MATERIALS AND METHODS

Melting points were determined on electrothermal capillary melting point apparatus and are uncorrected. TLC was performed on glass plates coated with silica gel (silica gel 60 F<sub>254</sub>) and compounds were visualized using an UV lamp. Proton magnetic resonance spectra were obtained with Varian EM-360A spectrophotometer and Varian Gemini 200 MHz (solution in dimethyl-sulfoxide-d<sub>6</sub> with tetramethyl-silane as internal standard). Ultraviolet spectral data were measured with Hitachi 124 spectrometer. The organic solvents and chemicals were obtained from the commercial and purified by the appropriate methods before use.

#### 5-[(2-Aminoethyl)amino|methyl uracil (2a)

5-Chloromethyl uracil (0.321 g, 2.00 mmol) in methanol (40 ml) was added under  $N_2$  to a stirred solution of dry ethylenediamine (20 ml) and heated at 75°C for 48hrs. The reaction mixture was evaporated *in vaccuo* 

to give light yellow solids which were recrystallized from CH<sub>3</sub>OH, brown solid (62%): m.p  $170\sim173^{\circ}$ C; Mass m/z 184 (M<sup>+</sup>); IR (KBr) 3303 cm<sup>-1</sup> (N-H), 3157 cm<sup>-1</sup> (C-H), 1677 cm<sup>-1</sup> (C=O); <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>)  $\delta$  2.5~2.7 (m, 4H, CH<sub>2</sub>CH<sub>2</sub>),  $\delta$  3.4 (s, 2H, CH<sub>2</sub>Cl),  $\delta$  3.37 (t, 2H, NHCH<sub>2</sub>),  $\delta$  7.8 (s, 1H, H6)

### 6-[(2-Aminoethyl)amino]methyluracil (2b)

The same procedure described above in compound **2a** was employed for the preparation of **2b** to give a yellow solid (81%): m.p  $130\sim134^{\circ}$ C; Mass m/z 184 (M<sup>+</sup>); IR (KBr) 3299.6 cm<sup>-1</sup> (N-H), 2988.6 cm<sup>-1</sup> (C-H); <sup>1</sup>H-NMR (D<sub>2</sub>O)  $\delta$  3.41 (t, 2H, NHCH<sub>2</sub>),  $\delta$  3.23 (t, 2H, CH<sub>2</sub>NH<sub>2</sub>),  $\delta$  5.72 (s, 1H, H5).

### [N-(Uracil-5-yl-methyl)ethane-1,2-diamine]dichloroplatinum (II) (3a)

To a stirred solution of  $K_2PtCl_4$  (0.21 g, 0.5 mmol) in distilled  $H_2O$  (2 ml) was added under  $N_2$  **2a** (0.11 g, 0.6 mmol) in deionized  $H_2O$  (2 ml). The basic, homogenous reaction mixture (pH=8) was continuously stirred at 70°C for 48 hours (pH=4 was achieved), then 5% aqueous KCl (10 ml) was added and the mixture was stirred for an additional one hour. The precipitate was collected, washed several times with deionized water, and dried to give grey solid (58%). An analytical sample was obtained by chromatography on silica gel and elution with  $CH_2Cl_2-CH_3OH$  (10:1): m.p >300°C; Mass m/z 451 ( $M^+$ ); IR (KBr) 3160 cm $^{-1}$  (N-H), 3050 cm $^{-1}$  (C-H);  $^1$ H-NMR (DMSO-d<sub>6</sub>)  $\delta$  3.46 (t, 2H, NHCH<sub>2</sub>),  $\delta$  3.28 (t, 2H, CH<sub>2</sub>NH<sub>2</sub>),  $\delta$  7.80 (s, 1H, H6); UV  $\lambda_{max}$  289.0 nm.

### [N-(Uracil-6-yl-methyl)ethane-1,2-diamine]dichloroplatinum (II) (3b)

The same procedure described above in compound **3a** was employed for the preparation of 3b to give a grey solid (63%): m.p >300°C; Mass m/z 451 (M<sup>+</sup>); IR (KBr) 3134 cm<sup>-1</sup> (N-H), 2939 cm<sup>-1</sup> (C-H); <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>)  $\delta$  3.54 (t, 2H, NHCH<sub>2</sub>),  $\delta$  3.37 (t, 2H, CH<sub>2</sub>-NH<sub>2</sub>),  $\delta$  5.60 (s, 1H, H5); UV  $\lambda_{max}$  282.8 nm.

### 5-Chloromethyl-2',3',5'-tri-O-acetyluridine (4a)

To a stirred mixed solution of 5-chloromethyl uracil (0.3 g, 1.87 mmol) and 1,2,3,5-tetra-O-acetyl-β-D-ribo-furanose (0.594 g, 1.87 mmol) in anhydrous CH<sub>3</sub>CN (30 ml) under  $N_2$  was added hexamethyldisilazane and trimethylsilyl chloride, followed by SnCl<sub>4</sub>. The reaction mixture was stirred at room temperature for 24 hours, and evaporated *in vaccuo* to give yellow sirupy residues. The residues were dissolved in CH<sub>2</sub>Cl<sub>2</sub> (30 ml) and washed with saturated NaHCO<sub>3</sub> (2×20 ml) and H<sub>2</sub>O (2×20 ml), and the organic CH<sub>2</sub>Cl<sub>2</sub> solution was dried over anhydrous MgSO<sub>4</sub>. Filtration, eva-

poration and chromatography on silica gel (CH<sub>2</sub>Cl<sub>2</sub>-CH<sub>3</sub>OH (20:1)) gave yellow syrupy residue (51%): IR (KBr) 3177 cm<sup>-1</sup> (N-H); <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ 2.11 (s, 9H, CH<sub>3</sub>CO<sub>2</sub>-), δ 3.6~4.4 (m, 3H, H<sub>1'</sub>, H<sub>3'</sub>, H<sub>4'</sub>), δ 4.05~4.71 (m, 4H, H<sub>2'</sub>, H<sub>3'</sub>, H<sub>5'</sub>), δ 5.31 (m, 1H, H<sub>4'</sub>), δ 5.89 (d, 1H, H<sub>1</sub>), δ 7.28 (s, 1H, H<sub>6</sub>)

### 6-Chloromethyl-2',3',5'-tri-O-acetyluridine (4b)

The same procedure described above in compound **4a** was employed for the preparation of 4a to give yellow syrupy residue (68%): IR (KBr) 3177 cm<sup>-1</sup>; <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$  2.13 (s, 9H, CH<sub>3</sub>CO<sub>2</sub>-),  $\delta$  4.05~4.48 (m, 4H, H<sub>2</sub><sup>1</sup>, H<sub>3</sub><sup>1</sup>, H<sub>5</sub><sup>1</sup>),  $\delta$  5.31~5.97 (m, 3H, H<sub>11</sub>, H<sub>41</sub>, H<sub>5</sub>).

# 5-[(Aminoethyl)amino]methy-2',3',5'-tri-O-acetyluridine (5a)

5-Chloromethyl-2',3',5'-tri-O-acetyluridine (**4a**) (0.9 g, 2.34 mmol) was added under  $N_2$  to a stirred solution of dry ethylenediamine (0.34 ml, 5.00 mmol) and refluxed for 24 hours. The reaction mixture was evaporated *in vaccuo* to afford yellow oily residues which were chromatographed on silica gel and elution with  $CH_2Cl_2$ - $CH_3OH$  (20:1) light brown solid (87%): IR (KBr) 3185 cm<sup>-1</sup> (N-H), 2943 cm<sup>-1</sup> (C-H); <sup>1</sup>H-NMR (DMSOd<sub>6</sub>)  $\delta$  2.14 (s, 9H,  $CH_3CO_2$ -),  $\delta$  3.12~3.31 (m, 4H,  $CH_2$ - $CH_2$ -),  $\delta$  4.06~4.20 (m, 4H,  $CH_2$ - $CH_3$ -),  $\delta$  5.07~5.28 (m, 2H,  $CH_3$ -),  $\delta$  7.24 (s, 1H, H6).

# 6-[(Aminoethyl)amino]methy-2',3',5'-tri-O-acetyluridine (5b)

The same procedure described above in compound **5a** was employed for the preparation of 5b to give yellow solid (81%): IR (KBr) 3179 cm<sup>-1</sup> (N-H), 3008 cm<sup>-1</sup> (C-H); <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>)  $\delta$  2.19 (s, 9H, CH<sub>3</sub>-CO<sub>2</sub>-),  $\delta$  2.97~3.15 (m, 4H, CH<sub>2</sub>CH<sub>2</sub>-),  $\delta$  3.24~4.18 (m, 4H, H<sub>2</sub>, H<sub>3</sub>, H<sub>5</sub>),  $\delta$  4.61~5.54 (m, 3H, H<sub>1</sub>, H<sub>4</sub>, H<sub>5</sub>).

# {[N-(2',3',5'-tri-O-acetyl)uridine-5-yl-methyl]ethane-1, 2-diamine}di-chloroplatinum (II) (6a)

To a stirred solution of  $K_2PtCl_4$  (2.29 g, 0.70 mmol) in deionzed  $H_2O$  (3 ml) was slowly added under  $N_2$  **5a** (0.31 g, 0.7 mmol) in deioniged  $H_2O$  (2 ml). The basic, homogenous reaction mixture was continously stirred at 70°C for 36 hours until pH of 4 was achieved, then 5% aqueous KCl (20 ml) was added and the mixture was stirred for an additional one hour. The precipitate was collected, washed several times with deioniged  $H_2O$  (10 ml), and dried to give brown solid (69%): m.p >300°C; Mass m/z 709 (M<sup>+</sup>); IR (KBr) 3188 cm<sup>-1</sup> (N-H), 2913 cm<sup>-1</sup> (C-H); <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>)  $\delta$  2.10 (s, 9H, CH<sub>3</sub>CO<sub>2</sub>-),  $\delta$  3.25~3.48 (m, 4H, CH<sub>2</sub>CH<sub>2</sub>-),  $\delta$  4.11~4.52 (m,  $H_2$ -,  $H_3$ -,  $\delta$ -10~5.35 (m, 2H,  $H_3$ -),  $\delta$ -7.24 (s, 1H, H6).

# {[N-(2',3',5'-tri-O-acetyl)uridine-6-yl-methyl]ethane-1, 2-diamine}di-chloroplatinum (II) (6b)

The same procedure described above in compound 6a was employed for the preparation of 6b to give light reddish solid (51%): m.p >300°C; Mass m/z 709 (M\*); IR (KBr) 3190 cm<sup>-1</sup> (N-H), 3009 cm<sup>-1</sup> (C-H);  $^{1}$ H-NMR (DMSO-d<sub>6</sub>)  $\delta$  2.10 (s, 9H, CH<sub>3</sub>CO<sub>2</sub>-),  $\delta$  3.25~ 3.41 (m, 4H, CH<sub>2</sub>CH<sub>2</sub>-)  $\delta$  3.82~4.41 (m, 4H, H<sub>2</sub>, H<sub>3</sub>, H<sub>5</sub>),  $\delta$  4.64~5.52 (m, 3H, H<sub>1</sub>, H<sub>4</sub>, H<sub>5</sub>).

# [N-(Uridine-5-yl-methyl)ethane-1,2-diamine]dichloroplatinum (II) (7a)

To a stirred solution of 6a (0.1 g, 0.25 mmol) in a mixed solvent of DMSO (10 ml) and CH<sub>3</sub>OH (10 ml) was added CH<sub>3</sub>ONa (0.25 mmol), and the reaction mixture was stirred at room temperature for 15 hours. On the end of the deacetylation, Dowex-50 (H<sup>+</sup>) ion exchange resin (5 ml) was added, and the mixture was filtered and the resin was washed with CH<sub>3</sub>OH (3×10 ml). The combined filtrate and washings were concentrated *in vaccuo* to give brown residues which were crystallized from CH<sub>3</sub>OH, brown solid (35%): m. p >300°C; <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>)  $\delta$  3.22~3.86 (m, 6H, -CH<sub>2</sub>CH<sub>2</sub>-, H<sub>2</sub>, H<sub>3</sub>,),  $\delta$  4.28~5.10 (m, 4H, H<sub>1</sub>, H<sub>4</sub>, H<sub>5</sub>),  $\delta$  7.26 (s, 1H, H<sub>6</sub>).

# [N-(Uridine-6-yl-methyl)ethane-1,2-diamine]dichloroplatinum (II) (7b)

The same procedure described above in compound **7a** was employed for the preparation of **7b** to give grey solid (38%): m.p >300°C;  $^1\text{H-NMR}$  (DMSO-d<sub>6</sub>)  $\delta$  3.15~3.81 (m, 6H, -CH<sub>2</sub>CH<sub>2</sub>-, H<sub>2'</sub>, H<sub>3'</sub>),  $\delta$  4.8 (s, -OH),  $\delta$  4.49~5.61 (m, 5H, H<sub>1'</sub>, H<sub>4'</sub>, H<sub>5'</sub>, H<sub>5</sub>).

#### **Evaluation of cytotoxic activity**

The cytotoxic effect of the synthesized compounds was determined by the modified method (Mosmann, et al., 1983; Carmichael et al., 1987; Kim, et al., 1994 a-c); MTT-Microculture Tetrazolium Assay. The assay is dependent on the cellular reduction of water-soluble MTT (Sigma Chemical Co., St. Louis, M.O) by the mitochondrial dehydrogenase of vial cells to a blue water-nonsoluble formazan crystal product which can be measured spectrophotometrically.

Following appropriate incubation of cells (P-388, FM-3A and J-82) in the presence or absence of synthesized-compounds, 3-(4,5-dimethylthiazol-2-yl) -2,5-dipheny-2H-tetrazolium bromide (MTT) was added to each well and incubated at  $37^{\circ}$ C for further 4 hours before processing as described below. For cell growth, serially increasing cell numbers were plated in different columns across 96-well microtiter plates. Well growing cells were harvested, counted and inoculated at the concentration of  $2\times10^4$  cells/ml into 96-well mi-

**Table I.**  $IC_{50}$  values for uracil-platinum (II) complexes **3a**, and **3b** and uridine-platinum (II) complexes **6a**, **6b**, **7a** and **7b** 

Compounds	IC <sub>50</sub> (μg/ml) <sup>a</sup>		
	FM-3A <sup>b</sup>	P-388 <sup>c</sup>	J-82 <sup>d</sup>
3a	28	17	20
3 <b>b</b>	31	21	21
6a	48	34	19
6b	54	53	15
7a	34	27	13
7 <b>b</b>	60	60	60

\*mean values of triplicate runs. The concentration of synthesized compounds required to reduce cell number to 50% of controls in a growth inhibition assay.

crotiter plates. After 24 hours, synthesized compounds were applied to triplicate culture wells and culture were incubated an 37°C for 3 days. Following this incubation, 20 µl of MTT solution (5 mg/ml in phosphate buffer solution; KCl 0.2 g, KH<sub>2</sub>PO<sub>4</sub> 0.2 g, NaCl 8.0 g, Na<sub>2</sub>HPO<sub>4</sub> 1.15 g, MgCl<sub>2</sub> 0.101 g/l, pH- 7.4) was added to microculture wells. After 4 h incubation at 37°C, the supernatant was removed from each well and 100 µl of 100% DMSO was added to solubilize the formazan crystals which were formed by the cellular reduction of MTT. After thorough mixing with a mechanical plate mixer, absorbance spectra was read on ELISA Processor Microplate Reader (Behring Co.) at a wavelength of 570 nm and a reference wavelength of 650 nmv(absorbance peak for DMSO). All measurements were carried out in triplicates. There was good reproducibility between replicate wells with standard errors +10% (Table I).

### **RESULTS AND DISCUSSION**

A number of uracil platinum (II) complexes, 3a and **3b**, and uridine nucleoside platinum (II) complexes 6a, 6b, 7a and 7b, have been synthesized by the treatment of the diamine-uracils, 3a and 3b, or diamine-uridine nucleosides, 5a and 5b with the appropriate molar ratio of potassium tetrachloroplatinate in deionized water at 78°C for 48 hours. The starting 5chloromethyluracil (1a) and 6-chloromethyluracil (1b) were reacted with ethylenediamine to afford the respective 5-[(2-aminoethyl)amino]methyluracil (2a), and 6-[(2-aminoethyl)amino]methyluracil (2b) outlined in Scheme. The reactions of the starting materials 1a and **1b** with 1,2,3,5-tetra-0-acetyl-β-D-ribofuranose under hexamethyldisilagane, and trimethylsilyl chloride followed by the addition of stannic chloride, afforded 5-chloromethyl-2',3',5'-tri-O-acethyluridine (4a) and 6-chloromethyl-2',3',5'-tri-O-acethyluridine (4b), respectively.

<sup>a</sup>Reagents: A ethylenediamine, reflux, B  $K_2PCl_4$ ,  $H_2O$ ,  $78^{\circ}C$ , 48 hrs,  $N_2$ ; C i) HMDS, TMS-Cl, CH $_3$ CN, ii) SnCl $_4$ ; D i) CH $_3$ ONa, DMSO ii) Dowex-50 (H $^{\dagger}$ )

**Scheme**<sup>a</sup>. Synthesis of cisplatin Complexes-bearing Uracils (**3a,b**) and Uridines (**6a,b**, **7a,b**)

The uridine nucleosides 4a and 4b were reacted with ethylenediamine to yield the respective 5-[(aminoethyl)amino]methyl-2',3',5'-tri-O-acetyluracil (5a) and 6-[(aminoethyl)amino]methyl-2',3',5'-tri-O-acetyluracil (**5b**). The uridine nucleoside ethylenediamine ligands 5a and 5b were reacted with potassium tetrachloroplatinate to yield the nucleoside uridine-platinum (II) complexes 6a and 6b, and the complexes were deacetylated with CH3ONa to afford [N-(uridine-5-ylmethyl)ethane-1,2-diamine]dichloroplatinum (II) (7a) and [N-(uridine-6-yl-methyl)ethane-1,2-diamine]dichloroplatinum (II) (7b) respectively. The synthesized compounds were identified by the FT-IR, <sup>1</sup>H-NMR, UV and mass spectra. Six heretofore unreported uracil-platinum (II) complexes 3a and 3b, and uridine nucleoside platinum (II) complexes 6a, 6b, 7a and 7b were evaluated for antitumor efficacy against the following three cell lines:

- (a) mouse lymphoid neoplasma cell (P-388)
- (b) mouse mammary carcinoma cell (FM-3A)
- (c) human bladder carcinoma cell (J-82)

and none of our synthesized compounds showed any siginificant antitumor activity against above the three cell lines (Table I).

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<sup>&</sup>lt;sup>b</sup>Mouse mammary carcinoma cell.

<sup>&</sup>lt;sup>c</sup>Mouse lymphoid neoplasma cell.

dHuman bladder carcinoma cell.

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