Structural characterization of LaTi_{0.8}V_{0.2}O₃ compounds by transmission electron microscopy

J.Y. Kim, E.J. Yun*, K.S. Park**, K.H. Shim***, S.Y. Ryou**** and Y.H. Kim*****

Department of Materials Engineering, Hoseo University, Asan 336-795, Korea

- *Department of Electrical, Electronics and Control Engineering, Hoseo University, Asan 336-795, Korea
- **Department of Materials Science and Engineering, Chung-Ju National University, Chungju 380-702, Korea
- *** Electronics and Telecommunications Research Institute, Semiconductor Division, Yusong 305-350, Korea
- **** Department of Materials Science and Engineering, Sunmoon University, Asan 336-840, Korea
- ***** Department of Chemistry, Dankook University, Cheonan 330-714, Korea

투과전자현미경에 의한 LaTio,8Vo,2O3 화합물의 결정구조 분석

김좌연, 윤의중*, 박경순**, 심규환***, 류선윤****, 김유혁*****

호서대학교 재료공학과, 아산, 336-795

- *호서대학교 전기전자제어공학과, 아산, 336-795
- **충주산업대학교 재료공학과, 충주, 380-702
- ***한국전자통신연구원 반도체연구소, 유성, 305-350
- ****선문대학교 재료공학과, 이산, 336-840
- *****단국대학교 화학과, 천안, 330-714

Abstract The crystalline structure of $LaTi_{0.8}V_{0.2}O_3$ solid solutions, prepared by arc-melting palletized mixtures of predried La_2O_3 , V_2O_3 , TiO_2 , and Ti, was investigated by transmission electron microscopy and computer image simulation. Computer image simulations were performed by the multislice method for a wide range of sample thickness and defocusing value. The structure of $LaTi_{0.8}V_{0.2}O_3$ was determined as a GdFeO₃-type orthorhombic (a \approx 5.58 Å, b \approx 7.89 Å, and c \approx 5.58 Å) with a space group P_{nma} . No evidence of ordering of Ti and V atoms in $LaTi_{0.8}V_{0.2}O_3$ was found.

요 약 아크 용융을 사용하여, 예비 건조된 La_zO_s , V_zO_s , TiO_z , TiO_z , TiO_z 한물로부터 제조된 $LaTi_{0s}V_{0z}O_s$ 고용체의 결정구조를 투과전자현미경과 컴퓨터 이미지 시뮬레이션을 이용하여 연구하였다. 컴퓨터 이미지 시뮬레이션은 multislice 방법으로 여러 시편 두께와 초점 거리에서 실시하였다. $LaTi_{0s}V_{0z}O_s$ 의 결정구조는 P_{nma} 공간군을 가진 $GdFeO_s$ 형태의 사방정계($a\approx5.58$ Å, $b\approx7.89$ Å, and $c\approx5.58$ Å)로 결정되었다. $LaTi_{0s}V_{0z}O_s$ 에서 Ti와 V 원자의 규칙화에 대한 중거를 찾아 볼 수 없었다.

1. Introduction

The study of metal-insulator (M-I) transitions in transition-metal oxide systems is of very attractive to scientists [1]. LaTi_{1-x}V_xO₃ is an unusual perovskite-related compound because the antiferromagnetic ordering of the end members, LaTiO₃ and LaVO₃, is disrupted and the metal-like conductivity is observed for concentration in the range $0.10 \le x \le 0.25$ [2]. LaTiO₃ was first reported in 1954 [3] but it was not until 1979 that its structure was

determined as the GdFeO₃-type orthorhombic structure with a space group P_{nma} by the X-ray diffraction (XRD) studies from a melt-grown, highly twinned crystal [4]. At present, it is established that LaTiO₃ is a semiconductor with a band gap of 0.01 eV [5], and that LaTiO₃ with an oxygen content of 3.0 is paramagnetic [6] or canted antiferromagnetic with a metal (M)-insulator (I) transition at 125 K [7]. There are several different reports about the crystal structure of LaVO₃, i.e., cubic perovskite structure [8], tetragonally distorted perovskite structure [9],

hexagonal distortion of the cubic perovskite structure [10], and isostructural to LaTiO₃ with the orthorhombic GdFeO₃-type structure (space group P_{nma}) [11]. Despite of the discrepancy in the crystal structures, it is agreed that LaVO₃ is a semiconductor with a band gap of 0.14 eV [9].

It is not obvious why solid solutions of antiferromagnetic LaTiO3 and LaVO3 would give metallike conductivity. This is especially surprising for B-site-substituted transition metal oxide systems. Formation of solid solutions through disordered substitutions on perovskite B sites in metallic ABO₃ phases (B = transition metal) invariably localizes electrons and results in insulating behavior [12, 13]. It is suspected that it could take place in two cases. First, if these compounds have a orthorhombic or tetragonal structure of lower symmetry, the ordering of Ti and V atoms within discrete layers of the unit cell could account for the unexpected loss of magnetic ordering and onset of metallic conductivity. Second, for rare earth oxide compounds, physical properties depend strongly on the size of the rare earth ion [14] or may be structurally controlled [15]. M-I transitions are associated with the change in structural parameters such as metal-oxide (M-O) distance or metal-oxide-metal (M-O-M) angle. The structural parameters control the band width of valence and conduction bands, and thus directly affect the ratio of electron correlationship energy to band width and ultimately the transport properties. In this work, We describe the structural properties of LaTi_{0.8}V_{0.2}O₃ phase for explain of an unusual M-I transition in the LaTi_{1-x}V_xO₃ phases. The crystalline structure of LaTi_{0.8}V_{0.2}O₃ was investigated by transmission electron microscopy and computer image simulation.

2. Experimental

LaTi_{0.8}V_{0.2}O₃ solid solutions were prepared by arcmelting palletized mixtures of predried La₂O₃, V₂O₃, TiO₂, and Ti. The oxygen contents of the mixtures were adjusted to be oxygen deficient (~1.0) to compensate for the slight oxidation in the dc arc furnace. The palletized mixtures and a button of zirconium metal were placed in separated cavities of a water-cooled copper hearth inside an arc furnace that was evacuated and purged with gettered Ar gas several times prior to the reaction. Samples

Table 1 Input microscope parameters used in the computer simulation of lattice images

were fired several times with repetitive turnings. Before each pellet was fired, the zirconium metal was melted in order to purify the atmosphere inside the furnace. The resulting phases were then pulverized in a percussion mortar and then finely ground. All single phase samples were stored in a vacuum atmosphere dry box to prevent oxidation.

The crystal structure of LaTi_{0.8}V_{0.2}O₃ was investigated by transmission electron microscopy (TEM). In order to confirm the crystal structure determined from TEM, experimental high-resolution lattice images were compared with computer simulated ones. The computer simulations of lattice images were performed using the multislice method [16-18], assuming a GdFeO₃-type orthorhombic structure with a space group Pnma. The sample thicknesses ranged from 50 to 200 Å with a step of 7.89 Å, and the defocusing values from -300 to -1100 Å with a step of 200 Å. The input microscope parameters used in the computer simulation of the images are given in Table 1.

3. Results and discussion

The selected area diffraction (SAD) technique was used to determine if there was ordering between the Ti and V atoms in the B sublattice, and to obtain the unit cell length and space group from several different zone axes of many different crystals. Then, it was determined if unexpected spots were the result of multiple diffraction. Diffraction due to double diffraction can be distinguished by tilting the crystal about the direction which contains the spot in question. The intensity of this spot will remain if it is an allowed reflection and it will disappear if it is due to double diffraction.

We investigated the diffraction pattern with $[10\overline{1}]$, [010], [001], [100] and $[01\overline{1}]$ zone axes. Figure 1(a)

shows a selected area diffraction (SAD) pattern from LaTi_{0.8}V_{0.2}O₃. This pattern can be indexed as the [101] zone axis of an orthorhombic structure (a \approx c \approx 5.58 Å and b \approx 7.89 Å). For checking the double diffraction, systematic absences of reflections were investigated by tilting the crystal about the direction which contains the reflection in question. When the sample was tilted away from the [101] zone axis with respect to the [010] axis, the (010) spot was disappeared as shown in Fig. 1(b). This indicates that the (010) spot is an additional spot arising from the double diffraction. In Fig. 1(b), the conditions for allowed (0k0) reflections are k= even integer. When the sample was tilted away with respect to the [111] axis, the (111), (121), and (101)

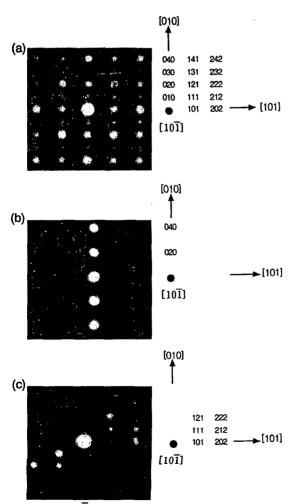


Fig. 1. (a) (101) SAD pattern from LaTi_{0.8}V_{0.2}O₃ with an orthorhombic structure. SAD patterns obtained from the sample tilted away from the [101] zone axis with respect to the (b) [010] and (c) [111] axes.

spots did not disappear as shown in Fig. 1(c). Therefore, these spots correspond to allowed reflections.

A different zone axis [010] of the same orthorhombic structure gave the SAD pattern shown in Fig. 2(a). The sample was tilted away from the [010] zone axis with respect to the [100], [001], and [201] axes, respectively, as shown in Figs. 2(b)-(d). The (100) and (300) spots were disappeared in Fig. 2(b), and the (001) and (003) spots were disappeared in Fig. 2(c). Hence the (100), (300),

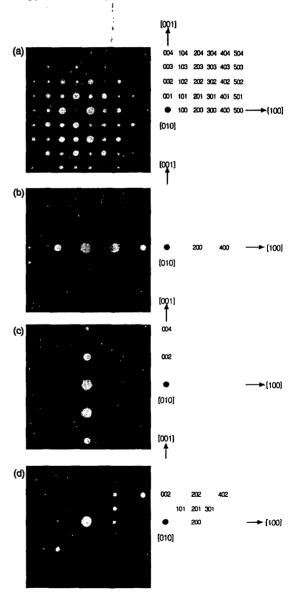


Fig. 2. (a) (010) SAD pattern from LaTi_{0.8}V_{0.2}O₃. SAD patterns obtained from the sample tilted away from the [010] zone axis with respect to the (b) [100], (c) [001], and (d) [201] axes.

(001), and (003) spots originated from the double diffraction. The (101), (201), and (301) spots, however, remained as shown in Fig. 2(d), indicating allowed spots. From these results, the conditions for allowed (h00) and (00l) reflections are h = eveninteger and l = even integer, respectively, and the conditions for allowed (h0l) reflections are h = evenor odd integer and l = even or odd integer. Also, the patterns from the [001], [100], and [011] zone axes of LaTi_{0.8}V_{0.2}O₃ were obtained, as shown in Figs. 3(a)-(c). The above results indicate that the most probable structure of LaTi_{0.8}V_{0.2}O₃ at room temperature is GdFeO₃-type orthorhombic (a \approx 5.58 Å, b \approx 7.89 Å, and $c\approx5.58$ Å) with a space group P_{nma} (reflection conditions; 0kl: k+1 = even integer, hk0:h = even, h00: h = even integer, 0k0: k = eveninteger, 001: 1 = even integer). This result agrees with the XRD and synchrotron X-ray diffraction (SXRD) data [19]. In addition, no evidence of ordering of Ti and V atoms in LaTi_{0.8}V_{0.2}O₃ was found (The

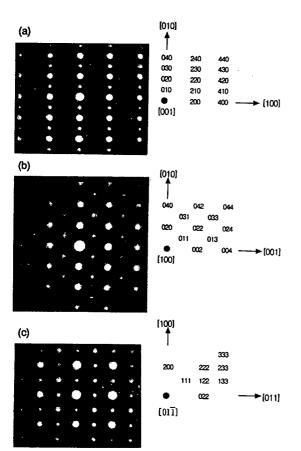


Fig. 3. (a) (001), (b) (100), and (c) (01 $\overline{1}$) SAD patterns from LaTi_{0.8}V_{0.2}O₃.

evidence of ordering of Ti and V atoms is an additional spots in SAD pattern with a lattice parameter of $2a_p$ where $2a_p$ is the unit cell parameter). From these results, the presence of metallic

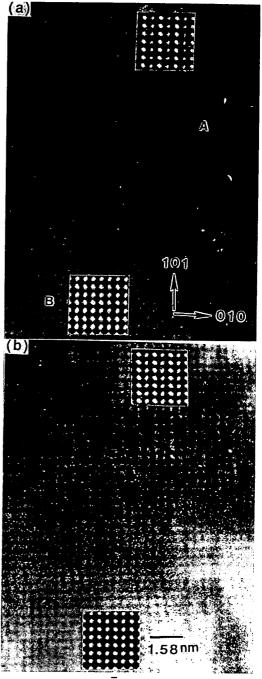


Fig. 4. Experimental (101) lattice images obtained from the same sample area under two different defocusing values, (a) -700 and (b) -900 Å.

conductivity in LaTi_{0.8}V_{0.2}O₃ is not obvious now. We can only suppose this behavior with the change in structurally parameters (Hubbard model) [15].

In order to confirm the orthorhombic structure of LaTi_{0.8}V_{0.2}O₃ determined from TEM, a visual comparison of experimental and computer simulated lattice images of LaTi_{0.8}V_{0.2}O₃ was carried. (101) experimental lattice images from the same sample area were obtained under two different defocusing values (Fig. 4). In Fig. 4, the periodicities along the [010] and [101] directions in area A correspond to the (010) and (101) interplannar spacings, respectively, and the periodicities along the [010] and [101] directions in area B to the (020) and (101) interplannar spacings, respectively. It is shown that the (010) spot was generated by double diffraction (see Figs. 1(a) and (b)), which contributes to the generation of the (010) lattice fringes in area A.

The computer simulations of lattice images were performed using the multislice method, assuming a GdFeO₃-type orthorhombic. The atomic positions and unit lengths obtained from the Rietveld refinement of neutron data are shown in Table 2. By a comparison of the experimental images (Fig. 4) and the simulated images (Fig. 5), the experimental images were in good agreement with the simulated images. The experimental images indicated by A in

Table 2 Refined structural parameters for LaTi_{0.8}V_{0.2}O₃ at 50 K

Atom	Site symmetry	x	у	z	B (iso)	Fraction
La Ti V O (1)	.m. 1 1 .m.	0.5 0.5 0.0886	0 0 0.25	0.9944 0 0 0.0771 0.7142	$0.18 \\ 0.18 \\ 0.69$	0.82 0.18 1.09

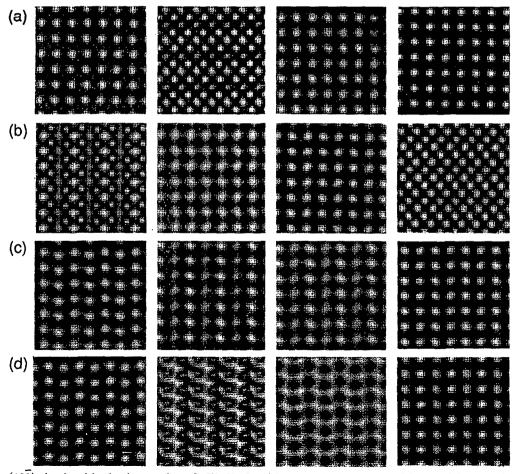


Fig. 5. (101) simulated lattice images from $LaTi_{0.8}V_{0.2}O_3$ with a sample thickness of (a) 94.8, (b) 126.4, (c) 158.0, and (d) 189.6 Å. (The defocusing values for each column are, from left to right, $\Delta f = -500$, -700, -900, and -1100 Å).

Figs. 4(a) and (b) are well matched with the simulated images obtained -700 and -900 Å defocusing values, respectively, at 158 Å sample thickness. Also, the experimental images indicated by B in Figs. 4(a) and (b) are well matched with the simulated images obtained -700 and -900 Å defocusing values, respectively, at 126 Å sample thickness. The good agreement between the experimental and simulated images confirms that LaTi_{0.8}V_{0.2}O₃ compound has a GdFeO₃-type orthorhombic (a \approx 5.58 Å, b \approx 7.89 Å, and c \approx 5.58 Å) with a space group P_{nma} .

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