

## Effect of O<sub>3</sub> and NO<sub>2</sub> on Net Photosynthesis, Transpiration and Accumulation of Nitrite in Sunflower Leaves

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Photosynthesis and transpiration rates were simultaneously measured in attached sunflower leaves (*Helianthus annuus* L. cv. Russian Mammoth) during exposure to NO<sub>2</sub> and O<sub>3</sub> to determine the effect of mixed gas on photosynthesis and the stomatal aperture.

The application of O<sub>3</sub> alone reduced both the net photosynthetic and transpiration rates. An analysis of the CO<sub>2</sub> diffusive resistances indicated that the main cause affecting photosynthesis reduction during O<sub>3</sub> exposure was not the internal gas phase of the leaf ( $rCO_2^{iq}$ ) but rather the liquid phase or mesophyll diffusive resistance ( $rCO_2^{liq}$ ), suggesting that there is a very concomitant relation between photosynthetic reduction and  $rCO_2^{liq}$ . The application of NO<sub>2</sub> alone caused a marked reduction of the net photosynthesis yet no significant reduction of transpiration, indicating that NO<sub>2</sub> affects the CO<sub>2</sub> fixation processes with no influence on the stomatal aperture. A greater reduction in the photosynthesis of sunflower plants was caused by the application of NO<sub>2</sub> alone as compared to a combination of NO<sub>2</sub> and O<sub>3</sub>. NO<sub>2</sub> alone reduced the photosynthetic rate by 90%, whereas a mixture of NO<sub>2</sub> and O<sub>3</sub> reduced it by 50%.

Key words : gas mixture, nitrite, NO<sub>2</sub>, O<sub>3</sub>, sunflower

### 1. Introduction

Recently, increasing attention has been given to the effects of air pollutant mixtures on plants due to the presence of numerous gaseous and particulate compounds in the lower atmosphere. Despite numerous studies dealing with the singular effects of sulphur dioxide (SO<sub>2</sub>), nitrogen dioxide (NO<sub>2</sub>), and ozone (O<sub>3</sub>), there are still insufficient reports on the combined effects of air pollutants on plant growth<sup>6,11,13,23</sup>. The responses of plants to air pollutant mixtures are frequently categorized for convenience into three types: additive, greater than additive, and less than additive<sup>7</sup>. Numerous studies have been conducted on visible injuries and growth reductions caused by phytotoxic air pollutant mixtures<sup>8,20</sup>. Air pollutants are the cause of a number of different biochemical effects on

exposed leaves, which often result in changes in the cellular structure. Angela *et al.*<sup>1</sup> reported that fumigation with SO<sub>2</sub> or NO<sub>2</sub> led to swollen thylakoids and a reduction in the number of grana stacks, when compared with the control. Ambient concentrations of O<sub>3</sub> in many areas can alter the permeability of plant cell membranes, disrupt the metabolism, decrease the foliar chlorophyll and photosynthesis, change the photosynthate allocation, and suppress the growth and yield<sup>12,15,17</sup>. Unfortunately, however, the results are often contradictory, probably due to the influence of various factors such as light, temperature, humidity, duration of exposure, and the concentration ratio of the air pollutants in the mixtures. The antagonistic effects of SO<sub>2</sub> towards O<sub>3</sub> were reported on by Ashmore and Oenal<sup>3</sup> for six cultivars of barley, and indicated that leaf injury was higher after ex-

posure to O<sub>3</sub> alone than after exposure to both pollutants. Treatment with a mixture of O<sub>3</sub> and SO<sub>2</sub> induced a higher stomatal resistance compared to a single action<sup>5)</sup>, thereby suggesting antagonistic effects, i.e. the pollutant flux into the leaves was reduced.

These reported results have focused on the effects of SO<sub>2</sub> and O<sub>3</sub> or NO<sub>2</sub> and SO<sub>2</sub> mixtures. Furukawa<sup>9)</sup> indicated that the most ubiquitous air pollutants contaminated in the atmosphere of urban and suburban areas have changed from SO<sub>2</sub> into NO<sub>2</sub> and O<sub>3</sub>. However, there is little information concerning the effects of NO<sub>2</sub> and O<sub>3</sub> mixtures on the photosynthesis of higher plants. The present paper demonstrates the possible effects of NO<sub>2</sub> and O<sub>3</sub> and their mixtures on net photosynthesis and transpiration, even though the concentration of NO<sub>2</sub> applied was relatively higher than that found in the field.

## 2. Materials and Methods

### 2.1. Plant materials

Sunflower seedlings (*Helianthus annuus* L. cv. Russian Mammoth) were grown at 25°C with 70% relative humidity in a growth chamber. The plants were cultivated for 4 weeks in plastic pots (11 x 15 cm) filled with a mixture of vermiculite, perlite, and gravel (2:2:1, v/v/v). Each pot contained 5 g of Magamp-K and 15 g of magnesia lime. No additional nutrients were supplied to increase the susceptibility to NO<sub>2</sub><sup>21)</sup>.

### 2.2. Fumigation system

The plants were exposed to NO<sub>2</sub> and/or O<sub>3</sub> using an acrylic assimilation chamber (125 liter, cubic) which was placed in a controlled environment room (1.7 x 2.3 x 2.0 m). The field air was passed in succession through activated charcoal and a catalyst containing MnOx and CuO filters to remove any ambient air pollutants and then fed into the controlled environment room. This filtration system was able to remove O<sub>3</sub> and SO<sub>2</sub> almost perfectly, however a trace amount of NO<sub>2</sub> (below 5 nl l<sup>-1</sup>) remained in the room. NO<sub>2</sub> gas from a compressed cylinder containing 2 ml l<sup>-1</sup> NO<sub>2</sub> in N<sub>2</sub> was injected

through a solenoid valve into the air stream. The concentration of NO<sub>2</sub> in the room was regulated by a thermal mass-flow controller equipped with a controlling system consisting of a chemiluminescent NO-NO<sub>2</sub>-NO<sub>x</sub> analyzer (Thermo Electron, Model 14). Ozone was generated by a silent electrical discharge in dry oxygen and regulated by a system similar to that described for NO<sub>2</sub> equipped with a controlling system consisting of a chemiluminescent O<sub>3</sub> analyzer (Kimoto, Model 806). Recordings of the pollutant concentrations in the room indicated that on starting the fumigation, the concentration reached 90 % of the fixed level within 5 min. Pollutant concentrations were regulated within ± 1% of the desired levels.

### 2.3. Measurement of gas exchange

Fully expanded leaves were accommodated in the assimilation chamber. The stem was led through a port at the bottom of the chamber, so that the leaves were inside whereas the roots and pot were outside. Measurements were performed at 28 ± 0.5 °C with 75% relative humidity. Two small fans (10 cm diameter) placed on the inner wall stirred the chamber air. The air was continuously sucked by a pump through a suction pipe on the upper side of the chamber. The air flow rate was measured by a rotameter and was adjusted to 551 min<sup>-1</sup>. The wind speed inside the chamber was 0.4 m s<sup>-1</sup>. This magnitude of wind speed minimized the diffusive resistance of the boundary layer to water vapor and a CO<sub>2</sub> transfer. Using wet filter paper of a similar size and orientation to the leaves, the resistance of the boundary layer to water vapor transfer was determined as 0.06 to 0.1 s cm<sup>-1</sup>.

The illumination system consisted of 24 metal halide lamps (Yoko Lamp, Toshiba, 400 W). The light was filtered through a heat-absorbing glass filter, which removed any radiation above 800 nm. The quantum flux density in the assimilation chamber was 500 mmol m<sup>-2</sup> s<sup>-1</sup>. The transpiration rate was determined by the gravimetric method using an electronic top-loading balance (Mettler, Model PL-3000). The transpirational water loss was continuously recorded using a thermal data acquisition system. The pots were enclosed in plastic bags to prevent the evaporation of water from the pot

surfaces. The net photosynthetic rate was determined in an open circuit system by measuring the CO<sub>2</sub> concentrations in the inlet and outlet of the chamber using an infra-red CO<sub>2</sub> analyzer (Shimazu, Model URA-2S).

#### 2.4. Estimation of diffusive resistance

The photosynthesis and transpiration in the assimilation chamber were measured simultaneously, and the diffusive resistances to the CO<sub>2</sub> transfer from the bulk air to the CO<sub>2</sub> fixation site were also determined. The resistances to CO<sub>2</sub> diffusion through the boundary layer and internal gas-phase of the leaf ( $rCO_2^{gas}$ ) and from the surface of the mesophyll cells to the site of CO<sub>2</sub> fixation ( $rCO_2^{iq}$ ) were calculated from the net photosynthesis ( $\mu\text{mol m}^{-2}\text{ s}^{-1}$ ) and transpiration ( $\text{mol m}^{-2}\text{ s}^{-1}$ ) according to the method developed by Gastra<sup>10</sup>. The degree of stomatal aperture was designated as the stomatal diffusive conductance. The diffusion coefficient of CO<sub>2</sub> was related to that of water vapor, and a conversion factor of 1.56<sup>18</sup> was applied to convert the gas-phase diffusive resistance for the flux of water vapor into stomatal diffusive conductance.

#### 2.5. Nitrite assay

The nitrite content was determined using a modified version of the method developed by Yoneyama *et al.*<sup>26</sup>. After measuring the fresh weight of the leaves, they were frozen and stored at -80 °C until used. The frozen leaves were ground in a cold mortar with liquid nitrogen and then with a 0.1 M potassium phosphate buffer (pH 7.5). The homogenate was squeezed through a nylon cloth and the filtrate was centrifuged at 18,000 x g for 20 min. To 0.1 ml of the supernatant, 1.5 ml of 1.0% (w/v) sulfamide in 1.5 N HCl and 1.5 ml of 0.02% (w/v) N-(1-naphthyl) ethylenediamine dihydrochloride were added and the mixture was left for 20 min. Thereafter, the optical density was measured at 540 nm.

#### 2.6. Statistical analysis

To test if there was a significant difference between the two pollutant combinations, F statistics

were applied. If the difference was insignificant at a 95% confidence interval, the effect was designated as additive. In contrast, if the difference was significant at the same confidence interval, the effect was noted as greater or less than additive.

### 3. Results

#### 3.1. Effect of O<sub>3</sub> on photosynthesis, transpiration, and diffusive resistance

Treatment for 2 h with 10 nmol O<sub>3</sub> l<sup>-1</sup> caused no significant changes in either net photosynthesis or transpiration in the sunflower leaves (Figs. 1A and 1B). The exposure to 20 nmol O<sub>3</sub> l<sup>-1</sup> resulted in a progressive decline of photosynthesis and transpiration during the exposure periods. A 2 h exposure to 20 nmol O<sub>3</sub> l<sup>-1</sup> reduced the net rates of photosynthesis and transpiration to 65 and 75 % of the pre-exposure rates, respectively (Figs. 1A and 1B). Fig. 1C shows the effect of O<sub>3</sub> on  $rCO_2^{gas}$  and  $rCO_2^{iq}$ . The net photosynthesis response to O<sub>3</sub> was largely reflected by changes in the  $rCO_2^{iq}$ .

If the inhibition of net photosynthesis resulted solely from the stomatal closure, the  $rCO_2^{iq}$  should remain constant during exposure to O<sub>3</sub>. However, 20 nmol O<sub>3</sub> l<sup>-1</sup> caused a gradual increase of  $rCO_2^{iq}$  immediately after exposure and finally the increase reached 1.5 times the initial value.

#### 3.2. Effect of NO<sub>2</sub> on photosynthesis, transpiration, and diffusive resistance

Net photosynthesis was found to be more sensitive to NO<sub>2</sub> than transpiration (Figs. 2A and 2B). Exposure to NO<sub>2</sub> resulted in a rapid decline of the net photosynthetic rate. The inhibition of transpiration by NO<sub>2</sub> was much smaller than that of O<sub>3</sub>. Exposure to 97 or 195 nmol of NO<sub>2</sub> for 2 h reduced the net photosynthetic rate by 20 and 90%, respectively, however, no significant reduction in transpiration was detected. Even when the net photosynthetic rate was reduced to 10% of the initial rate by exposure to 195 nmol of NO<sub>2</sub> l<sup>-1</sup> for 2 h, no significant increase in  $rCO_2^{gas}$  was detected

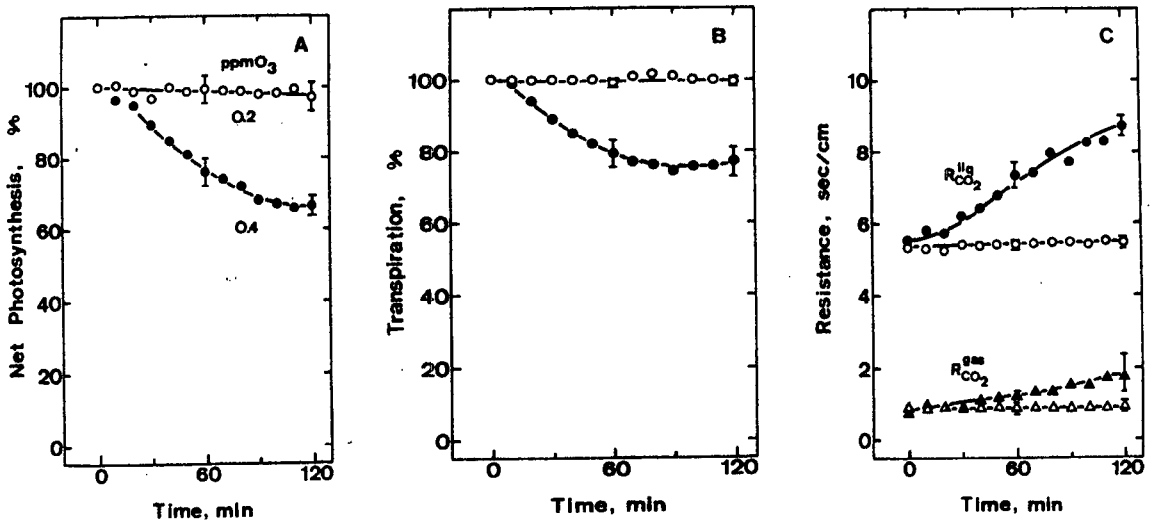


Fig. 1. Effect of O<sub>3</sub> on net photosynthesis(A), transpiration(B), and CO<sub>2</sub> diffusive resistances(C)(○△, 0.2; ●▲, 0.4) in sunflower leaves. Rates of net photosynthesis and transpiration are expressed as percentages of the pre-exposure rates. Gas-phase ( $rCO_2^{gas}$ ) and liquid-phase ( $rCO_2^{liq}$ ) diffusive resistances were estimated using the data of the net photosynthesis and transpiration rates. O<sub>3</sub> concentrations: 10 nmol l<sup>-1</sup> (0.2, ○), 20 nmol l<sup>-1</sup> (0.4, ●). Each point is the mean of at least three replicates. Error bars indicate representative standard deviations.

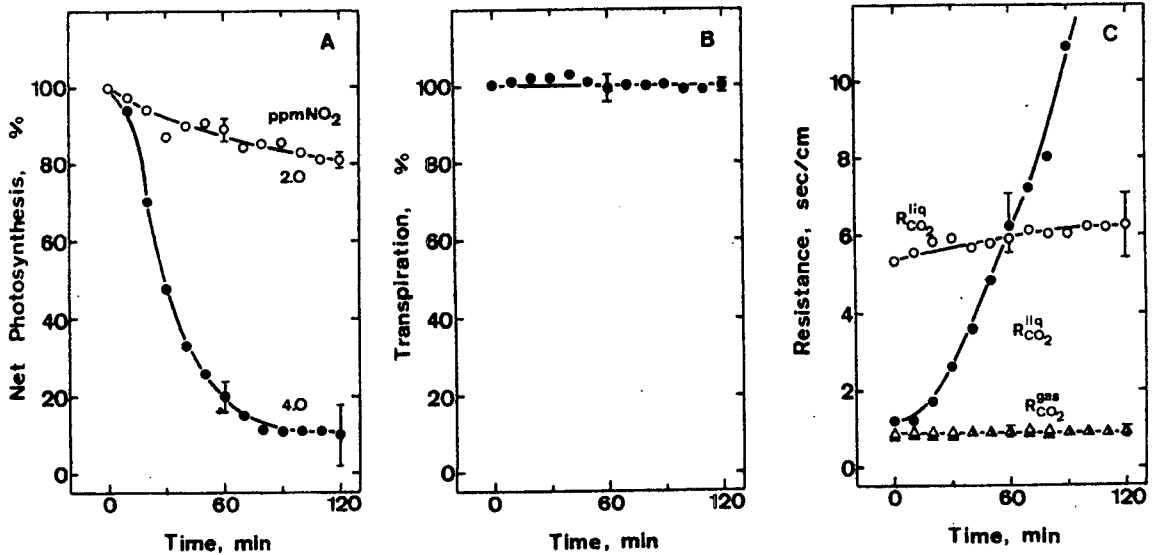


Fig. 2. Effect of NO<sub>2</sub> on net photosynthesis(A), transpiration(B), and CO<sub>2</sub> diffusive resistances(C)(○△, 2.0; ●▲, 4.0) in sunflower leaves. NO<sub>2</sub> concentrations: 97 nmol l<sup>-1</sup> (2.0, ○), 195 nmol l<sup>-1</sup> (4.0, ●). Each point is the mean of at least three replicates. Error bars indicate representative standard deviations.

(Fig. 2C). These results suggest that the threshold concentration required to cause stom-atal closure

with a 2 h treatment of NO<sub>2</sub> is above 195 nmol of NO<sub>2</sub> l<sup>-1</sup> in sunflower leaves.

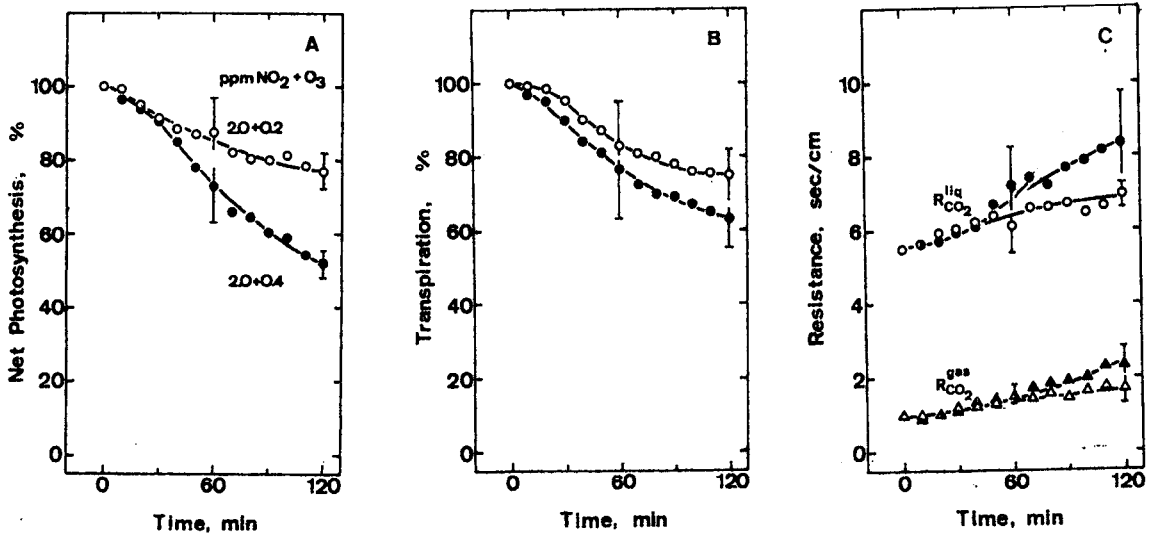


Fig. 3. Effect of O<sub>3</sub> and NO<sub>2</sub> mixtures on net photosynthesis(A), transpiration(B), and CO<sub>2</sub> diffusive resistances(C)(○△, 2.0+0.2; ●▲, 2.0+0.4) in sunflower leaves. The NO<sub>2</sub> concentration was 97 nmol l<sup>-1</sup> and the O<sub>3</sub> concentration was 10 (2.0+0.2, ○) or 20 nmol l<sup>-1</sup> (2.0+0.4, ●). Each point is the mean of at least three replicates. Error bars indicate representative standard deviations.

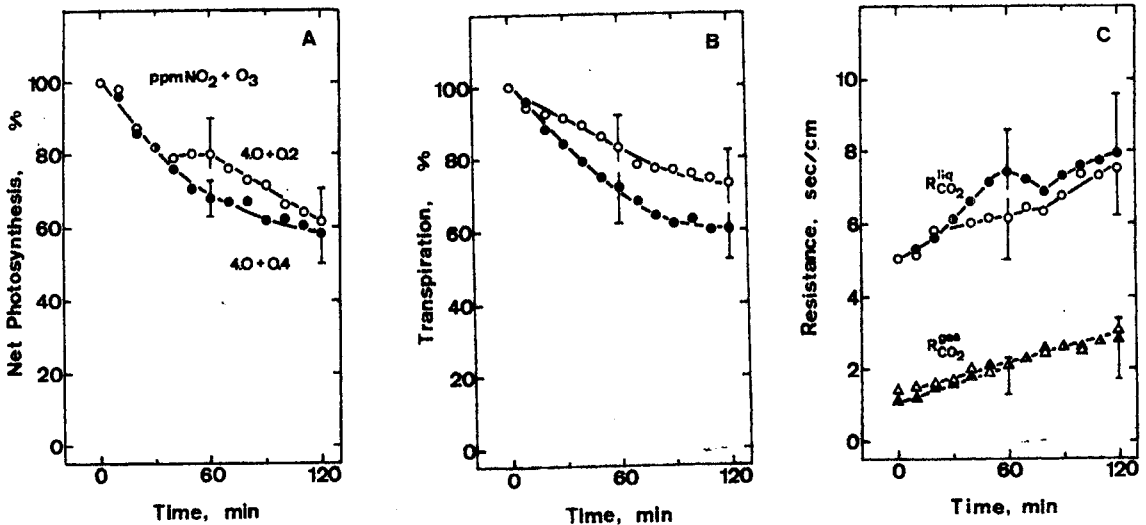


Fig. 4. Effect of O<sub>3</sub> and NO<sub>2</sub> mixtures on net photosynthesis(A), transpiration(B), and CO<sub>2</sub> diffusive resistances(C)(○△, 4.0+0.2; ●▲, 4.0+0.4) in sunflower leaves. The NO<sub>2</sub> concentration was 195 nmol l<sup>-1</sup> and the O<sub>3</sub> concentration was 10 (4.0+0.2, ○) or 20 nmol l<sup>-1</sup> (4.0+0.4, ●). Each point is the mean of at least three replicates. Error bars indicate representative standard deviations.

### 3.3. Effect of NO<sub>2</sub> and O<sub>3</sub> mixture on photosynthesis, transpiration, and diffusive resistance

The reduction of net photosynthesis induced by a mixture of NO<sub>2</sub> and O<sub>3</sub> was additive when

the concentration of NO<sub>2</sub> in the mixture was low (Fig. 3A). Treatment with a mixture of 97 nmol of NO<sub>2</sub> l<sup>-1</sup> plus 10 nmol of O<sub>3</sub> l<sup>-1</sup> for 2 h reduced the net photosynthesis to 77 % of the initial rate, which was not significantly different from the

inhibition caused by  $\text{NO}_2$  or  $\text{O}_3$  gas alone. Exposure to a mixture of 97 nmol of  $\text{NO}_2 \text{ l}^{-1}$  plus 20 nmol of  $\text{O}_3 \text{ l}^{-1}$  also caused an additive reduction of net photosynthesis. In contrast to these additive effects, when the concentration of  $\text{NO}_2$  in the mixture was high enough (195 nmol of  $\text{NO}_2 \text{ l}^{-1}$ ) to almost perfectly inhibit net photosynthesis by  $\text{NO}_2$  alone, the inhibition induced by 195 nmol of  $\text{NO}_2 \text{ l}^{-1}$  plus 10 nmol of  $\text{O}_3 \text{ l}^{-1}$  was less than additive (Fig. 3A). The behavior of transpiration differed considerably from that of photosynthesis during exposure to a mixture. A mixture of  $\text{NO}_2$  and  $\text{O}_3$  at any combination of concentrations of each gas produced a significantly greater reduction of transpiration than would be anticipated from summing the effect due to each gas alone (Figs. 3B and 4B). Since the reduction of transpiration and net photosynthesis occurred nearly simultaneously with the duration of the exposure, the degree to which net photosynthesis is affected by stomatal closure can be estimated. Figs. 3C and 4C show that  $r\text{CO}_2^{\text{gas}}$  and  $r\text{CO}_2^{\text{liq}}$  both increased just after the initiation of the mixed treatment. Exposure to a  $\text{NO}_2$  and  $\text{O}_3$  mixture caused a greater increase in  $r\text{CO}_2^{\text{gas}}$  than that caused by  $\text{O}_3$  or  $\text{NO}_2$  alone. The most significant increase in  $r\text{CO}_2^{\text{gas}}$  was observed when the leaves were treated with a mixture of 195 nmol of  $\text{NO}_2 \text{ l}^{-1}$  plus 10 or 20 nmol of  $\text{O}_3 \text{ l}^{-1}$ .

### 3.4. Accumulation of nitrite

To examine the nitrate accumulation in the  $\text{NO}_2$ -treated leaves, the nitrite content in the sunflower leaves treated with  $\text{NO}_2$  was measured. Fig. 5 represents the relationship between the accumulated amount of nitrite and the relative rate of net photosynthesis. On a semilogarithmic semilogarithmic scale, a linear relationship between these two factors was observed in the  $\text{NO}_2$ -treated leaves.

## 4. Discussion

The present results suggest that the response of photosynthesis to a mixture of  $\text{NO}_2$  and  $\text{O}_3$  is quite different from that of transpiration. Bender *et al.*<sup>(6)</sup> have shown that the growth suppression of beans caused by  $\text{O}_3$  alone did not occur when the plants were treated with  $\text{NO}_2$ , probably the stimulatory

effect of  $\text{NO}_2$  counteracts the negative effects of  $\text{O}_3$ . They indicated that the growth stimulation during vegetative growth by  $\text{NO}_2$  and  $\text{O}_3+\text{NO}_2$  was accompanied by corresponding changes in the plant nitrogen metabolism. Similar results were shown in the present study when the concentration of  $\text{NO}_2$  was high enough to almost perfectly inhibit net photosynthesis (Fig. 2). The effect of a mixture was less than additive despite a significant inhibition of net photosynthesis by higher concentrations of  $\text{O}_3$  (Fig. 3). When the concentration of  $\text{NO}_2$  was low and the inhibition of net photosynthesis was moderate, the effect of a mixture was always additive, whether the concentration of  $\text{O}_3$  was high or low. In contrast, the transpiration inhibition induced by a mixture was always greater than additive irrespective of the  $\text{O}_3$  concentration. Even when the transpiration was reduced by half by 20 nmol of  $\text{O}_3 \text{ l}^{-1}$  alone, the effect of a mixture was greater than additive. These results indicate that when the concentration of  $\text{NO}_2$  is below the threshold to inhibit transpiration, the effect of a mixture is greater than additive even when treatment with  $\text{O}_3$  inhibits transpiration.

The analysis of the  $\text{CO}_2$  diffusion processes exhibited markedly different photosynthetic responses to individual and mixed gas treatments with  $\text{NO}_2$  and  $\text{O}_3$ . The photosynthetic decline caused by  $\text{NO}_2$  or  $\text{O}_3$  alone was mainly attributed to the increase in  $r\text{CO}_2^{\text{liq}}$ , although  $r\text{CO}_2^{\text{gas}}$  increased slightly during treatment with  $\text{O}_3$  alone (Figs. 1 and 2). In contrast, mixed gas treatments affected  $r\text{CO}_2^{\text{gas}}$  and  $r\text{CO}_2^{\text{liq}}$  simultaneously, suggesting that the contribution of the increase in  $r\text{CO}_2^{\text{gas}}$  to the decrease in net photosynthesis was comparable to that of  $r\text{CO}_2^{\text{liq}}$  in a mixed treatment (Figs. 3 and 4). The increase in  $r\text{CO}_2^{\text{liq}}$  reflects alterations in the available enzyme levels, since  $r\text{CO}_2^{\text{liq}}$  is influenced by enzymatic activity. Accordingly, these results indicate that both stomatal closure and photosynthesis inhibition occur concurrently during mixed treatments. This is also indicated by the observation that the increase in  $r\text{CO}_2^{\text{gas}}$  was roughly parallel to the increase in  $r\text{CO}_2^{\text{liq}}$  over the exposure period applied in the present experiment. These observations suggest that the less than additive inhibition of net photosynthesis induced by a  $\text{NO}_2$  and  $\text{O}_3$  mixture is caused by stomatal closure, resulting in the reduced in-

corporation of phytotoxic gas into leaves.

NO<sub>2</sub> has been shown to increase the nitrate reductase activity of conifer needles<sup>23,24</sup>. The flux of NO<sub>2</sub> into a spruce branch is linearly correlated with the stomatal conductance of the needles<sup>23</sup> and the nitrate reductase activity responds linearly to the uptake of NO<sub>2</sub>-N<sup>24</sup>. According to a recently revised model<sup>1,4</sup>, NO<sub>2</sub> is converted to NO<sub>3</sub><sup>-</sup> in the apoplast of the leaves and subsequently incorporated in amino acids and proteins<sup>19</sup>. Bender *et al.*<sup>6</sup> indicated that NO<sub>2</sub> treatments stimulate growth until anthesis, suggesting that plants are able to detoxify and utilize the additional nitrogen source.

The physiological data of Angela *et al.*<sup>2</sup> from NO<sub>2</sub>-exposed trees is not consistent with the current results. They reported that after NO<sub>2</sub>-exposure at a concentration relevant to ambient conditions, the photosynthetic capacity increases together with a slight increase in pigmentation, however, the current results show a marked inhibition of photosynthesis. These results indicate that the effect of NO<sub>2</sub> fumigation depends greatly on the duration of the NO<sub>2</sub> exposure, NO<sub>2</sub> concentration, plant species, and ability of plant to detoxify NO<sub>2</sub>. The linearity between the accumulated amount of nitrite and the magnitude of the inhibition of net photosynthesis caused by the stomatal opening and absorption of NO<sub>2</sub> during treatment with NO<sub>2</sub> results in an increased nitrite concentration in leaves (Fig. 5). Yoneyama and Sasakawa<sup>26</sup> also identified a nitrite accumulation in NO<sub>2</sub>-treated leaves. The exposure of plant leaves to NO<sub>2</sub> would seem to result in the leaves absorbing NO<sub>2</sub> through stomata and converting it into nitrite in the leaf tissues. Nitrite, which is accumulated in plant cells, can penetrate into chloroplasts and affect the photosynthetic processes. The inhibition of carbonic anhydrase by nitrite may be one of the mechanisms that inhibit CO<sub>2</sub> fixation, since CO<sub>2</sub> transfer conductance initially depends on carbonic anhydrase activity and this activity may be a limiting factor to photosynthesis under stress conditions<sup>14,22</sup>. Mohammad *et al.*<sup>16</sup> showed that a decreased carbonic anhydrase activity of about 18% can affect CO<sub>2</sub> diffusion towards carboxylation sites in the chloroplast. As a result, the CO<sub>2</sub> concentration in the chloroplast in the ribulose-1,5-bisphosphate carboxylase/oxygenase vicinity is reduced. Accordingly, further investigations are necessary to under-

stand plant responses to the correlation between the accumulated amount of nitrite, stomatal opening, and photosynthesis.

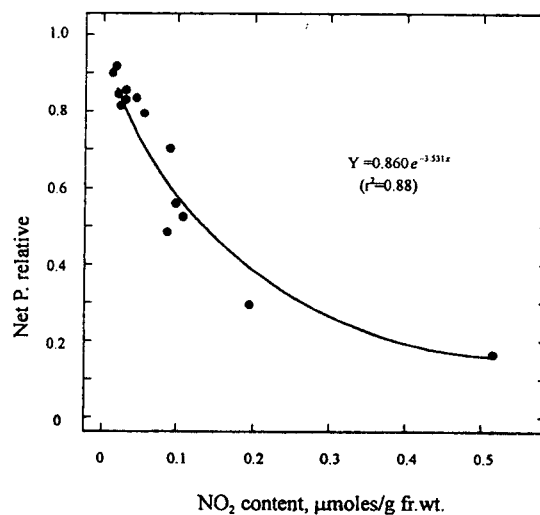


Fig. 5. Relationship between accumulated amount of nitrite in leaf and net photosynthesis (Net P). 195 nmol NO<sub>2</sub> l<sup>-1</sup> was treated for one hour then the accumulated amount of nitrite and Net P was measured in the leaves. The Net P shows the relative values to a control (1.0 means 100%), which was not treated with NO<sub>2</sub>.

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