

Synthesis of 7,8-Dichloro-6-Nitro-1*H*-1,5-Benzodiazepine-2,4-(3*H*, 5*H*)-dione as a potential NMDA Receptor Glycine Site Antagonist

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An efficient procedure for the preparation of 7,8-dichloro-6-nitro-1*H*-1,5-benzodiazepine-2,4-(3*H*, 5*H*)-dione(7) as a potential lead compound for the NMDA receptor glycine binding site antagonist, starting from readily available 4,5-dichloro-2-nitroaniline(8), is described. The key step in the synthesis involves the cyclization of malonic ester amide **10** to compound **11**.

Key words: NMDA Receptor, Glycine site, Quinoxalinedione, Benzodiazepine.

INTRODUCTION

The ionotropic *N*-methyl-*D*-aspartate(NMDA) receptor antagonists acting at the glycine site have broad therapeutic potential and offer a highly attractive target for CNS drug discovery such as Alzheimer disease, stroke, head injury, epilepsy and schizophrenia (Cotman *et al.*, 1981, Meldrum *et al.*, 1990, Carling *et al.*, 1993, Iverson *et al.*, 1994 and Leeson *et al.*, 1994). In comparison with NMDA receptor antagonists acting competitively at the glutamate site or noncompetitively as channel blockers, glycine antagonists may have significantly improved side-effect profiles (Kemp *et al.*, 1993). Many classes of glycine antagonists with high affinity and selectivity have now been synthesized, and they can be categorized mainly as kynurenic acid **1** (Leeson *et al.*, 1991), 2-quinolone **2** (McQuaid *et al.*, 1992), 2-carboxytetrahydroquinoline **3** (Leeson *et al.*, 1992), quinoxalinedione **4** (Honore *et al.*, 1988, Epperson *et al.*, 1993 and Cai *et al.*, 1997), 2-carboxyindole **5** (Salituro *et al.*, 1992 and Fabio *et al.*, 1997) and benzazepine-2,5-dione **6** (Swartz *et al.*, 1992). Their representative chemical structures are depicted in Fig. 1.

In spite of many classes of compounds mentioned above, most of these lack activity in the central nervous

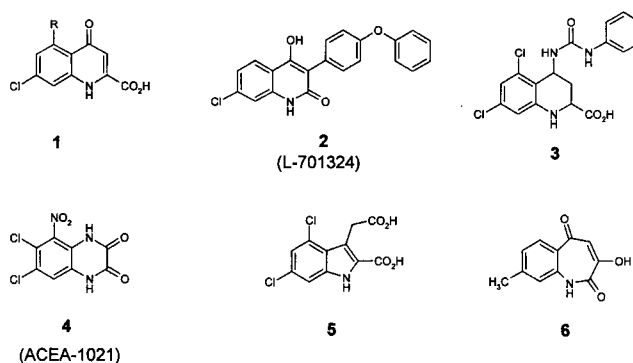


Fig. 1. Structure of Various Glycine Site Antagonist

system following systemic dosing. Evidently, significant improvements in blood-brain barrier permeability and bioavailability are important factors to be considered as a good drug candidate. The recently described compound ACEA 1021 (**4**), (Lessler *et al.*, 1989, Kessler *et al.*, 1989, Woodward *et al.*, 1993 and Cai *et al.*, 1997) and L-701324 (**3**), (Kulagowski *et al.*, 1994) appear to have the best systemic activity to date and are expected to provide further impetus for the evaluation of the glycine site *in vivo* (Leeson *et al.*, 1994).

In connection with designing new lead molecules for NMDA glycine site antagonist, we were interested in benzodiazepine type compound **7**; ring expansion product of ACEA 1021 and yet having similar structure to benzazepine **6** (Fig. 2).

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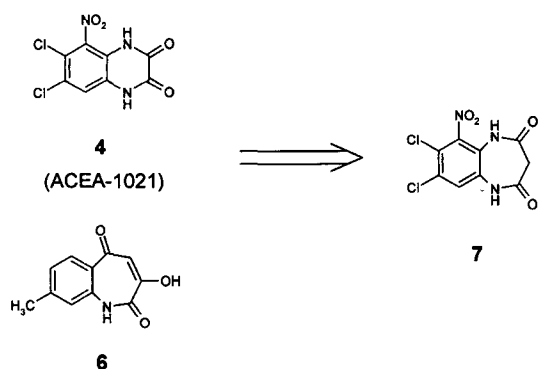
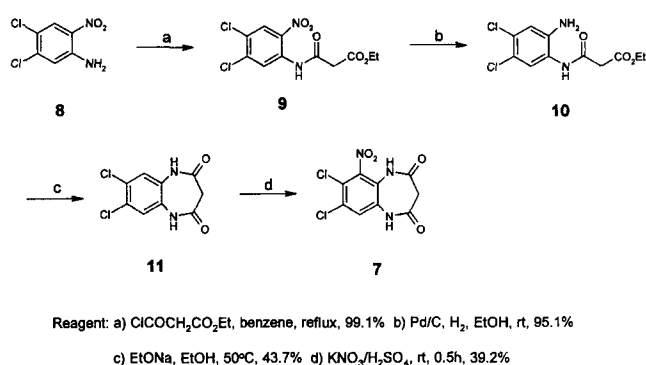


Fig. 2. A conceptual design for target compound 7



Scheme 1. Synthesis of 7,8-dichloro-6-nitro-1H-1,5-benzodiazepine-2,4-(3H,3H)-dione

The present paper describes synthesis of 7,8-dichloro-6-nitro-1H-dihydro-1,5-benzodiazepine-2,4-(3H,5H)-dione (7) as a potential lead compound for the NMDA receptor glycine binding site antagonist, starting from readily available 4,5-dichloro-2-nitroaniline (8) (Scheme 1).

MATERIALS AND METHODS

Mass spectra were recorded on a Shimadzu QP-1000 spectrometer (20eV). High Resolution Mass Spectra (HRMS) were obtained on a JEOL JMS-DX-305 high resolution mass spectrometer. $^1\text{H-NMR}$ spectra were recorded either on a JEOL-60 or on a Varian Gemini-200 MHz spectrometer. Chemical shifts were expressed in ppm downfield from TMS used as an internal standard. Melting point was determined on an electrically heated Thomas-Hoover capillary melting point apparatus and uncorrected. All chromatographic separations were performed on Merck silica gel (Kieselgel 60, 230~400 mesh).

N-(4,5-Dichloro-2-nitrophenyl)malonic acid ethyl ester amide(9)

Ethyl malonyl chloride (7.26 ml, 51 mmol) was added to a well stirred solution of 4,5-dichloro-2-nitro aniline (10 g, 46.4 mmol) in anhydrous toluene (120 ml) at 15~20°C.

The reaction mixture was heated at 100~110°C for 4 h, cooled to room temperature and then washed with sat. NaHCO_3 solution (20 ml) and brine (30 ml). The organic layer was concentrated at reduced pressure to give product 9 (14.75 g, 99.1%). The crude product was practically pure and used for the next step without purification. An analytical sample was obtained by recrystallization from EtOAc/Hexane; mp 192-194°C; $^1\text{H-NMR}$ (200MHz, DMSO-d_6) δ 1.35(t, 3H), 3.64(s, 2H), 4.34(q, 2H), 8.35(s, 1H), 9.04(s, 1H), 11.40(bs, 1H); MS: m/z 320(M^+).

N-(2-Amino-4,5-dichlorophenyl)malonic acid ethyl ester amide(10)

A stirring mixture of compound 9 (14.6 g, 45.5 mmol) and Pd/C (1.5 g) in ethanol (250 ml) was hydrogenated under normal pressure at room temperature for 24 h. The resulting reaction mixture was filtered through celite and the organic solution was concentrated at reduced pressure to give the desired product 10 (12.5 g, 95.1%). The crude product was directly used for the next reaction and the analytical sample was obtained by recrystallization from ethanol; mp 186-188°C; $^1\text{H-NMR}$ (200 MHz, DMSO-d_6) δ 1.20(t, 3H), 3.45(s, 2H), 4.15(q, 2H), 5.40(bs, 2H), 6.89(s, 1H), 7.46(s, 1H), 9.40(bs, 1H); MS: m/z 290(M^+).

7,8-Dichloro-1H-1,5-benzodiazepine-2,4-(3H,5H)-dione (11)

Sodium ethoxide (86 mg, 1.2 mmol) was added to a stirred solution of compound 10 (291 mg, 1 mmol) in absolute ethanol (10 ml). The reaction mixture was heated at 50°C for 8 h, cooled to room temperature, and then concentrated at reduced pressure. The resulting slurry was diluted with water (20 ml), then acidified with 6N-HCl to give white precipitate as crude product. The crude product was purified by flash column chromatography with 30 % hexane in ethyl acetate as an eluent to get pure product 11 (107 mg, 43.7%); mp 292-293°C; $^1\text{H-NMR}$ (200MHz, DMSO-d_6) δ 3.35(s, 2H), 7.32(s, 2H), 10.51(bs, 2H); MS: m/z 245(M^++1); HRMS : calc. for $\text{C}_9\text{H}_6\text{N}_2\text{O}_2\text{Cl}_2$ 243.9806, found 243.9797.

7,8-Dichloro-6-nitro-1H-1,5-benzodiazepine-2,4-(3H,5H)-dione(7)

Compound 11 (122.5 mg, 0.5 mmol) was dissolved in $\text{c-H}_2\text{SO}_4$ (3 ml) at 0°C. To this solution was added KNO_3 (303 mg, 0.75 mmol) over 5 min and the resulting slurry was stirred at room temperature. After 30 min, the

reaction mixture was poured into ice-cold water (20 ml), and whole mixture was extracted with ethyl acetate (3 × 10 ml).

The combined extracts were washed with brine (20 ml), dried over Na₂SO₄, concentrated at reduced pressure to give crude product. The crude product was purified by flash column chromatography with 30% hexane in ethyl acetate as an eluent to get pure product **7** (100 mg, 39.2%); mp 270-271°C; ¹H-NMR (200MHz, DMSO-d₆) δ 3.35(s, 2H), 7.59(s, 1H), 10.80(s, 2H); MS: m/z 290 (M⁺+1); HRMS: calcd. for C₉H₅N₃O₄Cl₂ 288.9657, found 288.9661.

RESULTS AND DISCUSSION

Quinoxaline-2,3-diones were introduced as the first antagonists of the AMPA-subtype of Non-NMDA excitatory amino acid receptor (Honore, *et al.*, 1988) and were subsequently shown to have comparable affinities for the glycine site. Efforts to improve the glycine site selectivity in this series have focused on both aromatic substitution and on modification of the heterocyclic ring. Several reports have appeared showing very significantly improved affinity and 10³ fold selectivity (Woodward *et al.*, 1993) for the glycine site vs the AMPA receptor with the trisubstituted quinoxaline-2,3-dione ACEA 1021 (**4**). It is interesting to note that the benzazepine **6**, which also has b-dicarbonyl functionality, has been reported active in the gerbil model of global ischaemia (Swartz *et al.*, 1992).

With this background, it was envisioned that the 1,5-benzodiazepine **7**, which is ring expanded product of the ACEA 1021 and yet having similar structure to benzazepine **6**, might be potential lead compounds as NMDA glycine site antagonist.

At the beginning stage of the synthesis, it was thought that the target compound **7** could be easily prepared by condensing 4,5-dichloro-ortho-phenyldiamine, with either malonic acid or malonic acid ester. Several attempts, however, were proved to be unsuccessful. The synthesis had to be started with 2-nitro-4,5-dichloro aniline (**8**).

Treatment of compound **8** with ethyl malonyl chloride provided amide **9** in 99% yield. The nitro group of amide **9** was then reduced by hydrogen gas with Pd/C catalyst to give amino compound **10**. The amino compound **10** could be cyclized to compound **11** in both acidic and basic condition albeit in low yield (43.7%). The best condition for the cyclization was to utilize sodium ethoxide in ethanol solution. Final nitration to get target product **7** from compound **11** was accomplished by KNO₃/H₂SO₄, which was utilized by Cai *et al.* (Cai, *et al.*, 1997). The normal nitration conditions (HNO₃/H₂SO₄, HNO₃) were not successful. With this synthetic strategy to prepare various 1,5-benzodia-

zepine-2,4-dione derivatives, the biological activities of the new synthetic compounds for the NMDA receptor site will be anticipated, hoping to develop new lead molecules with improved pharmacological properties.

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