Ring Closure of N-(2-Hydroxyethyl)-N'-phenylthioureas: One-Pot Synthesis of 2-Phenylaminothiazolines

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The cyclization reaction of N-(2-hydroxyethyl)-N'-phenylthioureas 2 containing ambident nucleophile was examined in the combination of a variety of bases and p-toluenesulfonyl chloride (TsCl). N-(2-Hydroxyethyl)thioureas 2 were readily obtained in high yields from the reaction of the corresponding 1,2-aminoalcohols with phenyl isothiocyanate, avoiding the need for O-protection. The use of a one-pot reaction (NaOH/TsCl) was found to be most effective in producing the requisite 2-phenylaminothiazolines (S-cyclization) 3 in the case of thioureas 2a-2e derived from N-unsubstituted aminoalcohols, while in the thioureas 2f and 2g prepared from N-substituted aminoalcohols the combination of Et₃N and TsCl led to the S-cyclization products.

Introduction

The 2-aminothiazoline ring system has gained much interest as biologically active molecules such as potent inhibitors of human nitric oxide synthase.\(^1\) octopaminergic-agonists.\(^2\) anthelmintics.\(^3\) and anti-inflammatory agents.\(^4\) These compounds are usually prepared by the hydrochloric acid-catalyzed cyclization of \(N-(2-\)hydroxyethyl)\) thioureas\(^{2a,2b,3,5}\) or the cyclization of hydrogen sulfate of thioureas\(^{2a,2b,3,5}\) or the cyclization of hydrogen sulfate of thioureas\(^{2a,2b,3,5}\) or the formation of 2-aminothiazolines and are not applicable to acid sensitive or racemization-prone substrates due to the vigorous acidic reaction conditions. Alternatively, treatment of aromatic amines with 2-haloalkyl isothiocyanates gives 2-aminothiazolines.\(^7\) This method, however, has some limitations in the scope of aromatic amines.\(^{7b}\)

Recently, we reported that 2-methylaminothiazolines were synthesized selectively from N-(2-hydroxyethyl)-N'-methylthioureas by the intramolecular Mitsunobu reaction conditions (DEAD/TPP).8a To obtain the requisite 2-phenylaminothiazolines, we applied this Mitsunobu reaction conditions to the substrates such as N-(2-hydroxyethyl)-N-phenylthioureas 2. However, with thioureas 2a-2e, only small amount of 2-phenylaminothiazolines 3 were produced along with unknown mixtures of products. With thioureas 2f-2h. 2-imidazolidinethiones 4 were mainly obtained.9 In addition, in the course of our work in the evolization reaction of N-(2hydroxyethyl)-N'-phenylureas, we found that one-pot reaction of N-(2-hydroxyethyl)ureas proceeded in the presence of TsCl and some bases to give N-cyclized products in good yields. 86 On the basis of this reaction conditions, we preliminarily described a successful access to 2-phenylaminothiazolines 3 from the corresponding N-(2-hydroxyethyl)-N'phenylthioureas as a more convergent approach. Thioureas 2 proceeded through mild nucleophilic attack upon the tosylate intermediate in the presence of a base either by the sulfur atom to provide 3 or by the nitrogen to give the 2imidazolidinethiones 4 depending on the structure of thioureas (Scheme 1). In this article the synthetic method of 2phenylaminothiazolines 3 from the corresponding N-(2-

S-alkylation Shape
$$R^2$$

PhNH NR³
HO R²
 R^2

N-alkylation PhN NR³
 $R^3 = H$

Scheme I

hydroxyethyl) thioureas 2 is described in detail (eq. 1).

Results and Discussion

N-(2-Hydroxyethyl)thioureas **2** were readily obtained in high yields from the reaction of the corresponding 1.2-aminoalcohols with phenyl isothiocyanate, which provided exclusively the desired products under mild conditions, thus avoiding the need for O-protection (Table 2). Surveys of one-pot reactions by the combination of TsCl (1.1 equiv) with various basic metallic (*I*-BuOK, NaOH, and NaH) or non-metallic (Et₃N and Et₃N/DMAP) reagents were performed to **2a** in THF (Table 1). In the present reaction, the use of NaOH was found to be most effective in producing 2-phenylaminothiazoline **3a**. The NaOH was added to a mixture of the TsCl and **2a** at room temperature. The reactions were complete within 30 min at room temperature.

The one-pot reaction of various substrates 2a-2h was examined and the results are shown in Table 2. With thioureas 2a-2e prepared from N-unsubstituted aminoalcohols (R³ = H). S-cyclization to 3 was mainly observed with a trace amount of the N-cyclized products. As we expected all reactions proceeded in good yields with regiocontrol (S-cyclization > N-cylization) to give 2-phenylaminothiazolines. However, the thioureas 2f and 2g prepared from N-

Table 1. Effect of Base in One-pot Reaction of 2a

Entry	Base	Time, Temp.	Yield (%)" of 3a
l	t-BuOK (2.5 equiv)	30 min, 0 °C	78
2	NaOH (2.5 equiv)	30 mm, rt	94
3	Nall (2.5 equiv)	30 min, rt	62
4	Et ₃ N (5.0 equiv)	17 hr, reflux	62
5	Et ₃ N/DMAP	17 hr, reflux	33

[&]quot;Isolated yields by column chromatography.

Table 2. Preparations and Cyclizations of N-(2-Hydroxyethyl)-thioureas 2

Entry	R ¹	R ²	R ³	Yield (%) ^a of 2	Yield (%) ^b of 3
a	Me	Me	H	71	94
b	Me	H	H	98	77
c	Et	H	H	99	78
d	(S)-PhCH ₂	H	H	86	70
e	(S)- <i>i</i> -Pt	H	H	85 ^h	79
f	H	H	Me	93	29 (40)°
g	H	H	Et	91	27 (72)°
h	H	H	H	95	d

[&]quot;Recrystallized yields. ^bIsolated yields by column chromatography. The yield from the use of Et₃N is in parenthesis. ^aThe chlorinated thiourea was mainly obtained in 64% yield.

substituted aminoalcohols ($R^3 = Me$, Et) gave a mixture of 2-iminothiazolidines (S-alkylation products) and 2-imidazolidinethiones (N-alkylation products) in the ratio of 29/54 and 27/65, respectively. Thiourea 2h prepared from 2aminoethanol gave mainly a chlorinated thiourea in a 64% yield, containing a small amount of tosylate (6% yield). To improve the yields of S-evelized products in the case of 2f and 2g, various bases employed above were also applied to 2g in THF. Contrary to above result, the refluxed reaction in the presence of 5 equiv of Et₃N gave the most effectively Sevelized product with almost complete regioselectivity. With thiourea 2f using Et₃N also afforded S-cyclized product in a 40% improved yield. Thus, the use of Et₃N was the most effective to the thioureas 2f and 2g derived from N-substituted aminoalcohols. Although a further investigation such as the relationships of the pKa values of thioureas and regioselectivity is needed to understand this reaction, the Sevelization selectivity is remarkably affected by the base employed depending on the nucleophilicity of thioureas.

In conclusion, we developed a mild synthetic method for 2-phenylaminothiazolines from the corresponding 1,2-aminoalcohols and phenyl isocyanate using one-pot reaction with NaOH/TsCl or Et₃N/TsCl depending on the structure of 1,2aminoalcohols.

Experimental Section

General. ¹H NMR and ¹³C NMR spectra were recorded using 300 MHz and 75 MHz NMR spectrometer; chemical shifts are reported in ppm using TMS as an internal standard.

Melting points were determined on a capillary apparatus and uncorrected. Mass spectra were recorded on a HP 5983B GC/Mass spectrometer. Analytical TLC was performed on 0.25 mm precoated silica gel plates. Flash chromatography was carried out with 230-400 mesh silica gel.

General procedure for the preparation of thiourea 2. To a stirred solution of 1.2-aminoalcohol (4.59 mmol) in THF (10 mL) under nitrogen at room temperature was added a solution of phenyl isothiocyanate (0.50 mL, 4.18 mmol) in THF (5 mL) dropwise for 5 min with a syringe. The reaction mixture was stirred for 30 min and evaporated. The crude products except 2e were recrystallized to give the requisite product.

N-(1,1-Dimethyl-2-hydroxy)ethyl-*N*'-phenylthiourea (2a). mp 127-128 °C (toluene); 1R (CDCl₃, cm⁻¹) 3262, 1278; ¹H NMR (CDCl₃) δ 7,23-7.43 (5H, m), 6,16 (1H, bs), 3.79 (2H, s), 1.40 (6H, s).

N-(2-Hydroxy-1-methyl)ethyl-*N*'-phenylthiourea (2b). mp 83-84 °C (hexane/acetone); 1R (KBr. cm⁻¹) 3346, 1242; ¹H NMR (DMSO) δ 9.50 (1H, s), 7.51 (1H, d, J = 7.9 Hz), 7.04-7.46 (5H, m), 4.87 (1H, t, J = 5.1 Hz), 4.30 (1H, bs), 3.47 (1H, dd, J = 4.8, 10.6 Hz), 3.40 (1H, dd, J = 5.2, 10.6 Hz), 1.10 (3H, d, J = 6.7 Hz); ¹³C NMR (DMSO) δ 179.9, 139.7, 128.7, 124.1, 123.0, 64.0, 51.2, 17.0.

N-(1-Ethyl-2-hydroxy)ethyl-*N*'-phenylthiourea (2c). mp 145-146 °C (toluene): IR (CDCl₃, cm⁻¹) 3243, 1248: ¹H NMR (CDCl₃) δ 7.80 (1H, bs), 7.23-7.50 (5H, m), 6.20 (1H, d, J = 7.8 Hz), 4.50 (1H, bs), 3.65-3.91 (2H, m), 2.36 (1H, bs), 1.58 (2H, dq, J = 7.5, 14.0 Hz), 0.96 (3H, t, J = 7.5 Hz).

N-[(1S)-2-Hydroxy-1-phenylmethyl]ethyl-*N*'-phenylthiourea (2d). mp 103-104 °C (hexane/acetone): IR (CDCl₃, cm⁻¹) 3366, 1248; ¹H NMR (CDCl₃) δ 7.61 (1H, bs), 6.90-7.40 (10H, m), 6.20 (1H, d, J = 7.9 Hz), 4.85 (1H, bs), 3.80 (1H, dd, J = 4.0, 10.9 Hz), 3.63 (1H, dd, J = 4.8, 10.9 Hz), 2.86-2.99 (2H, m), 2.16 (1H, bs); ¹³C NMR (CDCl₃) δ 180.3, 137.1, 135.7, 130.2, 129.2, 128.7, 127.4, 126.8, 125.1, 63.7, 57.4, 36.6.

N-[(1S)-2-Hydroxy-1-(1-methylethyl)]ethyl-*N*'-phenylthiourea (2e). The crude product was purified by flash column chromatography. R_f = 0.5 (ethyl acetate/hexane 1 : 1); mp 93-95 °C; IR (CDCl₃, cm⁻¹) 3273, 1313; ¹H NMR (CDCl₃) δ 8.83 (1H, bs), 7.19-7.38 (5H, m), 6.45 (1H, d., J = 5.8 Hz), 4.38 (1H, bs), 3.74 (1H, dd, J = 3.8, 11.2 Hz), 3.62 (1H, dd, J = 5.8, 11.2 Hz), 3.39 (1H, bs), 2.98 (1H, bs), 1.80-1.93 (1H, m), 0.90 (3H, d, J = 6.8 Hz), 0.88 (3H, d, J = 6.8 Hz); ¹³C NMR (CDCl₃) δ 181.0, 136.2, 129.4, 127.5, 125.2, 63.7, 62.1, 29.4, 19.5, 18.9.

N-(2-Hydroxyethyl)-*N*-methyl-*N*'-phenylthiourea (2f). mp 134-135 °C (hexane); IR (CDCl₃, cm⁻¹) 3256, 1346; ¹H NMR (CDCl₃) δ 7,26-7,38 (5H, m), 3,79-3,87 (4H, m), 3,36(3H, s).

N-Ethyl-*N*-(2-hydroxyethyl)-*N*'-phenylthiourea (2g). mp 158-159 °C (chloroform/acctone): IR (KBr. cm⁻¹) 3258. 1352; ¹H NMR (DMSO) δ9.43 (1H, bs), 7.05-7.31 (5H, m), 5.53 (1H, bs), 3.79 (2H, q, J = 7.0 Hz), 3.71 (4H, bs), 1.17 (3H, t, J = 7.0 Hz); ¹³C NMR (DMSO) δ 181.2, 141.1, 128.2, 125.1, 124.3, 60.1, 52.5, 46.5, 12.3.

N-(2-Hydroxyethyl)-*N*'-phenylthiourea (2h). mp 138-139 °C (chloroform/acetone); IR (CDCl₃, cm⁻¹) 3362, 1250; ¹H NMR (CDCl₃) δ 7.72 (1H, bs), 7.19-7.48 (5H, m), 6.47 (1H, bs), 3.77-3.85 (4H, m), 1.99 (1H, bs).

General procedure for the preparation of 2-pheny-laminothiazolines 3. To a stirred solution thiourea 2 (0.88 mmol) in THF (10 mL) under nitrogen at room temperature was added a solution of NaOH (88 mg. 2.2 mmol) in water (3 mL) and TsCl (0.18 g. 0.97 mmol) in THF (5 mL) dropwise for 5 min with a syringe. The reaction mixture was stirred for 30 min, added with water (30 mL), and extracted with ether (50 mL × 3). The organic layer was dried, filtered, evaporated, and purified by flash column chromatography to give 3.

4,5-Dihydro-4,4-dimethyl-*N***-phenyl-2-thiazolamine (3a)**. mp 114-116 °C; R_f = 0.3 (ethyl acetate); IR (CDCl₃, cm⁻¹) 1687; ¹H NMR (CDCl₃) δ 6,93-7,25 (5H, m), 4,02 (2H, s), 1.33 (6H, s); ¹³C NMR (CDCl₃) δ 156.0, aromatics omitted, 78.7, 61.1, 28.0; HRMS (EI) calcd for $C_{11}H_{14}N_2S$ 206,0878, found 206,0864.

4,5-Dihydro-4-methyl-*N***-phenyl-2-thiazolamine (3b)**. mp 104-105 °C; R_f = 0.3 (ethyl acetate); IR (CDCl₃, cm⁻¹) 1687; ¹H NMR (CDCl₃) δ 6.95-7.28 (5H, m), 4.43 (1H, t, J = 8.0 Hz), 4.10-4.17 (1H, m), 3.87 (1H, t, J = 7.5 Hz), 1.27 (3H, d, J = 6.3 Hz); ¹³C NMR (CDCl₃) δ 156.8, aromatics omitted, 73.6, 55.3, 21.2; HRMS (EI) calcd for C₁₀H₁₂N₂S 192.0721, found 192.0733.

4,5-Dihydro-4-ethyl-*N***-phenyl-2-thiazolamine (3c).** mp 97-98 °C: R_f = 0.2 (ethyl acetate); IR (CDCl₃, cm⁻¹) 1703; ¹H NMR (CDCl₃) δ 6.94-7.27 (5H, m), 4.37-4.43 (1H, m), 3.88-3.99 (2H, m), 1.49-1.67 (2H, m), 0.93 (3H, t, J = 7.4 Hz); ¹³C NMR (CDCl₃) δ 156.8, aromatics omitted, 71.9, 60.8, 28.5, 9.7; HRMS (EI) calcd for $C_{11}H_{14}N_2S$ 206.0878, found 206.0876.

(4S)-4,5-Dihydro-*N*-phenyl-4-phenylmethyl-2-thiazolamine (3d). R_f = 0.3 (ethyl acetate/hexane 1 : 1); IR (CDCl₃, cm⁻¹) 1687; ¹H NMR (CDCl₃) δ 6.95-7.33 (10H, m), 4.25-4.30 (2H, m), 4.13-4.04 (1H, m), 3.00 (1H, dd, J = 5.2, 13.5 Hz), 2.73 (1H, dd, J = 7.4, 13.5 Hz); ¹³C NMR (CDCl₃) δ 157.8, 138.5, 130.1, 129.7, 129.5, 127.5, 123.2, 120.9, 72.4, 63.0, 42.9; HRMS (EI) calcd for $C_{16}H_{16}N_2S$ 268.1034, found 268,1018.

(4S)-4,5-Dihydro-4-(1-methylethyl)-*N*-**phenyl-2-thiazolamine (3e)**. mp 65-67 °C; $R_f = 0.3$ (ethyl acetate/hexane 1:1); lR (CDCl₃, cm⁻¹) 1680; ¹H NMR (CDCl₃) δ 6.89-7.24 (5H, m), 4.28 (1H, dd, J = 8.2, 8.2 Hz), 3.99 (1H, dd, J = 6.6, 8.2 Hz), 3.64-3,71 (1H, m), 1.60-1,71 (1H, m), 0.88 (3H, d, J = 6.7 Hz), 0.81 (3H, d, J = 6.8 Hz); ¹³C NMR (CDCl₃) δ 156.7, 143.3, 128.8, 122.2, 120.8, 70.3, 65.7, 33.0, 18.6, 17.9; HRMS (EI) calcd for $C_{12}H_{16}N_2S$ 220.1034, found 220,1035,

3-Methyl-2-phenyliminothiazolidine (3f). mp 88-89 °C: R_f = 0.5 (ethyl acetate/hexane 3 : 7); IR (CDCl₃, cm⁻¹) 1616: ¹H NMR (CDCl₃) δ 6.92-7.29 (5H, m), 3.56 (2H, t, J = 6.8 Hz), 3.13 (2H, t, J = 6.8 Hz), 3.03 (3H, s): ¹³C NMR (CDCl₃) δ 159.4, aromatics omitted, 53.0, 33.8, 26.8; EIMS 192 (M, 32), 167 (46), 149 (100), 71 (25), 57 (55); HRMS

(EI) calcd for C₁₀H₁₂N₂S 192.0721, found 192,0768,

1-Methyl-3-phenyl-2-imidazolidinethione (4f). mp 88-89 °C; R_f = 0.4 (ethyl acetate/hexane 3 ; 7); IR (CDCl₃, cm⁻¹) 1336; ¹H NMR (CDCl₃) δ 7.21-7.54 (5H. m), 4.01 (2H. m), 3.73 (2H. m), 3.24 (3H. s); ¹³C NMR (CDCl₃) δ 182.0. aromatics omitted, 48.7, 48.5, 35.2; EIMS 192 (M. 11), 91 (38), 77 (100), 57 (60), 51 (54); HRMS (EI) calcd for C₁₀H₁₂N₂S 192.0721, found 192,0749.

3-Ethyl-2-phenyliminothiazolidine (3g). R_f = 0.5 (ethyl acetate/hexane 3:7); IR (CDCl₃, cm⁻¹) 1624; ¹H NMR (CDCl₃) δ 6.91-7.28 (5H, m), 3.58 (2H, t, J = 6.9 Hz), 3.57 (2H, q, J = 7.2 Hz), 3.10 (2H, t, J = 6.9 Hz), 1.22 (3H, J = 7.2 Hz); ¹³C NMR (CDCl₃) δ 158.4, aromatics omitted, 50.2, 41.1, 26.7, 11.9; EIMS 206 (M, 1), 77 (100), 56 (57), 51 (45); HRMS (EI) calcd for $C_{11}H_{14}N_2S$ 206.0878, found 206.0840.

1-Ethyl-3-phenyl-2-imidazolidinethione (4g). mp 65-67 °C; R_f = 0.4 (ethyl acetate/hexane 3 : 7); IR (CDCl₃, cm⁻¹) 1346; ¹H NMR (CDCl₃) δ 7.19-7.55 (5H, m), 3.94-4.00 (2H, m), 3.78 (2H, q, J = 7.2 Hz), 3.65-3.72 (2H, m), 1.23 (3H, t, J = 7.2 Hz); ¹³C NMR (CDCl₃) δ 181.1, aromatics omitted. 48.8, 45.5, 42.4, 11.8; EIMS 206 (M, 91), 106 (97), 77 (100), 51 (45); HRMS (EI) calcd for C₁₁H₁₄N₂S 206.0878, found 206.0864,

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- 9. Thioureas 2f, 2g, and 2h furnished a mixture of 2-iminothiazolidines and 2-imidazolidinethiones in the ratio of

20/80, 12/88, and 31/69, respectively. The separation and purification of products was not convenient due to the byproducts, triphenylphosphine oxide and 1,2-dicarbethoxyhydrazine.