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Communications

Palladium-catalyzed Acylation of β , β -Diphenyl- α -(trifluoromethyl)vinylstannane as a Novel Route to 1,3-Disubstituted 2-(Trifluoromethyl)indene Derivatives

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Although a various types of α -(trifluoromethyl)vinylmetal reagents such as lithium^{1,2} and zinc^{3,6} have been synthesized and utilized previously, the preparation and synthetic utility of α -(trifluoromethyl)vinylstannane reagent have been quite limited. Only a couple of papers described about chemistry of α -(trifluoromethyl)vinylstannane reagent. The α -(trifluoromethyl)vinylstannane reagent bearing only hydrogens at β position has been synthesized from the reaction of 2-bromotrifluoroisopropene with lithium tributylstannate in the presence of CuI and utilized for the cross-coupling reactions with acyl chlorides in the presence of catalytic amount of PdCl(Bn)(PPh₃)₂ in HMPA at 65 °C to give α-(trifluoromethyl)vinyl ketone derivatives. ⁷ Ichikawa also carried out the reaction of α -(trifluoromethyl)vinylstannane reagent with α,β -unsaturated acyl chlorides in the presence of catalytic amount of Pd(PPh₃)₂Cl₂ and CuCN in toluene at 55-75 °C to give the desired NaZarov substrates.8 Recently, we reported about the preparation of a novel α -(trifluoromethyl)vinvlstannane reagent bearing two phenyl groups at β -position and the cross coupling reactions of it with aryl iodides to give trifluoromethylated triphenylethene derivatives which are important framework of many nonsteroidal antiestrogens. 10 As a part of our continuing studies on the synthetic utility of β . β -diphenyl- α -(trifluoromethyl)vinylstannane reagent, we examined palladium-promoted acylation of this reagent with acyl chlorides to give $\beta.\beta$ -diphenyl- α -trifluoromethylated enone derivatives which are useful intermediates for the formation of novel 1.3-disubstituted 2-(trifluoromethyl)indene derivatives via Friedel-Craft's type of the cyclization. Since nonfluorinated 1.3-disubstituted indene derivative such as Indenestrol A exhibited mammary tumor inhibiting antiestrogen

activity. ^{11,12} it is expected that 1,3-disubstituted 2-(trifluoromethyl)indene derivatives also have a potential similar activity. Herein, we describe the palladium-promoted acylation of $\beta.\beta$ -diphenyl- α -(trifluoromethyl)vinylstannane reagent and the formation of 1.3-disubstituted 2-(trifluoromethyl)indene derivatives from the acylated adduct .

A starting material, β , β -diphenyl- α -(trifluoromethyl)vinylstannane reagent 1, was prepared via several steps from 2.3,3.3-tetrafluoro-1-phenyl-1-phenylthiopropene. First of all, the acylation reaction of 1 with acetyl chloride was carried out in the presence of several palladium catalyst. The use of Pd catalyst such as Pd(PPh₃)₄ or Pd(PPh₃)₂Cl₂ in THF, DMF, toluene or HMPA did not provide any acylated product. However, acylation reaction to give acylated product 2a was successfully accomplished by using a mixture of 10 mol% Pd(PPh₃)₂Cl₂ and 10 mol% CuCN in toluene at 50 °C for 6 h. The higher temperature (80 °C) was needed for the completetion of acylation of 1 with types of benzoyl chlorides. Therefore, starting material 1 underwent the acylation reaction with a various types of acyl chlorides, such as ethyl chloroformate, furoyl chloride, naphthoyl chloride and benzoyl chlorides bearing a bromo, methoxy, methyl, or nitro on the benzene ring, to give the corresponding trifluoromethylated enone derivatives 2 at 80 °C for 6 h. The experimental results of the acylation reactions are summarized in Table 1.

Reduction of 2a with LiAlH₄ (1.5 equiv.) in ether at reflux temperature for 3 h afforded the corresponding allylic alcohols 3a¹³ in 71% yield. The use of NaBH₄ did not provide the desired product, whereas the starting material was always recovered. The Fridel-Craft's type of cyclization of 3a was successfully accomplished to give 2-trifluoromethyl-3-methyl-

Table 1. The acylation reactions of 1 with acyl chlorides

$$\begin{array}{c} CF_3 \\ Bu_3Sn \end{array} + \begin{array}{c} O \\ R \end{array} \begin{array}{c} Pd(PPh_b)_2Cl_2(10 \text{ mol}\%)/CuCN(10 \text{ mol}\%) \\ \text{toluene, T } ^{\circ}C, 6 \text{ h} \end{array} \begin{array}{c} CF_3 \\ R \end{array} \begin{array}{c} \\ O \end{array}$$

1			2
Compound No.	T (°C)	R	Yield (%) ^a
2a	50	CH ₃	68
2b	80	C_6H_5	80
2c	80	$(o-CH_3)-C_6H_4$	78
2d	80	$(p\text{-}CH_3)\text{-}C_6H_4$	88
2e	80	$(p\text{-OCH}_3)\text{-C}_6\text{H}_4$	85
2f	80	$(m-NO_2)-C_6H_4$	72
2g	80	$(m-{ m CH_3})-{ m C_6H_4}$	81
2h	80	$(m-Br)-C_6H_4$	77
2i	80	C_2H_5O	62
2j	80	2-furanyl	75
2k	80	2-naphthyl	86

[&]quot;Isolated yields.

1-phenylindene (4a)¹⁴ by using AlCl₃ (1.2 equiv.) in methylene chloride at -78 °C, followed by the slowly warming to room temperature. The use of dilute H₂SO₄ instead of AlCl₃ at reflux temperature caused not only to decrease the yield of indene derivatives 4a, but also to extend the reaction time. The more excess of AlCl₃ (2.0 equiv.) was used to carry out the acylation of 3d and 3f because of possible coordination of oxygen with AlCl₃. Reduction of other types of enone derivatives 2, followed by treatment with AlCl3 under the same reaction condition also provided the corresponding 1,3disubstituted 2-(trifluoromethyl)indene derivatives 4 in good yields. The experimental results of reduction and cyclization reactions are summarized in Table 2. Although 2-(trifluoromethyl)indene has been prepared in a previous literature, this method is a lack of generality and provides low yield preparation.15

A typical reaction procedure for the preparation of **2a** is as follows. To a toluene (5 mL) solution of acetyl chloride (0.088 g. 1.12 mmol) and $\beta\beta$ -diphenyl- α -(trifluoromethyl)vinyl-

Table 2. The synthesis of 1,3-disubstituted 2-trifluoromethylated indene derivatives 4

Compound No.	R	Yield of 3 (%) ^a	Yield of 4 (%)*
3a, 4a	CH ₃	71	76
3b, 4b	C_6H_5	7 4	78
3c, 4c	(p-CH ₃)-C ₆ H ₄	69	7 1
3d, 4d	(p-OCH ₃)-C ₆ H ₄	73	68
3e, 4e	$(m\text{-Br})\text{-C}_6\mathrm{H}_4$	70	73
3f, 4f	$(m-NO_2)-C_6H_4$	67	66
3g, 4g	2-furany1	62	63
3h, 4h	2-naphthyl	71	74

[&]quot;Isolated yields.

stannane (0.402 g, 0.75 mmol) was added Pd(PPh₃)₂Cl₂ (10 mol%) and CuCN (10 mol%), and the reaction mixture was heated at 50 °C for 6 h under argon atmosphere. After the reaction mixture was quenched with water and then washed with 5% KF solution and brine, solution was extracted with ether twice. The ether solution was dried and chromatographed on SiO2 column. Elution with a mixture of hexane and ethyl acetate (20:1) provided 0.145 g of 2-trifluoromethyl-1,1-diphenyl-1-buten-3-one (2a) in 68% yield. 2a: mp 65-66 °C: ¹H NMR (100 MHz, CDCl₃) δ 7.41-7.18 (m, 10H). 2.03 (s. 3H): ¹⁹F NMR (100 MHz, CDCl₃) δ -55.42 (s. 3F); MS, m/z (relative intensity) 290 (M⁻, 81), 289 (100), 275 (12), 255 (22), 227 (29), 213 (30), 207 (17), 178 (29), 176 (19), 151 (13), 127 (19), 105 (33), 77 (13), 51 (10); IR (KBr) 3082, 3059, 3028, 2927, 2855. 1704, 1617, 1608, 1492, 1445, 1418, 1357, 1322, 1261, 1216. 1145, 1111, 1078, 1042, 998, 951, 759, 699, 656 cm⁻¹. Anal. Calcd for C₁₇F₃H₁₃O; C, 70.32; H. 4.52. Found: C, 70.49; H. 4.60.

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- 13. Spectroscopic data of **3a**: oil; ¹H NMR (100 MHz, CDCl₃) δ 7.38-7.11 (m, 10H), 4.58 (m, 1H), 1.87 (d, J = 8.0 Hz, 1H), 1.52 (d, J = 7.4 Hz, 3H); ¹⁹F NMR (100 MHz, CDCl₃) δ -52.64 (s, 3F); MS, m/z (relative intensity) 292 (M⁺, 44), 277 (15), 274 (19), 259 (15), 237 (13), 223 (16), 209 (26), 178 (18), 171 (21), 167 (100), 165 (33), 151 (29), 127 (59), 77 (22), 51 (17); IR (neat) 3395, 3059, 3028, 2963, 2932, 1623, 1492, 1445, 1318, 1260, 1130, 1077, 1018, 797, 759, 700 cm⁻¹. Anal. Calcd for C₁₇F₃H₁₅O: C, 69.84; H, 5.18. Found: C, 70.01; H, 5.26.
- Spectroscopic data of 4a: mp 75-77 °C; ¹H NMR (100 MHz, CDCl₃) δ 7.41-7.10 (m, 10H), 4.89 (q, J = 7.5 Hz, 1H), 1.76 (d, J = 7.5 Hz, 3H); ¹9F NMR (100 MHz, CDCl₃) δ -52.50 (s, 3F); MS, m/z (relative intensity) 274 (M⁻, 39), 233 (41), 205 (100), 203 (36), 196 (52), 190 (15), 177 (24), 127 (14), 101 (20), 91 (18), 77 (19), 51 (15); IR (KBr) 3058, 2929, 2855, 2360, 1598, 1492, 1445, 1320, 1262, 1194, 1137, 1034, 763, 732, 700 cm⁻¹. Anal. Calcd for C₁¬F₃H₁s: C, 74.43; H, 4.78. Found: C, 74.27; H, 4.89.
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