Improving the Regioselectivity in the Cyclic Iminocarbonate Rearrangement: Enantioselective Synthesis of the Paclitaxel Side Chain

Gae Y. Cho, Kyung Mi An, and Soo Y. Ko°

Department of Chemistry and Division of Molecular Life Sciences, Ewha Womans University, Seoul 120-750, Korea Received January 18, 2001

Keywords : Cyclic iminocarbonate rearrangement. Enantioselective synthesis. Paclitaxel side chain.

 β -Amino- α -hydroxy acids represent an important class of compounds with wide biological activities.¹ In particular. *N*-benzoyl-*syn*-phenylisoserine, the C-13 side chain of Paclitaxel has been the target of numerous synthetic efforts as the interest in and the paucity of the anti-cancer agent has prompted a wide search for alternative sources for the parent drug and its fragments.^{2,9} A key in the side chain synthesis is the stereocontrol of the C-2/C-3 vicinal carbons, and various strategies have been devised to address the relative (*syn*) as well as absolute (2*R*,3*S*) stereocontrol. These include asymmetric 1.2-induction,³ use of chiral auxiliary.⁴ enzymatic resolution.⁵ and those employing powerful asymmetric oxidation processes such as catalytic asymmetric epoxidation.⁶ dihydroxylation⁷ and aminohydroxylation⁸ (followed by stereospecific functional group transformations as necessary).

Employing Sharpless's asymmetric dihydroxylation (AD) as the key step, one faces two issues in the synthesis of the Paclitaxel side chain: the regiocontrol and the stereocontrol, particularly in a *relative* sense (*i.e.*, diastereocontrol). Thus, a (2R.3S)-cinnamate diol, available *via* the AD, needs to be converted to N-benzovl-(2R.3S)-phenvlisoserine with a substitution of N-benzoyl group regioselectively at the C-3 and with net-retention of configuration. A reaction sequence involving an N-benzovl substitution with inversion of configuration at C-3. a perhaps more straightforward synthetic plan that would require (2R,3R)-cinnamate diol starting material, is not applicable as anti(ervthro)-diols are not directly accessible via AD process.^{10,11} While the reported syntheses solved the issue of regiocontrol by taking advantage of the fact that the C-3 was activated as benzylic, this could cause the stereocontrol to be less than complete during the necessary double inversion (net-retention) N-substitution steps due to the participation of S_N 1-type reaction pathways. Thus, one AD-based synthesis endured an incomplete (6:1) regioselection.^{7a} while another suffered from a modest (4:1)diastereoselection. 76

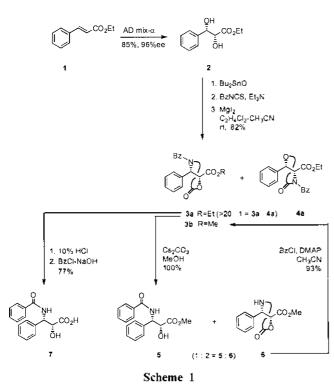
Our solution to this double-faced problem is to use the cyclic inninocarbonate rearrangement, which we reported recently (Scheme 1).¹² The rearrangement converts *syn*-diols (AD products) to protected *syn*-amino alcohols in a one-pot operation: the configurations at both carbinol carbons are retained. Thus, ethyl *trans*-cinnamate (1) was dihyroxylated using AD-mix- α to give the (2*R*,3*S*)-cinnamate diol (2, 85% yield, 96%ee¹³), which was subjected to the cyclic iminocarbonate rearrangement protocol. While the original procedure

(using tetrabutylammonium bromide as the nucleophile) had been initially reported to produce preferentially the C-2 nitrogen isomer (4a. the "wrong" regioisomer for the synthesis of the side-chain. 1:2.7 regioselectivity),128 subsequent works led to a discovery that using LiI nucleophile in acetonitrile/C₂H₄Cl₂ reversed the regioselection of the rearrangement toward the desired C-3(benzylic)-nitrogen isomer (3a, 3: 1 regioselectivity).¹²⁶ Further optimizations of the reaction conditions have now resulted in a great improvement in regioselectivity (>20:1 for the desired isomer 3a, 82%combined yield) when the rearrangement was performed with MgI₂ as the nucleophile in acetonitrile/ $C_2H_4Cl_2$ at room temp. No diastereomeric products were observed and the enantiomeric purity of the product was confirmed to be as high as that of the starting diol 2.13 The stereochemical integrity, therefore, was fully maintained during the rearrangement. The high diastereoselectivity is presumably due to the fact that the rearrangement $(2 \rightarrow 3a)$ involves an intramolecular (cyclic) nucleophilic substitution step and the svn-diastereomer, as obtained, is also thermodynamically more stable. Therefore, an S_{N} l-type reaction pathway, if present at all, would also yield the net retention diastereomer preferentially.14 For an additional advantage, the desired regioisomer 3a conveniently crystallized out from the crude product (hexane-ethyl acetate).

The rearranged product. ethyl 1-benzoyl-5-(S)-phenyloxazolidin-2-one-4-(R)-carboxylate (**3a**), is only a short distance away from the Paclitaxel side chain. It has been reported that various deprotection methods under basic or acidic solvolytic conditions usually remove the *N*-benzoyl group first.^{12a} Nevertheless, we sought to cleave selectively the carbonyl group of the oxazolidinone ring, while leaving the *N*-benzoyl group intact. as this would lead to a very atom-economic route to the Paclitaxel side chain. Although our endeavor was met with a partial success, we have devised an efficient synthetic process for our target compound, nonetheless.

When the *N*-benzoyl-oxazolidinone **3a** was treated with Cs_2CO_3 in anhydrous methanol, a mixture of products was obtained. consisting of the oxazolidinone ring-opened product **5** and the *N*-debenzoylated compound **6** in a 1 : 2 ratio (100% combined yield). *N*-Benzoyl-(2*R*.3*S*)-phenylisoserine methyl ester (**5**), the methyl ester of the Paclitaxel side chain, crystallized out from hexane-ethyl acetate (28% yield from **3a**, 97%ee). The *N*-debenzoylated compound, 5-(*S*)-phenyl-

Notes



oxazolidin-2-one-4-(R)-carboxylate methyl ester (6). recovered from the mother liquor, could be re-N-benzoylated to give compound **3b** (BzCl, DMAP. 93% yield), which may be recycled to furnish additional amounts of the product **5**.

Alternatively, the oxazolidinone **3a** was subjected to exhaustive hydrolytic conditions (10% HCl. reflux for 7 days), and without isolation of the resulting phenylisoserine, the crude hydrolyzed product was *N*-benzoylated under Schotten-Baumann conditions to yield *N*-benzoyl-(2R.3S)-phenylisoserine (7) in 77% (from **3a**). The stereochemical integrity was sustained through the harsh and prolonged reaction conditions.¹⁵

In conclusion, we have achieved a very efficient synthesis of the Paclitaxel side chain, N-benzoyl-(2R.3S)-phenylisoserine, in a three-step sequence employing as key steps the asymmetric dihydroxylation and the regio- and diastereoselective cyclic iminocarbonate rearrangement processes.

Experimental procedure for the MgI₂-catalyzed cyclic iminocarbonate rearrangement. To a solution of (2R,3S)cinnamate diol ethyl ester (213 mg, 1.0 mmol) in dichloroethane (15 mL), dibutyltin oxide (305 mg, 1.2 mmol) was added. The mixture was heated to reflux for 4 hr with a concomitant removal of water using a Dean-Stark trap. Benzovl isothiocvanate (0.187 mL, 1.4 mmol) and triethylamine (0.142 mL, 1.0 mmol) were added and the mixture was heated to reflux for further 2 hr. After it was cooled to rt, the mixture was diluted with MeCN (30 mL) and MgI₂ (304 mg, 1.0 mmol) was added. It was then stirred at rt overnight. Aqueous extractive work-up (EtOAc - aq. NaHCO₃, dil. HCl. then brine) yielded the crude product (277 mg. 82%, ≥ 20 : 1 regioisomeric mixture 3a : 4a). Crystallization from hexane-EtOAc produced pure ethyl 1-benzoyl-5-(S)-phenyloxazolidin-2-one-4-(R)-carboxvlate (3a, 75%).¹⁶

Acknowledgment. This work was supported by the KRF (2000-015-DP0271). G.Y.C. is grateful to the Brain Korea 21 project for a graduate fellowship.

References and Notes

- (a) Cole, D. C. Tetrahedron 1994, 50, 9517. (b) Cardillo, G.; Tomasini, C. Chem. Soc. Rev. 1996, 25, 117. (c) Enantioselective Synthesis of β-Amino Acids; Juaristi, E. Ed.; VCH Publishers: New York, 1996.
- Reviews: (a) Boa, A. N.; Jenkins, P. R.; Lawrence, N. J. Contemporary Org. Synth. 1995, 47. (b) Nicolaou, K. C.; Dai, W.-M.; Guy, R. K. Angew. Chem. Int. Ed. Engl. 1994, 33, 15. (c) Boge, T. C.; Georg, G. I. in reference 1(c), Chapter 1.
- (a) Kang, S. H.; Kim, C. M.; Youn, J.-H. Tetrahedron Lett. 1999, 40, 3581. (b) Tomasini, C.; Vecchione, A. Org. Lett. 1999, 1, 2153. (c) Cardillo, G.; Gentilucci, L.; Tolomelli, A.: Tomasini, C. J. Org. Chem. 1998, 63, 2351. (d) Nocioni, A. M.; Papa, C.; Tomasini, C. Tetrahedron Lett. 1999, 40, 8453. (e) Dondoni, A.; Perrone, D.: Semola, T. Synthesis 1995, 181. (f) Lee, K.-Y.; Kim, Y.-H.; Park, M.-S.; Ham, W.-H. Tetrahedron Lett. 1998, 39, 8129. (g) Denis, J.-N.; Correa, A.; Greene, A. E. J. Org. Chem. 1991, 56, 6939. (h) Hanessian, S.; Sancéau, J. Y. Can. J. Chem. 1996, 74, 621. (i) Palomo, C.; Cossio, F. P.; Aizpurua, J. M.; Mielgo, A.; Aurrekoetxea, N. Tetrahedron Lett. 1990, 31, 6429. (j) Davis, F. A.; Reddy, T.; Reddy, R. E. J. Org. Chem. 1992, 57, 6387.
- 4. (a) Georg, G. I.; Mashava, P. M.; Akgün, E.; Mistead, M. W. Tetrahedron Lett. 1991, 32, 3151. (b) Kanazawa, A. M.; Denis, J.-N.; Greene, A. E. J. Org. Chem. 1994, 59, 1238. (c) Kanazawa, A. M.; Denis, J.-N.; Greene, A. E. J. Chem. Soc., Chem. Commun. 1994, 2591. (d) Swindell, C. S.; Tao, M. J. Org. Chem. 1993, 58, 5889. (e) Hattori, K.; Yamamoto, H. Tetrahedron 1994, 50, 2785. (f) Hattori, K.; Miyata, M.; Yamamoto, H. J. Am. Chem. Soc. 1993, 115, 1151. (g) Bunnage, M. E.; Davies, S. G.; Goodwin, C. J. J. Chem. Soc., Perkin Trans. 1 1994, 2385. (h) Bunnage, M. E.; Davies, S. G.; Goodwin, C. J. J. Chem. Soc., Perkin Trans. 1 1993, 1375. (i) Commerçon, A.; Bézard, D.; Bernard, F.; Bourzat, J. D. Tetrahedron Lett. 1992, 33, 5185. (j) Farina, V.; Hauck, S. I.: Walker, D. G. Synlett 1992, 761. (k) Ojima, I.: Habus, I.: Zucco, M.: Park, Y. H.: Sun, C. M.: Brigaud, T. Tetrahedron 1992, 48, 6985. (1) Ojima, I.: Habus, I.: Zhao, M. J. Org. Chem. 1991, 56, 1681. (m) Mukai, C.; Kim, I. J.; Furu, E.; Hanaoka, M. Tetrahedron 1993, 49, 8323. (n) Jost, S.; Gimbert, Y.; Greene, A. E.; Fotiadu, F. J. Org. Chem. 1997, 62, 6672. (o) Ha, H.-J.; Park, G.-S.; Ahn, Y.-G.; Lee, G.-S. Bioorg. Med. Chem. Lett. 1998, 8, 1619.
- (a) Kaywer, M. M.; Mihoviolvic, M. D.; Kearns, J.; Feicht, A.; Stewart, J. D. J. Org. Chem. 1999, 64, 6603.
 (b) Kearns, J.; Kaywer, M. M. Tetrahedron Lett. 1994, 35, 2845. (c) Patel, R. N.; banerjee, A.; Howell, J. M.; McNamee, C. G.; Brozozowski, D.; Mirfakhrae, D.; Nanduri, V.; Thottathil, J. K.; Szarka, L. J. Tetrahedron Asymmetry 1993, 4, 2069. (d) Lee, D.; Kim, M.-J. Tetrahedron Lett. 1998, 39, 2163. (e) Brieva, R.; Crich, J. Z.; Sih, C. J. J. Org. Chem. 1993, 58, 1068. (f) Wuts, P. G. M.; Gu, R. L.; Northuis, N. M. Tetrahedron Asymmetry 2000, 11, 2117. (g) Gou, D.-

434 Bull. Korean Chem. Soc. 2001, Vol. 22, No. 4

M.; Liu, Y.-C.; Chen, C.-S. J. Org. Chem. **1993**, 58, 1287. (h) Barco, A.; Benetti, S.; Risi, C. D.; Pollini, G. P.; Romagnoli, R.; Zanirato, V. Tetrahedron Lett. **1994**, 35, 9289.

- (a) Denis, J. N.; Greene, A. E.; Serra, A. A.; Luche, M. J. J. Org. Chem. 1986, 51, 46. (b) Deng, L.; Jacobsen, E. N. J. Org. Chem. 1992, 57, 4320. (c) Jacobsen, E. N.; Furukawa, Y.; Martinez, L. E. Tetrahedron 1994, 50, 4323. (d) Righi, G.; Rumboldt, G. J. Org. Chem. 1996, 61, 3557. (e) Bonini, C.; Righi, G. J. Chem. Soc., Chem. Commun. 1994, 2767.
- (a) Wang, Z. M.; Kolb, H. C.; Sharpless, K. B. J. Org. Chem. 1994, 59, 5104. (b) Koskinen, A. M. P.; Karvinen, E. K.; Siirilä, J. P. J. Chem. Soc., Chem. Commun. 1994, 21. (c) Denis, J. N.; Correa, A.; Greene A. E. J. Org. Chem. 1990, 55, 1957. (d) Song, C. E.; Lee, S. W.; Roh, E. J.; Lee, S. Tetrahedron Asymmetry 1998, 9, 983.
- (a) Li, G.; Sharpless, K. B. Acta Chemica Scandinavica 1996, 50, 649. (b) Bruncko, M.; Schlingloff, G.; Sharpless, K. B. Angew. Chem. Int. Ed. Engl. 1997, 36, 1483. (c) Song, C. E.; Oh, C. R.; Roh, E. J.; Lee, S.; Choi, J. H. Tetrahedron Asymmetry 1999, 10, 671.
- Kobayashi, S.; Ishitani, H.; Ueno, M. J. Am. Chem. Soc. 1998, 120, 431.

- (a) Ko, S. Y.; Malik, M. *Tetrahedron Lett.* **1993**, *34*, 4675.
 (b) Ko, S. Y.; Malik, M.; Dickinson, A. F. J. Org. Chem. **1994**, *59*, 2570.
 (c) Ko, S. Y.; Lerpiniere, J. *Tetrahedron Lett.* **1995**, *36*, 2101.
- cf. Wang, L.; Sharpless, K. B. J. Am. Chem. Soc. 1992, 114, 7568.
- (a) Cho, G. Y.; Ko, S. Y. J. Org. Chem. 1999, 64, 8745. (b) Cho, G. Y.; Park, J. N.; Ko, S. Y. Tetrahedron Lett. 2000, 41, 1789. (c) Park, J. N.; Ko, S. Y.; Koh, H. Y. Tetrahedron Lett. 2000, 41, 5553.
- The enantiomeric purity was determined by chiral HPLC (Pirkle column; hexane-PrOH gradient eluent).
- 14. When an *anti(erythro)*-diol was subjected to the MgI₂catalyzed rearrangement protocol, the corresponding *syn(trans)*-diastereomeric oxazolidinone was obtained, via an apparent *inversion* of configuration at one carbinol carbon. This observation implies a thermodynamic control of stereoselectivity (as opposed to stereospecificity).
- [α]_D -39.4 (c 0.83, EtOH): Lit.^{7a} [α]_D 35.5 (c 1.07, EtOH). Anal. Caled for C₁₆H₁₄NO₄: C, 67.36; H, 5.30; N, 4.91. Found: C, 67.01; H, 5.36; N, 4.91.
- 16. A full characterization of this compound has been reported elsewhere (see reference 12a).

Notes