

Crystal Structure of Byakangelicin ($C_{17}H_{18}O_7$)

Yang Bae Kim, Yong Ho Oh, Il Yeung Park¹, and Kuk Hyun Shin²

College of Pharmacy, Seoul National University, Seoul 151-742, Korea, ¹College of Pharmacy, Chungbuk National University, Cheongju 361-763, Korea, and ²Natural Products Research Institute, Seoul National University, Seoul 110-460, Korea

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The crystal structure of byakangelicin, one of furanocoumarin aldose reductase inhibitors, was determined by X-ray diffraction method. The crystal is triclinic, with $a = 8.114(1)$, $b = 10.194(1)$, $c = 11.428(1)\text{\AA}$, $\alpha = 111.50(1)$, $\beta = 95.57(1)$, $\gamma = 112.52(1)^\circ$, $D_x = 1.41$, $D_m = 1.39 \text{ g/cm}^3$, space group P1 and $Z = 2$. The intensity data were collected by ω -2 θ scan method with CuK α radiations. The structure was solved by direct method and refined by full matrix least-squares procedure to the final R -value of 0.056. There are two molecules with different conformations in an asymmetric unit. The molecules are kept by two intermolecular O-HO type hydrogen bonds and van der Waal's forces in the crystal. The absolute configuration of the molecules was estimated to S-form by the 'Eta refinement' procedure.

Key words: Byakangelicin, Aldose reductase inhibitor (ARI), Absolute configuration, X-ray diffraction

INTRODUCTION

Several furanocoumarins isolated from *Angelicae* plants, most frequently prescribed in Chinese medicines, have been found to cause a profound inhibitory activity of the hepatic microsomal cytochrome *P-450* dependent mixed-function oxidase (MFO) system, and this activity was accompanied by the loss of cytochrome *P-450* itself (Shin and Woo, 1986, 1990).

Furanocoumarins such as byakangelicin, neobyakangelicin and isopimpinellin strongly inhibit insulin-stimulated lipogenesis. Therefore, the crude drug *Byakuchi* may activate the actions of lipolytic hormones and selectively inhibit the effects of antilipolytic hormones (Kimura and Arichi, 1982).

The enzyme aldose reductase (AR) which brings about intracellular accumulation of sorbitol or galactitol in the polyol pathway of aldose metabolism has been demonstrated to play important roles not only in the cataract formation in the lens (Van Heyningen, 1959; Pirie and Van Heyningen, 1964) but also in the pathogenesis of

diabetic complications such as neuropathy (Ward, 1973), retinopathy (Engerman and Kern, 1984) and nephropathy (Beyer-Mears et al., 1984).

Evidence, therefore, suggests that compounds which inhibit AR are expected to be effective in preventing the sugar cataract formation as well as in the diabetic complications. Recently, a number of structurally diverse AR inhibitors (ARI) not only of synthetic but also of natural origin, have been extensively studied to clarify their *in vivo* effects for prevention of cataract formation as well as diabetic complications in experimental animals (Dvornic et al., 1973) and even in clinical trials (Handelman and Turtle, 1981).

Several synthetic ARs are currently available and many have been tested for their clinical use, albeit with limited success (Raskin and Rosenstock, 1987), i.e.; Synthetic compounds with diverse structures such as sorbinil (Beyer-Mears and Cruz, 1985), epalrestat (Terashima et al., 1982), flavonoids (Shimazu, 1984) and isoliquiritigenin (Aida et al., 1990) from natural origin have been extensively studied and reported to inhibit AR. Shin and co-workers carried out the research for a new potential AR inhibitor useful for the treatment of galactosemic as well as for the diabetic cataract from *Angelica dahurica* roots and they found that byakangelicin is the most promising active principles utilizable as a lead compound,

Correspondence to: Yang Bae Kim, College of Pharmacy, Seoul National University, San 56-1, Shinlim-Dong, Kwanak-Ku, Seoul 151-742, Korea
E-mail: ybkim@plaza.snu.ac.kr

because this compound not only inhibited AR *in vitro* but also prevented formation of galactosemic cataract *in vivo*, and furthermore, it was a main component peculiar to the plant part.

In this work, the three dimensional structure of byakangelicin,7-(2,3-dihydroxy-3-methylbutoxy)-6-hydroxy-4-methoxy-5-benzofuranacrylic acid, extracted from the roots of *Angelica dahurica* (*Umbelliferae*), was determined by X-ray diffraction method in order to provide useful information for the understanding of the structure-activity relationship.

METHODS

Yellowish prismatic crystals were grown by slow evaporation from a mixture of methanol and ether at room temperature.

Crystal data

$C_{17}H_{18}O_7$
 $M_r = 334.33$
Triclinic

CuK radiation
 $\lambda = 1.5418$
Cell parameters from 25

$P1$ reflections (range 11.7-18.7°)
 $a = 8.114(1) \text{ \AA}$
 $b = 10.194(1) \text{ \AA}$ $m = 0.872$
 $c = 11.428(1) \text{ \AA}$ $T = 293 \text{ K}$
 $\alpha = 111.50(1)^\circ$ Colourless Prism
 $\beta = 95.57(1)^\circ$ $0.3 \times 0.3 \times 0.2 \text{ mm}^3$
 $\gamma = 112.52(1)^\circ$
 $V = 780.8(2) \text{ \AA}^3$ $Z = 2$
 $D_x = 1.41 \text{ g/cm}^3$
 $D_m = 1.39 \text{ g/cm}^3$ (measured by flotation in KI solution)

Data collection

<i>Enraf-Nonius CAD4</i>	$R_{\text{int}} = 0.025$
diffractometer	$q_{\text{max}} = 70.0^\circ$
w/2q scans	$h = -9 \rightarrow 9$
Absorption correction:	$k = -11 \rightarrow 11$
none	$l = 0 \rightarrow 13$
2927 measured reflections	3 standard reflections
2731 unique reflections	frequency: 60 min
2584 observed reflections	intensity decay: 1.2%
[$F > 3\sigma(F)$]	

Table I. Fractional atomic coordinates and equivalent isotropic displacement parameters(Å², X10⁴) with their e.s.d's.
 $U_{eq} = (1/3) \sum_i \sum_j U_{ij} a_i a_j$

Atom	[Molecule A]				[Molecule A]			
	x/a	y/b	z/c	Ueq	x/a	y/b	z/c	Ueq
O(1)	9780(6)	3145(5)	1277(4)	63(2)	-204(5)	6459(5)	8421(4)	61(2)
O(2)	11084(6)	5954(5)	3492(4)	76(2)	-1362(6)	3754(5)	6226(4)	75(2)
O(3)	13752(6)	8719(4)	3414(4)	55(2)	-4090(6)	928(5)	6318(4)	67(2)
O(4)	13767(6)	5638(5)	-878(4)	73(2)	-4019(7)	3978(5)	10581(4)	76(2)
O(5)	8035(8)	627(6)	358(7)	111(3)	1621(7)	9059(6)	9385(5)	92(3)
O(6)	7745(8)	4872(6)	4401(5)	89(2)	-931(8)	2694(7)	3806(5)	103(3)
O(7)	9444(8)	9140(5)	5863(6)	86(2)	2 08(7)	491(5)	3882(5)	79(2)
C(8)	15012(8)	9698(7)	3016(6)	63(2)	-5380(9)	-53(8)	6713(7)	71(3)
C(9)	15111(9)	8921(7)	1793(7)	72(3)	-5390(8)	714(7)	7910(6)	58(3)
C(10)	13773(7)	7278(5)	1338(5)	42(2)	-4074(7)	2346(7)	8375(6)	57(2)
C(11)	12987(7)	7238(6)	2383(5)	46(2)	-3342(8)	2400(6)	7331(5)	53(2)
C(12)	11699(7)	5919(7)	2419(5)	56(2)	-2042(8)	3735(7)	7299(5)	53(2)
C(13)	11164(6)	4519(6)	1287(5)	42(2)	-1457(7)	5116(7)	8398(5)	59(3)
C(14)	9172(9)	1698(7)	239(7)	69(3)	4 75(8)	7946(8)	9476(6)	67(3)
C(15)	9943(10)	1640(8)	-831(7)	64(3)	-286(10)	8010(8)	10576(7)	70(3)
C(16)	11199(8)	2942(7)	-845(6)	59(3)	-1547(9)	6720(7)	10580(5)	57(3)
C(17)	11844(7)	4456(6)	205(5)	48(2)	-2190(7)	5174(6)	9472(5)	48(2)
C(18)	13130(8)	5818(7)	214(5)	59(2)	-3497(7)	3770(6)	9481(5)	46(2)
C(19)	14548(13)	6906(11)	-1194(8)	82(4)	-4847(11)	2693(8)	10905(7)	79(3)
C(20)	9812(9)	6593(8)	3695(5)	61(3)	-187(9)	2942(7)	5999(6)	71(3)
C(21)	9122(7)	6510(6)	4838(5)	53(2)	608(8)	3184(7)	4894(5)	55(2)
C(22)	8125(8)	7500(7)	5280(5)	58(2)	1625(8)	2122(6)	4427(5)	55(2)
C(23)	7202(10)	7214(10)	6322(7)	80(4)	2398(13)	2387(10)	3335(9)	78(4)
C(24)	6633(11)	7241(11)	4162(7)	84(4)	3114(13)	2427(12)	5528(9)	99(5)

Table II. The bond lengths(Å) of byakangelicin with their e.s.d's.

	[Mol. A]	[Mol. B]		[Mol. A]	[Mol. B]
O(1)-C(13)	1.416(6)	1.346(6)	O(1)-C(14)	1.368(7)	1.395(7)
O(2)-C(12)	1.361(6)	1.396(6)	O(2)-C(20)	1.412(6)	1.469(6)
O(3)-C(8)	1.375(7)	1.387(8)	O(3)-C(11)	1.374(6)	1.361(7)
O(4)-C(18)	1.374(6)	1.335(6)	O(4)-C(19)	1.401(9)	1.425(7)
O(5)-C(14)	1.193(7)	1.209(7)	O(6)-C(21)	1.462(7)	1.456(7)
O(7)-C(22)	1.435(7)	1.453(7)	C(8)-C(9)	1.359(9)	1.306(9)
C(9)-C(10)	1.454(8)	1.440(8)	C(10)-C(11)	1.413(7)	1.394(7)
C(10)-C(18)	1.418(7)	1.394(8)	C(11)-C(12)	1.370(8)	1.382(8)
C(12)-C(13)	1.401(7)	1.364(8)	C(13)-C(17)	1.392(7)	1.407(7)
C(14)-C(15)	1.420(1)	1.445(9)	C(15)-C(16)	1.336(9)	1.324(9)
C(16)-C(17)	1.413(8)	1.462(7)	C(17)-C(18)	1.377(8)	1.423(7)
C(20)-C(21)	1.488(7)	1.529(7)	C(21)-C(22)	1.503(7)	1.576(7)
C(22)-C(23)	1.525(8)	1.514(8)	C(22)-C(24)	1.541(8)	1.506(9)

Table III. The bond angles(°) of byakangelicin with their e.s.d's.

C(14)-O(1)-C(13)	120.7(5)	123.9(4)	C(20)-O(2)-C(12)	114.6(4)	111.6(4)
C(11)-O(3)-C(8)	105.6(4)	105.8(5)	C(19)-O(4)-C(18)	121.0(5)	121.2(5)
C(9)-C(8)-O(3)	12.9(5)	111.6(6)	C(10)-C(9)-C(8)	105.7(6)	107.9(6)
C(11)-C(10)-C(9)	105.0(5)	104.4(5)	C(18)-C(10)-C(9)	138.3(5)	137.8(5)
C(18)-C(10)-C(11)	16.7(5)	117.6(5)	C(10)-C(11)-O(3)	110.7(5)	110.2(5)
C(12)-C(11)-O(3)	123.2(5)	123.7(5)	C(12)-C(11)-C(10)	126.0(5)	126.1(5)
C(11)-C(12)-O(2)	122.9(5)	124.7(5)	C(13)-C(12)-O(2)	123.1(5)	119.5(5)
C(13)-C(12)-C(11)	113.9(4)	115.8(5)	C(12)-C(13)-O(1)	115.1(4)	118.5(5)
C(17)-C(13)-O(1)	121.1(4)	119.7(5)	C(17)-C(13)-C(12)	123.7(5)	121.7(5)
O(5)-C(14)-O(1)	114.5(7)	117.6(6)	C(15)-C(14)-O(1)	117.5(5)	116.5(5)
C(15)-C(14)-O(5)	128.0(7)	126.0(6)	C(16)-C(15)-C(14)	121.8(6)	121.4(6)
C(17)-C(16)-C(15)	122.0(6)	120.6(6)	C(16)-C(17)-C(13)	116.6(5)	117.8(5)
C(18)-C(17)-C(13)	120.1(5)	120.9(5)	C(18)-C(17)-C(16)	123.2(5)	121.3(5)
C(10)-C(18)-O(4)	124.3(5)	127.1(5)	C(17)-C(18)-O(4)	116.1(5)	115.0(4)
C(17)-C(18)-C(10)	119.5(5)	117.9(5)	C(21)-C(20)-O(2)	111.5(4)	106.2(4)
C(20)-C(21)-O(6)	106.5(5)	108.0(5)	C(22)-C(21)-O(6)	106.1(4)	108.9(4)
C(22)-C(21)-C(20)	113.1(4)	109.7(5)	C(21)-C(22)-O(7)	109.1(5)	106.5(4)
C(23)-C(22)-O(7)	106.8(5)	107.9(5)	C(23)-C(22)-C(21)	111.9(5)	108.4(5)
C(24)-C(22)-O(7)	107.0(5)	109.6(6)	C(24)-C(22)-C(21)	113.8(5)	112.6(5)
C(24)-C(22)-C(23)	107.9(5)	111.6(6)			

RefinementRefinement on F $R = 0.056$ $wR^2 = 0.081$ $S = 0.605$

2534 reflections

563 parameters

All H-atom parameters refined

Unit weights applied

 $(\Delta/\sigma)_{\text{max}} = 0.17$ $\Delta\rho_{\text{max}} = 0.250 \text{ \AA}^{-3}$ $\Delta\rho_{\text{min}} = -0.020 \text{ \AA}^{-3}$ atomic scattering factors from *SHELX76*

(Sheldrick, 1976)

Intensity data were collected with a scan width of $\Delta\omega = (1.2 \pm 0.15\theta)^\circ$ (estimated from $\omega-\theta$ plots). The counter aperture was also adjusted as a function of θ . The horizontal aperture width ranged from 2.0 to 3.1 mm, while the vertical one was set at 4 mm. All reflections were corrected for the usual L_p effects and decay compensation was applied.

Data collection: CAD-4 software (Enraf-Nonius, 1989).

Cell refinement: CAD-4 software (Enraf-Nonius, 1989).

Data reduction: MolEN (Fair, 1990). Program(s) used to

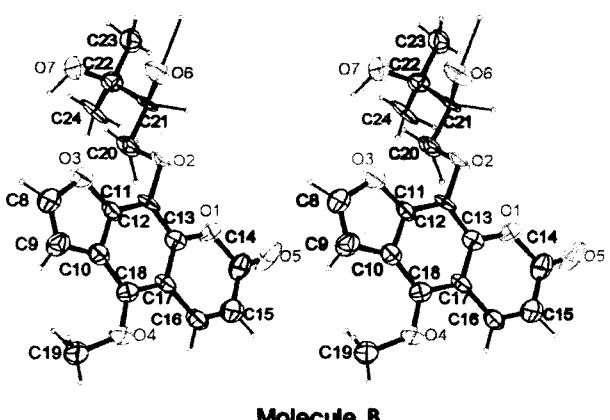
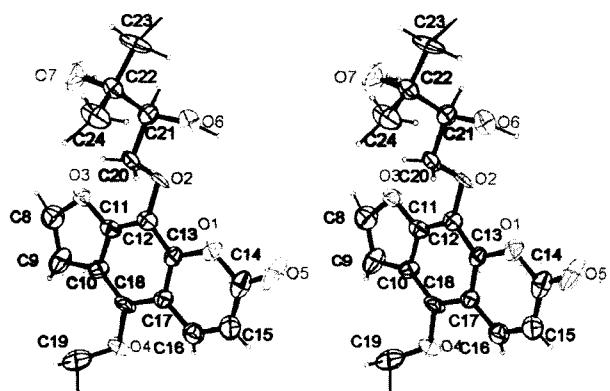


Fig. 1. ORTEPII (Johnson, 1976) drawing of byakangelicin with the atomic numbering scheme. The displacement ellipsoids are drawn at the 50% probability level. H atoms are drawn as small circles of arbitrary radii.

solve structure : MULTAN84 (Main, Germain & Woolfson, 1984). Program(s) used to refine structure: SHELX76 (Sheldrick, 1976). Molecular graphics: ORTEP (Johnson, 1976). Software used to prepare material for publication: MolEN (Fair, 1990).

RESULTS AND DISCUSSION

There are two byakangelicin molecules in an asymmetric unit, which are tentatively named as molecule A and molecule B. The atomic parameters are presented in Table I.

Bond lengths and angles of this molecule are within the range of chemically reasonable values (Table II, III). The stereoscopic views of both The conformations of both molecules are slightly different. The torsion angles of O(6)-C(21)-C(22)-C(24) are -67.12° (gauche) in molecule A and molecules, drawn by ORTEP II (Johnson, C.K., 1976), are presented in Fig.1.

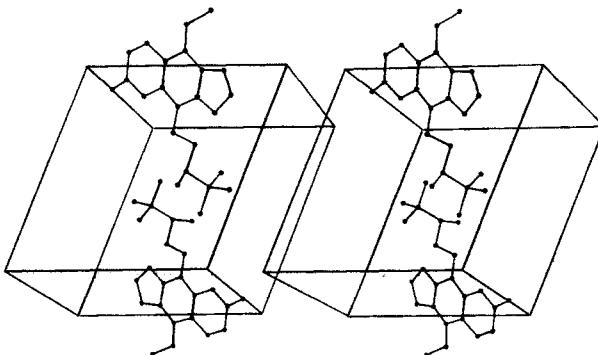


Fig. 2. Crystal packing of byakangelicin

-173.93° (*trans*) in molecule B. However, those of O(6)-C(21)-C(22)-O7 are 173.45° (*trans*) in molecule A and -53.78° (*gauche*) in molecule B.

The molecules in crystal state are kept by two intermolecular O-HO type hydrogen bonds. The distance of O6(A)(x,y,z)-HO6(B)(x+1,y,z) is 2.694 Å and that of O7(A)(x,y,z)H-O7(B)(x+1,y+1,z) is 3.079 Å. The molecules are kept by *van der Waal's* forces in crystal.

The absolute configuration was estimated by using the 'Eta refinement' in the MolEN program package to refine the final atomic coordinates together with the eta parameter. After Eta refinement with the value of Eta parameter ; -1, the final R value was 7.6%. This result shows that present configuration is reasonable and both molecules have S forms.

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