

Conformal Al₂O₃ Nanocoating of Semiconductor Nanowires by Atomic Layer Deposition

Joo-Won Hwang, Byung-Don Min and Sang-Sig Kim

Abstract - Various semiconductor nanowires such as GaN, GaP, InP, Si₃N₄, SiO₂/Si, and SiC were coated conformally with aluminum oxide (Al₂O₃) layers by atomic layer deposition (ALD) using trimethylaluminum (TMA) and distilled water (H₂O) at a temperature of 200°C. Transmission electron microscopy (TEM) revealed that Al₂O₃ cylindrical shells conformally coat the semiconductor nanowires. This study suggests that the ALD of Al₂O₃ on nanowires is a promising method for preparing cylindrical dielectric shells for coaxially gated nanowire field-effect transistors.

Keywords: nanowire, atomic layer deposition (ALD), Al₂O₃ coating

1. Introduction

A variety of semiconductor nanowires have been investigated due to their great potential in the application of optical and electronic devices [1, 2]. To realize the optical and electronic devices utilizing semiconductor nanowires, protection of the nanowires from oxidation, nitrification, and wetting is crucial; in ambient air, the surface of Si nanowires are oxidized to form SiO₂ layers surrounding themselves and GaP nanowires are nitrified [3]. Coating of semiconductor nanowires is therefore required to retain their optical, electrical, and physical properties.

Aluminum oxide (Al₂O₃) exhibits many excellent electrical and optical properties including high dielectric constant, very low permeability of alkali ions, high thermal conductivity, and transparency over a wide range of wavelengths [4-6]. Due to these characteristic properties, extensive attempts have been made to utilize Al₂O₃ in the fields of capacitor dielectrics and gate oxides in memory devices, insulating layers in electroluminescent display devices, waveguide, and corrosion-resistant coating films on metals [7]. More recently, Al₂O₃ barrier layers deposited on Si by atomic layer deposition (ALD) technique have been investigated to study electrical and reliability characteristics of Si/Al₂O₃/Si capacitor structures for the future application of DRAM [8-10].

Lieber's group has recently realized the fabrication of a coaxially gated nanowire field-effect transistor (FET) [11]. The coaxially gated nanowire FET has drawn much attention due to its capacitance enhancement compared to planar gates utilized for nanowire FETs. In this FET, the active

channel is a Ge shell and the gate dielectric is a SiO_x shell. The concept of the coaxially gated nanowire FET is applicable to other semiconductor nanowires. For the fabrication of the coaxially gated nanowire FETs using other semiconductor nanowires including GaN, GaP, InP, Si₃N₄, SiO₂/Si, SiC, and ZnO semiconductor nanowires, the cylindrical gate dielectrics should be formed on these nanowires; for the SiO₂/Si nanowire, the SiO₂ shell also could be a cylindrical gate dielectric. The motive of this study is therefore to suggest a straightforward method for the formation of the cylindrical dielectric with a dielectric constant higher than SiO₂ on any of semiconductor nanowires; dielectric constant is 3.9 and 10.5-12 for SiO₂ and Al₂O₃, respectively.

In this study, GaN, GaP, InP, Si₃N₄, SiO₂/Si, SiC, and ZnO semiconductor nanowires were first synthesized by a variety of growth methods and then deposited conformally with Al₂O₃ by ALD. The structural and chemical properties of the deposited Al₂O₃ were investigated by transmission electron microscopy (TEM), selected area electron diffraction (SAED) pattern, and energy-dispersive X-ray (EDX) spectroscopy. In this paper, the experimental results of the Al₂O₃-deposited GaN semiconductor nanowires are described in detail as a representative case and other Al₂O₃-deposited nanowires are briefly described and compared.

2. Experimental procedure

Semiconductor nanowires have been synthesized using a variety of growth methods. The synthesis procedures of the GaN, GaP, SiC, and ZnO nanowires are briefly described as follows. (1) GaN nanowires were synthesized from Fe nanoparticles deposited on a Si substrate during the evapo-

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ration from a mixture of Ga and GaN powders at temperatures of 950–1100°C [12]. (2) GaP nanowires were prepared from the evaporation of ball-milled GaP powders at 1100°C. They were grown on an Ni nanoparticles-deposited alumina substrate at a temperature of 900°C [3]. (3) SiC nanowires were synthesized from the reaction of a Si (100) wafer with a mixture of Ga and GaN powders at a temperature of 1100°C under a flow of the CH₄ (99.95%) and H₂ (99.999%) mixture [13]. (4) ZnO nanowires were formed on an Si substrate from the evaporation of ball-milled ZnO powders at 1380°C for three hours while Ar was flown into the chamber at a rate of 500 sccm [14]. The detailed synthetic procedures of the InP, Si₃N₄, and SiO₂/Si nanowires will be reported elsewhere.

Al₂O₃ was deposited on the synthesized semiconductor nanowires at a growth temperature of 200°C using the ALD technique. Trimethylaluminum (TMA) and distilled water were utilized as precursors for deposition [4]. The process pressure was 250 and 230 mTorr for the dosing of chemical precursors and the Ar purging, respectively. Precursor elements must be alternately dosed to the semiconductor nanowires. The dosing time and purging time were 1 and 20 sec, respectively. 100 process cycles were performed for the deposition of Al₂O₃; one cycle of the deposition of Al₂O₃ consists of TMA dosing, Ar purging, H₂O dosing, and Ar purging. TEM images, EDX spectra, and SAED patterns of the semiconductor nanowires were taken by transmission electron microscopy (TEM, Philips CM30) or high-resolution transmission electron microscopy (HRTEM, JEOL JEM-2010).

3. Results and discussion

Fig 1 is a TEM image of a nanowire selected from GaN nanowires grown on an Si (100) substrate; its diameter and length are 60 nm and 20 μm, respectively. The XRD pattern taken for the GaN nanowires showed that its lattice structure is wurtzite (not shown here). The SAED pattern in the inlet of Fig. 1 indicates that the growth direction of the GaN nanowire is parallel to the [0001] lattice direction. Fig 2 is a TEM image of a selected GaN nanowire deposited with Al₂O₃ by ALD; the Al₂O₃ shell is marked by the arrows. The TEM image illustrates that an Al₂O₃ shell with the cap surrounds the GaN nanowire. The Al₂O₃ shell is 20 nm in thickness; its thickness is substantially uniform along the nanowire. The inset of Fig. 2 shows the SAED pattern of the Al₂O₃-deposited GaN nanowire. This pattern shows the halo overlapped with a set of single-crystal electron diffraction spots, while that of the bare GaN nanowire shows only the spots (see the inset of Fig. 1).

The presence of the halo in the SAED pattern indicates that Al₂O₃ deposited on GaN nanowire is amorphous. Fig 3

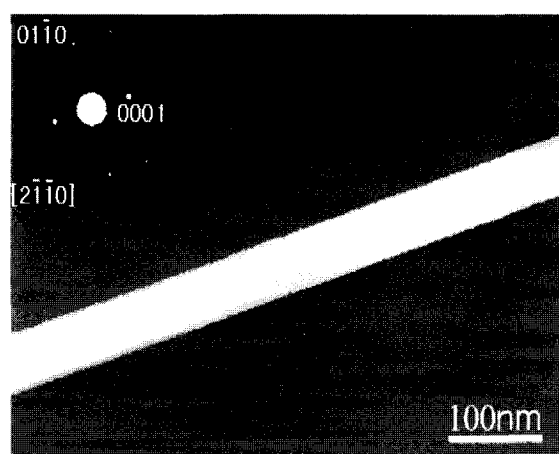


Fig. 1 TEM image of a selected GaN nanowire grown on a Si (100) substrate. The SAED pattern in the inset indicates that the growth direction is parallel to the [0001] lattice direction.

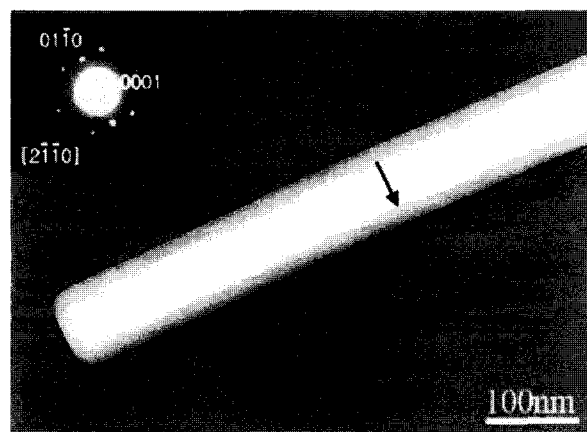


Fig. 2 TEM image of a selected GaN nanowire coated with Al₂O₃ by ALD. The Al₂O₃ shell is marked with an arrow. In the inset, the SAED pattern shows a halo overlapped with the single-crystal electron diffraction spots.

shows the representative EDX spectra taken from a (a) bare and (b) Al₂O₃-deposited GaN nanowires. The EDX spectrum taken from the Al₂O₃-deposited GaN nanowire exhibits Al- and O-related peaks, indicating that the deposited part is indeed Al₂O₃; the Cu-related peak is due to the presence of Cu grids. The TEM, SAED, and EDX illustrate that the GaN nanowire is deposited cylindrically with amorphous Al₂O₃ material.

The lattice structures, lattice constants, growth directions, and diameters (or thicknesses) of the semiconductor nanowires and Al₂O₃ shells are summarized in Table 1. Fig 4 shows TEM images of GaP, InP, Si₃N₄, SiO₂/Si, SiC, and ZnO nanowires deposited with Al₂O₃ by ALD; the Al₂O₃ shells are marked by arrows. The TEM images of the deposited nanowires illustrate that the cylindrical Al₂O₃

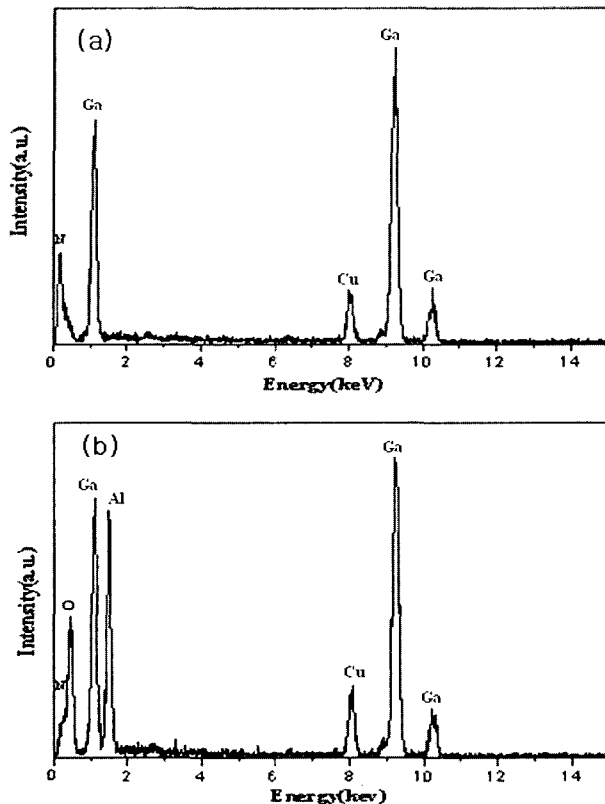


Fig. 3 EDX spectra taken from (a) bare and (b) Al₂O₃-deposited GaN nanowires.

shells surround conformally the six different semiconductor nanowires, as observed for the Al₂O₃-coated GaN nanowire. For these Al₂O₃-coated nanowires, the Al₂O₃ shells are 20 nm in thickness. The cross sections of the cylindrical shells surrounding the GaP, InP, Si₃N₄, SiO₂/Si, SiC, and ZnO nanowires are circular. The thicknesses of the Al₂O₃ shells are uniform along the nanowires. This observation reveals that these six different semiconductor nanowires all are deposited conformally with Al₂O₃. SAED patterns of these six different nanowires coated by Al₂O₃ showed the halo patterns that are the same as for the Al₂O₃-deposited GaN nanowire, indicating that the Al₂O₃ shells on all the semiconductor nanowires are amorphous. The deposition of the amorphous Al₂O₃ on semiconductor nanowires by ALD is associated with the relatively low growth temperature (200°C); Iizuka et al. report that growth temperatures above 800°C are required to obtain the crystalline Al₂O₃ layers [15].

The thicknesses of Al₂O₃ coating shells (on the seven different semiconductor nanowires) prepared with five different cycles were precisely measured by TEM. The thicknesses of the Al₂O₃ shells for each of the five different cycles are the same for all semiconductor nanowires.

The deposition rate of the Al₂O₃ coating may be determined to be 0.20 nm/cycle. This value agrees with the

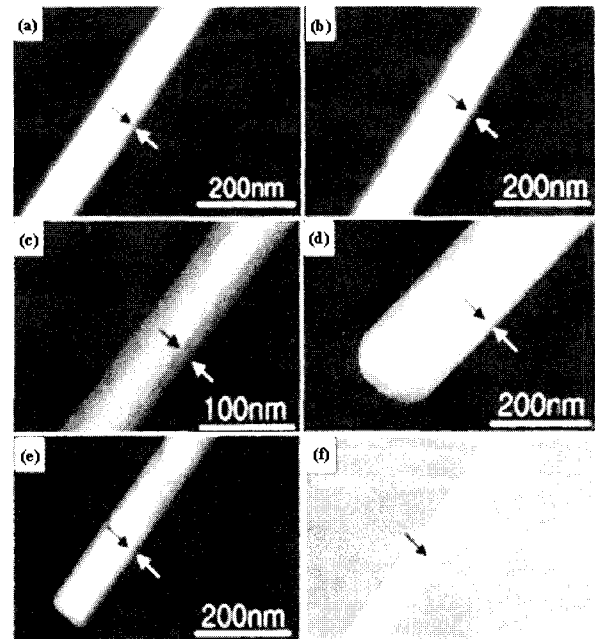


Fig. 4 TEM images of (a) GaP, (b) InP, (c) Si₃N₄, (d) SiO₂/Si, (e) SiC, and (f) ZnO nanowires deposited with Al₂O₃ using ALD. The Al₂O₃ shells are marked by arrows.

growth rate for flat Si surfaces [4]. These indicate that the growth rate of the Al₂O₃ nano-shells deposited by ALD does not depend on the kind of semiconductor material as well as crystal structure. Moreover, the thickness of a Al₂O₃ coating layer is substantially uniform for semiconductor nanowires having different chemical components, different lattice structures, and different lattice constants (see the details in Table 1). The excellent uniformity is due to the amorphous properties of the Al₂O₃ coating material and its excellent sticking capability. If the Al₂O₃ coating layer is not amorphous, the thickness of the Al₂O₃ coating layer

Table 1 Characteristics of semiconductor nanowires and Al₂O₃.

Material	Lattice structure*	Lattice constant (Å)	Growth direction	Diameter (D) or thickness (T) (nm)
GaN	W	a=3.190, c=5.189	[0001]	60(D)
GaP	Z	a=5.451	[111]	140(D)
InP	Z	a=5.861	[111]	50(D)
Si ₃ N ₄	W	a=7.754, c=5.622	[2 $\bar{1}$ 10]	40(D)
SiO ₂	A	-	-	50(T)
Si	D	a=5.392	[111]	40(D)
SiC	Z (3C)	a=4.359	[111]	65(D)
ZnO	W	a=3.250, c=5.205	[1120]	20(D)
Al ₂ O ₃	A	-	-	20(T)

*wurtzite (W), zinc blende (Z), amorphous (A), and diamond (D)

would depend on the structural and compositional properties and therefore would be nonuniform. For instance, the TiO₂ and ZnO films deposited by ALD are microcrystalline and these deposited layers are nonuniform on the nanometer scale [16–18]. The uniform thickness observed for all the nanowires also reveals the excellent sticking capability of the Al₂O₃ material, which our observation indicates is independent of semiconductor substances.

5. Conclusion

GaN, GaP, InP, Si₃N₄, SiO₂/Si, SiC, and ZnO semiconductor nanowires were deposited cylindrically with amorphous Al₂O₃ via ALD. TEM images and SAED patterns of the deposited semiconductor nanowires reveal that 20 nm-thick Al₂O₃ cylindrical shells surround the semiconductor nanowires. Comparison of these Al₂O₃ shells deposited on the seven different semiconductor nanowires illustrates that the crystal structure and chemical components of the shells deposited by ALD using TMA and distilled water are independent of the chemical component, lattice structure, and growth direction of the nanowire. Our study shows that the ALD has an excellent capability to deposit Al₂O₃ cylindrically on any semiconductor nanowires, suggesting that the ALD of Al₂O₃ on nanowires is a promising methods to prepare cylindrical dielectric shells in coaxially gated nanowire field-effect transistors.

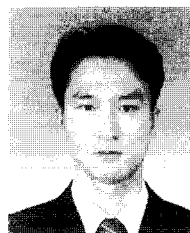
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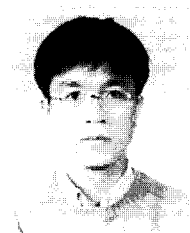
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