# Polyketide and Sesquiterpenediol Metabolites from a Marine-Derived Fungus 

Xifeng Li, Se-Kwon Kim, Jung Sook Kang,' Hong Dae Choi, ${ }^{\text {T }}$ and Byeng Wha Son ${ }^{\text {* }}$<br>Deparment of Chemistry, Pukyong National Universit: Busan 608-737, Korea<br>${ }^{\dagger}$ College of Dentistry. Pusam National University, Busan 602-739. Korea<br>${ }^{\ddagger}$ Department of Chemistry, Dongeni Unversith, Busan 6/4-714, Korea<br>Received March 2. 2004

Kcy Words: Marine-derived fungus, Brominated alkenoate, Methyl 2,4-dibromo-5-oxo-2-decenoate, Methyl 2.4-dibromo-5-oxo-3-decenoate, Cyclonerodiol

Marine microorganisms such as bacteria and fungi inhabit virtually any environment in the sea, and they are rich sources of chemically and biologically diverse compounds. ${ }^{1.2}$
In our search for bioactive compounds in marine microorganisms, ${ }^{3}$ two new halogenated alkenoates, methyl 2.4-dibromo-5-oxo-2-decenoate (1) and methyl 2.4-dibromo-5-oxo-3-decenoate (2), and the known sesquiterpenediol. cycloneroidol (3), were isolated from the broth of an unidentified fungus, which was separated from the surface of the marine red alga Gracillaria vernueosa collected at Jinha, Ulsan in 2002.


The fungus was cultured ( 10 L ) in a seawater-based medium. ${ }^{4}$ The resulting broth and mycelium were extracted separately to afford crude extracts of 0.7 g and 6.5 g , respectively. The broth extract (EtOAc) was subjected to a combination of column chromatography on silica gel ( $n$ hexane/EtOAc) and octadesyl silica (ODS) gel ( $\mathrm{H}_{2} \mathrm{O} / \mathrm{MeOH}$ ) to furnish the fractions containing compounds 1 and 2 ( 20 mg ), and compound $\mathbf{3}$ ( 25 mg ). Further purification of each fraction by HPLC (YMC ODS-A, MeOH- $\mathrm{H}_{2} \mathrm{O}=5$ : 1) yielded compounds $\mathbf{1}(5.5 \mathrm{mg}), 2(8.0 \mathrm{mg})$, and $\mathbf{3}(11 \mathrm{mg})$, respectively.

Compound $2^{5}$ was isolated as a yellow oil which was thought to have a molecular composition of $\mathrm{C}_{11} \mathrm{H}_{16} \mathrm{Br}_{2} \mathrm{O}_{3}$ from the high resolution (HR) FABMS and ${ }^{15} \mathrm{C}$ NMR data.

Three degree of unsaturation in HRFABMS implied that 2 contained two carbonyls and one double bond. The quasimolecular ions were observed at $\mathrm{m} / \mathrm{z} 355,357$, and 359 with the ratio $1: 2: 1$, indicating that compound 2 has two bromine atoms. The IR spectrum of 2 showed absorptions for ester ( $1743,1137 \mathrm{~cm}^{-1}$ ) and enone (1697, $1265 \mathrm{~cm}^{-1}$ ) functionality. The UV spectrum also exhibited the presence of an enone chromophore [257 nm $(\log \varepsilon 3.5)]$.

In the ${ }^{1} H$ NMR spectrum, the presence of an ester methyl
proton $\mid \delta 3.85\left(3 \mathrm{Il}\right.$, s. $\left.1-\mathrm{OCH}_{3}\right) \mid$, an olefinic proton $\mid \delta 7.43$ ( $1 \mathrm{II}, \mathrm{d}, J-9.8 \mathrm{~Hz}, \mathrm{ll}-3$ )|, an allyl proton $[5.28$ ( $1 \mathrm{HI}, \mathrm{d}, J-$ $9.8 \mathrm{~Hz}, \mathrm{H}-2$ ) J, and $n$-pentyl protons was inferted. Detailed analyses of the ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra of 2 , including the results from COSY, DEPT, IIMQC, and IIMBC experiments, revealed signals ascribable to a methyl ester $\mid \delta 3.85$ ( 3 H , s, $\left.1-\mathrm{OCl}_{5}\right)$, $167.5(\mathrm{C}-1), 53.7$ ( 1 -OMc)]. 1,2.4,4-tetrasubstituted-2-buten-1-one $[\delta 5.28$ (111, d, $J-9.8 \mathrm{~Hz}, \mathrm{HI}-2), 7.43$ (11I, d, $J-9.8 \mathrm{~Hz}, \mathrm{H}-3)], 40.9(\mathrm{C}-2), 135.4(\mathrm{C}-3), 130.4(\mathrm{C}-4)$. 193.7 (C-5)], and n-pentyl moiety [ $\delta 2.82(2 \mathrm{H}, \mathrm{t}, J-7.3 \mathrm{~Hz}$. $\left.\mathrm{H}_{2}-6\right), 1.66\left(2 \mathrm{H}, \mathrm{m}, \mathrm{H}_{2}-7\right), 1.32\left(4 \mathrm{H}, \mathrm{m}, \mathrm{I} \mathrm{I}_{2}-8 / 9\right), 0.91(3 \mathrm{H}, 1$, $\left.J-6.6 \mathrm{HL}, \mathrm{H}_{\mathrm{Y}}-10\right), 38.8(\mathrm{C}-6), 23.9(\mathrm{C}-7), 31.2(\mathrm{C}-8), 22.4$ (C-9), 13.9 (C-10)| (Table 1).
The connection of the functional groups in 2, which led to the planar structure, was achieved on the basis of HMQC and HMBC data. Key HMBC correlations from $1-\mathrm{OCH}_{3}$ to $\mathrm{C}-1$, from I1-2 to $\mathrm{C}-1, \mathrm{C}-3$, and $\mathrm{C}-4$, from 11-3 to $\mathrm{C}-1$ and $\mathrm{C}-$ 5, from IJ-6 to C-5, and from I1-7 to C-5 were critical in establishing the planar structure of 2 .

Two bromines were confirmed to attach to C-2 and C-4 by the HMBC correlations between $\mathrm{H}-2$ and $\mathrm{C}-1, \mathrm{C}-3$, and $\mathrm{C}-4$, as well as the characteristic mass fragments of $m / 29$ $\left[\mathrm{C}_{5} \mathrm{II}_{1 /} \mathrm{CO}\right]^{\prime}$ and $203 \mid \mathrm{M}-\mathrm{Cl}_{3} \mathrm{OCOCl} \mathrm{BBr}^{-}$.

Compound $1^{6}$ was oblained as a yellow oil, and IIRFABMS

Table 1. ' $\mathrm{H}(\delta$, mult. $J)$ and ${ }^{13} \mathrm{C}(\delta$. mult) NMR Data for Methy] 2.4-Dibromo-5-oxo-2-decenoatc (1) and Its 3-decenoate (2) $)^{\text {s }}$

| $\begin{gathered} \text { Carbon } \\ \text { No. } \end{gathered}$ | 1 |  | 2 |  |
| :---: | :---: | :---: | :---: | :---: |
|  | $\delta_{H}$ | $\delta$ | $\delta_{H}$ | $\delta$ |
| 1 |  | 161.9 (s) |  | 167.5 (s) |
| 2 |  | 120.5 (5) | 5.28 (d. 9.8) | 40.9 (d) |
| 3 | 7.63 (d. 9.9) | 137.7 (d) | 7.43 (d. 9.8) | 135.4 (d) |
| 4 | 5.22 (d. 9.9) | 47.4 (d) |  | 130.4 (s) |
| 5 |  | 199.9 (s) |  | 193.7(s) |
| 6 | 2.85 (dt. 17.3.7.3) | 39.8 (t) | 2.82 (t. 7.3 ) | 38.8 (1) |
|  | 2.61 (dt. 17.3.7.3) |  |  |  |
| 7 | 1.66 (m) | 23.5 (t) | 1.66 (m) | 23.9 (t) |
| 8 | 1.32 (m) | 31.1 (t) | 1.32 (m) | 31.2 (t) |
| 9 | 1.32 (m) | 22.3 (t) | 1.32 (m1) | 22.4 (t) |
| 10 | 0.90 (t. 6.9) | 13.9 (q) | 0.91 (t. 6.6) | 13.9 (q) |
| 1-OMe | 3.87 (s) | 53.8 (q) | 3.85 (s) | 53.7 (q) |

[^0]and ${ }^{13} \mathrm{C}$ NMR methods established the molecular formula to be $\mathrm{C}_{11} \mathrm{H}_{16} \mathrm{Br}_{3} \mathrm{O}_{3}$. The general features of its UV. IR and NMR spectra (Table 1) closely resembled those of compound 2. except that the coupling pattern of NMR signal assigned to the methy lene $\left(\mathrm{H}_{2}-6\right)$ was changed from triplet [ $\delta 2.82$ ( 2 H . $\left.\left.\mathrm{t} . J=7.3 \mathrm{~Hz} . \mathrm{H}_{-}-6\right)\right]$ for 2 to doublet of triplet $[\delta 2.61(\mathrm{lH}, \mathrm{dt}$. $\left.J=17.3,7.3 \mathrm{~Hz}, \mathrm{H}_{a}-6\right)$ and $2.85(\mathrm{LH} . \mathrm{dt} . J=17.3 .7 .3 \mathrm{~Hz}$. $\mathrm{H}_{\mathrm{b}}-6$ )] for 1 (Table I).
Detailed analyses of the ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra of 1 . including the results from DEPT. COSY. HMQC. and HMBC experiments. suggested that the metabolite $\mathbf{1}$ is the positional isomer of double bond of compound 2 .
The location of double bond of the metabolite 1 was determined by the HMBC data, in which diagnostic correlations from $\mathrm{H}-4$ to $\mathrm{C}-2, \mathrm{C}-3$. and $\mathrm{C}-5$. and from $\mathrm{H}-6$ to $\mathrm{C}-5, \mathrm{C}-7$. and $\mathrm{C}-8$ showed the $\mathrm{C} 2-\mathrm{C} 3$ double bond in 1
On the basis of all of the foregoing evidence, the structures of compounds 1 and 2 were determined as methyl 2.4-dibromo-5-oxo-2-decenoate and methyl 2.4-dibromo-5-oxo3 -decenoate. respectively.
Cyclonerediol (3). ${ }^{7}$ a sesquiterpenediol, was first reported as a metabolite of the fungus Trichothecium reseum. ${ }^{8.9}$ Subsequent isolations were made from Gibberella fiyikwoi, ${ }^{10}$ Fusorium culmonum, ${ }^{11}$ and Trichoderma koningil as the plant growth regulatory active constituent. ${ }^{12}$ The biosynthetic pathway has been specifically established with the cell-free extracts of $G$.fijikuroi ${ }^{13}$ for cyclonerodiol.

Acknowledgements. Mass spectral data was kindly provided by the Korea Basic Science Institute. This work was supported by the Brain Korea 21 Project in 2003 (F020).

## References and Notes

1. Faulkner. D. J. Nat. Prod. Rep. 2002, 19. 1-47.
2. Pietra. F. Aiat. Prod. Rep. 1997, 14. 453-464.
3. Li. X.: Lee, S. M.: Choi. H. D.: Kang. J. S.: Son. B. W. Chem. Pham. Bull. 2003. 51. 1458-1459.
4. The fungus was cultured for 30 days (static) at $29^{\circ} \mathrm{C}$ in SWS medium: soytone $\left(0.1^{\circ} \%\right.$ ), soluble starch $\left(1.0^{\circ} \%\right)$ and seawater
$\left(100^{\circ} \%\right)$
5. Methyl 2.4-dibromo-5-oxo-3-decenoate (2) was isolated as a yellow oil which showed: $\left[(x]_{\mathrm{D}}-40{ }^{\circ} \mathrm{C}\left(c \quad 0.2 . \mathrm{CHCl}_{3}\right)\right.$ : IR $(\mathrm{KBr})$ : 2952. 2930. 2856. 1743. 1697. 1437. 1265. $1137 \mathrm{~cm}^{-1}$ : UV (MeOH): $203(\log \varepsilon 3.7) .257$ (3.5) nm: LREIMS $n: z 359[\mathrm{M}+\mathrm{H}]^{+}$ (0.4), $357[\mathrm{M}+\mathrm{H}]^{+}(0.9), 355[\mathrm{M}+\mathrm{H}]^{+}(0.4), 333(0.6) .331$ (1.1). 329 (0.7), $302\left[\mathrm{M}-\mathrm{H}-\mathrm{C}_{4} \mathrm{H}_{4}\right]^{-}$(3). $300\left[\mathrm{M}+\mathrm{H}_{4} \mathrm{C}_{4} \mathrm{H}^{+}\right.$(6). 298 $\left[\mathrm{M}+\mathrm{H}-\mathrm{C}_{4} \mathrm{H}_{9}\right]^{+}$(3). $277[\mathrm{M}-\mathrm{Br}]^{+}$(8). $275[\mathrm{M}-\mathrm{Br}]^{+}$(8). 259 [M$\left.\mathrm{C}_{4} \mathrm{H}_{9}-\mathrm{CH}_{2} \mathrm{CO}\right]^{-}(1.9) .257$ [M-C. $\left.\mathrm{H}_{9}-\mathrm{CH}_{2} \mathrm{CO}\right]^{+}(3.3) .255$ [M$\left.\mathrm{C}_{4} \mathrm{H}_{9}-\mathrm{CH}_{2} \mathrm{CO}\right]^{-}$(1.7). 245 (14). 243 (15). 221 (22). 219 (21). 203 (12). 189 (26), 187 (27), 149 (49), 99 (100). 71(76): LRFABMS mz $355[\mathrm{M}-\mathrm{H}]^{-} .357[\mathrm{M}+\mathrm{H}]^{+} .359[\mathrm{M}+\mathrm{H}]^{+}$, HRFABMS m/z 354.9544 (caled for $\mathrm{C}_{11} \mathrm{H}_{1} \because \mathrm{O}_{3}{ }^{24} \mathrm{Br}_{2}, 354.9545$ ), 356.9521 (calcd for $\mathrm{C}_{11} \mathrm{H}_{1}=\mathrm{O}_{3}{ }^{-9} \mathrm{Br}^{81} \mathrm{Br}$. 356.9524 ). 358.9506 (caled for $\mathrm{C}_{11} \mathrm{H}_{1}: \mathrm{O}_{3}$ ${ }^{81} \mathrm{Br}_{2} .358 .9504$ ): See Table 1 for NMR spectral data.
6. Methyl 2.4-dibromo-5-oxo-2-decenoate (1) was isolated as a yellow oil which showed: $[\alpha]_{\mathrm{II}}-10^{\circ} \mathrm{C}\left(c \quad 0.2, \mathrm{CHCl}_{3}\right) ;$ IR $(\mathrm{KBr})$ : $2952,2930,2856.1734,1436.1240,1040.751 \mathrm{~cm}^{-1}, \mathrm{UV}(\mathrm{MeOH})$ 203 ( $\log \varepsilon 3.8$ ), 248 (3.3) nm: LRFABMS mz 355 [M+H]-. 357 $[\mathrm{M}+\mathrm{H}]^{+} .359[\mathrm{M}+\mathrm{H}]^{+}$: HRFABMS $m z 354.9543$ (caled for $\mathrm{C}_{11} \mathrm{H}_{1}: \mathrm{O}_{3}{ }^{79} \mathrm{Br}_{2} .354 .9544$ ). 356.9522 (caled for $\mathrm{C}_{11} \mathrm{H}_{1}-\mathrm{O}_{3}{ }^{79} \mathrm{Br}^{31} \mathrm{Br}$. 356.9524 ). 358.9506 (caled for $\mathrm{C}_{11} \mathrm{H}_{1} 0 \mathrm{O}_{3}{ }^{81} \mathrm{Br}_{2} .358 .9506$ ): See Table 1 for NM spectral data.
7. Cyclonerodiol (3) was isolated as a yellow oil which showed spectral data virtually identical to those reported in the literature ${ }^{12}$ except for the assignment of NMR data. The NMR data were reassigned as follow: 'H-NMR (CDCl: $) \delta .1 .05(3 \mathrm{H} . \mathrm{d} . J=7.0 \mathrm{~Hz}$. $\left.\mathrm{H}_{3}-1\right) .1 .59(1 \mathrm{H} . \mathrm{m} . \mathrm{H}-2) .1 .57 .1 .59$ (each $\left.1 \mathrm{H} . \mathrm{m} . \mathrm{H}_{2}-4\right) .1 .86(2 \mathrm{H}$. m. $\mathrm{H}_{2}-5$ ), 1.83 ( $1 \mathrm{H} . \mathrm{m} . \mathrm{H}-6$ ), 1.49 (2H. t. $J=8.3 \mathrm{~Hz} \mathrm{H}_{2}-8$ ), 2.05 $\left(2 \mathrm{H} . \mathrm{m} . \mathrm{H}_{2}-9\right), 5.12(1 \mathrm{H}, \mathrm{t} . J=7.0 \mathrm{~Hz} . \mathrm{H}-10) .1 .69\left(3 \mathrm{H}, \mathrm{s}, \mathrm{H}_{3}-12\right)$. $1.26\left(3 \mathrm{H}, \mathrm{s}, \mathrm{H}_{3}-13\right), 1.17\left(3 \mathrm{H} . \mathrm{s} . \mathrm{H}_{3}-14\right) .1 .63$ (3H. s, $\left.\mathrm{H}_{3}-15\right)^{13}{ }^{13} \mathrm{C}-$ NMR (CDCl $)^{2}$ : 14.5 (C-1). 44.2 (C-2). 81.3 (C-3). 40.4 (C-4). 24.3 (C-5). 54.2 (C-6). 74.8 (C-7). 40.4 (C-8). 22.6 (C-9). 124.5 (C-10). 131.7 (C-11). 25.7 (C-12). 26.1 (C-13). 25.0 (C-14). 17.7 (C-15).
8. Nozoe. S.; Goi. M.; Morisaki, N. Tetrahedron Lett. 1970. 15. 1293.
9. Evans. R.: Hanson. J. R.: Nyfeler. R. J. Chem. Soc. Perkin Troms. I 1976. 1214.
10. Cross. B. E.: Markwell. R. E.: Stewart. T. C. Tetrahedron 1971. 27. 1663.
11. Hanson. J. R.: Hitchcock. P. B.: Nyteler, R. J. Chem. Soc., Perkin Trans. 1 1975, 1586.
12. Cutler. H. G.: Tacyno. J. M.: Phillips. R. S.: VonTerseh. R. L.: Cole. P. D.: Montemurro. N. Agric. Biol. Chem. 1991. 55. 243.
13. Cane. D. E.: Iyengar. R.: Shiao. M.S. J. An. Chem. Soc. 1981. 103.914

[^0]:    "Recorded in CDCl at $400 \mathrm{MHz}\left({ }^{1} \mathrm{H}\right)$ and $100 \mathrm{MHz}\left({ }^{13} \mathrm{C}\right)$

