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Palladium-Catalyzed Synthesis of 3-Substituted Indoles from 2-Iodoaniline and Aldehydes

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It is known that many indole containing compounds exhibit a wide spectrum of pharmacological and biological activities. Thus, besides conventional named routes, homogeneous transition metal-catalyzed synthetic methods have also been developed as alternative methods for the construction of indole framework because of the wide availability of substrates.1 We have also reported on ruthenium-catalyzed synthesis of indoles via an alkanol group transfer from alkanolamines to anilines (amine exchange reaction)² and ring-opening of epoxides by anilines.³ In connection with this report, palladium-catalyzed direct or step-by-step annulation of 2-haloanilines with carbonyl compounds such as ketones,⁴ 1,3-dikctones⁵ and α -keto esters⁶ also give access to a variety of indoles.7 However, in contrast to the annulation with such carbonyl compounds, there is no precedent on the similar annulation of 2-haloanilines with aldehydes leading to indoles. Under these circumstances, herein we report on palladium-catalyzed synthesis of 3substituted indoles from 2-iodoaniline and aldehydes.

Based on our recent report on palladium-catalyzed synthesis of indoles from 2-iodoaniline (1) and ketones.^{4b} several annulations between 1 and octanal (2a) are attempted (Eq. 1 and Table 1).⁸ Treatment of equimolar amounts of 1 and 2a in the presence of a catalytic amount of $Pd(dba)_2$ (dba – dibenzylideneacetone) and 1,1'-bis(di-*iso*-propylphosphino)-ferrocene (dipf) along with NaO'Bu in DMF at 120 °C for 40 h afforded 3-hexylindole (3a) in 35% yield (25% for 20 h). A slight increase in the yield of 3a was observed with the

molar ratio of [2a]/[1] = 2 (44% yield; 31% for 20 h). The addition of molecular sieves, 4A as dehydrating agent did not give any significant change under the employed conditions and lower reaction temperature (80 °C) resulted in a lower yield of **3a** (24% for 20 h). Among examined solvents (for 20 h) DMF was revealed to be the solvent of choice (23% in MeCN; 12% in dioxane; 10% in toluene).

From the reaction of 1 with an array of aldehydes 2 under the controlled reaction conditions, the corresponding 3substituted indoles were formed in moderate yields, and several representative results are summarized in Table 1. Aldehydes (**2a-2c**) which have aliphatic straight chain were similarly cyclized with 1 irrespective of chain length (runs 1-3). The reaction proceeds likewise with aldehyde 2d which has a branched chain to give the corresponding indole 3d (run 4). With β -phenyl substituted aldehydes (**2e** and **2f**) the product yield was generally higher than that when other examined aldehydes were used (runs 5 and 6). However, the reaction between 1 and 3,3-dimethylbutyraldehyde (**2g**) resulted in a very complicated unidentifiable mixture without the formation of indole **3g** (run 7).

In summary, we have demonstrated that 2-iodoaniline is cyclized with various aldehydes under Pd(dba)₂/1,1'-bis(di*iso*-propylphosphino)ferrocene/NaO'Bu/DMF to give the corresponding 3-substituted indoles in moderate yields. The present reaction is a straightforward synthetic method of

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| Run | Aldehydes 2 | Indoles 3 | Yield (%) |
|-----|--------------------|------------|-----------|
| 1 | CHO 2a | N 3a | 44 |
| 2 | СНО 2b | N 3b | 48 |
| 3 | CHO | | 40 |
| 4 | сно 2d | N 3d | 40 |
| 5 | Ph CHO 2e | N 3e | 58 |
| 6 | Ph CHO | Ph N 3f | 53 |
| 7 | сно 2g | N 3g | 0 |

Table 1. Palladium-catalyzed synthesis of indoles 3 from 2-iodoaniline (1) and aldehydes 2^{μ}

indoles from readily available 2-iodoaniline and aldehydes.

Typical experimental procedure is as follows. A mixture of 1 (0.219 g, 1 mmol), hexanal (**2b**) (0.200 g, 2 mmol), $Pd(dba)_2$ (0.029 g, 0.05 mmol), 1,1'-bis(di-*iso*-propylphosphino)ferrocene (0.025 g, 0.06 mmol), and NaO'Bu (0.211 g, 2.2 mmol) in DMF (10 mL) was placed in a 50 mL pressure vessel. After the system was flushed with argon, the mixture was allowed to react at 120 °C for 40 h. The reaction mixture was filtered through a short silica gel column (ethyl acetate-chloroform mixture) to eliminate inorganic salts. Removal of the solvent left a crude mixture, which was separated by thin layer chromatography (ethyl acetate/hexane mixture = 1/5) to give 3-butylindole (**3b**) (0.083 g, 48%). The compounds prepared by the above procedure were characterized spectros-copically.⁹

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- We have also reported on similar catalytic systems for intramolecular carbon-nitrogen bond forming reactions: Cho, C. S.; Lim, D. K.; Heo, N. H.; Kim, T.-J.; Shim, S. C. Chem. Commun. 2004, 104.
- 9. **3-Butylindole (3a)**.¹⁰ Yellow oil: ¹H NMR (400 MHz, CDCl₃): δ 0.94 (t, J = 7.0 Hz, 311), 1.37-1.46 (m, 211), 1.65-1.73 (m, 211), 2.75 (t, J = 7.5 Hz, 2H), 6.92 (d, J = 2.0 Hz, 1H), 7.08-7.12 (m, 1H), 7.15-7.21 (m, 1H), 7.31 (d, J = 8.0 Hz, 1H), 7.61 (d, J = 8.0 Hz, 1H), 7.80 (br s. 1H); ¹³C NMR (100 MHz, CDCl₃): δ 14.0, 22.7, 24.8, 32.3, 111.0, 117.1, 119.0 (<2), 121.0, 121.8, 127.6, 136.3, **3-Hexylindole (3b)**.¹¹ Pale yellow oil: ¹¹H NMR (400 MHz, CDCl₃) δ 0.89 (t, J = 7.0 Hz, 3H), 1.30-1.42 (m, 6H), 1.67-1.74 (m, 2H), 2.75 (t, J = 7.6 Hz, 2H), 6.96 (d, J = 2.0 Hz, 1H), 7.09-7.12 (m, 1H), 7.16-7.20 (m, 1H), 7.35 (d, J = 8.6 Hz, 1H), 7.61 (d, J = 8.0 Hz, 1H), 7.90 (br s. 1H); ¹³C NMR (100 MHz, CDCl₃) δ 14.5, 23.1, 25.6, 29.7, 30.5, 32.2, 111.4, 117.6, 119.4 (<2), 121.4, 122.2, 128.0, 136.7.

3-Decylindole (3c).¹¹ Yellow oil: ¹H NMR (400 MHz, CDCl₃) δ 0.87-0.90 (m, 3H), 1.26-1.40 (m, 14H), 1.66-1.73 (m, 2H), 2.73 (t, J = 7.5 Hz, 2H), 6.87 (d, J = 2.0 Hz, 1H), 7.08-7.12 (m, 1H), 7.15-7.19 (m, 1H), 7.27 (d, J = 8.5 Hz, 1H), 7.60 (d, J = 8.0 Hz, 1H), 7.69 (br s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 14.1, 22.7, 25.2, 29.4, 29.6, 29.7, 30.2, 31.9, 1H10, 117.1, 119.0, 121.0, 121.7, 127.6, 136.3, several peaks are eclipsed.

3-Isopropylindole (3d).¹² Light yellow oil: ¹H NMR (400 MHz, CDCl₃) δ 1.36 (dd. *J* = 7.0 and 1.5 Hz, 6H). 3.16-3.27 (m. 1H). 6.93 (d. *J* = 2.0 Hz, 1H). 7.09-7.13 (m. 1H). 7.16-7.20 (m. 1H). 7.33 (d. *J* = 8.0 Hz, 1H). 7.66 (d. *J* = 7.5 Hz, 1H). 7.84 (br s, 1H): ¹³C NMR (100 MHz, CDCl₃) δ 23.3, 25.4, 111.1, 119.0, 119.2, 119.4, 121.8, 124.0, 126.7, 136.5.

3-Benzylindole (3e),¹³ Solid (hexane): mp 103-106 °C (lit. 105-106 °C): ¹H NMR (400 MHz, CDCI₃) δ 4.08 (s. 2H), 6.79 (s. 1H), 7.04-7.08 (m. 1H), 7.14-7.20 (m. 2H), 7.23-7.27 (m. 5H), 7.50 (d. J = 8.0 Hz, 1H), 7.71 (br s. 1H); ¹³C NMR (100 MHz, CDCI₃) δ 32.1, 111.6, 116.2, 119.6, 119.8, 122.5, 122.9, 126.4, 127.9, 128.8, 129.2, 136.9, 141.7.

- **3-(1-Phenylethyl)indole (3f).**¹⁴ Viscous brown oil; ¹H NMR (400 MHz, CDCl₃) δ 1.66 (d, J = 7.0 Hz, 3H), 4.33 (q, J = 7.0 Hz, 1H), 6.85 (d, J = 1.5 Hz, 1H), 6.95-6.99 (m. 1H), 7.09-7.16 (m. 2H), 7.21-7.28 (m. 5H), 7.34 (d, J = 8.0 Hz, 1H), 7.69 (br s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 21.4, 35.8, 110.0, 118.1, 118.6, 120.1, 120.2, 120.8, 124.8, 125.7, 126.4, 127.2, 135.5, 145.8.
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[&]quot;Reaction conditions: 1 (1 mmol), 2 (2 mmol), Pd(dba)_2 (0.05 mmol), dipf (0.06 mmol), NaO'Bu (2.2 mmol), DMF (10 mL), 120 $^\circ\!C$, for 40 h.