

Characteristics of Volatile Components from *Magnolia obovata* Thunb. by Different Analysis Methods

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ABSTRACT : This study was carried out to establish an optimum method for identifying the volatile components of *Magnolia obovata* Thunb. using the dynamic headspace (Purge & Trap) and simultaneous distillation and extraction (SDE) method. Between the two different identification analysis, the volatile components were more easily detected in the SDE than the Purge & Trap method. Among the identified volatile components, the 12 compounds were detected to have similar retention times and match quality within the 45 minutes in both identification methods. The maximum values of the major volatile components were detected differently by SDE and (Purge & Trap) method such as α -pinene (3.4, 18.2%), β -pinene (3.5, 10.3%), l-limonene (5.2, 15.4%). These results indicated that the Dynamic Headspace (Purge & Trap) was much more reliable method for identifying the volatile components of *Magnolia obovata* Thunb. as compared to the SDE method.

Key words : *Magnolia obovata*, volatile components, α -pinene, β -pinene, limonene, β -caryophyllene

INTRODUCTION

Magnolia obovata Thunb. is a deciduous ornamental tree and is well grown in the central region of Korean peninsula (Bae, 2001). It was introduced from Japan in 1920s as a name of "hubak" and has been miscalled ever. A deciduous large tree, up to 22 m high has a smooth light rust-ash color and aromatic bark. The creamy-white flower blooms in May and emits a strong odoriferous scent. Petal numbers from 9 to 15, with the outer 3 being pale-green tinged pink on the outside, 6~12 cm wide, 8~10 cm log, which is 3 cm longer than inner part.

The red purple fruits ripen in autumn and it gives off sweet smells just like fresh baked bread. The fruit corns are 10~12 cm long, oblong, apex truncate and rounded base. Its pericarp has some several useful

fatty acids, namely, palmitic (20.4%), steric (1.3%), oleic (53.3%), linoleic (20.4%), and linolenic (2.6%) (Anonymous, 2002). The quantity and quality of fatty acid composition is similar to peanut cultivar "Iksan 17" which contained palmitic (10.1%), steric (2.9%), oleic (53%), linoleic (29.3%), and linolenic acid (1.0%). The bark, consists of rough, thick and tight cylinders, has an aromatic taste and strong pungent smell. It is often used as a drug in Koran folk medicine. Specially, in the southern region of the country, the bark has been used for chest and abdominal pains, indigestion, diarrhea, cough, and sputum (Bae, 2001). Mass spectra are widely used in order to identify the peaks resulting from a chromatographic separation. The most common and available approach for knowing its structural information is the reference mass spectra database

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(Kang *et al.*, 1998). This study was investigated the volatile components from its aromatic flowers with the different two analysis methods, the dynamic headspace (here after called purge trap) and simultaneous distillation and extraction (SDE) method, to find the preferable isolation method in further studying the fragrant of volatile components of natural resource plants.

MATERIALS AND METHODS

Plant materials

Sample was originally collected in the medicinal plant garden of the Division of Genseng & Medicinal Crop, National Institute of Crop Science, RDA during full flowering period in may 2002. Sample were dried under natural conditions and kept 4 °C refrigerator until use.

Oil isolation

Air dried were subjected to simultaneous distillation and extraction method (SDE) for two hours using a cleveger apparatus, The distilled essential oil was collected and dried over dehydrate (sodium sulfate) and stored at 4 °C, the colorless Oil yield was 0. 4% (w/w).

GC/MS analysis

The analysis was carried out using a Hewlett Packard 5890 GC/MS system operating in EI mode at 70 eV, equipped with a DB-5 capillary column (30 m × 0.25 mm; film thickness 0.25 μm). The initial temperature of the oven was 50 °C and subsequently

raised to 230 °C with 3 °C/min. rate. Helium was used as the carrier gas at the rate of 1 ml min⁻¹.

Identification

Chemical constituents was identified by comparing of their relative retention times and mass spectrum with those obtained from authentic samples and Willey library.

RESULTS AND DISCUSSION

The two different analysis methods, the Purge & Trap and simultaneous distillation and extraction (SDE) method were used. The 43 volatile components were identified on the SDE method but 25 compounds were detected on the Purge & Trap instrument (Table 2, 3). Usually, volatile components of plants can be detected during the retention time from 10 to 50 minutes. This result showed slight differences as the α -Pinene and nonadecane were 9.80 and 94.94, respectively, which were not common at retention times observed in plants. A total of 43 volatile components were detected the SDE method where 12 compounds were selected as the major components, which were compared with the Purge & Trap method. The volatile components of the SDE method were mostly monoterpene and sesquiterpene derivatives. α -Pinene (3.4%), β -pinene (3.6%), α -terpineol (2.3%), myrcene (0.7%), limonene (5.2%), camphene (2.1%), bornyl acetate (5.3%), were the monoterpene. β -Caryophyllene (4.4%) and caryophyllene oxide (4.6%) were detected as the sesquiterpene derivatives. The compounds with high content were bornyl acetate (5.3%), limonene (5.2%), bornyl (5.1%), β -Caryophyllene (4.4%), β -asarone (4.6%), and T-muurolol (3.8%), Among these components, limonene which has been well-known for anticancer and β -asarone for its antifungal effects, were the concerned components for further studying. The volatile components identified using of the Purge & Trap were also mostly mono- and sesquiterpene derivatives. The content of most monoterpene was higher more than two times as compared to the content identified using SDE. Especially, the content of α -pinene (18.2%), β -pinene (10.3%), myrcene (2.5%), limonene (15.4%) and camphene (6.8%) were higher

Table 1. Conditions of Gas Chromatography/Mass spectrometry for analyzing the volatile components of *Magnolia ovobata* Thunb.

Instrument	HP 5890 GC/HP 5973 MSD
Column	DB-5 (crossed linked methyl siloxan, 30 m × 0.25 mm × 0.33 μm)
Inject temp	250 °C
Oven temp	50 °C (3 min.) → 8 °C/min. → 230 °C (3 min.)
Auxiliary temp	230 °C
Carrier gas	
(flow rate)	He (0.8 ml/min.)
Split ratio	1:10
Ionization mode	Electric impact (70 eV)
Ms source temp	230 °C

Table 2. Identification of the volatile components in essential oil of *Magnolia ovobata* Thunb. using GC/MS.

Peak no.	Retention time	Volatile components	Peak	Match (%)	Contents (%)
1	9.80	α -pinene	162	97	3.43
2	10.85	camphene	98	98	2.08
3	13.21	β -pinene	172	97	3.55
4	14.49	mycrene	3	97	0.06
5	17.12	benzene	48	96	1.01
6	17.75	limonene	247	95	5.21
7	20.85	<i>trans</i> -sabinene hydrate	59	98	1.25
8	23.52	<i>cis</i> -sabinene	5	98	0.16
9	24.00	linalool	132	96	2.74
10	25.21	2-cyclohexen	1	97	0.03
11	27.03	l-camphene	113	98	2.37
12	29.18	1-borneol	243	91	5.14
13	29.95	terpinen-4-ol	92	97	1.94
14	31.06	β -terpineol	106	91	2.25
15	35.96	gerany acetate	80	83	1.69
16	36.40	chavicol	150	98	3.09
17	38.33	bornyl acetate	251	99	5.33
18	44.78	geranyl acetate	26	91	0.55
19	47.00	β -caryophyllene	205	99	4.36
20	49.00	β -selinene	88	98	1.86
21	51.57	α -selinene	73	98	1.54
22	51.90	oxetane	6	91	0.13
23	53.37	calacorene	53	46	1.13
24	54.31	caryophyllene oxide	14	80	0.29
25	56.85	humulene epoxide	221	81	4.68
26	58.21	β -asarone	45	81	0.93
27	59.54	bicyclo(4.4.0)dec-1-en	219	99	4.61
28	60.12	t-muurolol	67	95	1.41
29	61.08	5-oxatric	181	96	3.79
30	62.27	nonadecene	678	70	1.40
31	73.67	palmitic acid	14	97	0.28
32	77.05	eicosane	38	99	0.77
33	78.38	<i>trans</i> -2-nonadecane	20	98	0.38
34	79.78	10-heneicosane	17	83	0.32
35	81.68	11-hydroxy-7.8.9.10-tetrahydrobebzo(b)xa	9	99	0.19
36	82.22	heneicosane	15	83	0.31
37	83.11	linoleic acid	110	97	2.32
38	84.60	magnolol	18	96	0.36
39	86.90	N-docosane	9	95	0.16
40	87.87	<i>trans</i> -2-nonadecane	13	97	0.27
41	92.28	<i>cis</i> -2-nonadecane	4	91	0.07
42	92.81	<i>cis</i> -2-nonadecane	6	87	0.11
43	94.94	nonadecane	135	91	2.85

Table 3. Identification of the volatile components from dried flowers of *Magnolia ovobata* Thunb. using the Purge & Trap.

Peak no.	Retention time	Components	Peak	Match (%)	Contents (%)
1	7.18	hexanal	39	90	1.06
2	9.08	flomamidinium acetate	6	74	0.14
3	10.04	3-hexen-1-ol	17	50	0.42
4	13.98	α -phellandrene	28	90	0.78
5	14.39	α -pinene	593	97	18.17
6	15.23	camphene	226	98	6.82
7	17.18	β -pinene	339	97	10.33
8	18.53	myrcene	86	95	2.54
9	19.22	α -phellandrene	17	86	0.41
10	20.11	α -terpinene	22	98	0.58
11	20.70	benzene, 1-methy1-3-(1-methylethy1)	137	96	4.17
12	21.08	1-limonene	502	98	15.40
13	21.22	1,8 cineol	59	95	1.70
14	23.25	\sqrt -terpinene	29	96	0.86
15	25.43	benzene, 1-methy1-4-[1-methylethenyl]	22	98	0.63
16	26.57	α -terpinolene	80	95	2.42
17	26.80	cyclopropyl carbind	9	25	0.23
18	29.34	camphore	33	98	0.92
19	31.09	endo-Borneol	79	90	2.38
20	36.73	carvol	45	95	1.31
21	39.81	bornyl acetate	40	90	1.19
22	45.75	propanoic acid	26	72	0.72
23	48.68	β -caryophyllene	103	99	3.13
24	49.88	<i>trans</i> - α -bergamotene	18	90	0.50
25	50.82	α -Humulene	24	97	0.71
26	51.36	<i>trans</i> - β -farnesene	18	91	0.47
27	55.35	delta-cardinene	16	95	0.42

Table 4. Differences of the major volatile components content between the two analysis methods in *Magnolia ovobata* Thunb.

Volatile Components	Purge & Trap			SDE**		
	*R.T	Quality (%)	Content (%)	R.T	Quality (%)	Content (%)
α -pinene	14.39	97	18.17	9.80	97	3.43
camphene	15.23	98	6.82	10.85	98	2.18
β -pinene	17.18	97	10.33	13.21	97	3.55
myrcene	18.53	95	2.54	14.49	97	0.67
benzene, 1-methylethyl	20.70	96	4.17	17.12	96	1.01
1-limonene	21.08	98	15.40	17.75	95	5.21
1,8 cineol	21.22	95	1.70	—***	—	—
α -terpinolene	26.57	95	2.42	31.06	91	2.25
borneol	31.09	90	2.38	29.18	91	5.14
carvol	36.73	95	1.31	—	—	—
bornyl acetate	39.81	90	1.19	38.33	99	5.33
β -caryophyllene	48.68	99	3.13	47.00	99	4.36

* Retention time. ** Steam distillation method. *** Non-detected.

several times in the Purge & Trap method. However, the content of bornyl acetate (1.2%), was lower three times, comparing the sesquiterpene content of the Purge & Trap to SDE method, β -caryophyllene (3.1%) and α -humulene content was less in the Purge & Trap (SDE). The content of α -Pinene (18.2%), limonene (15.4%), β -pinene (10.3%), camphene (6.8%), β -Caryophyllene (3.1%) and myrcene (2.5%) were conspicuously higher in the Purge & Trap. The same components were identified with both

Purge & Trap and SDE method were α -Pinene, β -pinene, α -terpinene, myrcene, limonene, camphore, bornylacetate and β -Caryophyllene. However, the magnolol compound in bark which is well-known for anti-germs effect (Bae, 2001), was also detected at the retention time 86.90 in the SDE. In the isolation spectrum of chromatography, the separation was shown more clearly and exactly in the Purge & Trap than the SDE method. Meanwhile, the volatile components were easily isolated in the SDE method

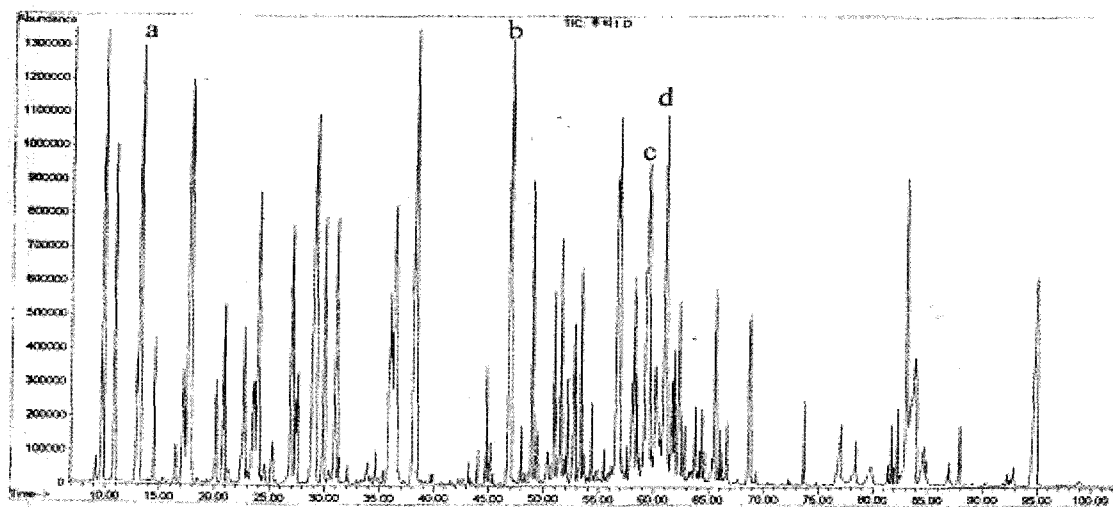


Fig. 1. Total ion chromatogram spectrum of GC/MS in essential oil of *Magnolia ovobata* thunb. (a: β -pinene, b: β -caryophyllene, c: β -asarone, d:t-muurolol).

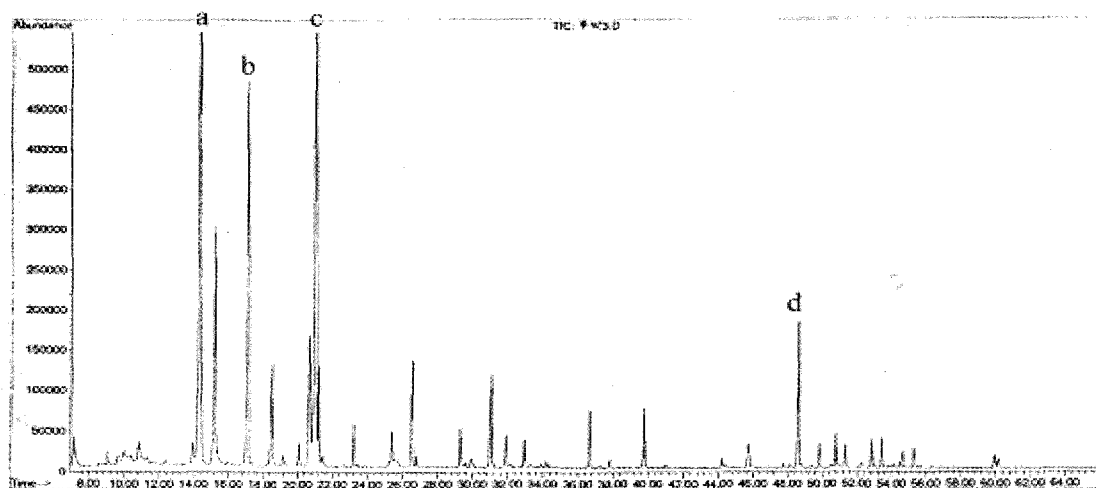


Fig. 2. Total ion chromatogram spectrum of Dynamic headspace (Purge & Trap) of *Magnolia ovobata* thunb. (a: β -pinene, b: β -pinene, c:1-limonene, d: β -caryophyllene).

than the Purge & Trap method in *Magnolia obovata* Thunb. (Fig. 1 & 2). This result was opposite to the study of Lee *et al.* (1997) where the Purge & Trap was found as an adequate analysis method for low polar volatile components than the SDE method and other solvent extraction methods.

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