

## Synthesis of Hybrid Sol Based on $ZrO_2$ - $SiO_2$ System and their Coating Properties

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### ABSTRACT

Organic-inorganic hybrid sol based on  $ZrO_2$ - $SiO_2$  system was prepared by sol-gel process. Firstly,  $ZrO_2$  non-aqueous precursor sol was synthesized and then organosilane compounds which include epoxy silane (GPTS; 3-glycidoxypropyl tri-methoxysilane) and acryl silane (ACS; (3-(tri-methoxysilyl)propylmethacrylate)) were added to  $ZrO_2$  precursor sol for hybridization. Finally, commercial silica sol was added to improve the mechanical properties. Synthesized organic-inorganic Zr-hybrid sol was coated on polycarbonate substrate for enhancing its mechanical properties, especially hardness. Vicker's hardness of polycarbonate substrate was increased from 13.6 to 17.8 MPa and its pencil hardness was increased from 2 to 7 H, respectively, after coating and drying at 100°C for 30 min.

**Key words :** Hybrid sol, Non-Aqueous  $ZrO_2$  precursor sol, Non-Scratching, Polycarbonate

### 1. Introduction

Since Schmidt,<sup>1-4)</sup> sol-gel technology has been widely used for synthesis of hybrid materials. Most of these papers and patents<sup>5-8)</sup> have been focused on  $SiO_2$ -,  $Al_2O_3$ -, and  $TiO_2$ -organosilane hybrid fabrication. Anyway, these new materials offer many opportunities for applications in different fields such as optical devices, catalysis, electronics etc on.

Recently, thermoplastic substrates such as polycarbonate sheet and film are used to the construction field for shielding window and electronic component because of their lightness and economical needs. While polycarbonate resin is a tough transparent material, it is generally desirable to improve the abrasion resistance and weathering resistance. Also, their mechanical properties, especially, hardness is not so good comparing to inorganic materials. Accordingly, many efforts to improve these properties have been done by coating using the above mentioned sol-gel hybrid compositions.

Stabilized- $ZrO_2$  ceramics, however, is well-known material which has been widely used in ceramic field due to its superior mechanical properties. So, in this present work, we have focused on  $ZrO_2$  oxide to fabricate hybrid sol and enhance the mechanical properties of the polycarbonate substrate and investigated coating properties of its hybrid film.

### 2. Experimental Procedures

#### 2.1. Non-Aqueous $ZrO_2$ Precursor

At first, non-aqueous  $ZrO_2$  precursor has been prepared before hybridizing with organic silane compounds. Non-aqueous  $ZrO_2$  sol fabrication is based on Yoldas process,<sup>9)</sup> which have been widely used and very easy method to prepare aqueous oxide sol and bulk ceramic, for example,  $Al_2O_3$ . In Yoldas process, water as solvent and acids for peptizing precipitated sol were used without controlling water contents. Comparing to Yoldas process, here, non-aqueous  $ZrO_2$  sol uses methanol as solvent replacing of water and excess acid for peptizing the precipitates in solvent. By using this process, transparent  $ZrO_2$  sol without precipitates was obtained by refluxing at 50°C for 24 h, in which  $Zr(O^iPr)_4$  concentration range from 0.71 mole/L(Zr-10) to 2.12 mole/L(Zr-30) in methanol solvent adding nitric acid (63 wt%). Subsequently, organic silane compound are added to this sol for hybridization as mentioned in the next section.

#### 2.2. Hybridization of $ZrO_2$ -Based Sol with Organic Silane

Organic silane used in this works for hybridization with the above mentioned  $ZrO_2$  non-aqueous sol are epoxy silane (GPTS; 3-glycidoxypropyl tri-methoxysilane) and acryl silane (ACS; (3-(tri-methoxysilyl)propylmethacrylate)). And the  $Zr(O^iPr)_4$ /silane mole ratio is 1.99, 2.98, 3.38, 3.98, 5.98, which sols are named as Zr-10, Zr-15, Zr-17, Zr-20, and Zr-30, respectively. Here, GPTS/ACS mole ratio is fixed as 2.2. GPTS and ACS is added to  $ZrO_2$  precursor sol in turns and refluxed at 50°C for 24 h, respectively. Finally,

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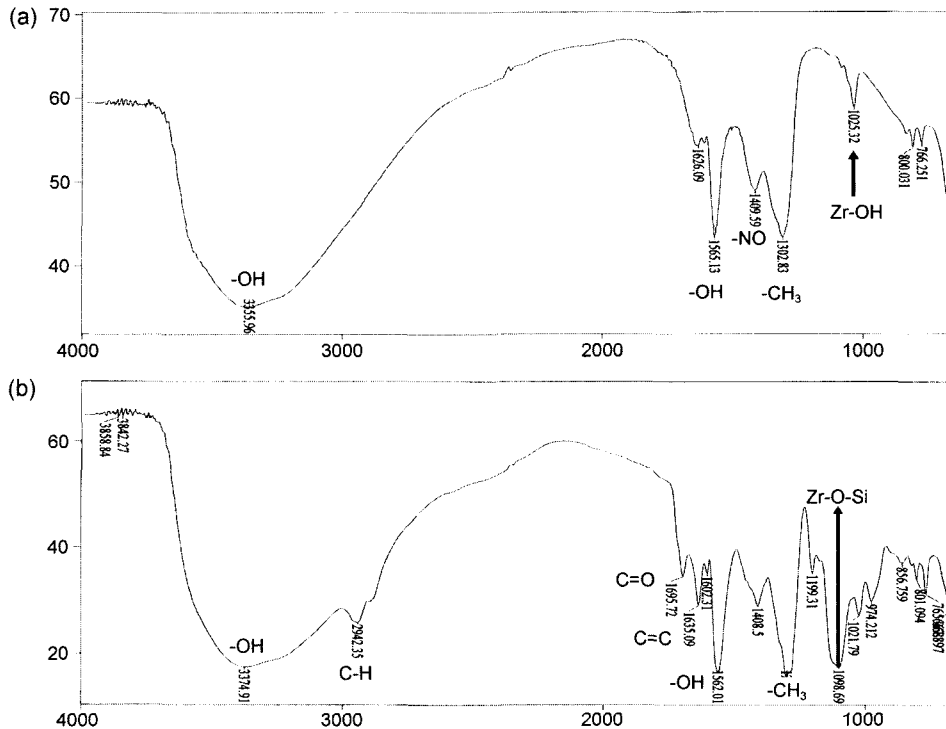


Fig. 1. FT-IR spectra of (a) Zr-15; non-aqueous  $ZrO_2$  precursor sol and (b) Zr-15+ silane; hybrid coating sol.

commercial silica sol (Ludox, HS-40, Dupont Co.) was added to the above hybrid sol by 10% volume to improve the mechanical properties. Obtained sol is transparent and viscous liquid and coated on polycarbonate substrate by rolling method and dried at  $100^\circ C$  for 30 min in drying oven.

FT-IR (Bomen MB-104, CANADA) is used to confirm the hybridization of  $ZrO_2$  sol with silane and hardness measurement of coating film on polycarbonate was done about Vickers hardness (Fv-7e, Future-Tech Corp., load : 0.3/Kg, \*Dwell time : 15 sec) and Pencils hardness (100 g load).

\*Dwell time : Current load pressurization time.

### 3. Results and Discussion

Fig. 1 shows the FR-IR spectra of non-aqueous  $ZrO_2$  sol and Zr-15 hybrid sol. In Fig. 1(a), absorption bands around  $3355\text{ cm}^{-1}$  and  $1565\text{ cm}^{-1}$  are assigned to OH vibration of methanol and water and their first overtone absorption peak, respectively. Absorption band at  $1302\text{ cm}^{-1}$  is attributed to  $CH_3$  stretching and  $1409\text{ cm}^{-1}$  and  $1025\text{ cm}^{-1}$  are due to NO and ZrOH, respectively. In fig. (b), absorption band of  $2942\text{ cm}^{-1}$  is C-H stretching of  $CH_3$  and  $CH_2$ - group and  $1695\text{ cm}^{-1}$  and  $1635\text{ cm}^{-1}$  is assigned to C=O and C=C of stretching of an ester carbonyl in silane compound, respectively.<sup>10</sup> Especially, it is shown that  $1098\text{ cm}^{-1}$  new absorption peak seems to be attributed to Zr-O-Si metaloxane bond. Generally, it is well-known that, Si-O, that is, metaloxane bonds absorption peak appears around in the range  $1000 - 1100\text{ cm}^{-1}$  in case of  $SiO_2$  sol derived by sol-gel process. Accordingly, it

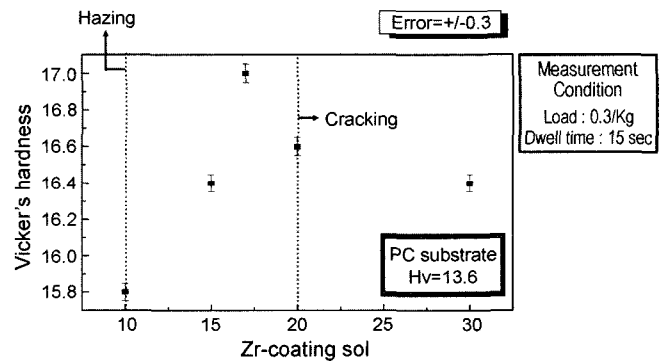
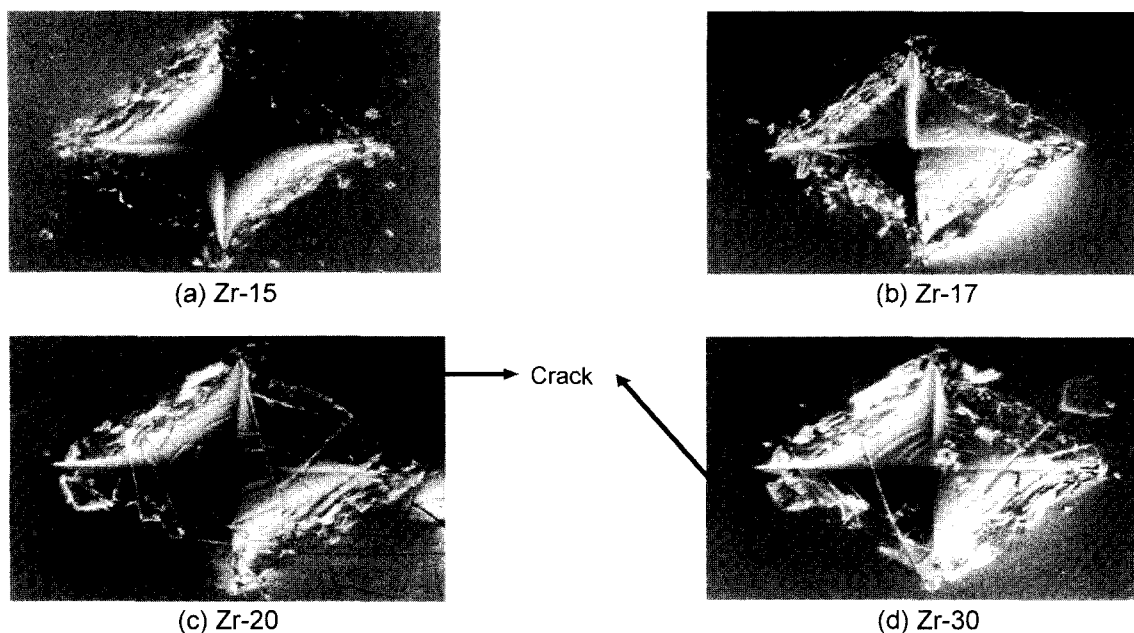


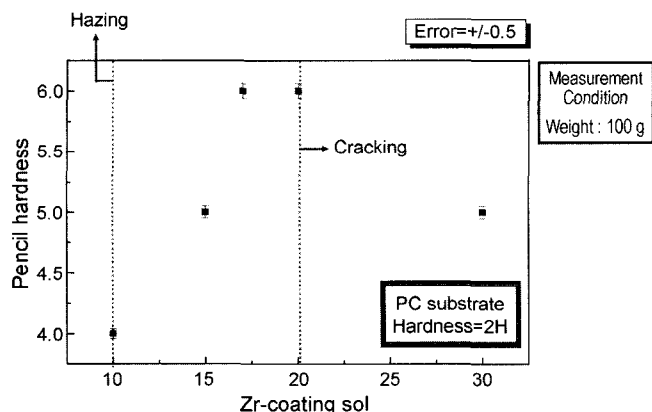
Fig. 2. Vicker's hardness of each Zr-hybrid coating film on PC substrate.

was confirmed that hybrid sol successively was prepared by this process.

Fig. 2 shows the results of measured Vicker's hardness of coating film on substrate comparing to bare polycarbonate which value is  $13.6\text{ MPa}$  and Fig. 3 shows their resulting optical photographs. As shown in Fig. 2, Vicker's hardness value of each coating film is  $15.8$  (Zr-10),  $16.4$  (Zr-15),  $17.2$  (Zr-17),  $16.6$  (Zr-20),  $16.4\text{ MPa}$  (Zr-30), respectively, and the value was averaged by 10 time measurements. However, in Fig. 3, cracks were observed in Zr-20 and Zr-30 coating film before and after the measurement. In case of lower  $ZrO_2$  concentration in hybrid sol (Zr-10;  $Zr(O^iPr)_4$ /Silane mole ratio is 1.99), hazing of coating film on the polycarbonate substrate occurred after drying at  $100^\circ C$  for 30 min. The other compositions besides Zr-10 in this



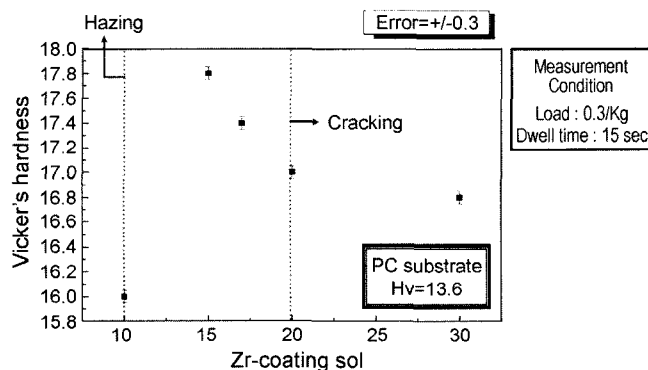
**Fig. 3.** Optical photographs after Vicker's hardness measurement. (a) Zr-15, (b) Zr-17, (c) Zr-20, and (d) Zr-30 hybrid coating films.



**Fig. 4.** Pencil hardness of each Zr-hybrid coating film on PC substrate.

work, became a transparent coating film on the substrate after drying. But as shown in Figs. 2 and 3, cracks appeared at the  $Zr(O^iPr)_4$ /silane mole ratio compositions over 3.98 (Zr-20 and Zr-30).

As soon as sol was coated on the substrate, hazing did not occur. But it appeared just after dried neither at room temperature nor at  $100^\circ C$ . At present, its not clear whether it is attributed to adhesion problem between coated film and substrate or bonding problem between  $ZrO_2$  precursor and silane. Anyway, its vickers hardness (15.8 MPa) shows the lower value comparing to Zr-30 (16.4 MPa) which has many cracks in coating film, even if the value was higher than bare polycarbonate substrate (13.6 MPa). Therefore, it is thought that hazing is probably attributed to the adhesion problem between coated film and substrate.

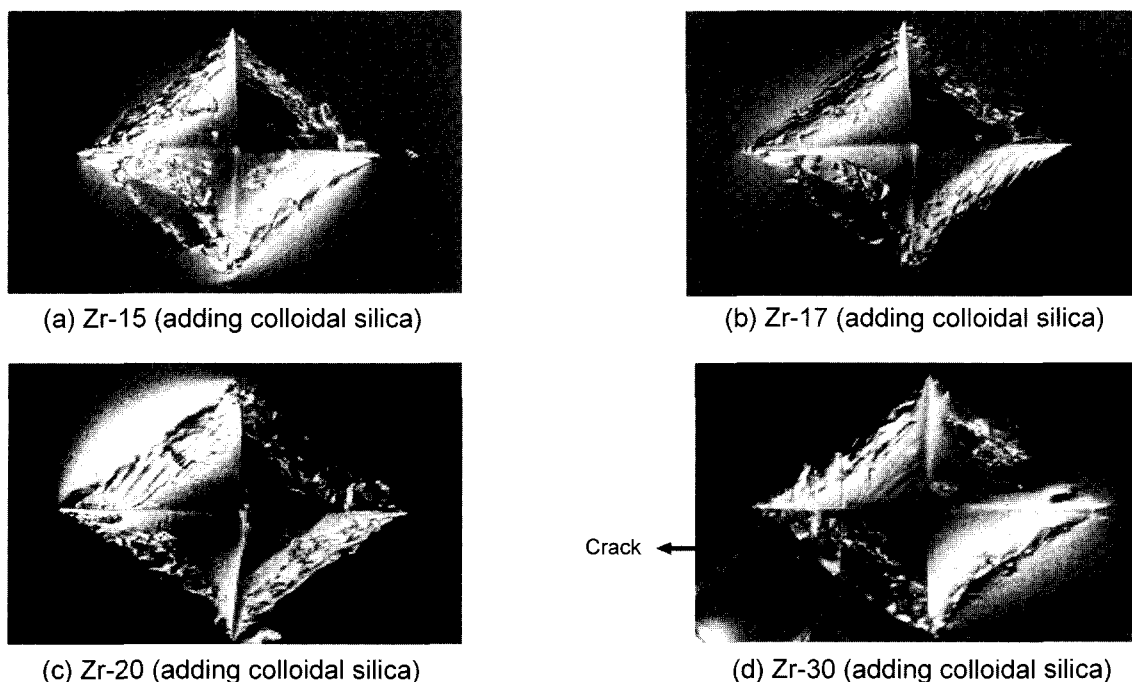


**Fig. 5.** Vicker's hardness of each Zr-hybrid coating film on PC substrate with adding 10 vol% colloidal silica (HS-40).

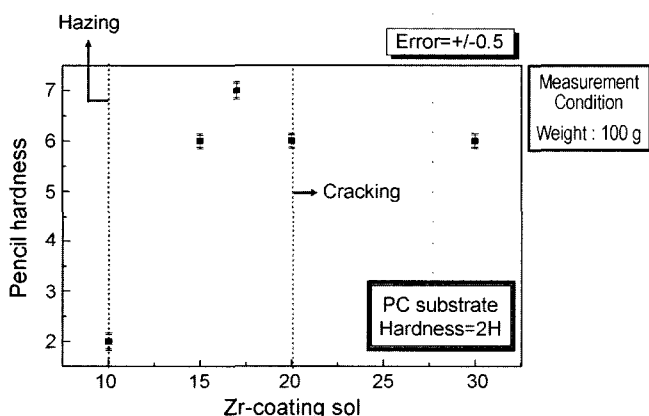
Cracking in Zr-20 and Zr-30 coating film seems to be caused by  $ZrO_2$  content increase, that is, oxide content portion added up because  $ZrO_2$  has no flexibility.

Fig. 5 shows the result of measurement of Vicker's hardness of coating film, which included the additional silica sol (HS-40, Dupont Co). In this case, vickers hardness values of each coating film increased higher than non-adding hybrid sols, which value were 16.0 (Zr-10), 17.8(Zr-15), 17.4(Zr-17), 17.0 (Zr-20), 16.8 MPa(Zr-30), respectively.

Fig. 6 shows their optical photographs after Vickers measurement. As shown in the Fig. 5, by addition of silica sol to the Zr-hybrid sol, hazing was not avoid, but in the Fig. 6, cracking in the coating film was prohibited by silica sol addition, especially, in Zr-20 coating film and also pencil hardness became higher than non-adding Zr-hybrid sols value, as shown in Fig. 7. Its attributed to colloidal silica particles pin effect to block crack propagation.



**Fig. 6.** Optical photographs after Vicker's hardness measurement. (a) Zr-15, (b) Zr-17, (c) Zr-20, and (d) Zr-30 hybrid coating films with adding 10 vol% colloidal silica (HS-40).



**Fig. 7.** Pencil hardness of each Zr-hybrid coating film on PC substrate with adding 10 vol% colloidal silica (HS-40).

#### 4. Conclusions

$ZrO_2-SiO_2$  hybrid coating sol successively was prepared by sol-gel process and coated on the polycarbonate substrate by rolling method. Then, Vicker's hardness of polycarbonate on substrate was increased from 13.6 to 17.7 MPa and its pencil hardness was increased from 2 to 7 H, respectively, after coating and drying at 100°C for 30 min.

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