# Solution-phase Synthesis and Preliminary Evaluation of 1,6,8-Trisubstituted Tetrahydro-2H-pyrazino[1,2-a]pyrimidin-4,7-dione Derivatives as a NF-kB Inhibitor 

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Key Words : Bicylic $\beta$-tım mimetics. NF-kB inlubitor

To develop a potent fom of NF-kB inhibitors. $\beta$-turn peptidomimetics with a new scaffold (1) ${ }^{1-6}$ as shown in Figure 1 were designed.

Previously. ${ }^{7}$ we reported the synthesis and structureactivity relationships of new 1.6.8-trisubstituted tetrahydro2 H -pyrazino[1.2-a] pyrimidin-4.7-dione derivatives to find the correlation between the polarity of the C-6 substituent and the inhibitory activity: However, we failed to introduce the carboxylic acid group at the C-6 position by solid phase method
In this study to investigate the effect of the carboxylic acid moiety at $\mathrm{C}-6$ position of the bicyclic ring. bicyclic $\beta$-turn mimetics 7a-g were synthesized using solution phase. and their NF-kB inhibitory activities are discussed.

## Chemistry

The $\beta$-tum mimetics were prepared from solution-phase synthesis. according to our previous solid-phase synthetic protocol. ${ }^{7}$ Benzaldehyde (1) was reacted with ammoacetaldehỵde dimethyl acetal. and subsequently treatment with sodium borohydride in MeOH gave the secondary amine 2. which was then coupled with the cbz-Asp(OBut)-OH with HOBT/DIC in DMF to give 3. Deprotection of the Cbz group 3 by catalytic hydrogenation in EtOH gave the amine compound. which was then coupled with Cbz - $\beta$-alanine to afford 4 . After cleavage of the Cbz group of 4 by catalytic hydrogenation the resulting compound was treated with the
$\mathrm{R}_{1}=\mathrm{OCH}_{3},-\mathrm{H},-\mathrm{F}$
$\mathrm{R}_{2}=$-Isopropyl, isobutyl
$\mathrm{R}_{3}=$ Benzyl derivatives
$\mathrm{X}=\mathrm{O}, \mathrm{S}, \mathrm{N}$


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p-nitrophenyl chloroformate in the presence of DIEA to produce 5 . The urea type compounds 6a-g were accomplished by treatment of compound 5 with the corresponding amines.

Cleavage of the acetal of $6 \mathbf{a}-\mathbf{g}$ followed by stereoselective tandem acyliminium cyclization by treatment with formic acid at room temperature was carried out to give the $6.6-$ bicyclic $\beta$-turn mimetics $7 \mathrm{a}-\mathrm{g}$. All final products were purified by preparative TLC (silica gel) to afford the pure products.

## Biological studies

All new 1.6.8-trisubstituted tetrahydro-2 $H$-pyrazino[1.2-a]pyrimidin-4.7-dione derivatives $7 \mathrm{a}-\mathrm{g}$ subjected to preliminary in vitro NF-kB inhibitory activity screening ${ }^{8}$ exhibited different biological properties. depending on the kind of substituents at $\mathrm{N}-1$ position of the main bicyclic system. According to the results assembled in Figure 2. compounds 7d and 7e, which contain the fluorobenzyl groups at $\mathrm{N}-1$ position, exhibited slightly better activity than their methoxybenzyl group 7b and benzyl group 7a. Tested at a concentration of $10 \mu \mathrm{M}$. both compounds showed a $40 \%$ inhibition against the target NFkB 549. The compounds $7 \mathrm{a}-\mathrm{g}$. having a carboxylic acid group at C-6 position. showed slight differences to their isobutyl group 7a*-g*.

We found that introduction of carboxylic acid at the C-6 position of bicyclic $\beta$-turn mimetics did not affect biological activity compared with the alkyl group. It is of interest to investgate the fluoro substituent and this is in progress.

## Summary

The solution-phase synthesis of a new series of 1.6.8trisubstituted tetrahydro-2H-pyrazino[1.2-a]pyrimidin-4.7diones as bicyclic $\beta$-turn mimetics is described herewith. Their NF-kB inhibitory activities were tested and the effect of substituents of the bicyclic ring was investigated. Among these compounds. 7d and 7e showed the most potent activity.

## Experimental Part

Melting point (mp): Thomas Hoover apparatus, uncorrected. ${ }^{l}$ H NMR spectra: Varian Gemini 300 spectrometer. tetra-


Scheme 1. i) Aminoacetaldehyde dimethyl acetal, toluene; ii) $\mathrm{NaBH}_{\perp}$, MeOH; iiii) Cbz -ASP (OBut)-OH, 1,3 -diisopropylcarbodiimide, DMF: iv) $10 \% \mathrm{Pd} / \mathrm{C}, \mathrm{THF}: \mathrm{EtOH}=1: 1 ;$ v) Cbz-b-Ala-OH, HOBT, DMF; $10 \% \mathrm{Pd} / \mathrm{C}, \mathrm{THF}: \mathrm{EtOH}=1: 1$; vii) $p$-Nitrophenyl chloroformate, NA-disopropy lethyl amine, $\mathrm{CH}_{2} \mathrm{Cl}: \mathrm{THF}=1: \mathrm{l}$; viii) Coresponding amines, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ : ix) Formic acid

(-) control: None (+) control: phorbol myristate acetate. FkB A549@10 10 M

$7 \mathrm{a}^{*}-\mathrm{g}^{*}$


Figure 2. In witro NFkB A549 inhibitory activity of $7 \mathbf{a - g}$ and $7 \mathbf{a}^{*}-\mathrm{g}_{\mathrm{g}}{ }^{*}$. 8
methylsilane (TMS), as an internal standard. The mass spectrometry system was based on a HP5989A MS Engine (Palo Alto. CA. USA). IR spectra: Perkin Elmer 16F-PC FTIR.
N -(2,2-Dimethoxyethyl)benzylamine (2). To a stirred solution of aminoacetaldehyde dimethyl acetal ( 48.8 mmol . 5 mL ) in dry toluene ( 60 mL ) was added dropwise benzaldehyde ( 1.48 .8 mmol .4 .9 mL ) and the reaction mixture was stirred for 3 h at $80^{\circ} \mathrm{C}$. Evaporation of the solvent in vorou gave a crude residue, which was dissolved with MeOH $(50 \mathrm{~mL})$. To the resulting solution was added dropwise $\mathrm{NaBH}_{+}(51.8$ nmol. 2.0 g$)$ at $0^{\circ} \mathrm{C}$ and was stirred for 24 h at room temperature. The mixture was diluted with $\mathrm{H}_{2} \mathrm{O}$ (40 $\mathrm{mL}), 1 N-\mathrm{HCl}$ and ethyl acetate ( 100 mL ). The organic layer was dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, concentrated, and the
resulting residue was purified by silica gel column clromatography with EtOAc/hexane ( $1: 1.5$ ) to give $2(8.8 \mathrm{~g} .92 \%)$ as a pale yellow oil. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 2.76(2 \mathrm{H}, \mathrm{d}, J=5.4$ $\mathrm{Hz}), 3.37(6 \mathrm{H} . \mathrm{s}), 3.82(2 \mathrm{H}, \mathrm{s}) .4 .50(1 \mathrm{H}, \mathrm{t}, J=5.4 \mathrm{~Hz}) .7 .37$ ( $5 \mathrm{H} . \mathrm{m}$ ).
$N$-Benzyl- $N$-(2,2-dimethoxyethyl)-3-benzyloxycarbonylaminosuccinamic acid $\boldsymbol{t}$-butyl ester (3). A solution of Cbz-Asp(OBut)-OH ( 5.6 mmmol .1 .80 g ). HOBT ( 5.6 mmol .0 .86 g). DIC ( 5.6 mmol .0 .9 mL ) in dry-DMF ( 20 mL ) was added to the solution of $2(5.1 \mathrm{mmol} .1 .0 \mathrm{~g})$ in dry-DMF $(20 \mathrm{~mL})$ at room temperature and was stirred for 12 h at same temperature. The reaction mixture was poured into cold water and extracted with ethyl acetate. The organic layer was successively' washed with water and dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$. Evaporation of the solvent in vocho gave a crude residue.

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which was purified by silica gel column chromatography with EtOAc/hexane ( $1: 4$ ) to give $3(2.1 \mathrm{~g}, 70 \%)$ as a pale yellow oil. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 0.85(3 \mathrm{H}$. dd. $J=6.6$ and $13.8 \mathrm{~Hz}) .0 .99(3 \mathrm{H}$. dd. $J=6.6$ and 16.5 Hz$) .1 .32(1 \mathrm{H} . \mathrm{m})$. $1.68(2 \mathrm{H} . \mathrm{m}) .3 .37(6 \mathrm{H} . \mathrm{m}) .3 .56(2 \mathrm{H} . \mathrm{m}) .4 .57(1 \mathrm{H} . \mathrm{t} . J=5.2$ $\mathrm{Hz}) .4 .76(2 \mathrm{H}, ~ \mathrm{~s}) .4 .94(1 \mathrm{H} . \mathrm{m}) .5 .10(2 \mathrm{H}, \mathrm{d} . J=7.5 \mathrm{~Hz})$. 7.27 ( $10 \mathrm{H} . \mathrm{m}$ ).

N -Benzyl- N -(2,2-dimethoxyethyl)-3-(3-benzyloxycarbonylamino)propionylaminosuccinamic acid $t$-butyl ester (4). Compound 3 ( 13.4 mmol. 6.7 g ) and 1.5 g of $\mathrm{Pd} / \mathrm{C}$ ( $10 \%$ ) were dissolved in THF and was hydrogenated at 50 psi for 2 h . The solution was filtered through celite and was evaporated to give a residue, which was used without further purification. A solution of $\mathrm{Cbz}-\beta-\mathrm{Ala}-\mathrm{OH}(20.0 \mathrm{mmol} .4 .46$ $\mathrm{g})$, HOBT ( 20.0 mmol .3 .06 g ) and DIC ( 20.0 mmol .3 .13 mL ) in dry-DMF ( 20 mL ) was added to the above solution in dry-DMF ( 20 mL ) at room temperature and was stirred for 12 h at same temperature. The reaction mixture was poured into cold water and extracted with ethyl acetate. The organic layer was successively washed with water and dried over anhy drous $\mathrm{Na}_{2} \mathrm{SO}_{4}$. Evaporation of the solvent in vacuo gave a crude residue. which was purified by silica gel column chromatography with EtOAc/hexane (1:4) to give $4(6.4 \mathrm{~g}$. $83 \%$ ) as a pale yellow oil. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 1.40(9 \mathrm{H}, \mathrm{d} . J$ $=4.5 \mathrm{~Hz}) .1 .64(2 \mathrm{H} . \mathrm{m}), 2.41(2 \mathrm{H}, \mathrm{m}) .3 .36(6 \mathrm{H} . \mathrm{m}), 3.45$ $(2 \mathrm{H}, \mathrm{m}) .3 .57(2 \mathrm{H}, \mathrm{m}) .3 .83(1 \mathrm{H}, \mathrm{m}) .4 .50(2 \mathrm{H} . \mathrm{m}) .4 .99(1 \mathrm{H}$. m). $5.08(2 \mathrm{H} . \mathrm{s}) .7 .24(10 \mathrm{H} . \mathrm{m})$.

N -Benzyl- N -(2,2-dimethoxyethyl)-3-(p-nitrophenoxycarbonylamino)propionylaminosuccinamic acid $t$-butyl ester (5). Compound 4 ( 11.2 mmol. 6.4 g ) and 1.5 g of $\mathrm{Pd} / \mathrm{C}$ ( $10 \%$ ) were dissolved in THF and was hydrogenated at 50 psi for 2 h . The solution was filtered through celite and was evaporated to give a residue. which was used without further purification. To above solution of triethylamine ( 20.6 mmol . 3.6 mL ) in dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}(60 \mathrm{~mL})$ was added slowly $p$ nitrophenyl chloroformate ( 20.6 mmol .4 .3 g ) at $0^{\circ} \mathrm{C}$ and was stirred for 1 h at same temperature. The mixture was diluted with $\mathrm{H}_{3} \mathrm{O}(30 \mathrm{~mL}), \mathrm{CH}_{2} \mathrm{Cl}_{-}(50 \mathrm{~mL})$, and the organic layer was dried over anhyydrous $\mathrm{MgSO}_{4}$. The organic solvent was concentrated in vacto to give a residue. which was used without further purification.
N -Benzyl- N -(2,2-dimethoxyethyl)-3-(3-benzylureido)propionylaminosuccinamic acid $t$-butyl ester (6a). To the solution of $5(0.7 \mathrm{mmol} .0 .4 \mathrm{~g})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$ was added benzylamine ( $2.1 \mathrm{mmol}, 0.23 \mathrm{~mL}$ ) and was stirred for 2 h at room temperature. The reaction mixture was neutralized with $1 N-\mathrm{HCl}$. diluted with water ( 20 mL ) and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30$ mL ), and washed with brine. The organic layer was dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$. Evaporation of the organic solvent in vocto gave a crude residue. which was purified by silica gel column chromatography with ethyl actate to give 6 a $(0.16 \mathrm{~g} .40 \%)$ as a pale yellow oil. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 1.40$ $(9 \mathrm{H} . \mathrm{d}, J=4.5 \mathrm{~Hz}) .1 .61(2 \mathrm{H}, \mathrm{m}), 2.36(2 \mathrm{H}, \mathrm{m}), 3.30(6 \mathrm{H}$. m). $3.54(4 \mathrm{H} . \mathrm{m}) .4 .32(2 \mathrm{H}, \mathrm{m}) .4 .50(1 \mathrm{H} . \mathrm{m}) .4 .93(2 \mathrm{H} . \mathrm{m})$. 5.41 ( $1 \mathrm{H} . \mathrm{q} . ~ J=8.1 \mathrm{~Hz}$ ). $7.24(10 \mathrm{H} . \mathrm{m})$.

The synthesis of compounds $6 \mathrm{~b}-\mathrm{g}$ from 5 was carried out by the same procedure as described for the preparation of $6 \mathbf{a}$.

6b: Yield $40 \% .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 1.40(9 \mathrm{H} . \mathrm{d}, J=4.5$ $\mathrm{Hz}) .1 .29(2 \mathrm{H}, \mathrm{m}) .2 .38(2 \mathrm{H}, \mathrm{m}) .3 .37(6 \mathrm{H}, \mathrm{s}) .3 .51(2 \mathrm{H}, \mathrm{m})$. $3.76(2 \mathrm{H} . \mathrm{d} . J=1.5 \mathrm{~Hz}), 3.80(3 \mathrm{H}, \mathrm{s}), 4.26(2 \mathrm{H} . \mathrm{t} . J=3.5$ $\mathrm{Hz}) .4 .72(\mathrm{lH.m} .4 .96(2 \mathrm{H} . \mathrm{m}) .5 .19(1 \mathrm{H} . \mathrm{q} . J=8.1 \mathrm{~Hz})$. $6.86(4 \mathrm{H}, \mathrm{m}) .7 .27(5 \mathrm{H}, \mathrm{m})$.

6c: Yield $37 \%{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 1.40(9 \mathrm{H}$, d. $J=4.5$ $\mathrm{Hz}) .1 .61(2 \mathrm{H}, \mathrm{m}) .2 .36(2 \mathrm{H} . \mathrm{m}) .3 .30(6 \mathrm{H}, \mathrm{s}) .3 .51(2 \mathrm{H}, \mathrm{m})$. $3.76(2 \mathrm{H} . \mathrm{m}) .4 .52(2 \mathrm{H} . \mathrm{m}) .4 .89(1 \mathrm{H} . \mathrm{m}) .4 .98(2 \mathrm{H} . \mathrm{m}) .5 .04$ ( $1 \mathrm{H} . \mathrm{q} . J=8.1 \mathrm{~Hz}$ ). $7.19(3 \mathrm{H} . \mathrm{m}), 7.24(5 \mathrm{H} . \mathrm{m})$.

6d: Yield $35 \%$. ${ }^{1}$-NMR ( $\mathrm{CDCl}_{3}$ ) $\delta 1.40(9 \mathrm{H} . \mathrm{d}, J=4.5$ $\mathrm{Hz}) .1 .61(2 \mathrm{H}, \mathrm{m}) .2 .36(2 \mathrm{H}, \mathrm{m}) .3 .31(6 \mathrm{H}, \mathrm{s}) .3 .51(2 \mathrm{H}, \mathrm{m})$. $3.76(2 \mathrm{H} . \mathrm{m}) .4 .52(2 \mathrm{H} . \mathrm{m}) .4 .89(1 \mathrm{H} . \mathrm{m}) .4 .96(2 \mathrm{H} . \mathrm{m}) .5 .49$ $(1 \mathrm{H} . \mathrm{q}, ~ J=8.1 \mathrm{~Hz}) .7 .02(4 \mathrm{H} . \mathrm{m}), 7.29(5 \mathrm{H} . \mathrm{m})$.

6e: Yield $37 \%$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 1.40(9 \mathrm{H}, \mathrm{d} . J=4.5$ $\mathrm{Hz}) .1 .60(1 \mathrm{H} . \mathrm{m}) .2 .33(3 \mathrm{H} . \mathrm{s}) .2 .69(2 \mathrm{H} . \mathrm{m}) .2 .95(2 \mathrm{H} . \mathrm{m})$. $3.38(6 \mathrm{H} . \mathrm{s}) .3 .45(2 \mathrm{H} . \mathrm{m}) .3 .51(2 \mathrm{H} . \mathrm{m}) .4 .52(2 \mathrm{H} . \mathrm{m}) .4 .87$ ( $1 \mathrm{H} . \mathrm{m}$ ). $4.95(2 \mathrm{H}, \mathrm{m}) .5 .29(1 \mathrm{H}, \mathrm{m}) .7 .12$ ( $9 \mathrm{H} . \mathrm{m}$ ).

6 f: Yield $32 \% .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 1.40(9 \mathrm{H}, \mathrm{d}, J=4.5$ $\mathrm{Hz}) .1 .60(2 \mathrm{H} . \mathrm{m}) .2 .37(2 \mathrm{H}, \mathrm{m}) .2 .70(4 \mathrm{H}, \mathrm{m}) 3.29(2 \mathrm{H} . \mathrm{m})$, $3.53(4 \mathrm{H} . \mathrm{m}) .4 .38(3 \mathrm{H} . \mathrm{m}) .5 .33(2 \mathrm{H} . \mathrm{m}) .6 .03(\mathrm{lH} . \mathrm{q} . J=$ $2.31 \mathrm{~Hz}) .7 .29(5 \mathrm{H} . \mathrm{m})$.

6 g : Yield $50 \%$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 1.40(9 \mathrm{H}, \mathrm{d}, J=4.5$ $\mathrm{Hz}) .1 .60(2 \mathrm{H}, \mathrm{m}) .1 .64(4 \mathrm{H}, \mathrm{m}) .2 .48(2 \mathrm{H}, \mathrm{m}) .3 .02(2 \mathrm{H}, \mathrm{m})$, $3.34(4 \mathrm{H} . \mathrm{m}) .3 .37(4 \mathrm{H}, \mathrm{m}), 4.35(2 \mathrm{H} . \mathrm{m}) .5 .32(1 \mathrm{H}, \mathrm{dd} . J=$ 3.0 and 9.0 Hz$) .6 .01(1 \mathrm{H} . \mathrm{q} . ~ J=6.0 \mathrm{~Hz}) .7 .27(5 \mathrm{H} . \mathrm{m})$.
( $(6 \mathrm{~S})$-8-Benzyl-1-[(benzylamino)carbonyl]tetrahydro2 H -pyrazino[1,2-a]pyrimidin-4,7-dione-6-ly\}acetic acid (7a). A solution of $6 \mathrm{a}(0.14 \mathrm{mmol} .82 \mathrm{mg})$ and formic acid ( 7 mL ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(300 \mathrm{~mL}$ ) was stirred for 12 h at room temperature. Evaporation of the solution in vacuo gave a crude residue. which was purified by silica gel column chromatography with EtOAc/acetone (3:1) to give 7a (19.0 $\mathrm{mg} .30 \%$ ) as a foamy solid. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 1.78(2 \mathrm{H}$. $\mathrm{m}) .2 .37(2 \mathrm{H}, \mathrm{m}) .3 .29(4 \mathrm{H}, \mathrm{m}), 4.38(3 \mathrm{H} . \mathrm{m}) .5 .33(2 \mathrm{H}, \mathrm{m})$. $6.03(1 \mathrm{H} . \mathrm{q} . ~ J=2.3 \mathrm{~Hz}) .7 .29(10 \mathrm{H} . \mathrm{m})$. $-\mathrm{HRMS}(\mathrm{FAB})$ Calcd. for $\mathrm{C}_{24} \mathrm{H}_{26} \mathrm{~N}_{4} \mathrm{O}_{5} 450.1903$, Found ( $\mathrm{M}^{+}$) 450.1907 .

The synthesis of compounds $\mathbf{7 b}-\mathrm{g}$ was carried out by the same procedure as described for the preparation of 7a.

7b: Yield $35 \%$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 1.82(2 \mathrm{H}, \mathrm{m}), 2.53$ $(2 \mathrm{H}, \mathrm{m}), 3.31(4 \mathrm{H} . \mathrm{m}), 3.80(3 \mathrm{H}, \mathrm{s}) .4 .35(4 \mathrm{H}, \mathrm{m}) .5 .35(1 \mathrm{H}$, dd. $J=3.0$ and 9.0 Hz$) .5 .99(1 \mathrm{H} . \mathrm{q} . J=9.0 \mathrm{~Hz}) .6 .86(2 \mathrm{H} . \mathrm{d}$ $J=6.0 \mathrm{~Hz}) .7 .30(7 \mathrm{H.m}) .-\mathrm{HRMS}(\mathrm{FAB}) \mathrm{Calcd}$ for $\mathrm{C}_{23} \mathrm{H}_{28} \mathrm{~N}_{4} \mathrm{O}_{6}$ 480.2009 , Found $\left(\mathrm{M}^{+}\right) 480.2005$.

7c: Yield $38 \%$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta \mathrm{L} .82(2 \mathrm{H} . \mathrm{m}), 2.43$ $(2 \mathrm{H}, \mathrm{m}) .3 .33(4 \mathrm{H} . \mathrm{m}), 4.35(4 \mathrm{H}, \mathrm{m}), 5.32(\mathrm{lH}, \mathrm{dd} . J=3.0$ and 9.0 Hz$), 6.01(1 \mathrm{H} . \mathrm{q} . ~ J=6.0 \mathrm{~Hz}) .7 .17(8 \mathrm{H}, \mathrm{m}) .-\mathrm{HRMS}$ (FAB) Calcd for $\mathrm{C}_{26} \mathrm{H}_{3} \mathrm{~N}_{4} \mathrm{O}_{5} 478.2216$, Found ( $\mathrm{M}^{+}$) 478.2220 .

7d: Yield $37 \%$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 1.82(2 \mathrm{H}, \mathrm{m}), 2.48$ $(2 \mathrm{H}, \mathrm{m}), 3.34(4 \mathrm{H} . \mathrm{m}), 4.35(4 \mathrm{H}, \mathrm{m}), 5.32(\mathrm{lH}, \mathrm{dd} . J=3.0$ and 9.0 Hz$), 6.01(1 \mathrm{H} . \mathrm{q} . J=6.0 \mathrm{~Hz}) .7 .00(2 \mathrm{H}, \mathrm{t} . J=8.7$ $\mathrm{Hz}) .7 .27(7 \mathrm{H}, \mathrm{m})$. -HRMS (FAB) Calcd. for $\mathrm{C}_{2} \mathrm{H}_{2} ; \mathrm{FN}_{4} \mathrm{O}_{5}$ 468.1809 , Found $\left(\mathrm{M}^{+}\right) 468.1808$.

7e: Yield $38 \%$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta \mathrm{L} .82(2 \mathrm{H} . \mathrm{m}), 2.34$ $(3 \mathrm{H}, \mathrm{s}) .2 .39(2 \mathrm{H}, \mathrm{m}) .2 .80(2 \mathrm{H} . \mathrm{t} . J=6.6 \mathrm{~Hz}) .3 .30(2 \mathrm{H} . \mathrm{m})$. $3.48(4 \mathrm{H}, \mathrm{m}) .4 .74(2 \mathrm{H} . \mathrm{m}) .5 .32(1 \mathrm{H}, \mathrm{dd} . J=3.0$ and 9.0 $\mathrm{Hz}) .5 .99(1 \mathrm{H} . \mathrm{q} . J=6.0 \mathrm{~Hz}) .7 .09(4 \mathrm{H}$. dd. $J=7.8$ and 21.9 Hz ). 7.28 ( $5 \mathrm{H} . \mathrm{m}$ ). -HRMS (FAB) Calcd. for $\mathrm{C}_{24} \mathrm{H}_{2} \mathrm{~F}_{2} \mathrm{~N}_{4} \mathrm{O}_{5}$
486.1715. Found $\left(\mathrm{M}^{-}\right) 486.1717$.

7f: Yield $40 \%$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 1.78(2 \mathrm{H} . \mathrm{m}) .2 .37$ $(2 \mathrm{H} . \mathrm{m}), 2.70(4 \mathrm{H} . \mathrm{m}) 3.29(2 \mathrm{H} . \mathrm{m}), 3.53(4 \mathrm{H}, \mathrm{m}) .4 .38(3 \mathrm{H}$. m). $5.33(2 \mathrm{H} . \mathrm{m}) .6 .03(1 \mathrm{H} . \mathrm{q} . J=2.3 \mathrm{~Hz}) .7 .29(5 \mathrm{H}, \mathrm{m})$. -HRMS (FAB) Calcd. for $\mathrm{C}_{2} \mathrm{H}_{2} \mathrm{~N}_{4} \mathrm{O}_{5} \mathrm{~S} 446.1624$, Found $\left(\mathrm{M}^{+}\right) 446.1630$
7 g : Yield $38 \%$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 1.80(2 \mathrm{H} . \mathrm{ml}) .1 .64$ $(4 \mathrm{H} . \mathrm{m}) .2 .48(2 \mathrm{H}, \mathrm{m}) .3 .02(2 \mathrm{H} . \mathrm{m}) .3 .34(4 \mathrm{H} . \mathrm{m}) .3 .37(4 \mathrm{H}$. $\mathrm{m}) .4 .35(2 \mathrm{H} . \mathrm{m}), 5.32(1 \mathrm{H}$. dd $J=3.0$ and 9.0 Hz$) .6 .01$ ( $1 \mathrm{H} . \mathrm{q} . J=6.0 \mathrm{~Hz}$ ). $7.27(5 \mathrm{H} . \mathrm{m})$. -HRMS (FAB) Calcd. for $\mathrm{C}_{2} 2 \mathrm{H}_{2}{ }_{8} \mathrm{~N}_{4} \mathrm{O}_{6} 444.2009$, Found (M $\mathrm{M}^{-}$4 44.2003.

Acknowledgements. We would like to thank Dr. Kalun and Dr. Masa for their helpful discussions and providing biological data the duration of this work. Finally, we wish to thank Hawon Pharmaceuticals co. which was supported with fund.

## References

1. Kahnı. M. Swhett. 1993. M. 821.
2. (a) Souers. A. T.: Virgilio. A. A.: Schürer. S. S.: Ellman. J. A.: Kogant. T. P.: Ankener. H. W.: Vanderslice. P. Bioorg Med. Chen. Left. 1998. 8. 2297. (b) Souers. A. J.: Virgilio. A. A.: Rosenquist. A.'; Fenuik, J. A.; Ellman, A. d. Am. Chem. Soc. 1999. 121, 1817.
3. Masakatsu. E.: Lee. M. S.; Hiroshi, N.; Marcin, S.; Scott, L.; Kahn. M. J. Am. Chem. Soc. 1999, 121. 12204.
4. Masakatsu. E.: Lee. M. S.: Richard. Y. W.: Shea. T. P.: Marcin1. S.: Scot. L.: Kahn. M. J. Med Chent 2002. +5. 1395.
5. Masakatsu. E.: Lee. M. S.: Stasiak. M.: Kahn. M. Tetrohedron Lett. 2001, 42.1237.
6. Kee. K. S.: Jois. D. S. Curvent Phamaceutical Design 2003, 9. 1209.
7. Kim. T.-W.: Cho. J.-H.: Han. T. H.: Lee. J. B.: Oh. C.-H. Bull. Korean Chem. Soc. 2006. 27. 484.
8. Jeong. D.-W.: Yoo. M.-H.: Kim. T. S.: Kim. T.-H.: Kim. I. Y. J. Biol. Chem. 2002, 277. 17871.
