

Synthesis of β -Aryl Substituted *N*-Tosyl *Aza*-Baylis-Hillman Adducts: Heck Reaction of *N*-Tosyl *Aza*-Baylis-Hillman Adducts

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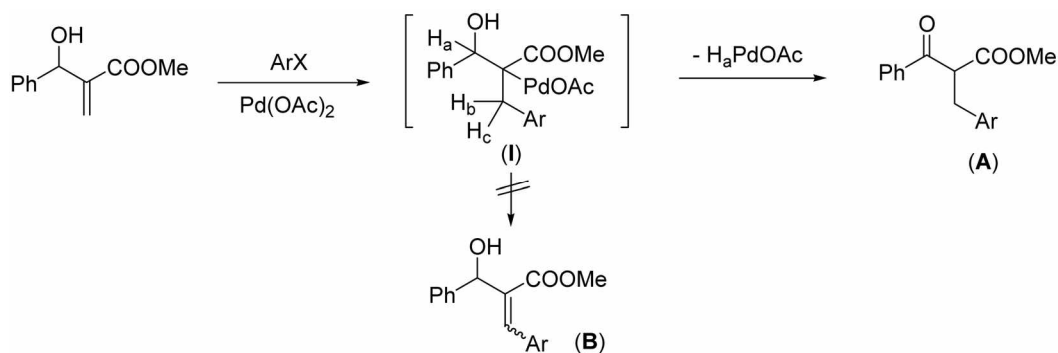
Key Words : Baylis-Hillman adducts, Heck reaction, *Aza*-Baylis-Hillman adducts

During the last two decades notable improvements in Baylis-Hillman chemistry have been achieved in view of the reaction rate and synthetic applications of Baylis-Hillman or *aza*-Baylis-Hillman adducts.¹ However, the general and efficient synthesis of β -branched *aza*-Baylis-Hillman adducts has remained unsolved. Although many approaches have been examined, most of the methods suffer from low yields and lack of generality.^{2,3} Thus, development of an efficient synthetic method of these compounds would be helpful in chemical transformations of Baylis-Hillman adducts.¹⁻⁴

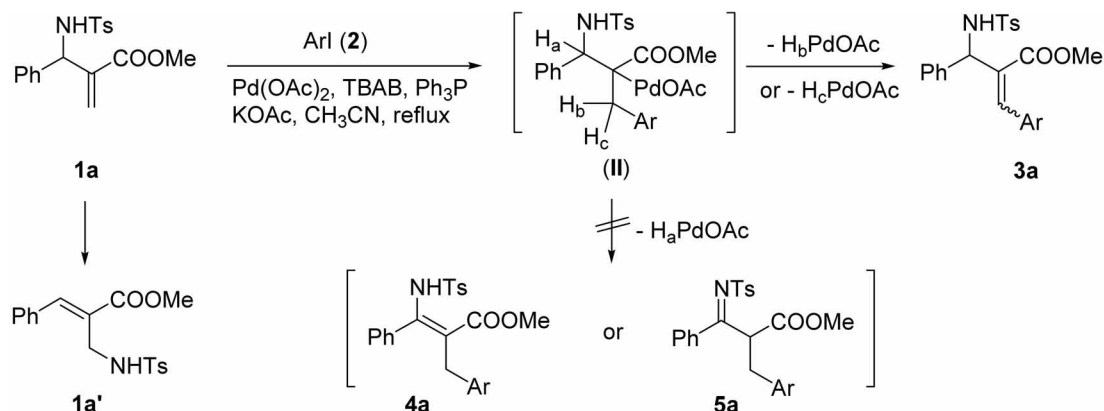
The most simple and convenient method for the preparation of β -aryl-substituted Baylis-Hillman adducts could be the palladium-mediated Heck reaction with aryl halides. Actually intermolecular Heck type arylation of Baylis-

Hillman adducts has been examined by some research groups.⁵ However, the reaction gave benzyl-substituted β -keto ester (**A**) as the major product instead of β -aryl-substituted Baylis-Hillman type adduct (**B**) as shown in Scheme 1.⁵ The compound (**A**) was generated *via* the *syn*-elimination of H_aPdOAc from the intermediate (**I**) and the following keto-enol tautomerization.^{5c} This unfavorable result might be the principle reason for the lack of any trials on the synthesis of β -aryl *aza*-Baylis-Hillman adducts *via* the Heck type arylation strategy.

Three types of compounds including **3a**, **4a** and **5a** could be produced from the Heck reaction of *N*-tosyl *aza*-Baylis-Hillman adduct **1a** as in Scheme 2. However, we expected that the conformation of the intermediate (**II**, Scheme 2) might be differ with that of the corresponding intermediate



Scheme 1



Scheme 2

of Baylis-Hillman alcohol (**1**, Scheme 1) due to the increased steric hindrance around H_a. Thus, we expected that the final *syn*-elimination of palladium could occur with H_b/H_c instead of H_a to produce desired **3a** as the major product. With the

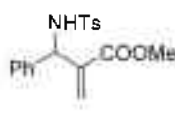
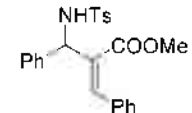
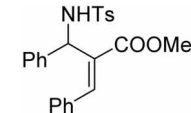
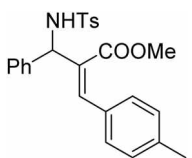
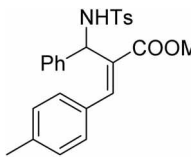
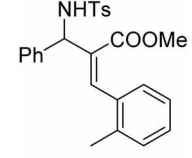
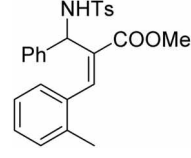
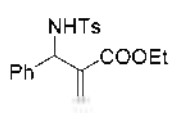
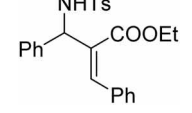
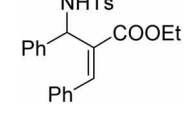
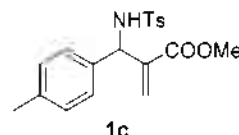
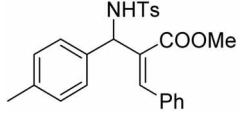
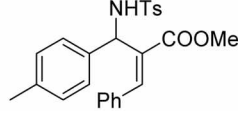
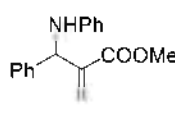
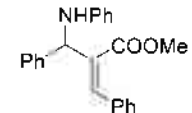
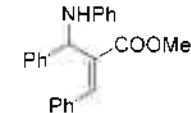
expectation we examined the reaction of **1a** and iodobenzene (**2a**). To our delight we obtained β -phenyl *N*-tosyl *aza*-Baylis-Hillman adduct **3a** in good yield (67%) as *E/Z* mixture and we wish to report herein the results. To the best

Table 1. Optimization of reaction conditions for the synthesis of **3a**

| Entry | Conditions | Results (% Yield) ^a |
|-------|--|---------------------------------|
| 1 | Pd(OAc) ₂ (5 mol%), Et ₃ N (3.0 equiv), PPh ₃ (20 mol%), DMF, 90-100 °C, 5 h | 3a (45), 1a (33) |
| 2 | Pd(OAc) ₂ (10 mol%), Et ₃ N (3.0 equiv), PPh ₃ (20 mol%), CH ₃ CN, reflux, 25 h | 3a (50), 1a (20) |
| 3 | Pd(OAc) ₂ (10 mol%), <i>n</i> -Bu ₄ NBr (0.5 equiv), Et ₃ N (3.0 equiv), PPh ₃ (20 mol%), CH ₃ CN, reflux, 20 h | 3a (60), 1a (15) |
| 4 | Pd(OAc) ₂ (5 mol%), <i>n</i> -Bu ₄ NBr (1.0 equiv), KOAc (2.0 equiv), PPh ₃ (10 mol%), CH ₃ CN, reflux, 18 h | 3a (73), 1a (11) |
| 5 | Pd(OAc) ₂ (5 mol%), <i>n</i> -Bu ₄ NBr (1.0 equiv), K ₂ CO ₃ (3.0 equiv), H ₂ O/DMF, 50-60 °C, 5 h | 3a (0), 1a' (95) |
| 6 | Pd(OAc) ₂ (5 mol%), K ₂ CO ₃ (3.0 equiv), PPh ₃ (20 mol%), CH ₃ CN, reflux, 20 h | 3a (30), 1a' (25) |

^aThe yield of **3a** is a combined yield of *E* and *Z* isomers. In some cases **3a** was contaminated with small amount of **1a**.

Table 2. Synthesis of β -aryl *aza*-Baylis-Hillman adducts

| Entry | Substrate | Conditions ^a | Products (%) |
|-------|--|---|---|
| 1 |  1a | 1. C ₆ H ₅ I (2a), 18 h 2. K ₂ CO ₃ , reflux, 8 h |  3a-Z (31)  3a-E (36) |
| 2 | 1a | 1. 4-MeC ₆ H ₄ I (2b), 24 h 2. K ₂ CO ₃ , reflux, 8 h |  3b-Z (37)  3b-E (41) |
| 3 | 1a | 1. 2-MeC ₆ H ₄ I (2c), 15 h |  3c-Z (26)  3c-E (48) |
| 4 |  1b | 1. 2a , 20 h 2. K ₂ CO ₃ , reflux, 8 h |  3d-Z (32)  3d-E (40) |
| 5 |  1c | 1. 2a , 20 h 2. K ₂ CO ₃ , reflux, 8 h |  3e-Z (35)  3e-E (41) |
| 6 |  1d | 1. 2a , 10 h 2. K ₂ CO ₃ , reflux, 6 h |  3f-Z (27)  3f-E (31) |

^aConditions: step 1: compound **1** (1.0 mmol), compound **2** (2.0 mmol), Pd(OAc)₂ (5 mol%), PPh₃ (10 mol%), TBAB (1.0 mmol), KOAc (2.0 mmol), CH₃CN, reflux; step 2: K₂CO₃ (1.0 mmol), reflux.

of our knowledge this is the first successful results for the synthesis of β -aryl *aza*-Baylis-Hillman adduct *via* palladium-mediated Heck reaction.⁵

The reactions of **1a** and **2a** under various Pd-mediated Heck reaction conditions were examined and the results are summarized in Table 1. In most cases (entries 1-4 and 6) we observed the formation of desired product **3a** in variable yields (30-73%) with some remaining starting material **1a**. When we used Et₃N the reaction was sluggish (entries 1-3). Among the conditions the use of Pd(OAc)₂/TBAB/KOAc/PPh₃ in refluxing CH₃CN (entry 4) was found to be the best. It is interesting to note that rearranged tosylamide derivative **1a'** was obtained almost quantitatively when we used K₂CO₃ as a base (entry 5).^{6,7}

Initially, we isolated **3a-E** (35%) and **3a-Z** (38%) under the conditions of entry 4 in Table 1. However, unfortunately, **3a-Z** was contaminated with small amount of starting material **1a**, which could not be separated easily by column chromatography due to their similar mobility. Thus we used K₂CO₃ in order to convert remaining **1a** into **1a'** completely according to the results of entry 5 in Table 1. In this manner we obtained analytically pure **3a-E** (36%) and **3a-Z** (31%), which were identified by comparison with the reported data (*vide infra*, entry 1 in Table 2).²

Encouraged by the successful results, we prepared starting materials **1b-d** according to the reported methods,⁸ and synthesized analogous compounds **3b-f** similarly under the optimized conditions and the results are summarized in Table 2. 4-Iodotoluene (**2b**) and 2-iodotoluene (**2c**) showed similar reactivity (entries 2 and 3). Other *N*-tosyl- (**1b** and **1c**) and *N*-phenyl- (**1d**) derivatives also showed same reactivity (entries 4-6). In most cases except entry 3, we observed some remaining starting materials **1a-d** and we treated the reaction mixture with K₂CO₃ before separation (*vide supra*).

In summary, we prepared some β -aryl *N*-tosyl *aza*-Baylis-Hillman adducts *via* the Heck type reaction of *aza*-Baylis-Hillman adduct and aryl iodide under the influence of Pd(OAc)₂/TBAB/KOAc/PPh₃ in refluxing CH₃CN in moderate yield as *E/Z* mixture.

Experimental Section

Typical procedure for the synthesis of 3a: A mixture of **1a** (345 mg, 1.0 mmol), **2a** (408 mg, 2.0 mmol), Pd(OAc)₂ (11 mg, 0.05 mmol), *n*-Bu₄NBr (322 mg, 1.0 mmol), KOAc (196 mg, 2.0 mmol), PPh₃ (26 mg, 0.1 mmol) in CH₃CN (3 mL) was heated to reflux for 18 h. To the reaction mixture K₂CO₃ (138 mg, 1.0 mmol) was added and maintained refluxing for 8 h. After the usual aqueous workup and column chromatographic purification process (hexanes/CH₂Cl₂/ether, 5:1:2) we obtained **3a-Z** (131 mg, 31%) and **3a-E** (152 mg, 36%) as white solids. The selected spectroscopic data of prepared compounds **3a** and **3f** are as follows.

Compound **3a-Z**: 31%; white solid, mp 117-119 °C; IR (film) 3290, 2924, 1711, 1163 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 2.20 (s, 3H), 3.39 (s, 3H), 5.32 (d, *J* = 9.3 Hz, 1H),

5.99 (d, *J* = 9.3 Hz, 1H), 6.63 (s, 1H), 6.94-6.97 (m, 2H), 7.12 (d, *J* = 8.1 Hz, 2H), 7.21-7.33 (m, 8H), 7.72 (d, *J* = 8.1 Hz, 2H); ¹³C NMR (CDCl₃, 75 MHz) δ 21.24, 51.62, 61.52, 126.46, 127.20, 127.81, 127.94, 128.33, 128.55 (2C), 129.53, 130.18, 134.58, 137.74, 138.12, 138.22, 143.34, 168.22; ESIMS *m/z* 422 (M⁺+1). Anal Calcd for C₂₄H₂₃NO₄S: C, 68.39; H, 5.50; N, 3.32. Found: C, 68.58; H, 5.77; N, 3.23.

Compound **3a-E**: 36%; white solid, mp 153-155 °C; IR (film) 3292, 3061, 1718, 1161 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 2.39 (s, 3H), 3.66 (s, 3H), 5.85 (d, *J* = 10.5 Hz, 1H), 6.34 (d, *J* = 10.5 Hz, 1H), 7.09 (d, *J* = 8.4 Hz, 2H), 7.16-7.19 (m, 2H), 7.26-7.43 (m, 10H), 7.69 (s, 1H); ¹³C NMR (CDCl₃, 75 MHz) δ 21.46, 52.09, 53.94, 126.26, 127.02, 127.57, 128.57, 128.78, 128.94 (2C), 129.23, 129.56, 133.65, 137.68, 139.07, 142.75, 142.92, 166.89; ESIMS *m/z* 422 (M⁺+1). Anal Calcd for C₂₄H₂₃NO₄S: C, 68.39; H, 5.50; N, 3.32. Found: C, 68.64; H, 5.46; N, 3.15.

Compound **3b-Z**: 37%; white solid, mp 126-128 °C; IR (film) 3292, 2960, 2918, 1699 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 2.21 (s, 3H), 2.33 (s, 3H), 3.41 (s, 3H), 5.29 (d, *J* = 9.3 Hz, 1H), 5.94 (d, *J* = 9.3 Hz, 1H), 6.57 (s, 1H), 6.86 (d, *J* = 8.4 Hz, 2H), 7.05-7.16 (m, 4H), 7.22-7.32 (m, 5H), 7.71 (d, *J* = 8.4 Hz, 2H); ¹³C NMR (CDCl₃, 75 MHz) δ 21.26 (2C), 51.61, 61.70, 126.46, 127.21, 127.76, 128.47, 128.53, 128.67, 129.18, 129.52, 131.61, 137.79, 138.29, 138.44, 138.75, 143.29, 168.39; LCMS *m/z* 435 (M⁺).

Compound **3b-E**: 41%; white solid, mp 161-163 °C; IR (film) 3309, 2952, 2924, 1697 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 2.38 (s, 3H), 2.39 (s, 3H), 3.64 (s, 3H), 5.88 (d, *J* = 10.5 Hz, 1H), 6.35 (d, *J* = 10.5 Hz, 1H), 7.06-7.10 (m, 4H), 7.17 (d, *J* = 8.1 Hz, 2H), 7.27-7.43 (m, 7H), 7.63 (s, 1H); ¹³C NMR (CDCl₃, 75 MHz) δ 21.40, 21.49, 52.02, 54.08, 126.34, 127.05, 127.55, 128.05, 128.57, 129.10, 129.19, 129.56, 130.84, 137.83, 139.17, 140.00, 142.89, 142.96, 167.03; LCMS *m/z* 435 (M⁺).

Compound **3c-Z**: 26%; white solid, mp 146-148 °C; IR (film) 3288, 2924, 1707 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 2.21 (s, 3H), 2.29 (s, 3H), 3.30 (s, 3H), 5.37 (d, *J* = 9.3 Hz, 1H), 5.99 (d, *J* = 9.3 Hz, 1H), 6.56-6.59 (m, 1H), 6.86 (s, 1H), 7.00-7.05 (m, 1H), 7.12-7.22 (m, 4H), 7.23-7.34 (m, 5H), 7.75 (d, *J* = 8.1 Hz, 2H); ¹³C NMR (CDCl₃, 75 MHz) δ 19.84, 21.34, 51.51, 61.22, 125.13, 126.41, 127.17, 127.79 (2C), 128.38, 128.60, 129.62, 129.66, 131.23, 134.62, 135.67, 137.93, 138.50, 138.81, 143.39, 167.87; LCMS *m/z* 435 (M⁺).

Compound **3c-E**: 48%; white solid, mp 140-142 °C; IR (film) 3309, 2952, 1703 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 2.17 (s, 3H), 2.35 (s, 3H), 3.70 (s, 3H), 5.69 (d, *J* = 10.2 Hz, 1H), 6.35 (d, *J* = 10.2 Hz, 1H), 7.05-7.12 (m, 4H), 7.21-7.29 (m, 7H), 7.42 (d, *J* = 8.4 Hz, 2H), 7.86 (s, 1H); ¹³C NMR (CDCl₃, 75 MHz) δ 19.86, 21.40, 52.17, 53.82, 126.13, 126.22, 126.92, 127.38, 127.75, 128.44, 129.30, 129.40, 129.95, 130.32, 132.90, 137.30, 137.69, 139.48, 141.59, 142.87, 167.00; LCMS *m/z* 435 (M⁺).

Compound **3d-Z**: 32%; white solid, mp 88-90 °C; IR (film) 3292, 2918, 1699 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz)

δ 0.83 (t, $J = 7.2$ Hz, 3H), 2.21 (s, 3H), 3.81-3.93 (m, 2H), 5.31 (d, $J = 9.3$ Hz, 1H), 5.99 (d, $J = 9.3$ Hz, 1H), 6.62 (s, 1H), 6.95-6.98 (m, 2H), 7.13 (d, $J = 8.1$ Hz, 2H), 7.23-7.33 (m, 8H), 7.73 (d, $J = 8.1$ Hz, 2H); ^{13}C NMR (CDCl_3 , 75 MHz) δ 13.33, 21.26, 60.87, 61.64, 126.50, 127.22, 127.77, 127.84, 128.40, 128.45, 128.49, 129.53, 130.54, 134.69, 137.81, 138.02, 138.20, 143.33, 167.72.

Compound **3d-E**: 40%; white solid, mp 148-149 °C; IR (film) 3311, 2964, 1693, 1261 cm^{-1} ; ^1H NMR (CDCl_3 , 300 MHz) δ 1.19 (t, $J = 7.2$ Hz, 3H), 2.39 (s, 3H), 4.06-4.16 (m, 2H), 5.85 (d, $J = 10.2$ Hz, 1H), 6.37 (d, $J = 10.2$, 1H), 7.09 (d, $J = 8.4$ Hz, 2H), 7.16-7.19 (m, 2H), 7.23-7.43 (m, 10H), 7.67 (s, 1H); ^{13}C NMR (CDCl_3 , 75 MHz) δ 14.02, 21.47, 53.99, 61.15, 126.27, 127.02, 127.52, 128.53, 128.77, 128.94, 129.19, 129.24, 129.50, 133.72, 137.75, 139.20, 142.48, 142.87, 166.43.

Compound **3e-Z**: 35%; white solid, mp 86-88 °C; IR (film) 3294, 2924, 1703 cm^{-1} ; ^1H NMR (CDCl_3 , 300 MHz) δ 2.25 (s, 3H), 2.33 (s, 3H), 3.44 (s, 3H), 5.31 (d, $J = 9.3$ Hz, 1H), 5.88 (d, $J = 9.3$ Hz, 1H), 6.66 (s, 1H), 6.97-7.00 (m, 2H), 7.10-7.22 (m, 7H), 7.28-7.30 (m, 2H), 7.75 (d, $J = 8.4$ Hz, 2H); ^{13}C NMR (CDCl_3 , 75 MHz) δ 21.01, 21.27, 51.64, 61.36, 126.37, 127.25, 127.94, 128.35, 128.52, 129.28, 129.53, 130.32, 134.67, 135.15, 137.60, 137.79, 138.03, 143.32, 168.30.

Compound **3e-E**: 41%; white solid, mp 142-143 °C; IR (film) 3311, 2952, 1699 cm^{-1} ; ^1H NMR (CDCl_3 , 300 MHz) δ 2.32 (s, 3H), 2.39 (s, 3H), 3.66 (s, 3H), 5.80 (d, $J = 10.5$ Hz, 1H), 6.31 (d, $J = 10.5$ Hz, 1H), 7.07-7.18 (m, 6H), 7.25 (d, $J = 7.8$ Hz, 2H), 7.32-7.42 (m, 5H), 7.66 (s, 1H); ^{13}C NMR (CDCl_3 , 75 MHz) δ 20.98, 21.48, 52.07, 53.77, 126.20, 127.05, 128.77, 128.97, 129.06, 129.21, 129.31, 129.53, 133.72, 136.07, 137.31, 137.74, 142.63, 142.88, 166.94.

Compound **3f-Z**: 27%; pale yellow oil; IR (film) 3402, 3026, 1712, 1601 cm^{-1} ; ^1H NMR (CDCl_3 , 300 MHz) δ 3.53 (s, 3H), 4.27 (br s, 1H), 5.41 (s, 1H), 6.66-6.74 (m, 3H), 6.92 (s, 1H), 7.12-7.19 (m, 2H), 7.22-7.45 (m, 10H); ^{13}C NMR (CDCl_3 , 75 MHz) δ 51.72, 61.66, 113.52, 118.03, 127.70, 127.99, 128.12, 128.15, 128.28, 128.82, 129.19, 133.85, 134.29, 135.48, 139.89, 146.62, 169.32; ESIMS m/z 344 ($M^+ + 1$). Anal Calcd for $\text{C}_{23}\text{H}_{21}\text{NO}_2$: C, 80.44; H, 6.16; N, 4.08. Found: C, 80.67; H, 6.05; N, 3.93.

Compound **3f-E**: 31%; pale yellow oil; IR (film) 3402, 3057, 1709, 1601 cm^{-1} ; ^1H NMR (CDCl_3 , 300 MHz) δ 3.70 (s, 3H), 5.17 (br s, 1H), 5.91 (s, 1H), 6.37-6.41 (m, 2H), 6.62-6.68 (m, 1H), 7.02-7.09 (m, 2H), 7.25-7.43 (m, 10H), 7.96 (s, 1H); ^{13}C NMR (CDCl_3 , 75 MHz) δ 51.88, 53.99, 113.42, 117.59, 126.44, 127.05, 128.44, 128.74, 128.91, 129.08, 129.21, 132.17, 134.82, 141.20, 141.72, 146.81, 167.26; ESIMS m/z 344 ($M^+ + 1$). Anal Calcd for $\text{C}_{23}\text{H}_{21}\text{NO}_2$: C, 80.44; H, 6.16; N, 4.08. Found: C, 80.76; H, 6.35; N, 4.02.

References and Notes

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- The reaction of **1a** and bromobenzene under the optimized conditions (entry 4 in Table 1) was examined, but we observed no reaction. Most of the starting material **1a** was remained (70-80%) and we observed the formation of small amounts (< 20%) of **1a'**, which might be produced via the Pd-mediated rearrangement. For the related reference, please see: Park, J. B.; Ko, S. H.; Kim, B. G.; Hong, W. P.; Lee, K.-J. *Bull. Korean Chem. Soc.* **2004**, *25*, 27-28.
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