# A New Naphthoquinone with Anti-inflammatory Activity from An Egyptian Collection of *Echiochilon fruticosum*

Hoda M. Fathy, Mohamed I. Aboushoer, Azza Baraka<sup>1</sup>, Maged S. Abdel-Kader\*, and Abdallah A. Omar

Department of Pharmacognosy, Faculty of Pharmacy

<sup>1</sup>Departmentof Pharmacology, Faculty of medicine, Alexandria University, Alexandria 21215, Egypt

**Abstract** – Phytochemical investigation of the roots of *Echiochilon fruticosum* resulted in the isolation of two naphthoquinone derivatives. Compound 1 was identified as anhydroalkanin while compound 2 was identified as a new derivative 5-hydroxy 8-methoxy 2-(4-methylpent-1,3-dienyl) naphthalene-1,4-dione named as echiochiloquinone. In addition, caffeic acid 3, caffeic acid methyl ester 4 were isolated. The structures were determined by physical, chemical and spectral methods. The anti-inflammatory activity of the root extracts and compound 2 was evaluated utilizing both cotton pellet-induced and carragenin-induced rat paw edema. The ulcerogenic effect was also studied. **Keywords** – *Echiochilon fruticosum*, Naphthoquinones, Echiochiloquinone, Caffeic acid, Anti-inflammatory, Ulcerogenic

## Introduction

Several members of the family Boraginaceae are used in folk medicine as anti-inflammatory, antiseptic, and remedy for wounds (Li, 2002; Roeder, 1995). Antitumor, antimicrobial, anti-inflammatory, contraceptive and hypotriglyceridemic activities were reported for members of the family (Leyva et al., 2000; Karyagina et al., 2001; Kourounakis et al., 2002; Singh et al., 2003; Zhen et al., 2002; Mats et al., 1982). Echiochilon fruticosum is distributed in the sandy and stony ground of North Africa, Sinai, Syria and Arabia (Täckholm, 1974). The nutritive value of E. fruticosum has been investigated and the possibility of its use as forage for domestic animals were demonstrated (Boshra). A variety of chemical entities has been reported from E. fruticosum, including volatile components, Eugenol glycoside, vomifoliol and trans syringin. Several flavonoid derivatives such as 5-deoxy vitexin, 5,7dimethoxy isoflavone, dihydro robinetin, garbanzol, 5,6,7trihydroxy flavanone, naringenin 5-methyl ether and pincomberin-7-glycoside were also isolated (Hashem, 2003; Bergaoui et al., 2004; Hammami et al., 2004).

# Experimental

General – Silica gel (230 - 400 mesh) and  $C_{18}$ -silica

\*Author for correspondence

Tel: +966-1-467-7264; E-mail: mpharm101@hotmail.com

gel, E. Merck were used for column chromatography. Precoated TLC plates 0.25 mm, silica gel 60 (GF-254), and precoated TLC 0.20 mm, RP18  $F_{254}$ , E. Merck, were used for TLC analysis. Melting points were determined on a Sturat SMP heating stage microscope. 1D and 2D NMR analyses were obtained using jeol 500 MHz spectrometer for <sup>1</sup>H and 125 MHz for <sup>13</sup>C NMR. Residual peaks of the deuterated solvents were used to reference the spectrum. EIMS was obtained on a Delsi-Nermag R30-10, while electrospray-ionization mass spectra were carried on a Finnigan LCQ. UV-VIS spectra were carried on a Helios  $\alpha$  thermo spectronic, England, supported with software Vision 32.

**Plant materials** – *Echiochilon fruticosum* Desf. was collected from Omayed biosphere, 90 km west Alexandria, Egypt. The plant material was identified by Prof. Dr. Lotfy Boulos, National Research Center, Dokki, Cairo, Egypt. A voucher was kept in the Department of Pharmacognosy, Faculty of Pharmacy, Alexandria University, Egypt.

**Extraction and Isolation** – The powdered air-dried roots (400 g) were extracted by maceration with light pet.  $(1 \times L 2)$ . The combined extract was evaporated under reduced pressure to give 1.7 g of dark red semisolid residue. The marc left after extraction with light pet. was air dried then macerated in 70% EtOH (1 L × 2) followed by 50% EtOH (1 L × 2). Each extract was evaporated under reduced pressure to yield 3 and 6 g respectively.

Part of the dried light pet. residue (1.5 g) was fractionated on a silica gel column  $(50 \text{ g}, 30 \times 2 \text{ cm})$  elution started with light pet.; polarity was gradually increased by CH<sub>2</sub>Cl<sub>2</sub> followed by EtOAc. Fractions of 200 ml were collected, evaporated under reduced pressure and screened by TLC. The fraction eluted with 10% CH<sub>2</sub>Cl<sub>2</sub> in light pet. (16 mg) was purified on C<sub>18</sub> silica gel column (25 g,  $3.5 \times 20 \text{ cm}$ ) eluted with 95% MeOH to afford **1** (8 mg). Fraction eluted with 25% CH<sub>2</sub>Cl<sub>2</sub> in light petroleum (100 mg) was crystallized from benzene to give **2** (80 mg).

The powdered air-dried aerial parts (2.2 Kg) were extracted at room temperature with 70% EtOH (7 L × 2). The combined EtOH extracts were evaporated under reduced pressure to give 62 g of dark green semisolid residue. The extract was dissolved in hot MeOH and allowed to settle to give 1.9 g precipitate. This precipitate was dissolved in CH<sub>2</sub>Cl<sub>2</sub> to give 0.9 g CH<sub>2</sub>Cl<sub>2</sub> soluble part. The CH<sub>2</sub>Cl<sub>2</sub> soluble part was fractionated on a silica gel column. Gradient elution was performed using light pet. with increasing concentrations of EtOAc. Fraction eluted with 10% EtOAc in light pet. (400 mg) yielded 50 mg of compound **3** after crystallization from MeOH. Fraction eluted with 20% EtOAc in light pet. (30 mg) was purified on a silica gel column (15 g, 1 × 20 cm) elution was carried out using CH<sub>2</sub>Cl<sub>2</sub> to yield 8 mg of **4**.

Table 1. <sup>1</sup>HNMR and <sup>13</sup>CNMR spectral data of 1 and 2 (CDCl<sub>3</sub>).

**Anhydroalkannin (dehydroxy shikonin)**  $1 - C_{16}H_{16}O_4$ , orange red oil, UV-VIS (MeOH)  $\lambda_{max}$  264, 283, 489, 510 and 558. <sup>1</sup>H NMR and <sup>13</sup>C NMR: Table 1. ESIMS *m/z* (rel. int.): 272 (M<sup>+</sup>, 100), 203 (50).

5-Hydroxy 8-methoxy 2-(4-methylpent-1,3-dienyl) naphthalene-1,4-dione (Echiochiloquinone) 2 –  $C_{17}H_{16}O_4$ , orange red crystals, m.p. 158 - 160°C, UV -VIS (MeOH)  $\lambda_{max}$  300, 510, 539, and 584 nm. <sup>1</sup>H NMR and <sup>13</sup>C NMR: Table 1. EIMS *m/z* (rel. int.): 284 (M<sup>+</sup>, 100), 269 (284 -CH<sub>3</sub>) (42), 256 (5), 253 (284 - OCH<sub>3</sub>) (21), 225 (18), 203 (5).

**Caffeic acid 3** – C<sub>9</sub>H<sub>8</sub>O<sub>4</sub>, yellowish white amorphous solid; m.p. 194°C; <sup>1</sup>H NMR ( $\delta_{H}$ , CDCl<sub>3</sub>): 7.07 (1H, s, H-3), 6.87 (1H, d, J = 8 Hz, H-5), 7.01 (1H, br d, J = 8.4 Hz, H-6), 7.57 (1H, d, J = 16 Hz, H-7), 6.26 (1H, d, J = 15.3 Hz, H-8), 5.8 (1H, br s, OH), 5.7(1H, br s, OH); <sup>13</sup>C NMR ( $\delta_{c}$ , CD<sub>3</sub>OD): 146.3 (C-1), 143.8 (C-2), 115.6 (C-3), 127.7 (C-4), 122.5 (C-5), 115.9 (C-6), 144.8 (C-7), 114.5 (C-8), 167.9 (C-9). EIMS m/z (rel. int.): 180 (M<sup>+</sup>, 100), 136 (18), 163 (180) (66).

**Caffeic acid methyl ester**  $4 - C_{10}H_{10}O_4$ , white amorphous solid, m.p 159°C. <sup>1</sup>H NMR ( $\delta_H$ , CDCl<sub>3</sub>): 7.02 (1H, d, J = 1.5 Hz, H-3), 7.07, (1H, dd, J = 1.5, 7.6 Hz, H-5), 6.90 (1H, d, J = 8.4 Hz, H-6), 7.61 (1H, d, J = 15.3Hz, H-7), 6.26 (1H, d, J = 16 Hz, H-8), 3.90 (3H,s). EIMS m/z (rel. int.): 194 (M<sup>+</sup>, 91), 179 (23), 177 (100).

| 1                |                                 | 2                     |                                    |                       |
|------------------|---------------------------------|-----------------------|------------------------------------|-----------------------|
| H/C              | $\delta_{\mathrm{H}}$           | $\delta_{\mathrm{C}}$ | δ <sub>H</sub>                     | $\delta_{\mathrm{C}}$ |
| 1                |                                 | 183.2                 |                                    | 189.4                 |
| 2                |                                 | 134.6                 | 6.11 (1H, d, <i>J</i> =10.7 Hz)    | 144.2                 |
| 3                | 6.84 (1H, s)                    | 151.6                 | 6.82 (1H, d, <i>J</i> = 10.7 Hz)   | 120.2                 |
| 4                |                                 | 183.16                |                                    | 183.9                 |
| 5                |                                 | 162.8                 |                                    | 152.1                 |
| 6                | 7.20 (2H, s)                    | 131.8                 | 7.41 (1H,s)                        | 125.8                 |
| 7                | 7.20 (2H, s)                    | 130.9                 |                                    | 140.8                 |
| 8                |                                 | 162.2                 |                                    | 158.9                 |
| 9                |                                 | 111.8                 |                                    | 122.7                 |
| 10               |                                 | 112.0                 |                                    | 113.5                 |
| 11               | 2.63 (2H, t, <i>J</i> = 7.6 Hz) | 26.6                  | 6.72 (1H, d, J=16 Hz)              | 121.6                 |
| 12               | 2.31 (2H, q, <i>J</i> = 7.6 Hz) | 29.8                  | 7.21 (1H, dd, <i>J</i> =16, 11 Hz) | 132                   |
| 13               | 5.1 (1H, m)                     | 129.2                 | 6.87 (1H, d, <i>J</i> = 11 Hz)     | 137.3                 |
| 14               |                                 | 133.8                 |                                    | 141.9                 |
| 15               | 1.69 (3H, s)                    | 17.9                  | 1.89 (3H,s)                        | 26.6                  |
| 16               | 1.60 (3H, s)                    | 25.8                  | 1.90 (3H,s)                        | 19.0                  |
| OCH <sub>3</sub> | _                               | _                     | 3.79 (3H,s)                        | 61.7                  |
| OH               | 12.47 (1H,s)                    | _                     | 12.48 (1H,s)                       | _                     |
| ОН               | 12.63 (1H,s)                    | _                     | _                                  | _                     |

## Anti-inflammatory activity

**Animals** – Adult male Sprague-Dawley rats (120 - 150 g) were used. They were acclimatized one week prior to use and allowed unlimited access to rat chow and water. Prior to the start of experiment, the animals were randomly divided into groups (six rats each).

Cotton pellet- induced granuloma bioassay - Cotton pellet  $(35 \text{ mg} \pm 1 \text{ mg})$  were impregnated with 0.2 ml solution of the test samples in CHCl<sub>3</sub> or acetone (containing 10 µmol) and the solvent was allowed to evaporate. Each cotton pellet was subsequently injected with 0.2 ml of an aqueous solution of antibiotics (1 mg of penicillin G and 1.3 mg of dihydrostreptomycin/ml). Two pellets were implanted subcutaneously, one in each axilla of the rat, under mild general anesthesia. One group of animals has received the standard reference indomethacin (ind.) and the antibiotics at the same doses and serves as positive control. Pellets containing only the antibiotics were similarly implanted in the negative control group. After seven days the animals were sacrificed and the two cotton pellets, with the adhering granulomas, were removed, dried for 48 hours at 60°C and weighed. The increment in dry weight (difference between the initial and final weights) was taken as a measure of granuloma  $\pm$ SE. This was calculated for each group and the percentage reduction in dry weight of granuloma from control value was also calculated. The ED<sub>50</sub> values were determined through dose response curve, using doses of 4, 7, 10 and 15 µmol of the test samples (Suleyman *et al.*, 1999) (Table 2).

**Carrageenin-induced paw edema in rats** – The paw edema was induced by subplantar injection of 50  $\mu$ l of 2% carrageenan solution in saline (0.9%). A similar volume of sterile saline was injected into the same region of the right hind paw. Ind. and the test materials were dissolved in DMSO and were injected subcutaneously in a dose of 10  $\mu$ mol/kg body weight, one hour prior to carrageenan injection. DMSO was injected to the control group. Paw volume was determined before each treatment (basal volume) and 2, 4 and 6 h after carrageenin injection using a plethysmometer (Model7150, Ugo Basile, Varese, Italy). The drugs were administered *po* or *ip* at 2 h before the injection of carrageenin (N = 6 for each group) (Winter *et al.*, 1962).

The percentage protection against inflammation was calculated as follows:

 $V_{c} - V_{d} / V_{c} \times 100$ 

Where  $V_c$  is the increase in paw volume in the absence of test compound (control) and  $V_d$  is the increase of paw

#### **Natural Product Sciences**

**Table 2**. The anti-inflammatory activity  $(ED_{50}, \mu mol)$  using Cotton pellet- induced granuloma and ulcerogenic effects of the root extracts and **2**.

| Test samples | ED <sub>50</sub> , μmol | % ulceration |
|--------------|-------------------------|--------------|
| Control      | -                       | 0.0          |
| Ind.         | 9.17                    | 100          |
| Light pet.   | 9.82                    | 0            |
| 70% EtOH     | 9.85                    | 0            |
| 50% EtOH     | 10.8                    | 0            |
| 2            | 9.91                    | 0            |

**Table 3.** The anti-inflammatory activity using Carrageenin-induced paw edema in rats of the root extracts and **2**.

| Test material | Increase in paw<br>edema (ml) ±SE <sup>1,2</sup> | % Protection | Activity relative to ind. |
|---------------|--|--------------|---------------------------|
| Control       | $0.87\pm0.041$                                   | 0.0          | 0.0                       |
| Ind.          | $0.21\pm0.021$                                   | 75.9         | 100                       |
| Light pet.    | $0.24\pm0.031$                                   | 72.4         | 95.5                      |
| 70% EtOH      | $0.28\pm0.030$                                   | 67.8         | 89.4                      |
| 50% EtOH      | $0.34\pm0.012$                                   | 60.9         | 80.3                      |
| 2             | $0.31\pm0.040$                                   | 64.4         | 84.8                      |

<sup>1</sup> SE: standard error.

<sup>2</sup> All data are significantly different from control (P < 0.001)

volume after injection of test compound. Data were expressed as the mean  $\pm$  S.E. Significant difference between the control and the treated groups was performed using ANOVA test and P values. The difference in results was considered significant when p < 0.05. The anti-inflammatory activity of the test compounds relative to that of ind. was also calculated. The results are recorded in Table 3.

**Ulcerogenic effects** – Ulcerogenic effect was determined by the reported method of (Komatsu *et al.*, 1973). The tested materials were administered to fasted rats having free access to drinking water. Four hours after administration of the materials, the rats were sacrificed, the stomach was removed and, after incision along the lesser curvature, rinsed with a tap soaked in warm ( $37^{\circ}$ C) saline and spread on a corkboard and pinned down. The mucosa of the glandular part of the stomach was inspected using a binocular microscope (10-fold magnification). The mucosal lesions were evaluated.

## **Results and Discussion**

The molecular formula of compound **1** ( $C_{16}H_{16}O_4$ , *m/z* 272 EIMS) together with the UV-VIS spectra in methanol and the bathochromic shift after addition of NaOAc indicated a naphthazarin core with a close resemblance to

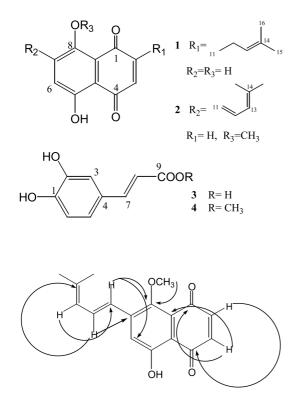


Fig. 1. HMBC correlations of 2.

alkanin moiety (Khatoon and Mehrota, 2000; Thomson, 1971). In the<sup>1</sup>HNMR spectrum of **1**, the two singlets at  $\delta_{\rm H}$ 12.63 and  $\delta_{\rm H}$  12.47 were assigned to two free hydroxyl groups at C-5 and C-8. The two singlet at  $\delta_{\rm H}$  7.2 (2H) and  $\delta_{\rm H}$  6.84 (1 H) were assigned to monosubstituted naphthazarin moiety. <sup>1</sup>H NMR spectrum also showed signals at  $\delta_{\rm H}$  5.1 (1H, m), 2.63 (2H, t, J = 7.6 Hz), 2.31 (2H, q, J = 7.6 Hz), 1.59 (s, 3H) and 1.69 (s, 3H) assigned for a methylated prenyl side chain. The alkyl substituent was further supported by the presence of a peak at m/z 203 (272-69) in the MS/MS spectra. The data of 1 were identical with those published for anhydroalkannin (dehydroxy shikonin) previously isolated from Alkanna hirsutissima, Alkanna tinctoria, Macrotemia cephalotes and Onosma heterophylla (Afzal and Tofeeg, 1975; Inoue et al., 1985; Papageorgiou, 1979; Han and Kaishim, 2008).

Compound **2** was isolated as orange rosette. Its UV-VIS spectrum in methanol showed maxima at  $\lambda_{max}$  300, 510, 539, and 584 nm consistent with a naphthazarin nucleus (Khatoon and Mehrota, 2000; Thomson, 1971). Compared with **1** one of the hydroxyl group signals was replaced by a methoxyl group [ $\delta_{\rm H}$  3.79 (3H, s),  $\delta_{\rm C}$  61.7] in the <sup>1</sup>H NMR spectrum of **2**. In the EIMS of **2** the fragments at m/z 269 (284 - CH<sub>3</sub>) and 253 (284 - OCH<sub>3</sub>) were in full support of the methoxyl group. The aromatic signals at  $\delta_{\rm H}$  6.11, 6.82 (each 1H, d, J = 10.7 Hz), and

7.41 (1H, s) were assigned for H-6, H-7, and H-3, respectively. These assignments were supported by HMBC correlation (Fig. 1). The extra carbon and proton signals were assigned to alkyl substituent (Table 1). The 3-bonds HMBC correlations of H-12 at  $\delta_{\rm H}$  7.21 with C-7 at  $\delta_C$  140.8; H-11 at  $\delta_H$  6.72 with both C-6 at  $\delta_C$  125.8 and C-8 at  $\delta_{C}$  158.9 (Fig. 1) unambiguously indicated the attachment of the alkyl moiety to C-7. The  $J_{11.12} = 16$  Hz indicated their trans-orientation. On the other hand the  $J_{12-13} = 11$  Hz was diagnostic for *cis*-orientation. The M<sup>+</sup> at m/z 284 in the EIMS spectrum was in full agreement with the proposed structure. Furthermore, the fragments at 203 (284-81) and 81 were in support for the presence of the alkyl substituent. The above discussion enable the identification of 2 as a new naphthoquinone derivative 5hydroxy 8-methoxy 2-(4-methylpent-1,3-dienyl)naphthalene-1,4-dione named as echiochiloquinone.

Evaluation of the anti-inflammatory activity of the different extracts and **2** was preformed applying two models of induced inflammations in experimental rats using ind. as a reference standard. In both the Cotton pellet- and carrageenin-induced paw edema all tested materials exhibited moderate anti-inflammatory activity as revealed from their  $ED_{50}$  values and percentage protection against inflammation. The light petroleum extract was the most active in the two assays followed by the 70% ethanol extract. The fact that **2** was less active than these two extracts clearly suggest that other active compounds still to be identified in the plant.

The tested extracts and compound 2 were evaluated for their ulcerogenic potential in rats. All the tested materials revealed a superior GI safety profiles (0 - 10% ulceration) in the experimental animals at oral doses of 30 mmol/kg/ day, when compared with ind., the reference standard drug, which was found to cause 100% ulceration under the same experimental conditions (Table 2). Gross observation of the isolated rat stomachs showed normal stomach architecture for all compounds.

## References

- Afzal, M. and Tofeeq, M., 5,8-dihydroxy-2,4-(4-methylpent-3-enyl)-1,4naphthoquinone and its 2-[4-methyl-1-(2-methylcrotonyloxy)pent-3enyl] analogue (shikonin angelate) from *Alkanna hirsutissima*. J. *Chem. Soci. Perkin I.* 1334-1335 (1975).
- Bergaoui, A., Hammami, S., Ben Janet, H., Ciavatta, L., Cimino, G., and Mighri, Z., Isolation of two O-heterosides from Echiochilon fruticosum plant growing in Tunisia. Journal de la Société Algériénne de Chimie. 14, 235-244 (2004).
- Boshra, S., Omayed biosphere reserve and its hinterland, EGYPT, www.UNESCO.org/mab/ecosyst/dryland.
- Hammami, S., Ben Janet, H., Bergaoui, A., Ciavatta, L., Cimino, G., and

#### **Natural Product Sciences**

Mighri, Z., Isolation and structure elucidation of a flavanone glycoside and vomifoliol from *Echiochilon fruticosum* growing in Tunisia. *Molecules*. **9**, 602-608 (2004).

- Han, J. and Kaishim, B., Antioxidants from a chinese medicinal herb Lithospermum erythrorhizon. Food Chem. 106, 2-10 (2008).
- Hashem, A., Phytochemical study of *Echiochilon fruticosum* Desf. *Bulletin Of the National Research Centre* (EGYPT). 28, 151-162 (2003).
- Inoue, K., Akaji, M., and Inouye, H., Quinones and related compounds in higher plants. XXI. New findings on the proton and carbon-13 nuclear magnetic resonance spectra of shikonin. *Chem. Pharm. Bull.* 9, 3993-3997 (1985).
- Karyagina, T., Arumanyan, V., Timchenko, T., and Airamshyili, D., Antimicrobial activity of shikonin preparations. *Pharmaceutical Chemistry Journal, (translation of khimiko-farmateyticheskii Zhurnal).* 35, 435-436 (2001).
- Khatoon, S. and Mehrota, S., Pharmacognostical study of Japanese drug 'Nan-Shikon', root of *Arnebia euchroma* growing in India. *Natural medicines*. 54, 171-177 (2000).
- Komatsu, T., Awata, H., Sakai, Y., and Yamanoto H., Additional data on anti inflammatory agent, ID-955. Arzneim Forsch (Drug research). 23, 500-503 (1973).
- Kourounakis, A.P., Assimopoulou, A.A.N., Papageorgiou, A.G., Gavalas, A., and Kourounakis, P.N., Alkanin and shikonin effect on free radicle process and on inflammation-A preliminary pharmacochemical investigation. *Archiv der Pharmazie*. 6, 262-266 (2002).
- Leyva, A., Pessoa, C., Boogaert, F., Sokaroski, R., Lemos, T.L.G., Wetmore, L.A., Huruta, R.R., and Moraes, M.O., Oncocalyxones A and C, 1,4-anthracendiones from *Auxemma oncocalyx*: Comparison with anticancer 1,9-anthracenediones. *Anticancer Research*. 20(2A),

1029-1031 (2000).

- Li, T.S.C., Chinese and related North American herbs: Phytopharmacology and therapeutic values. CRC press, Boca Raton, Florida (2002).
- Mats, M.N., Korkhov, V.V., Bogatkina, V.F., Ovacharenko, S.N., Kozhina, I.S., Shukhobodskii, B.A., and Nosova, L.I., Contraceptive properties of galenics from several plants of the family Boraginaceae. *Rastitel'nye Resursy.* 18, 87-91 (1982).
- Papageorgiou, V.P., <sup>1</sup>HNMR spectra of naturally occuring isohenylnaphthazarin pigments. *Planta Med.* 185-187 (1979).
- Roeder, E., Medicinal plants in Europe containing pyrrolizidine alkaloids. *Pharmazie*. **50**, 83-98 (1995).
- Singh, B., Sharma, M.K., Meghwal, P., and Singh, S., Anti inflammatory activity of shikonin derivatives from *Arnebia hispidissima*. *Phytomedicine*. **10**, 375-380 (2003).
- Suleyman, H., Demirezer, L.O., Kurunzum, A., Banoglu, Z.N., Gocer, F., and Ozbakir, G., Anti inflammatory effect of the aqueous extract from *Rumex patientia L.* roots. *J. Ethnopharmacol.* 65, 141-148 (1999).
- Täckholm, V., *Student's flora of Egypt*, 2<sup>nd</sup> Ed, Cairo University Press (1974).
- Thomson, R.H., Naturally occuring quinones. Academic Press (1971).
- Winter, C.A., Risley, E.A., and Nuss, G.W., Carrageenin induced oedema in hind paw of the rat as assay for anti inflammatory drugs. *Experimental Biology and Medicine*. 111, 544-547 (1962).
- Zhen, H., Liao, M., and Guo, J., Contraceptive constituents from Arnebia euchroma (Royle) Johnst. Tianran Chanwu Yanjui Yu Khaifa. 14, 1-4 (2002).

(Accepted March 16, 2009)