THE EFFECT OF THE REMOVAL OF CHONDROITIN SULFATE ON BOND STRENGTH OF DENTIN ADHESIVES AND COLLAGEN ARCHITECTURE

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ABSTRACT

Proteoglycan is highly hydrophilic and negatively charged which enable them attract the water. The objective of study was to investigate the effects of Proteoglycan on microtensile bond strength of dentin adhesives and on architecture of dentin collagen matrix of acid etched dentin by removing the chondroitin sulphate attached on Proteoglycan. A flat dentin surface in mid-coronal portion of tooth was prepared. After acid etching, half of the specimens were immersed in 0.1 U/mL chondroitinase ABC (C-ABC) for 48 h at 37°C, while the other half were stored in distilled water. Specimens were bonded with the dentin adhesive using three different bonding techniques (wet, dry and re-wet) followed by microtensile bond strength test. SEM examination was done with debonded specimen, resin-dentin interface and acid-etched dentin surface with/without C-ABC treatment.

For the subgroups using wet-bonding or dry-bonding technique, microtensile bond strength showed no significant difference after C-ABC treatment (p > 0.05). Nevertheless, the subgroup using rewetting technique after air dry in the Single Bond 2 group demonstrated a significant decrease of microtensile bond strength after C-ABC treatment. Collagen architecture is loosely packed and some fibrils are aggregated together and relatively collapsed compared with normal acid-etched wet dentin after C-ABC treatment. Further studies are necessary for the contribution to the collagen architecture of noncollagenous protein under the various clinical situations and several dentin conditioners and are also needed about long-term effect on bond strength of dentin adhesive. [J Kor Acad Cons Dent 35(3):211-221, 2010]

Key words: Chondroitin Sulphate, Collagen Architecture, Dentin Adhesive, Noncollagenous Protein, Microtensile Bond Strength

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I. Introduction

Easy-to-use, efficient and long durable bonding to dentin is a goal that has been endeavored by many researchers, manufacturers and clinicians for a long time. The modern dentin adhesive systems can be classified as the approach to the smear layer: Selfetch approach, Etch-and-Rinse approach.¹⁾ The self-

*Corresponding Author: **Kyoung-Kyu Choi** Department of Conservative Dentistry, School of Dentistry, Kyung Hee University 1 Hoegi-dong, Dongdaemun-gu, Seoul, 130-701, Korea Tel: 82-2-958-9337 Fax: 82-2-960-5108 E-mail: choikkyu@khu.ac.kr etch approach aims to modify the smear layer and incorporate it in the bonding process that involves either two or one-step application, whereas the Etchand-rinse adhesion strategy involves at least two or three steps with successive application of the acid etchant that result in complete removal of smear layer followed by the primer and application of the adhesive resin. At dentin, phosphoric acid treatment opens the dentin tubules by removing the smear plugs and exposes a microporous network of collagen that is almost free of hydroxyappatite, which the bonding monomers can impregnate. This impregnation process makes a complex hybrid layer formed by residual mineral crystals, collagen fibrils and noncollagenous protein.²⁾ Ideally, a similar volume of adhesive resin should replace the volume of mineral removed from dentin during the demineralization process. However, a decreasing gradient of resin monomer diffusion within the acid etched dentin results in incompletely infiltrated zones along the bottom of the hybrid layer in the demineralized zone of dentin by discrepancy between the depth of acid etching and resin infiltration.³⁾ This incompletely infiltrated zone produce the nanoleakage that may be the pathway that cause degradation of the resindentin interface by hydrolysis of collagen fibrils and polymerized resin.⁴⁻⁶⁾

In order to enhance the optimal formation of a hybrid layer, demineralized dentin has to be kept moist and collagen network maintained in a fully expanded state after acid etching or if dried, need to be re-wetting.⁷⁾ The presence of water play a crucial role in preventing the demineralized surface collagen from collapsing following dehydration.⁸⁾ When airdried demineralized dentin is re-wet with water, the collagen matrix re-expands and completely recovers the original dimensions to the levels of the original hydrated state⁷⁾ because spaces between fibrils is refilled with water and hydrogen bond between collagen fiber that restrain the adhesive resin infiltration is broken by water.^{10,11)}

Unlike enamel, of which the organic matrix makes up only about 0.4–0.6%, dentin contains approximately 20% organic component in weight.¹²⁾ Type I collagen constitutes about 90% of the dentin organic matrix and the remaining 10% of dentin organic matrix is composed by a number of non-collagenous proteins (NCPs). Proteoglycans (PGs) and phosphoproteins comprise of the main components of the dentin noncollagenous protein.¹³⁾

PGs are composed of a core protein and one or more glycosaminoglycans (GAGs) chains covalently attached to it. The side chains of GAGs attached to the core proteins demonstrate the functional characteristics of PGs¹⁴⁾ and two of the most prominent PGs in dentin are decorin and biglycan, which are relatively abundant and usually carry chondroin sulphate GAG chain.¹⁵⁾

PGs are highly hydrophilic and negatively charged which enable them attract the water due to carboxyl

and sulfate groups present in GAG chains and have electrostatic interaction with collagen,¹⁵⁻¹⁷⁾ also are susceptible to degradation both by matrix metalloproteinases (MMPs),¹⁸⁾ bacterial enzymes,²⁰⁾ and show the decreased antigenicity when 35% phosphoric acid was applied on dentin surface for more than 30s.²¹⁾ Those characteristics of PGs may play an important role in the maintenance of dentin collagen architecture and contribute to the bond strength of dentin adhesive and durability of bonding interface.

The aim of this study was to investigate the effects of Proteoglycan on microtensile bond strength of dentin adhesives and on architecture of dentin collagen matrix of acid etched dentin by removing the chondroitin sulphate that is most abundant glycosaminoglycan in dentin.

${\ensuremath{\mathbb I}}$. Materials and methods

1. Specimens Preparation

Seventy-two, caries-free, recently extracted human third molars were stored in a 0.5% chloramines T aqueous solution at 4° and used within 1 month following extraction. A flat dentin surface in mid-coronal portion of tooth was prepared perpendicular to the longitudinal axis of each tooth using a slow-speed Isomet diamond saw (Buehler Ltd., Lake Bluff, IL, USA) under copious water-cooling. Each dentin surfaces were then polished with a 600-grit silicon carbide paper under running water to creat a bonding surface with a standardized smear layer in mid-coronal dentine that was devoid of enamel. The teeth were randomly divided into two groups (N=36)according to the dentin adhesives used: Single Bond 2 (3M ESPE, St. Paul, MN, USA), a water-containing, ethanol-based etch-and-rinse adhesive, and One Step plus (Bisco Inc., Schaumburg, IL, USA), a water-free, acetone-based etch-and rinse adhesive. Composition, batch number and manufacturers of the tested adhesives are listed in Table 1. Each dentin surface to be bonded was etched with a 37% phosphoric acid gel (3M ESPE, St. Paul, MN) for 15 s to create a 5-8 µm thick zone of completely demineralized dentin. The etched dentin surface was then rinsed with deionized water for 10 s and kept visibly

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Adhesives	Composition	Batch #	Manufacturer	
Single Bond 2	Polyalkenoid acid copolymer, colloidal silica,	7KX	3M ESPE, St. Paul,	
	ethanol, water, photo-initiator		MN, USA	
One Step plus	BPDM, Bis-GMA, HEMA, acetone,	700005622	Bisco Inc., Schaumburg,	
	photo-initiator, fluoroaluminosilicate glass fillers	100003022	IL, USA	
			1	

Table 1. Composition, batch number and manufacturers of dentin adhesives tested

BPDM: biphenyl dimethacrylate Bis-GMA: bisphenol A diglycidyl methacrylate HEMA: 2-hydroxyethyl methacrylate CQ: camphoroquinone

moist to prevent collapse of the collagen matrix. Prior to the bonding procedures, the specimens of each group were randomly divided into two subgroups according to solution that was used for the pretreatment of acid-etched dentin surfaces: Half of the specimens in each group were immersed in 0.1 U/mL chondroitinase ABC (C-ABC) (Sigma chemical, St. Lois, MO.) in 0.1 M Tris/Acetate (pH 7.5) for 48 h (renewed after 24 h) at 37°C, while the other half were stored in distilled water under the same conditions. After incubation for 48 h, all specimens were thoroughly rinsed with distilled water for 5 min and bonded with the respective adhesive using three different bonding techniques (N=3 for each subgroup): I) moist dentin was blot dried; II) moist dentin was air dried for 15 s; and III) moist dentin was air dried for 15 s, re-wetted for 30 s and subsequently blot dried. The dentin adhesives were used in accordance with manufacturer's instructions. This was followed by incremental placement of a composite resin (Z-250, 3M ESPE, St. Paul, MN, USA) to a height of 5 mm. Light curing was performed with a Spectrum 800 halogen light-curing unit (Dentsply DeTrey, Konstanz, Germany) with power output of 600 mW/cm².

2. Microtensile Bond Strength Test

After water-storage at 37°C for 24 h, each tooth was sectioned occluso-gingivally into 1 mm thick serial slabs (N=6 for each teeth) by means of the slow speed diamond saw with water-cooling. The central slab was used for SEM examination, while the others were further sectioned into 1×1 mm beams, using the "non-trimming" technique of the microtensile testing. The two longest beams from each slab were selected for tensile testing, yielding 30 beams per subgroup. Specimens that failed prematurely during sectioning were recorded in each subgroup, and the beam dimension of each intact specimen was measured with digital caliper to the nearest 0.01 mm to provide the bonding surface area for calculating microtensile bond strength. Each beam was fixed by their ends to the microtensile bond testing device with cyanoacrylate glue (Zapit, Dental Ventures of America, Corona, CA, USA) and tested in tension using material tester (EZ tester, Shimadzu Co., Kyoto, Japan) at a crosshead speed of 1.0 mm/min. Microtensile bond strength was determined in MPa by calculating the quotient of maximum load and the its corresponding bonding area.

Statistical analysis was performed separately for each dentin adhesive since the intention of the study was not compare bond strengths of the two adhesives. Bond strength data was normally distributed (Kolmogorov-Smirnov test) and exhibited equal variances (Levene test). Thus, the data were analyzed using a two-way analysis of variance (ANOVA) for evaluating the effects of enzyme digestion and dentin condition, followed by Duncan test for post-hoc comparisons. Statistical analysis was performed using SPSS 11.0 for window (SPSS Inc., Chicago, IL, USA) with the level of significance set at 5%.

3. Field Emission-Scanning Electron Microscope (FE-SEM)

Two representative debonded beams of the mean bond strength of each subgroup were prepared for the observation of the fracture pattern. Moreover, the central slabs of each specimen were serially polished with 600, 1200, 1500 grit silicon carbide paper under running water, followed by a 5sec 37% phosphoric acid gel (3M ESPE, St. Paul, MN, USA) treatment and a 10-min 5.25% sodium hypochlorite treatment. The selected beams and slabs were dried for 24 h within the drierite desiccator (Hammond Drierite Company LTD., Xenia, Ohio, USA).

Another eight caries-free, human third molars were used to evaluate change of the collagen architecture after removal of the chondroitin sulfate in proteoglycan. Specimen preparation was carried out according to the method proposed by Perdigao et al.²²⁾ in order to avoid artifacts formation due to dehydration shrinkage of specimens and their observation under a high to ultra-high vacuum.²¹⁾ Fourteen 0.9 mm-thick dentin disks parallel to the occlusal surface were obtained from mid-coronal dentin that was devoid of enamel by low speed diamond saw. A transverse groove was made in the bottom surface of each dentin disk to facilitate splitting after specimen preparation, while its upper surface was etched with a 37% phosphoric acid gel for 15 s, rinsed with water for 10 s and blot dried. All the specimens were randomly distributed into two groups (N=4): one group was then treated with chondroitinase-ABC (C-ABC) and the other group was stored in distilled water as the same way described in the microtensile test section. After C-ABC/distilled water storage, specimens of each group were subdivided into two subgroups (N=2) according to dentin moisture, namely wet, and rewet. The specimens were immediately immersed in 2.5% glutaraldehyde in 0.1 M sodium cacodylate buffer at pH 7.4 for 4 hours at 4°C. After fixation, the disks were rinsed with 0.1 M sodium cacodylate buffer at pH 7.4 for 1 h, with three changes, followed by distilled water for 5 min. After dehydration in ascending ethanol series (50–100%), the specimens were HMDS-dried. Specimens for examination of sagittal surface of acid-etched dentin were split with forceps before mounting the stubs.

After dried, all of SEM specimens were mounted in aluminum stubs with carbon tape. They were sputter-coated with platinum by means of sputter-coating Unit (E-1045, Polaron, Germany) and examined with a FE-SEM (SUPRA 55, Zeiss, Germany) at an acceleration voltage of 5-10 kV.

I. Result

1. Microtensile Bond Strength

Premature failure (PF) was observed in all groups of air-dried dentin and Single Bond 2 groups pretreated with C-ABC. The data for microtensile bond strength and percentage of premature failure of each experimental group are summarized in Table 2. For One Step plus, the bond strengths in descending order were wed-bonding subgroups, subgroups using rewetting technique, and dry-bonding subgroups. The same was true when the Single Bond 2 group was taken into consideration. Two-way ANOVA indicated a significant difference in the bond strengths of each dentin adhesives among different moisture conditions of dentin ($p \langle 0.05$), whilst there was no significant difference between dentin with and without C-ABC treatment (p > 0.05), and showed that the interaction between dentin condition and C-ABC Treatment was significant for Single Bond 2 only (p = 0.002), not for One Step plus (p > 0.05).

For the subgroups using wet-bonding or dry-bonding technique, microtensile bond strength showed no significant difference after C-ABC treatment (p \rangle

Table 2. Results of microtensile bond strength value (MPa \pm SD) of dentin adhesives and premature failure (PF) percentages.

		-			*		
	Dentin without C-ABC Treatment			Dentin with C-ABC Treatment			
Adhesive	Wet (PF)	Dry (PF)	Rewet (PF)	Wet (PF)	Dry (PF)	Rewet (PF)	
One Step plus	$41.0\pm9.4^{\rm a}$	$11.0 \pm 3.5^{\text{b}}$ (69)	$35.3\pm8.4^{\rm a}$	$37.4 \pm 8.8^{\text{a}}$	$17.0 \pm 6.6^{\circ} (47)$	$36.2 \pm 5.8^{\text{a}}$	
Single Bond 2	$38.2\pm7.5^{\text{a}}$	$9.5\pm 6.2^{\rm d}$ (43)	$31.4\pm7.6^{\rm\scriptscriptstyle a,b}$	$33.0 \pm 8.5^{\circ}$ (36)	$15.8 \pm 7.9^{ m c,d}$ (31)	$23.7 \pm 9.0^{\text{b.c}}$ (11)	
Different superscript letters indicate statistically significant differences within the each dentin adhesive							



Figure 1. Results of microtensile bond strength value (MPa) of each dentin adhesives.

W: wet dentin not treated with enzyme (C-ABC) was blot dried. D: dentin not treated with enzyme (C-ABC) was air dried for 15 s. RW: dentin not treated with enzyme (C-ABC) was air dried for 15 s and then re-wetted for 30 s with wet gauze and blot dried. E-W: wet dentin treated with enzyme (C-ABC) was blot dried. E-D: dentin treated with enzyme (C-ABC) was air dried for 15 s. E-RW, dentin treated with enzyme (C-ABC) was air dried for 15 s and then re-wetted for 30 s with wet gauze and blot dried for 15 s. E-RW, dentin treated with enzyme (C-ABC) was air dried for 15 s and then re-wetted for 30 s with wet gauze and blot dried.

0.05). Nevertheless, the subgroup using rewetting technique after air dry in the Single Bond 2 group demonstrated a significant decrease of microtensile bond strength after C-ABC treatment compared with its counterpart without C-ABC treatment (p < 0.05).

Premature failure (PF) was observed in groups of dry-bonding technique in both tested dentin adhesives and C-ABC pretreated groups in Single Bond 2.

2. Field Emission-Scanning Electron Microscope (FE-SEM)

Representative FE-SEM images of fractured specimens were selected and shown in Figure 2. Images of resin-dentin interface were shown in Figure 3 and image of collagen architecture were shown in Figure 4, 5. For Single Bond 2, failure occurred within the adhesive resin to the bottom of the hybrid layer and dentinal tubules were occluded with the resin tag (Figure 2a). For One Step Plus, failure was occurred within the densely formed hybrid layer (Figure 2c). After C-ABC treatment, failures were occurred at the bottom of hybrid layer with resin tag exclusion from dentinal tubules at both dentin adhesives (Figure 2b, 2d). Regardless of dentin adhesive, thickness of hybrid layer was increased in C-ABC treated group compared with not treated group when adhesives were used to wet dentin (Figure 3).

Regarding FE-SEM observation with high magnification of the collagen architecture, the collagen of normal acid-etched wet dentin is uniformly arranged and compact interfibrillar space and fully expanded state (Figure 4a, 4b, Figure 5a). After C-ABC pretreatment, collagen architecture is loosely packed and some fibrils are aggregated together and relatively collapsed compared with normal acid-etched wet dentin (Figure 4c, 4d, Figure 5b, 5c).

IV. Disussion

In general, PGs are considered to play structural and metabolic functional roles in both soft and mineralized tissues of body, and are present in a wide variety of non-vertebrate and vertebrate species.²⁴⁾ They consist of glycosaminoglycan (GAG) chains, comprising repeating disaccharides units, covalently linked to a protein.¹⁵⁾ The GAG chains are mainly classified into five types: hyaluronan, heparin/heparan sulphate, chondroitin sulfate, der-

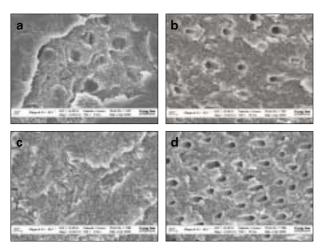


Figure 2. Field Emission–Scanning Electron Microscope (FE–SEM) observations of the fractured surfaces in dentin side after microtensile bond strength testing (Mag: $\times 10,000$).

a. Dentin side produced by Single bond 2 without C-ABC treatment. Failure was occurred within the adhesive resin to the bottom of the hybrid layer. Dentinal tubules were occluded with the resin tag.

b. Dentin side produced by Single bond 2 with C-ABC treatment.Failure was occurred at bottom of hybrid layer with resin tag exclusion.c. Dentin side produced by One Step Plus without C-ABC treatment.Failure was occurred from the top of the hybrid layer to the bottom.The hybrid layer is densely formed.

d. Dentin side produced by One Step Plus with C-ABC treatment. Failure was occurred at bottom of hybrid layer with resin tag exclusion.

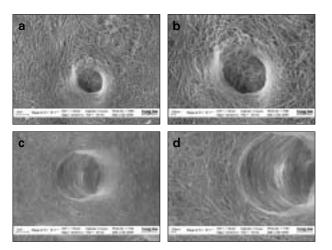


Figure 4. Field Emission–Scanning Electron Microscope (FE-SEM) observations of acid–etched dentin surfaces.

a. Collagen fibril architecture of acid-etched wet dentin without C-ABC pre-treatment.

b. A high-magnification view. Well-defined collagen fibrils are arranged with compact interfibrillar space.

c. Collagen fibril architecture of acid-etched wet dentin with C-ABC pre-treatment.

d. A high-magnification view. Collagen architecture is loosely packed with irregular diameter and some fibrils are fused together.

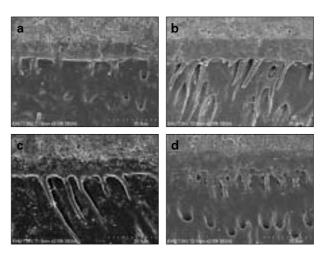


Figure 3. Field Emission–Scanning Electron Microscope (FE–SEM) observations of the resin–dentin interfaces (Mag: ×2,000).

a. Photomicrograph of resin-dentin interface of Single Bond 2 without C-ABC treatment. Hybrid layer of about 6μ thickness is observed.

b. Photomicrograph of resin-dentin interface of Single Bond 2 with C-ABC treatment. A thicker hybrid layer about $10\mu m$ thickness can be seen.

c. Photomicrograph of resin-dentin interface of One Step Plus without C-ABC treatment. Hybrid layer of about $4\sim 6\mu m$ thickness is observed. d. Photomicrograph of resin-dentin interface of One Step Plus with C-ABC treatment. A thicker hybrid layer about $10\mu m$ thickness can be seen like Single bond 2.

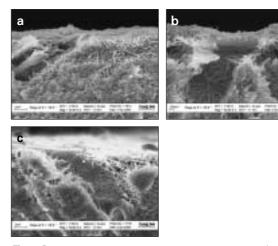


Figure 5. Field Emission–Scanning Electron Microscope (FE–SEM) observations of the vertically fractured dentin surface.

a. Collagen fibrils vertical architecture of acid-etched wet dentin without C-ABC treatment. Collagen network is fully expanded state.

b. Collagen fibrils vertical architecture of acid-etched wet dentin with C-ABC treatment. The collagen network is slightly collapsed compared with wet dentin without C-ABC treatment.

c. Collagen fibrils vertical architecture of acid-etched re-wetted dentin with C-ABC pre-treatment. Collagen network is relatively collapsed compared with normal acid-etched wet dentin.

matan sulphate and keratan sulphate, previously illustrated in predentin, dentin, 25,26) cementum, periodontal ligament²⁷⁾ and cementum-dentin junction²⁷⁾ of teeth. Biochemical and immunohistochemical studies on predentin and dentin PGs indicated that PGs predominantly belong to the small leucine-rich PG (SLRP) family¹⁴⁾ and the distinctive feature of the SLRP is the presence of between 7 and 24 leucinerich tandem repeats in the core protein. Most prominent member of dentin PGs are decorin and biglycan, which are identified in various tissues such as skin, tendon, gingival and bone. Decorin usually carries one GAG chain at the fourth amino acid at the Nterminal where chondroitin sulfate (CS) predominates within mineralized tissue and dermatan sulphate (DS) in soft tissues. Biglycan usually two GAG chains with CS being predominant in mineralized tissue like decorin. Because immunohistochemical labeling of decorin was more intensive than biglycan in dentin and significantly higher over the odontoblast process compared to the intertubular dentin¹⁴⁾ and decreasing gradient toward superficial dentin.²⁴⁾ The removing effect of chondroitinase ABC (C-ABC) that acts on chondroitin 4-sulfate, chondroitin 6-sulfate, and dermatan sulfate, and acts slowly on hyaluronate (Product information, Sigma chemical, St. Lois, MO.), to dentin might be more prominent in the deep dentin.

Dentin is composed of collagen based organic matrix with hydroxyapatite reinforcement that removed by acid etching in total-etch technique. Exposed collagen fibrils are linked by protein core of small leucine-rich proteoglycans (PGs) at the gap zone and they are interacting with each other via their glycosamonoglycans (GAGs) side chains, like a step of the ladder.^{29,30)} The unbranched polysaccharide chains of the GAG, which contain many carboxylate and sulfate group, make the GAG chains polyanionic with a high affinity for water molecules. These GAG-water hydrogel and orthogonal alignment of GAG with collagen fibrils make PGs behave like a stress transfer and distribute the functional tensile and compressive load.^{31,32)} But in current study, digestion of acid-etched dentin matrix with C-ABC for 48 h did not make a significant difference in microtensile bond strength of dentin adhesives regardless of dentin condition, except Single bond 2 with re-wet technique. Thus, it can be speculated that removal of chondroitin sulphate from a hydration state of acid-etched dentin collagen matrix may not have a significant influence on bond strength of dentin adhesives tested.

According to our knowledge, there have been few studies investigating the effect of C-ABC on dentin and it is still a matter of controversy.^{33,34)}

Mazzoni *et al.* showed the increased values of microtensile bond strength, reduced nanoleakage expression after C-ABC treatment of the dentin surface. Acetone based Prime and Bond NT increased almost doubled (92%) while ethanol-water based Adper Scotchbond Multi-Purpose increased about 28%. In their study, field emission in-lens scanning electron microscopy images of C-ABC treated dentin showed the wider empty spaces between the collagen fibrils, which were consistent with our findings (Figure 4d) and authors thought that those empty spaces act as the diffusion channels for more optimal monomer infiltration making the homogenous hybrid layer.

Elliott and co-workers studied the absence of PGs in transgenic mouse-tail tendons, and their result demonstrated that the collagen network was loosely packed with irregular contours and fibrils fused together.³⁵⁾

On the contrary, C-ABC treated dentin revealed a significant loss of collagen fibril architecture and bond strengths for the rewetted group in the study of Pereira *et al.*³⁴⁾ When dentin was kept moist, there is no significant difference of bond strength after C-ABC treatment like our study while bond strength to dentin digested with C-ABC had significantly lower bond strength compared to its control when the dentin was dried and rewetted. In the SEM analysis of pereira's experiment, loss of interfibrillar space was observed for the C-ABC rewetted group and this might be indicated the role of proteoglycan during the rewetting the dried dentin collagen. This decrease of bond strength in rewet dentin condition was coincided with our Single Bond 2 group. When the demineralized dentin was air-dried and rewetted by water, bond strength was returned to the range obtained on moist dentin even if there is a difference of rewetting time according to the type of solvent used in dentin adhesives.³⁶⁾ It could be possible because the water that is able to break the interpeptide H-bonds between the collapsed dentin collagen make the interfibrillar space that required as diffusion channels for resin infiltration re-expanded.³⁷⁾ During the re-expanding process with water, if hydrophilic proteoglycan between collagen networks was depleted by C-ABC or endogenous matrix metalloproteinase (MMP)^{18,19)} and bacterial enzyme²⁰⁾, this could restrict the breaking ability of interpeptide H-bond of water to the collapsed collagen network, thus, decreased the bond strength in rewetting dentin condition.

When the dentin was adhered with One-step plus, bond strength of C-ABC treated dentin after rewetting with water showed non-significant change compared with non C-ABC pretreated group. These differences of bond strength after rewetting the C-ABC treated dentin between dentin adhesives may be related with the solvent used in each dentin adhesives. Water-chasing ability of acetone³⁸⁾ may explain why acetone-based One step plus showed the more stable results regardless of the dentin condition except air-dried dentin surface than ethanol-based Single Bond 2.

Dehydration of dentin causes the collagen matrix collapse that results in the interfibrillar space that are needed as diffusion channels for dentin adhesive disappear. The collapsed dentin matrix does not expand when dentin adhesives are applied and eventually show the reduced bond strength and clinical symptoms of patients.¹⁰⁾ When the dentin surface was air-dried after the acid-etching procedure, all of dentin adhesives tested in our study also showed the significantly decreased bond strength compared with the bond strength obtained in the wet surface and re-wetted surface regardless of treatment of C-ABC.

A comparison of the FE-SEM of the dentin side of the failed bonds of dentin adhesives showed that adhesive failure including the resin tag exclusion mostly occurred in C-ABC treated wet dentin while mixed failure occurred in normal wet dentin even though there is no significant difference of bond strength between the normal wet dentin and C-ABC treated wet dentin. This may indicate the incomplete resin impregnation to the base of hybrid layer and into the dentinal tubule. Such a characteristic was more prominent for Sing Bond 2 that showed the premature failure for the wet group and re-wet group after C-ABC treatment. According to the atomic force microscope study of S.P. Ho et al.,¹⁷⁾ the use of C-ABC removed GAGs from CDJ resulting in breakdown of the collagen-PG network. This caused a collapse of the collagen network observed under dry condition and, under wet condition, the loss of structure resulted in an increase in height of the collagen architecture. This increasing height of the collagen network was shown indirectly through the examination of the representative specimens perpendicularly sectioned the dentin-resin interface in our experiment (Figure 3b, 3d). But vertical architecture of collagen after C-ABC pre-treatment is slightly reduced in the FE-SEM observation of fracture surface (Figure 5b). This difference might be caused by collapse of the chondroitin sulfate depleted collagen network which does not occupied by adhesive resin and does not connected by PGs between collagen fibers during the FE-SEM specimen processing.

Higher rate of premature failure of Single Bond 2 compared with One Step plus can be contributed to polyalkenoic acid copolymer as well as different solvents used in each system. Because this copolymer was found to provide better moisture stability to ScotchBond MultiPurpose (SBMP) (3M ESPE, St. Paul. MN. USA).³⁹⁾ it was added to the SBMP primer and Single Bond 2. Due to the significant difference in molecular weight between HEMA (MW = 130, Sigma, St. Louis, MO) and the copolymer (MW =14,000 to 20,000), while HEMA may diffuse more readily into the collagen network than the copolymer, polyalkenoic copolymer may have been kept at the surface of hybrid layer,⁴⁰⁾ resulting in amorphous electron-dense phase and consequently, may interfere with the diffusion of adhesive solution to optimize the hybridization, especially in C-ABC treated dentin which collagen network was partially coagulated.

In summary, within the limitation of this study, it is concluded that the selective removal of chondroitin sulphate in acid-etched dentin does not affect the bond strength of dentin adhesives used to wet dentin surface. But, once acid-etched dentin surface was dried, resulting in dentin collagen matrix collapse, and re-wetted with water in clinical situation, impairment of the GAGs may disturb the full reexpansion of collagen matrix and, as a result, the bond strength to dentin may be decreased according to the dentin adhesives.

The study about the role of noncollagenous protein in the adhesive dentistry is very few and just commenced. Further studies are necessary for the contribution to the collagen architecture of noncollagenous protein under the various clinical situations and several dentin conditioners and are also needed about long-term effect on bond strength of dentin adhesive.

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비교원성 단백질이 상아질 접착제의 결합강도와 교원질의 형태에 미치는 영향

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본 연구는 상아질의 비교원성 단백질을 chondroitinase ABC (C-ABC)를 이용하여 제거함으로써 비교원성 단백질의 제거 가 상아질 접착제의 미세인장결합강도와 교원질망의 형태에 미치는 영향을 상아질의 다양한 습윤상태에 따라 평가하고자 시행 하였다.

비교원성 단백질의 상아질접착제의 미세인장강도에 대한 영향을 평가하기 위해 제 3대구치의 상아질을 노출시키고, 두 군으 로 나누고 한 군은 C-ABC, 다른 군은 증류수를 37℃에서 48시간 동안 적용한 후, 상아질의 습윤상태(wet, dry 및 re-wet)와 상아질 접착제(Single Bond 2, One Step Plus)를 다르게 이용하여 복합레진을 수복하였다. 24시간 후 가로 1 mm, 세로 1 mm의 시편을 제작하고 미세인장강도를 측정하였다.

상아질 교원질의 형태변화를 관찰하기 위하여 상아질 시편에 산부식을 시행하고 C-ABC 적용 후, 시편을 제작하였고 미세인 장강도 측정후 파괴된 접착면의 파괴양상과 각 접착제의 접착계면 관찰을 위하여 FE-SEM 관찰하였다. C-ABC 처리여부와 관계없이 습윤한 상아질면에 접착한 군은 모든 접착제에서 통계학적으로 유의성있는 미세인장결합강도의 차이를 나타나지 않 았다(p > 0.05). C-ABC를 적용하였을 경우, Single Bond 2에서는 재수화한 상아질면에 접착한 군이 습윤한 상아질면에 접 착한 군에 비해 미세인장결합강도가 감소하였다(p < 0.05). FE-SEM 관찰결과, C-ABC를 적용후에는 접착성 파괴가 주로 일어났으며, 교원질 섬유간 거리가 증가하였으며 부분적으로 교원질 섬유들간에 응집된 양상이 관찰되었다.

주요단어: 비교원성 단백질, Chondroitin Sulphate, 교원질 구조, 상아질 접착제, 미세인장접착강도