

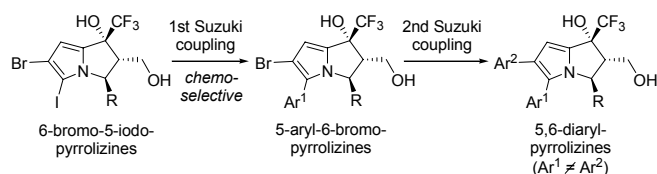
## Chemoselective Suzuki Cross-Coupling Reactions of Chiral Pyrrolizines

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Pyrrolizines have potent cytostatic effects and are thus potentially useful for the development of antitumor and antiviral agents.<sup>1</sup> In particular, chiral trifluoromethylated pyrrolizines would play a unique and significant role in medicinal chemistry because the introduction of a trifluoromethyl group into pharmaceuticals often enhances their physical and/or chemical properties.<sup>2</sup> Recently, we reported that the enantio- and diastereoselective organocatalytic cascade conjugate addition-aldol reactions of a series of 2-trifluoroacetylpyrroles to  $\alpha,\beta$ -unsaturated aldehydes afford highly functionalized chiral pyrrolizines that bear a trifluoromethyl group at the stereogenic center in good yields, high enantioselectivities, and excellent diastereoselectivities.<sup>3</sup> Using the chiral pyrrolizines as the cascade products, we planned the development of an efficient synthesis of unsymmetrically diarylsubstituted chiral pyrrolizines *via* two sequential Suzuki cross-coupling reactions with different arylboronic acids in a highly chemoselective controlled manner.<sup>4</sup> Specifically, in these two sequential Suzuki couplings of multiple halogenated heterocycles,<sup>5</sup> the use of pyrrolizines remains unexplored, in spite of the importance of pyrrolizines as pharmacophores in biologically active compounds. Here, we report the development of the efficient synthetic route of a series of unsymmetrically 5,6-diarylsubstituted chiral pyrrolizines *via* chemoselective Suzuki cross-coupling reactions of chiral 6-bromo-5-iodopyrrolizines, followed by the sequential 2nd Suzuki cross-coupling reaction (Scheme 1).

To explore the feasibility of chemoselective Suzuki cross-coupling reactions of 6-bromo-5-iodopyrrolizines, prepared according to our previous paper,<sup>3</sup> with various arylboronic acids, the Suzuki couplings of 6-bromo-5-iodopyrrolizine **1a** with 3-nitrophenylboronic acid (**2**) were carried out in the presence of Pd(PPh<sub>3</sub>)<sub>4</sub> as the catalyst and K<sub>2</sub>CO<sub>3</sub> as the base in 1,4-dioxane-H<sub>2</sub>O solution at elevated temperature (Table 1). The initial cross-couplings attempts using Pd(PPh<sub>3</sub>)<sub>4</sub> (5 mol %) and K<sub>2</sub>CO<sub>3</sub> (200 mol %) in 1,4-dioxane (0.2 M)-H<sub>2</sub>O (1.0 M) solution at 60 °C chemoselectively afforded the desired coupling product, 6-bromo-5-(3-nitrophenyl)pyrrolizines **3**, in 44% yield,



**Scheme 1.** Synthetic approach to unsymmetrically 5,6-diarylsubstituted chiral pyrrolizines *via* two sequential Suzuki cross-coupling reactions

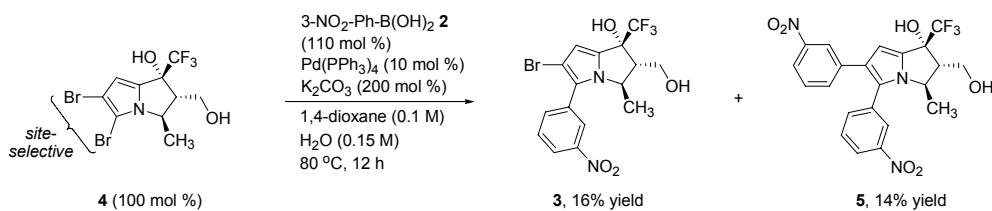
without 5-iodo-6-(3-nitrophenyl)pyrrolizine as the by-product (Table 1, entry 1). The chemoselective coupling product **3** was identified by <sup>1</sup>H NMR and high-resolution mass spectroscopy analyses. At 80 °C, the coupling product **3** showed an increased yield of 57% (Table 1, entry 2). Furthermore, when the loading of Pd(PPh<sub>3</sub>)<sub>4</sub> was increased to 10 mol %, the coupling product **3** was obtained in 63% yield (Table 1, entry 3). Remarkably, when the concentration of the 1,4-dioxane-H<sub>2</sub>O co-solvent was decreased to 0.1 M and 0.15 M, respectively, with the other reaction conditions remaining the same, the cross-coupling reaction afforded the coupling product **3** in 74% yield (Table 1, entry 4). At higher temperatures, under otherwise identical conditions, the coupling product **3** was obtained in a decreased yield of 56% (Table 1, entry 5).

In addition, we applied the optimized conditions of the chemoselective Suzuki cross-coupling reaction of 6-bromo-5-iodopyrrolizine **1a** to the site-selective Suzuki cross-coupling reaction<sup>6</sup> of 5,6-dibromopyrrolizine **4**, generally controlled by electronic and steric parameters, to obtain the desired product, 6-bromo-5-(3-nitrophenyl)pyrrolizines **3**; this product is the same as that obtained from chemoselective couplings (Scheme 2). However, the site-selective cross-coupling reaction afforded the desired coupling product **3** and the symmetrical 5,6-diarylpyrrolizine **5** as the by-product in low selectivity and poor yields. Hence, the chemoselective couplings proved superior in the case of 6-bromo-5-iodopyrrolizine **1a**.

**Table 1.** Optimization of chemoselective Suzuki cross-coupling reactions of 6-bromo-5-iodopyrrolizine **1a** with 3-nitrophenylboronic acid (**2**)<sup>a</sup>

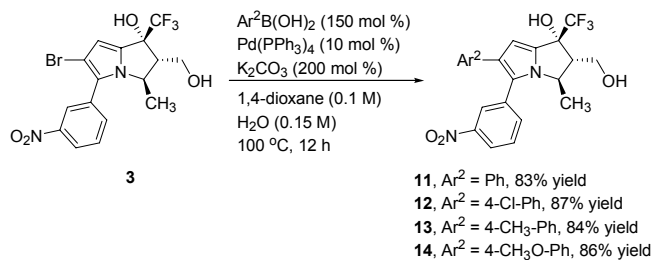
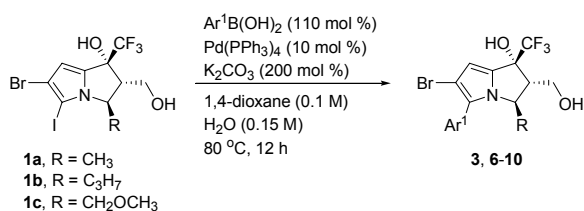
entry	Pd(PPh <sub>3</sub> ) <sub>4</sub> (mol %)	1,4-dioxane (M)	H <sub>2</sub> O (M)	temp (°C)	yield (%)
1	5	0.2	1.0	60	44
2	5	0.2	1.0	80	57
3	10	0.2	1.0	80	63
4	10	0.1	0.15	80	74
5	10	0.1	0.15	100	56

<sup>a</sup>Procedure: To a 1,4-dioxane (0.1 M) of **1a** (100 mol %) was added Pd(PPh<sub>3</sub>)<sub>4</sub> (10 mol %) at rt. After stirring for 20 min, K<sub>2</sub>CO<sub>3</sub> (200 mol %) in H<sub>2</sub>O (0.15 M) and arylboronic acid (110 mol %) were added. The mixture was stirred at 60, 80, or 100 °C for 12 h.

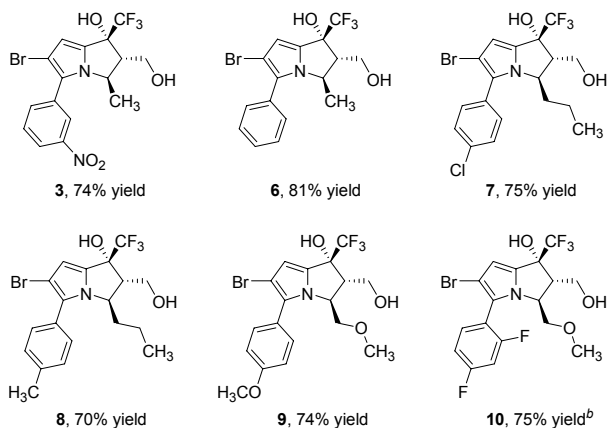


**Scheme 2.** Site-selective Suzuki cross-coupling reaction of 5,6-dibromopyrrolizine **4** with 3-nitrophenylboronic acid (**2**)

**Table 2.** Chemoselective Suzuki cross-coupling reactions of 6-bromo-5-iodopyrrolizines **1** with various boronic acids<sup>a</sup>



**Scheme 3.** Synthesis of unsymmetrical diarylpyrrolizines **11-14** via the sequential 2nd Suzuki couplings of 6-bromo-5-(3-nitrophenyl)pyrrolizines **3** with various arylboronic acids



<sup>a</sup>Procedure: See the Experimental Section. <sup>b</sup>1.5 eq of 2,4-difluorophenylboronic acid was used.

To expand the scope of the chemoselective Suzuki cross-coupling reactions of 6-bromo-5-iodopyrrolizines **1a-1c**, which bear aliphatic and methyl-protected hydroxymethyl substituents, the cross-couplings of the same compounds with a series of arylboronic acids having various electron-withdrawing and electron-donating substituents were explored under the optimized reaction conditions (Table 2). The corresponding coupling products **3** and **6-10** were chemoselectively obtained in high yields.

Next, we further explored the sequential 2nd Suzuki cross-coupling reactions of chiral 5-aryl-6-bromopyrrolizines with various arylboronic acids to obtain unsymmetrically 5,6-diaryl-substituted chiral pyrrolizines (Scheme 3). As a representative substrate in the 2nd Suzuki cross-coupling reactions, 6-bromo-5-(3-nitrophenyl)pyrrolizines **3** underwent the cross-couplings with a series of arylboronic acids (150 mol %) bearing electron-withdrawing and electron-donating substituents in the presence of Pd(PPh<sub>3</sub>)<sub>4</sub> (10 mol %) and K<sub>2</sub>CO<sub>3</sub> (200 mol %) in 1,4-dioxane (0.1 M)-H<sub>2</sub>O (0.15 M) solution at 100 °C to provide the desired unsymmetrical 5,6-diarylpyrrolizines **11-14** in high yields.

In summary, the chemoselective Suzuki cross-coupling reaction between chiral 6-bromo-5-iodopyrrolizines and a series of arylboronic acids has been achieved using Pd(PPh<sub>3</sub>)<sub>4</sub> as the catalyst and K<sub>2</sub>CO<sub>3</sub> as the base in 1,4-dioxane-H<sub>2</sub>O solution. The cross-coupling reaction provides chemoselectively various chiral 5-aryl-6-bromopyrrolizines in high yields. The sequential 2nd Suzuki cross-coupling reaction of 6-bromo-5-(3-nitrophenyl)pyrrolizines with arylboronic acids bearing electron-withdrawing and electron-donating substituents affords a variety of unsymmetrically 5,6-diarylsubstituted chiral pyrrolizines in high yields. In addition, this is the first example of two sequential Suzuki cross-coupling reactions of dihalopyrrolizines to provide unsymmetrical diarylpyrrolizines. Further studies on the development of two sequential Suzuki cross-coupling reactions are underway.

## Experimental Section

**General Procedure for the Chemoselective Suzuki Cross-coupling Reactions.** To a 1,4-dioxane solution (0.1 M) of 6-bromo-5-iodopyrrolizine **1a** (0.2 mmol, 100 mol %) was added Pd(PPh<sub>3</sub>)<sub>4</sub> (0.02 mmol, 10 mol %) at rt under argon atmosphere. After stirring for 20 min, K<sub>2</sub>CO<sub>3</sub> (0.4 mmol, 200 mol %) in H<sub>2</sub>O (0.15 M) and arylboronic acid (0.22 mmol, 110 mol %) were added. The mixture was stirred at 80 °C for 12 h. After cooling to rt, the mixture was diluted with EtOAc and washed with saturated NaHCO<sub>3</sub> followed by brine. And then, the mixture was dried over MgSO<sub>4</sub> and filtered through a short Celite pad. The solution was concentrated in vacuo and the residue was purified by flash column chromatography to provide the desired product, 5-aryl-6-bromopyrrolizine.

**General Procedure for the Sequential 2nd Suzuki Cross-coupling Reactions.** To a 1,4-dioxane solution (0.1 M) of 6-bromo-5-(3-nitrophenyl)pyrrolizine **3** (0.2 mmol, 100 mol %) was added Pd(PPh<sub>3</sub>)<sub>4</sub> (0.02 mmol, 10 mol %) at rt under argon atmosphere.

After stirring for 20 min,  $K_2CO_3$  (0.4 mmol, 200 mol %) in  $H_2O$  (0.15 M) and arylboronic acid (0.3 mmol, 150 mol %) were added. The mixture was stirred at  $100^\circ C$  for 12 h. After cooling to rt, the mixture was diluted with EtOAc and washed with saturated  $NaHCO_3$  followed by brine. And then, the mixture was dried over  $MgSO_4$  and filtered through a short Celite pad. The solution was concentrated in vacuo and the residue was purified by flash column chromatography to provide the desired product, 6-aryl-5-(3-nitrophenyl)pyrrolizine.

**The Spectroscopic Data of 3 and 5-14 are as Follows.**

**Compound 3:** yellow solid, mp  $123 - 124^\circ C$ ; IR (neat) 3446, 2948, 1716, 1533, 1349, 1283, 1179, 1081,  $740\text{ cm}^{-1}$ ;  $^1H$  NMR (400 MHz,  $CDCl_3$ )  $\delta$  8.35-8.34 (m, 1H), 8.25-8.22 (m, 1H), 7.84-7.82 (m, 1H), 7.64 (t,  $J = 8.0$  Hz, 1H), 6.36 (s, 1H), 4.89-4.83 (m, 1H), 4.23-4.20 (m, 1H), 4.00-3.94 (m, 1H), 3.63 (s, 1H), 2.75-2.71 (m, 1H), 2.48-2.45 (m, 1H), 1.11 (d,  $J = 6.4$  Hz, 3H);  $^{13}C$  NMR (100 MHz,  $CDCl_3$ )  $\delta$  148.2, 135.1, 133.4, 131.9, 129.6, 125.1, 123.8, 122.7, 124.4 (q,  $^1J_{CF} = 278.0$  Hz), 105.7, 100.4, 77.6 (q,  $^2J_{CF} = 32.0$  Hz), 59.8, 55.8, 55.1, 20.4; HRMS calcd for [M]  $C_{16}H_{14}BrF_3N_2O_4$  434.0089, found 434.0091.

**Compound 5:** light yellow solid, mp  $82 - 83^\circ C$ ; IR (neat) 3423, 2930, 1734, 1708, 1532, 1349, 1160, 1082,  $741\text{ cm}^{-1}$ ;  $^1H$  NMR (400 MHz,  $CDCl_3$ )  $\delta$  8.25-8.22 (m, 1H), 8.19-8.18 (m, 1H), 8.01-7.98 (m, 2H), 7.62-7.55 (m, 2H), 7.40-7.32 (m, 2H), 6.46 (s, 1H), 4.89-4.83 (m, 1H), 4.26-4.22 (m, 1H), 4.03-3.97 (m, 1H), 3.92 (s, 1H), 2.82-2.79 (m, 1H), 2.72-2.69 (m, 1H), 1.10 (d,  $J = 6.4$  Hz, 3H);  $^{13}C$  NMR (100 MHz,  $CDCl_3$ )  $\delta$  148.4, 148.2, 137.0, 135.8, 134.2, 133.9, 133.0, 130.0, 129.3, 126.6, 124.6 (q,  $^1J_{CF} = 281.0$  Hz), 124.5, 124.1, 123.0, 122.6, 120.9, 102.7, 77.2 (q,  $^2J_{CF} = 32.0$  Hz), 59.9, 55.6, 54.8, 20.6; HRMS calcd for [M]  $C_{22}H_{18}F_3N_3O_6$  477.1148, found 477.1151.

**Compound 6:** colorless oil; IR (neat) 3399, 2933, 1708, 1468, 1284, 1175, 1079,  $701\text{ cm}^{-1}$ ;  $^1H$  NMR (400 MHz,  $CDCl_3$ )  $\delta$  7.46-7.35 (m, 5H), 6.30 (s, 1H), 4.83-4.78 (m, 1H), 4.16-4.12 (m, 1H), 3.94 (s, 1H), 3.93-3.87 (m, 1H), 2.85-2.82 (m, 1H), 2.69-2.66 (m, 1H), 1.06 (d,  $J = 6.0$  Hz, 3H);  $^{13}C$  NMR (100 MHz,  $CDCl_3$ )  $\delta$  131.9, 130.2, 129.2, 128.4, 128.0, 127.8, 124.5 (q,  $^1J_{CF} = 284.0$  Hz), 105.1, 98.9, 77.6 (q,  $^2J_{CF} = 32.0$  Hz), 59.8, 55.5, 55.2, 20.1; HRMS calcd for [M]  $C_{16}H_{15}BrF_3NO_2$  389.0238, found 389.0239.

**Compound 7:** colorless oil; IR (neat) 3399, 2961, 1709, 1466, 1276, 1182, 1092, 833,  $722\text{ cm}^{-1}$ ;  $^1H$  NMR (400 MHz,  $CDCl_3$ )  $\delta$  7.42-7.05 (m, 5H), 6.30 (s, 1H), 4.70-4.66 (m, 1H), 4.14-4.09 (m, 1H), 3.89-3.83 (m, 1H), 3.85 (s, 1H), 2.89-2.83 (m, 2H), 1.43-1.30 (m, 2H), 1.19-1.06 (m, 1H), 1.04-0.90 (m, 1H), 0.67 (t,  $J = 7.6$  Hz, 3H);  $^{13}C$  NMR (100 MHz,  $CDCl_3$ )  $\delta$  134.0, 132.6, 130.2, 128.8, 128.7, 126.5, 124.5 (q,  $^1J_{CF} = 282.0$  Hz), 105.3, 99.4, 77.7 (q,  $^2J_{CF} = 32.0$  Hz), 60.9, 59.2, 51.8, 34.9, 16.7, 13.5; HRMS calcd for [M]  $C_{18}H_{18}BrClF_3NO_2$  451.0162, found 451.0160.

**Compound 8:** colorless oil; IR (neat) 3419, 2961, 1709, 1437, 1279, 1176, 1083, 822,  $722\text{ cm}^{-1}$ ;  $^1H$  NMR (400 MHz,  $CDCl_3$ )  $\delta$  7.32-7.30 (m, 2H), 7.25-7.21 (m, 2H), 6.28 (s, 1H), 4.69-4.65 (m, 1H), 4.13-4.06 (m, 1H), 4.01 (s, 1H), 3.88-3.81 (m, 1H), 2.90-2.85 (m, 2H), 2.39 (s, 3H), 1.38-1.21 (m, 2H), 1.14-1.04 (m, 1H), 1.03-0.92 (m, 1H), 0.64 (t,  $J = 7.2$  Hz, 3H);  $^{13}C$  NMR (100 MHz,  $CDCl_3$ )  $\delta$  137.9, 131.9, 129.2, 128.8, 127.9, 127.2, 126.0, 124.6 (q,  $^1J_{CF} = 281.0$  Hz), 123.2, 104.9, 98.7, 77.4 (q,

$^2J_{CF} = 32.0$  Hz), 60.9, 59.1, 51.8, 34.6, 21.3, 16.6, 13.5; HRMS calcd for [M]  $C_{19}H_{21}BrF_3NO_2$  431.0708, found 431.0706.

**Compound 9:** colorless oil; IR (neat) 3422, 2937, 1709, 1614, 1476, 1287, 1250, 1176, 1033, 836,  $726\text{ cm}^{-1}$ ;  $^1H$  NMR (400 MHz,  $CDCl_3$ )  $\delta$  7.39-7.35 (m, 2H), 6.99-6.95 (m, 2H), 6.30 (s, 1H), 4.78-4.74 (m, 1H), 4.15-4.12 (m, 1H), 3.93-3.87 (m, 1H), 3.86 (s, 3H), 3.69 (s, 1H), 3.24 (dd,  $J = 10.0, 3.2$  Hz, 1H), 3.19-3.16 (m, 1H), 3.08 (s, 3H), 3.04 (dd,  $J = 9.6, 6.8$  Hz, 1H), 2.79 (dd,  $J = 8.0, 3.6$  Hz, 1H);  $^{13}C$  NMR (100 MHz,  $CDCl_3$ )  $\delta$  159.9, 132.8, 130.9, 128.2, 125.1 (q,  $^1J_{CF} = 280.0$  Hz), 122.8, 114.4, 105.7, 99.3, 77.7 (q,  $^2J_{CF} = 32.0$  Hz), 71.3, 60.9, 59.4, 59.3, 55.7, 50.9; HRMS calcd for [M]  $C_{18}H_{19}BrF_3NO_4$  449.0450, found 449.0420.

**Compound 10:** light yellow oil; IR (neat) 3410, 2932, 1709, 1595, 1464, 1282, 1176, 954, 852,  $729\text{ cm}^{-1}$ ;  $^1H$  NMR (400 MHz,  $CDCl_3$ )  $\delta$  7.46-7.40 (m, 1H), 7.02-6.91 (m, 2H), 6.34 (s, 1H), 4.70-4.66 (m, 1H), 4.21 (s, 1H), 4.12 (dd,  $J = 11.6, 3.2$  Hz, 1H), 3.91-3.87 (m, 2H), 3.27 (dd,  $J = 10.0, 3.6$  Hz, 1H), 3.15-3.10 (m, 2H), 3.06 (s, 3H), 2.84-2.76 (m, 1H);  $^{13}C$  NMR (100 MHz,  $CDCl_3$ )  $\delta$  163.2 (dd,  $^1J_{CF} = 250.0$  Hz,  $^3J_{CF} = 12.0$  Hz), 160.3 (dd,  $^1J_{CF} = 249.0$  Hz,  $^3J_{CF} = 12.0$  Hz), 133.4, 133.3, 124.4 (q,  $^1J_{CF} = 280.0$  Hz), 121.2, 114.4 (dd,  $^2J_{CF} = 16.0$  Hz,  $^4J_{CF} = 4.0$  Hz), 111.6 (dd,  $^2J_{CF} = 22.0$  Hz,  $^4J_{CF} = 4.0$  Hz), 105.4, 104.3 (dd,  $^2J_{CF} = 26.0$  Hz,  $^2J_{CF} = 25.0$  Hz), 101.0, 77.6 (q,  $^2J_{CF} = 32.0$  Hz), 71.0, 60.2, 59.7, 58.9, 50.1; HRMS calcd for [M]  $C_{17}H_{15}BrF_5NO_3$  455.0155, found 455.0156.

**Compound 11:** yellow solid, mp  $171 - 172^\circ C$ ; IR (neat) 3419, 2933, 1735, 1532, 1351, 1287, 1151, 1107, 950, 772,  $702\text{ cm}^{-1}$ ;  $^1H$  NMR (400 MHz,  $CDCl_3$ )  $\delta$  8.19-8.16 (m, 2H), 7.60-7.58 (m, 1H), 7.51-7.47 (m, 1H), 7.24-7.17 (m, 3H), 7.14-7.12 (m, 2H), 4.90-4.83 (m, 1H), 4.25-4.20 (m, 1H), 4.03-3.96 (m, 1H), 3.41 (s, 1H), 2.81-2.77 (m, 1H), 2.54 (dd,  $J = 7.6, 4.4$  Hz, 1H), 1.09 (d,  $J = 6.4$  Hz, 3H);  $^{13}C$  NMR (100 MHz,  $CDCl_3$ )  $\delta$  148.3, 135.9, 135.2, 133.8, 133.6, 129.6, 129.3, 128.4, 128.2, 126.3, 124.6 (q,  $^1J_{CF} = 279.0$  Hz), 124.0, 123.8, 122.3, 102.9, 77.3 (q,  $^2J_{CF} = 32.0$  Hz), 60.0, 55.8, 54.7, 20.7; HRMS calcd for [M]  $C_{22}H_{19}F_3N_2O_4$  432.1297, found 432.1298.

**Compound 12:** yellow solid, mp  $148 - 149^\circ C$ ; IR (neat) 3418, 2888, 1734, 1533, 1497, 1349, 1168, 1152, 1045, 950, 838,  $725\text{ cm}^{-1}$ ;  $^1H$  NMR (400 MHz,  $CDCl_3$ )  $\delta$  8.19-8.16 (m, 2H), 7.58-7.49 (m, 2H), 7.17-7.15 (m, 2H), 7.04-7.02 (m, 2H), 6.34 (s, 1H), 4.87-4.80 (m, 1H), 4.22-4.19 (m, 1H), 4.01-3.95 (m, 1H), 3.84 (s, 1H), 2.79-2.77 (m, 2H), 1.08 (d,  $J = 6.4$  Hz, 3H);  $^{13}C$  NMR (100 MHz,  $CDCl_3$ )  $\delta$  148.4, 135.9, 133.8, 133.7, 133.5, 132.1, 129.7, 129.4, 128.6, 128.0, 124.6 (q,  $^1J_{CF} = 281.0$  Hz), 124.0, 123.9, 122.5, 102.7, 77.3 (q,  $^2J_{CF} = 31.0$  Hz), 59.9, 55.7, 54.7, 20.6; HRMS calcd for [M]  $C_{22}H_{18}ClF_3N_2O_4$  466.0907, found 466.0904.

**Compound 13:** yellow solid, mp  $90 - 91^\circ C$ ; IR (neat) 3394, 2888, 1532, 1349, 1285, 1170, 1152, 1043, 950, 797,  $695\text{ cm}^{-1}$ ;  $^1H$  NMR (400 MHz,  $CDCl_3$ )  $\delta$  8.19-8.14 (m, 2H), 7.60-7.57 (m, 1H), 7.51-7.47 (m, 1H), 7.03-6.99 (m, 4H), 6.35 (s, 1H), 4.88-4.82 (m, 1H), 4.23-4.18 (m, 1H), 4.01-3.95 (m, 1H), 3.54 (s, 1H), 2.80-2.76 (m, 1H), 2.67 (dd,  $J = 7.6, 4.4$  Hz, 1H), 2.30 (s, 3H), 1.09 (d,  $J = 6.4$  Hz, 3H);  $^{13}C$  NMR (100 MHz,  $CDCl_3$ )  $\delta$  148.3, 135.9, 134.0, 133.5, 132.2, 129.5, 129.4, 129.1, 128.1, 124.6 (q,  $^1J_{CF} = 283.0$  Hz), 124.0, 123.7, 122.2, 102.8, 77.3 (q,  $^2J_{CF} = 31.0$  Hz), 60.0, 55.8, 54.6, 21.0, 20.7; HRMS calcd for

[M] C<sub>23</sub>H<sub>21</sub>F<sub>3</sub>N<sub>2</sub>O<sub>4</sub> 446.1453, found 446.1456.

**Compound 14:** light brown solid, mp 63 - 64 °C; IR (neat) 3428, 2937, 1708, 1533, 1349, 1247, 1170, 1035, 836, 801 cm<sup>-1</sup>; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.19-8.14 (m, 2H), 7.59-7.57 (m, 1H), 7.51-7.47 (m, 1H), 7.06-7.03 (m, 2H), 6.77-6.75 (m, 2H), 6.33 (s, 1H), 4.91-4.82 (m, 1H), 4.23-4.20 (m, 1H), 4.02-3.95 (m, 1H), 3.77 (s, 3H), 3.59 (s, 1H), 2.80-2.76 (m, 1H), 2.70-2.64 (m, 1H), 1.09 (d, *J* = 6.4 Hz, 3H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 158.0, 148.3, 135.9, 133.9, 133.5, 129.5, 129.3, 128.9, 127.7, 124.7 (q, <sup>1</sup>*J*<sub>CF</sub> = 282.0 Hz), 123.9, 123.4, 122.1, 113.8, 102.7, 77.2 (q, <sup>2</sup>*J*<sub>CF</sub> = 31.0 Hz), 60.0, 55.7, 55.1, 54.6, 20.7; HRMS calcd for [M] C<sub>23</sub>H<sub>21</sub>F<sub>3</sub>N<sub>2</sub>O<sub>5</sub> 462.1403, found 462.1399.

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### References and Notes

- (a) Liedtke, A. J.; Keck, P. R. W. E. F.; Lehmann, F.; Koeberle, A.; Werz, O.; Laufer, S. A. *J. Med. Chem.* **2009**, *52*, 4968. (b) Michael, J. P. *Nat. Prod. Rep.* **2005**, *22*, 603. (c) Liddell, J. R. *Nat. Prod. Rep.* **2002**, *19*, 773. (d) Atwell, G. J.; Fan, J.-Y.; Tan, K.; Denny, W. A. *J. Med. Chem.* **1998**, *41*, 4744. (e) Laufer, S. A.; Augustin, J.; Dannhardt, G.; Kiefer, W. *J. Med. Chem.* **1994**, *37*, 1894. (f) Gruters, R. A.; Neeffjes, J. J.; Tersmette, M.; de Goede, R. A. Y.; Tulp, A.; Huisman, H. G.; Miedema, F.; Ploegh, H. L. *Nature* **1987**, *330*, 74. (g) Zalkow, L. H.; Glinski, J. A.; Gelbaum, L. T.; Fleischmann, T. J.; McGowan, L. S.; Gordon, M. M. *J. Med. Chem.* **1985**, *28*, 687.
- Purser, S.; Moore, P. R.; Swallow, S.; Gouverneur, V. *Chem. Soc. Rev.* **2008**, *37*, 320.
- Bae, J.-Y.; Lee, H.-J.; Youn, S.-H.; Kwon, S.-H.; Cho, C.-W. *Org. Lett.* **2010**, *12*, 4352.
- (a) Schneider, C.; Gueyrard, D.; Joseph, B.; Goekjian, P. G. *Tetrahedron* **2009**, *65*, 5427. (b) Guadagnin, R. C.; Suganuma, C. A.; Singh, F. V.; Vieira, A. S.; Cella, R.; Stefani, H. A. *Tetrahedron Lett.* **2008**, *49*, 4713. (c) Taylor, R. H.; Felpin, F.-X. *Org. Lett.* **2007**, *9*, 2911.
- For recent reviews of two sequential cross-coupling reactions of multiple halogenated heterocycles, see: (a) Schnürch, M.; Flasiak, R.; Khan, A. F.; Spina, M.; Mihovilovic, M. D.; Stanetty, P. *Eur. J. Org. Chem.* **2006**, 3283. (b) Schröter, S.; Stock, C.; Bach, T. *Tetrahedron* **2005**, *61*, 2245.
- (a) Hung, N. T.; Hussain, M.; Malik, I.; Villinger, A.; Langer, P. *Tetrahedron Lett.* **2010**, *51*, 2420. (b) Dang, T. T.; Ahmad, R.; Dang, T. T.; Reinke, H.; Langer, P. *Tetrahedron Lett.* **2008**, *49*, 1698. (c) Schröter, S.; Bach, T. *Synlett* **2005**, 1957.