Morita-Baylis-Hillman Route to Dimethyl 2,3-Dihydrobenzo[b]oxepine-2,4-dicarboxylates and Methyl 2-(2-Carbomethoxybenzo[b]furan-3-yl)propanoates from Salicylaldehydes

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A new synthetic method for dimethyl 2,3-dihydrobenzo[b]oxepine-2,4-dicarboxylates and methyl 2-(2-carbomethoxybenzo[b]furan-3-yl)propanoates by an intramolecular conjugate displacement reaction or an S_N2 reaction of acetates of Morita-Baylis-Hillman adducts of methyl (2-formylphenoxy)acetates has been described.

Key Words : Morita-Baylis-Hillman, S_N2-S_N2 ' Reaction, 2,3-Dihydrobenzo[b]oxepines, Benzo[b]furans, Methyl (2-formylphenoxy)acetates.

Introduction

Medium-sized oxacycles, 1-benzoxepine¹ and 2-benzoxepine² are important structural units present in numerous biologically active molecules.3 Especially 2,3-dihydro-1benzoxepines were investigated in order to develop orally active CCR5 antagonists as a novel scaffold.4 The several synthetic methods towards 2,3-dihydro-1-benzoxepines include ring-closing metathesis (RCM) of bis-olefins, 5 RCM of titanium-carbene complexes of diphenyl thioacetals having a carbon-carbon double bond, Wittig reaction of carboethoxycyclopropyltriphenylphosphonium fluoroborate with salicylaldehyde, ⁷ Dieckmann condensation of diester of salicylic acid, 8 Claisen-type condensation of o-formylphenoxybutyrate, ⁹ palladium-catalyzed [5+2] annulations from 2aroylmethoxyboronic acid and alkyne, 10 and palladiumcatalyzed intramolecular carboesterification of 3-(2-allyloxyphenyl)propiolic acid.11

Also, benzofuran-2-carboxylates, benzofuran-3-yl-acetates and their derivatives display interesting physiological activities and have found potential therapeutic applications. ¹² In addition, these were used to develop organic nonlinear optical cyan dyes for electro-optic devices. ¹³ The substituted benzofuran-2-carboxylates were prepared by the reaction of the appropriate salicylaldehyde with alkyl haloacetate in anhydrous dimethylformamide (DMF) at 130 °C in the presence of potassium carbonate (Equation 1). Benzofuran-3-yl-acetic acid derivatives were prepared by an base-assisted ring-opening and ring-closing process of 4-chloromethyl-2*H*-chromen-2-ones, which were readily obtained from the reaction of ethyl 4-chloroacetoacetate with phenols in acidic conditions. ¹⁵

The Morita-Baylis-Hillman (MBH) reaction ¹⁶ has attracted attention of organic chemists in recent years. This reaction provides synthetically useful multifunctional molecules which have been employed in various heterocycles syntheses. Among them MBH adducts of salicylaldehydes or *O*-benzyl-protected salicylaldehydes with acrylic acid esters or cycloalkenones were successfully used for the syntheses of 3-substituted coumarins, ¹⁷ tetrahydroxanthenones, ¹⁸ and 2-oxo-2,3-dihydrobenzo[*b*]oxepines. ¹⁹ During the continuing efforts for the development of MBH chemistry, ²⁰ we envisioned that we could synthesize the 2,3-dihydrobenzo[*b*]oxepines 5 and benzo[*b*]furan derivatives 9 from the acetates of MBH adducts of methyl (2-formylphenoxy)acetates 2 *via* an intramolecular conjugate displacement (ICD) reaction ²¹ or S_N2 mechanism, respectively, as shown in Scheme 1.

Results and Discussion

The required key starting material methyl (2-formylphenoxy)acetates 2 were prepared by the reaction of salicylaldehydes 1 with methyl chloroacetate in the presence of anhydrous K₂CO₃ in DMF at room temperature in 63-94% yields following the earlier procedure.²² The MBH reaction of 2 with methyl acrylate, 1,4-diazabicyclo[2,2,2]octane (DABCO), and triethanolamine without solvent at room temperature produced the MBH adducts 3 in 42-72% yields. The acetylation of 3 with acetic anhydride in the presence of N,N-dimethylaminopyridine (DMAP) in dichloromethane at room temperature gave MBH acetates 4 in 90-97% yields. With the substrate 4a we examined the ICD reaction under the basic conditions (Table 1). When we first tried to cyclize 4a to 5a, we used LDA in THF, but obtained a complex mixture (Entry 1). Next we then used NaH in DMSO at room temperature. This experiment gave the desired product dimethyl 2,3-dihydrobenzo[b]oxepine-2,4-dicarboxylate (5a) in very low yield (8%) along with the rearranged product methyl 2-acetoxymethyl-3-[2-(carbomethoxymeth-

yloxy)phenyl]propenoate (6a) (26%) (Entry 2). Sodium hydride-THF system was also unsuccessful at reflux temperature, only starting acetate 4a was recovered. Accordingly, we then tried a weaker base and selected Cs₂CO₃ in THF. Compound 4a did not react at room temperature, but at reflux temperature the reaction occurred to give the expected product 5a in 50% yield (Entry 5). We also explored K₂CO₃-

Table 1. Reaction of 4a under various basic conditions

Entry	Base	Solvent	Temp (°C)	Time (h)	Product (% yield) ^b
1	LDA	THF	-78 to -10	4	decomp.
2	NaH	DMSO	rt	24	5a (8)
					6a (26)
3	NaH	THF	reflux	48	n.r. ^c
4	Cs_2CO_3	THF	rt	24	n.r.
5	Cs_2CO_3	THF	reflux	26	5a (50)
6	K_2CO_3	THF	reflux	24	n.r.
7	Cs_2CO_3	CH ₃ CN	reflux	3	6a (34)
8	Cs_2CO_3	DMF	67-70	3	9a (31)
9	Cs_2CO_3	DMAc	67-70	17	9a (10)
10	Cs_2CO_3	DMSO	67-70	4	9a (8)

^aReaction conditions: **4a** (1 mmol), base (2.2 mmol), solvent (5 mL). ^bIsolated yields based on **4a**. ^cn.r = no reaction.

THF and Cs₂CO₃-CH₃CN systems at reflux temperature. No reaction occurred in K₂CO₃-THF system (Entry 6) and the only rearranged product 6a was obtained in 34% yield in Cs₂CO₃-CH₃CN system (Entry 7). Next we examined Cs₂CO₃ in aprotic polar solvents such as DMF, DMAc, and DMSO at 67-70 °C (Entries 8-10). In these reactions, 5a was not produced. Instead an unexpected benzofuran 9a was isolated in 31%, 10%, and 8% yields, respectively. As shown in Table 1, the yields are generally low, Cs₂CO₃ in THF and Cs₂CO₃ in DMF turned out to be preferred conditions giving 2,3-dihydrobenzo[b]oxepine 5a and benzo[b]furan 9a, respectively. With this result in hand, the reactions of other MBH acetates **4b-g** were carried out with Cs₂CO₃ in THF and Cs₂CO₃ in DMF, and the results are summarized in Table 2. The MBH acetates bearing electron-donating or electron-withdrawing groups are proceeded equally well and gave only the dihydrobenzo[b]oxepines **5b-f** in 20-58% yields under the Cs₂CO₃-THF system (Table 2, entries 1-6). The MBH acetate 4d derived from 3-ethoxy-substituted salicylaldehyde **1d** required a longer reaction time (96 h) and gave a lower yield (20%) of the product than those with 5-methoxy-substituted one 4c (54%) (Table 2, entries 3 and 4). Interestingly, the 3,5-dichloro-substituted MBH acetate **4g** gave only the benzo[b] furan **9g** in very low yield (21%) under the same reaction condition (Entry 7). It seems that the

Table 2. Synthesis of dimethyl 2,3-dihydrobenzo[b]oxepine-2,4-dicarboxylates **5** and methyl (2-carbomethoxybenzo[b]furan-3-yl)propanoates **9** from MBH acetates $\mathbf{4}^a$

$$\begin{array}{c} \text{OAc} & \text{[Method A]} \\ \text{CO}_2\text{Me} & \text{CS}_2\text{CO}_3, \text{THF} \\ \text{reflux} & \text{or [Method B]} \\ \text{CS}_2\text{CO}_3, \text{DMF} \\ \text{67-70 °C} & \text{5} & \text{9} \\ \end{array}$$

Entry	Apototo	\mathbb{R}^1	\mathbb{R}^2	Method	Time (h)	Product (% yield) ^b	
	Acetate					5	9
1	4-	Н	Н	A	26	5a (50)	n.d. ^c
	4a			В	3	n.d.	9a (31)
2	41.	Me	Н	A	24	5b (58)	n.d.
	4b			В	1	n.d.	9b (25)
3	4-	M-O	Н	A	23	5c (54)	n.d.
	4c	MeO		В	10	n.d.	9c (21)
4	43	Н	OEt	A	96	5d (20)	n.d.
	4d			В	48	n.d.	9d (11)
5	4-	Br	Н	A	24	5e (43)	n.d.
	4e			В	3	n.d.	9e (21)
6	46	Cl	Н	A	22	5f (53)	n.d.
	4f			В	5	n.d.	9f (23)
7		CI.	Cl	A	19	n.d.	9g (21)
	4g	Cl		В	3	n.d.	9g (24)

^aConditions: 4 (1 mmol), Cs₂CO₃ (2.2 mmol), THF or DMF (5 mL). ^bIsolated yields based on 4. ^cn.d. = not detected.

substituent group of 3-position plays an important role in governing the reactivity of the substrates by the stereo-electronic factors.

When we applied Cs₂CO₃ in DMF to MBH acetates **4b-g**, benzofurans **9b-g** were obtained in relatively poor yields (11-25%) along with a considerable amount of the decomposed materials. No dihydrobenzo[*b*]oxepines **5b-g** were produced. The formation of **5** could be explained by an S_N2' reaction in a 7-endo-trig mode. The formation of **9** could be regarded as an S_N2 reaction in a 5-exo-tet mode to form **7** followed by aromatization through sequential proton migration.²³ The different ring closure mode in THF and DMF solvents is uncertain.

We also examined the rearranged MBH allyl acetate 6a as a substrate, which was readily prepared from the MBH acetate 4a with DABCO in refluxing THF,²⁴ by the present protocol. On cyclization of 6a with Cs2CO3 in THF at reflux temperature for 48 hours, and with Cs₂CO₃ in DMF at 67-70 °C for 3 hours, the only benzofuran 9a was produced in poor yields (12% and 6%), respectively, presumably via 5-exotrig mode. Most of the starting material 6a was decomposed (Scheme 1). In order to increase the yield, we also investigated the reaction of MBH allyl bromide 10, which was readily obtained from the MBH adduct 3a with N-bromosuccinimide/dimethyl sulfide (NBS/DMS).²⁵ On cyclization of 10 with Cs₂CO₃ in THF at reflux temperature for 29 hours, 5a was produced in 23% yield. Under the influence of Cs₂CO₃ in DMF at 67-70 °C for 3 hours, the starting material was decomposed completely (Scheme 2). The use of some other basic conditions such as NaH/DMF, NaH/

DMSO, *t*-BuOK/THF, LDA/THF, and DBU/CH₃CN were also ineffective.

With the intent to introduce more diversity in the products employing this strategy, we carried out the MBH reaction of methyl (1-formyl-2-naphthalenyloxy)acetate (11a)²⁶ and methyl (2-formyl-1-naphthalenyloxy)acetate (11b) with methyl acrylate. Unfortunately the MBH adducts 12a and 12b were not produced under typical reaction conditions such as DABCO, DMAP or DBU in THF, DMF, and MeOH solvent systems. We also attempted to make the MBH alcohols *via* selenium-based route.²¹ Thus, methyl 2-phenyl-selanylpropanoate (13)²⁷ was deprotonated with LDA in THF, and a THF solution of the aldehyde 11a or 11b was added to the resulting carbanion. However, the alcohols 14a or 14b were not produced at all (Scheme 3).

Scheme 4

Finally, we examined the MBH reaction of thiosalicylaldehyde derivative, methyl (2-formylthiophenoxy)acetate (16) under typical condition, which was readily obtainable from PCC oxidation of methyl (2-hydroxymethylthiophenoxy)acetate (15).²⁸ But, the MBH reaction did not occur, instead an intramolecular condensation proceeded slowly under the reaction conditions to give the methyl benzo[*b*]-thiophene-2-carboxylate (17)²⁹ in 87% yield (Scheme 4).

Conclusion

A new method for the synthesis of dimethyl 2,3-dihydrobenzo[b]oxepine-2,4-dicarboxylates and methyl 2-(2-carbomethoxybenzo[b]furan-3-yl)propanoates by an intramolecular conjugate displacement reaction of acetates of Morita-Baylis-Hillman adducts of methyl (2-formylphenoxy)-acetates has been developed. The synthetic yields are generally low, dihydrobenzo[b]oxepines were produced in Cs₂CO₃-THF system and benzo[b]furans were given in Cs₂CO₃-DMF system, respectively.

Experimental Section

Silica gel 60 (70-230 mesh ASTM) used for column chromatography was supplied by E. Merck. Analytical TLC was carried out on Merck silica gel 60 F₂₅₄ TLC plates. Melting points were taken using an Electrothermal melting point apparatus and are uncorrected. Elemental analyses were obtained using a Thermo Electron Corporation Flash EA 1112 instrument. Low resolution mass spectra were recorded on a ThermoQuest Polaris Q mass spectrometer operating at 70 eV. IR spectra were recorded on a Nicolet Magna 550 FTIR spectrometer. The ¹H and ¹³C NMR spectra were measured on a Varian 300 spectrometer in CDCl₃. All chemical shifts are reported in ppm relative to TMS. The coupling constants (*J*) are expressed in Hz.

The known methyl (2-formylphenoxy)acetate (**2a**),²² methyl (2-formyl-4-methylphenoxy)acetate (**2b**),³⁰ methyl (4-bromo-2-formylphenoxy)acetate (**2e**),²² methyl (4-chloro-2-formylphenoxy)acetate (**2f**),³¹ methyl (1-formyl-2-naphthalenyloxy)acetate (**11a**),²⁶ methyl (2-formyl-1-naphthalenyloxy)acetate (**11b**),²⁶ and methyl (2-hydroxymethylthiophenoxy)acetate (**15**)²⁸ were prepared according to the literature procedures. Petroleum ether (PE) used refers to the fraction boiling in the range 30-60 °C.

Methyl (2-Formylphenoxy)acetates 2; General Procedure. A mixture of salicylaldehyde 1 (20 mmol), methyl chloroacetate (2.39 g, 22 mmol), and K_2CO_3 (4.14 g, 30 mmol) in anhydrous DMF (20 mL) was stirred at rt for 2-11 h. The reaction mixture was diluted with H_2O (30 mL) and extracted with Et_2O (3 × 40 mL). The combined organic layers were dried (MgSO₄) and the solvent was evaporated under reduced pressure. The resulting mixture was chromatographed on silica gel eluting with hexane-EtOAc (5:1) to produce 2 as a solid. The physical and spectral data of 2 prepared by this general method are listed as follows.

Methyl (2-Formyl-4-methoxyphenoxy)acetate (2c): Reaction time: 11 h; yield: 94%; white solid; mp 70-71 °C (Et₂O-PE); IR (KBr): 1764, 1685, 1617, 1494 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 3.81 (s, 6H, 2 × OCH₃), 4.73 (s, 2H, CH₂), 6.85 (d, J = 9.1 Hz, 1H, aromatic), 7.11 (dd, J = 9.1 and 3.2 Hz, 1H, aromatic), 7.35 (d, J = 3.5 Hz, 1H, aromatic), 10.53 (s, 1H, CH); ¹³C NMR (75 MHz, CDCl₃): δ 52.3, 55.8, 66.4, 110.5, 114.7, 123.2, 125.9, 154.5, 154.7, 168.9, 189.4; Ms m/z 224 (65, [M⁺]), 165 (70), 151 (100), 150 (99), 136 (28), 135 (48), 123 (57); Anal. Calcd for C₁₁H₁₂O₅: C, 58.93; H, 5.39. Found: C, 58.69; H, 5.52.

Methyl (2-Ethoxy-6-formylphenoxy)acetate (2d): Reaction time: 2 h; yield: 92%; white solid; mp 80-81 °C (Et₂O-PE); IR (KBr): 1758, 1687, 1585, 1473 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 1.47 (t, J = 7.0 Hz, 3H, CH₃), 3.77 (s, 3H, OCH₃), 4.10 (q, J = 7.0 Hz, 2H, CH₂), 4.87 (s, 2H, CH₂), 7.11-7.13 (m, 2H, aromatic), 7.40-7.45 (m, 1H, aromatic), 10.62 (s, 1H, CH); ¹³C NMR (75 MHz, CDCl₃): δ 14.7, 52.0, 64.6, 69.4, 118.8, 119.0, 124.3, 129.8, 150.1, 151.0, 169.8, 190.7; Ms m/z 238 (100, [M⁺]), 179 (28), 165 (18), 149 (17); Anal. Calcd for C₁₂H₁₄O₅: C, 60.50; H, 5.92. Found: C, 60.34; H, 5.79.

Methyl (2,4-Dichloro-6-formylphenoxy)acetate (2g): Reaction time: 2 h; yield: 63%; white solid; mp 115-116 °C (Et₂O-PE); IR (KBr): 1753, 1692, 1439 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 3.78 (s, 3H, OCH₃), 4.84 (s, 2H, CH₂), 7.61 (d, J = 2.6 Hz, 1H, aromatic), 7.75 (d, J = 2.6 Hz, 1H, aromatic), 10.52 (s, 1H, CH); ¹³C NMR (75 MHz, CDCl₃): δ 52.6, 70.1, 127.0, 128.7, 131.3, 132.1, 135.8, 155.3, 168.9, 188.4; Ms m/z 264 (7), 262 (18, [M⁺]), 232 (20), 230 (31), 205 (67), 203 (100), 191 (65), 190 (62), 189 (78), 188 (24); Anal. Calcd for C₁₀H₈Cl₂O₄: C, 45.66; H, 3.07. Found: C, 45.31; H, 3.35.

Morita-Baylis-Hillman Adducts 3; General Procedure. A mixture of 2 (10 mmol), methyl acrylate (2.70 mL, 30 mmol), DABCO (1.12 g, 10 mmol), and triethanolamine (1.19 g, 8 mmol) was stirred at rt for 6 h to 12 d. The reaction mixture was diluted with H_2O (20 mL) and extracted with CH_2Cl_2 (3 × 30 mL). The combined organic layers were dried (MgSO₄) and the solvent was evaporated under reduced pressure. The resulting mixture was chromatographed on silica gel eluting with hexane-EtOAc (2:1) to produce 3. The physical and spectral data of 3 prepared by this general method are listed as follows.

Methyl 3-[2-(Carbomethoxymethyloxy)]phenyl-3-hydroxy-2-methylenepropanoate (3a): Reaction time: 6 d; yield: 48%; colorless oil; IR (CH₂Cl₂): 3485, 1759, 1741, 1722, 1631, 1490, 1438 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): 8 3.72 (s, 3H, OCH₃), 3.80 (s, 3H, OCH₃), 4.05 (br s, 1H, OH), 4.70 (s, 2H, CH₂), 5.94 (s, 1H, CH), 5.98 (s, 1H, CH), 6.38 (s, 1H, CH), 6.77-6.80 (m, 1H, aromatic), 6.98-7.03 (m, 1H, aromatic); ¹³C NMR (75 MHz, CDCl₃): 8 51.6, 52.1, 65.2, 68.0, 111.8, 121.8, 125.6, 128.1, 128.7, 130.3, 140.9, 154.9, 166.6, 169.5; Ms *m/z* 248 (8), 230 (54), 220 (41), 202 (41), 193 (63), 165 (40), 147 (100), 135 (66); Anal. Calcd for C₁₄H₁₆O₆: C, 59.99; H, 5.75. Found: C, 59.73; H, 5.58.

Methyl 3-[2-(Carbomethoxymethyloxy)-5-methyl]-phenyl-3-hydroxy-2-methylenepropanoate (3b): Reaction time: 12 d; yield: 44%; white solid; mp 49-50 °C (Et₂O-PE); IR (KBr): 3483, 1760, 1741, 1721, 1631, 1499, 1438 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 2.27 (s, 3H, CH₃), 3.72 (s, 3H, OCH₃), 3.78 (s, 3H, OCH₃), 4.05 (d, J = 5.9 Hz, 1H, OH), 4.67 (s, 2H, CH₂), 5.93 (d, J = 5.6 Hz, 1H, CH), 5.98 (s, 1H, CH), 6.38 (s, 1H, CH), 6.69 (d, J = 8.5 Hz, 1H, aromatic), 7.03 (d, J = 8.2 Hz, 1H, aromatic), 7.13 (s, 1H, aromatic); ¹³C NMR (75 MHz, CDCl₃): δ 20.6, 51.8, 52.3, 65.7, 68.3, 112.0, 125.8, 128.9, 129.2, 130.2, 131.5, 141.0, 153.1, 166.8, 169.8; Ms m/z 294 (1, [M⁺]), 262 (6), 244 (44), 234 (17), 216 (49), 207 (81), 205 (73), 179 (47), 161 (100), 149 (58). Anal. Calcd for C₁₅H₁₈O₆: C, 61.22; H, 6.16. Found: C, 61.08; H, 5.84.

Methyl 3-[2-(Carbomethoxymethyloxy)-5-methoxy]-phenyl-3-hydroxy-2-methylenepropanoate (3c): Reaction time: 6 d; yield: 43%; colorless oil; IR (CH₂Cl₂): 3491, 1760, 1744, 1721, 1631, 1497, 1438 cm⁻¹; 1 H NMR (300 MHz, CDCl₃): δ 3.72 (s, 3H, OCH₃), 3.75 (s, 3H, OCH₃), 3.78 (s, 3H, OCH₃), 4.13 (d, J = 5.9 Hz, 1H, OH), 4.65 (s, 2H, CH₂), 5.94 (d, J = 5.6 Hz, 1H, CH), 5.97 (s, 1H, CH),

6.38 (s, 1H, CH), 6.74-6.75 (m, 2H, aromatic), 6.91-6.92 (m, 1H, aromatic); 13 C NMR (75 MHz, CDCl₃): δ 51.8, 52.3, 55.6, 66.3, 68.0, 113.5, 113.6, 113.9, 125.9, 131.9, 140.9, 149.3, 154.6, 166.7, 170.0; Ms m/z 310 (21, [M⁺]), 292 (9), 278 (12), 260 (18), 250 (12), 233 (12), 221 (47), 219 (38), 205 (23), 195 (25), 187 (27), 177 (100); Anal. Calcd for $C_{15}H_{18}O_7$: C, 58.06; H, 5.85. Found: C, 58.31; H, 5.92.

Methyl 3-[(2-Carbomethoxymethyloxy)-3-ethoxy]-phenyl-3-hydroxy-2-methylenepropanoate (3d): Reaction time: 8 d; yield: 42%; white solid; mp 62-63 °C (Et₂O-PE); IR (KBr): 3476, 1763, 1743, 1723, 1632, 1473, 1438 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 1.44 (t, J = 7.0 Hz, 3H, CH₃), 3.68 (s, 3H, OCH₃), 3.78 (s, 3H, OCH₃), 4.05 (q, J = 7.0 Hz, 2H, CH₂), 4.24 (d, J = 5.6 Hz, 1H, OH), 4.74 and 4.90 (AB quartet, J = 16.7 Hz, 2H, CH₂), 6.04 (s, 1H, CH), 6.06 (d, J = 5.6 Hz, 1H, CH), 6.38 (s, 1H, CH), 6.82-6.91 (m, 2H, aromatic), 6.98-7.03 (m, 1H, aromatic); ¹³C NMR (75 MHz, CDCl₃): δ 14.8, 51.8, 52.1, 64.2, 67.5, 69.0, 112.9, 119.7, 124.1, 125.3, 134.9, 141.5, 144.9, 150.6, 166.6, 171.2; Ms m/z 306 (2), 292 (5), 281 (5), 278 (9), 274 (15), 262 (10), 235 (100), 191 (76), 179 (17); Anal. Calcd for C₁₆H₂₀O₇: C, 59.25; H, 6.22. Found: C, 58.93; H, 6.08.

Methyl 3-[5-Bromo-2-(carbomethoxymethyloxy)]phenyl-3-hydroxy-2-methylenepropanoate (3e): Reaction time: 6 h; yield: 49%; white solid; mp 71-72 °C (Et₂O-PE); IR (KBr): 3480, 1759, 1742, 1720, 1630, 1484, 1437 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 3.74 (s, 3H, OCH₃), 3.79 (s, 3H, OCH₃), 3.98 (d, J = 5.9 Hz, 1H, OH), 4.68 (s, 2H, CH₂), 5.92 (d, J = 5.9 Hz, 1H, CH), 5.96 (s, 1H, CH), 6.40 (s, 1H, CH), 6.67 (d, J = 8.5 Hz, 1H, aromatic), 7.34 (dd, J = 8.5 and 2.6 Hz, 1H, aromatic), 7.49 (d, J = 2.6 Hz, 1H, aromatic); ¹³C NMR (75 MHz, CDCl₃): δ 52.0, 52.5, 65.6, 67.6, 113.7, 114.7, 126.6, 131.1, 131.6, 132.9, 140.2, 154.1, 166.6, 169.3; Ms m/z 343 (5), 341 (4), 328 (9), 326 (8), 310 (31), 308 (31), 282 (31), 280 (30), 271 (51), 227 (100), 225 (74), 213 (42), 201 (27), 199 (26), 157 (13), 155 (12); Anal. Calcd for C₁₄H₁₅BrO₆: C, 46.82; H, 4.21. Found: C, 46.65; H, 4.32.

Methyl 3-[2-(Carbomethoxymethyloxy)-5-chloro)]-phenyl-3-hydroxy-2-methylenepropanoate (3f): Reaction time: 8 h; yield: 64%; yellow oil; IR (CH₂Cl₂): 3472, 1756, 1741, 1720, 1632, 1486, 1438 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 3.74 (s, 3H, OCH₃), 3.79 (s, 3H, OCH₃), 4.01 (br s, 1H, OH), 4.68 (s, 2H, CH₂), 5.92 (s, 1H, CH), 5.96 (s, 1H, CH), 6.40 (s, 1H, CH), 6.72 (d, J = 8.8 Hz, 1H, aromatic), 7.20 (dd, J = 8.8 and 2.6 Hz, 1H, aromatic), 7.35 (d, J = 2.6 Hz, 1H, aromatic); ¹³C NMR (75 MHz, CDCl₃): δ 51.9, 52.4, 65.7, 67.6, 113.3, 126.5, 127.2, 128.3, 128.6, 132.5, 140.3, 153.6, 166.6, 169.3; Ms m/z 316 (1), 314 (2, [M⁺]), 264 (17), 254 (22), 236 (12), 225 (33), 209 (17), 199 (26), 195 (21), 183 (33), 181 (100), 171 (29), 169 (82), 167 (22), 165 (40), 157 (20), 155 (58); Anal. Calcd for C₁₄H₁₅ClO₆: C, 53.43; H, 4.80. Found: C, 53.19; H, 4.95.

Methyl 3-[2-(Carbomethoxymethyloxy)-3,5-dichloro)]-phenyl-3-hydroxy-2-methylenepropanoate (3g): Reaction time: 13 h; yield: 72%; yellow oil; IR (CH₂Cl₂): 3478, 1769, 1712, 1631, 1459, 1437 cm⁻¹; 1 H NMR (300 MHz, CDCl₃): δ 3.71 (s, 3H, OCH₃), 3.80 (s, 3H, OCH₃), 4.01 (d, J = 5.3

Morita-Baylis-Hillman Acetates 4; General Procedure. Ac₂O (1.53 g, 15 mmol) was added to a stirred solution of the MBH adduct 3 (10 mmol) and DMAP (0.34 g, 3 mmol) in CH_2Cl_2 (20 mL) at rt After stirring at the same temperature for 30 min-1 h, the reaction mixture was diluted with H_2O (5 mL) and extracted with CH_2Cl_2 (3 × 20 mL). The combined organic layers were dried (MgSO₄) and the solvent was evaporated *in vacuo*. The mixture was chromatographed on silica gel eluting with hexane-EtOAc (2:1) to produce 4. The physical and spectral data of 4 prepared by this general method are listed as follows.

Methyl 3-Acetoxy-3-[2-(carbomethoxymethyloxy)]-phenyl-2-methylenepropanoate (4a): Reaction time: 1 h; yield: 90%; colorless oil; IR (CH₂Cl₂): 1743, 1729, 1635, 1493, 1438, 1232 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 2.11 (s, 3H, CH₃), 3.73 (s, 3H, OCH₃), 3.78 (s, 3H, OCH₃), 4.67 (s, 2H, CH₂), 5.72 (s, 1H, CH), 6.43 (s, 1H, CH), 6.77 (d, J= 7.9 Hz, 1H, aromatic), 6.97-7.02 (m, 1H, aromatic), 7.10 (s, 1H, CH), 7.24-7.32 (m, 2H, aromatic); ¹³C NMR (75 MHz, CDCl₃): δ 21.0, 52.0, 52.1, 65.6, 68.1, 111.8, 121.6, 126.8, 127.5, 128.1, 129.5, 138.7, 155.2, 165.7, 169.1, 169.4; Ms m/z 304 (5), 279 (50), 248 (23), 247 (100), 231 (19), 220 (10), 219 (16), 215 (10), 194 (17), 187 (26), 165 (21); Anal. Calcd for C₁₆H₁₈O₇: C, 59.62; H, 5.63. Found: C, 59.39; H, 5.48

Methyl 3-Acetoxy-3-[2-(carbomethoxymethyloxy)-5-methyl]phenyl-2-methylenepropanoate (4b): Reaction time: 1 h; yield: 93%; white solid; mp 53-54 °C (Et₂O-PE); IR (KBr): 1743, 1726, 1635, 1501, 1437, 1231 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 2.11 (s, 3H, CH₃), 2.28 (s, 3H, CH₃), 3.73 (s, 3H, OCH₃), 3.77 (s, 3H, OCH₃), 4.65 (s, 2H, CH₂), 5.73 (s, 1H, CH), 6.43 (s, 1H, CH), 6.67 (d, J = 7.9 Hz, 1H, aromatic), 7.04-7.07 (m, 2H, aromatic), 7.07 (s, 1H, CH). ¹³C NMR (75 MHz, CDCl₃): δ 20.6, 21.0, 51.9, 52.1, 65.9, 68.1, 112.0, 126.4, 127.2, 128.6, 129.9, 131.0, 138.8, 153.2, 165.7, 169.3, 169.4; Ms m/z 336 (2, [M⁺]), 293 (56), 277 (11), 262 (29), 261 (100), 245 (20), 207 (25), 201 (56), 179 (25); Anal. Calcd for C₁₇H₂₀O₇: C, 60.71; H, 5.99. Found: C, 60.49; H, 5.70.

Methyl 3-Acetoxy-3-[2-(carbomethoxymethyloxy)-5-methoxy]phenyl-2-methylenepropanoate (4c): Reaction time: 1 h; yield: 92%; colorless oil; IR (CH₂Cl₂): 1743, 1726, 1634, 1499, 1436, 1229 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 2.11 (s, 3H, CH₃), 3.74 (s, 3H, OCH₃), 3.76 (s, 3H, OCH₃), 3.77 (s, 3H, OCH₃), 4.62 (s, 2H, CH₂), 5.71 (s, 1H, CH), 6.43 (s, 1H, CH), 6.76-6.78 (m, 2H, aromatic), 6.86-6.87 (m, 1H, aromatic), 7.06 (s, 1H, CH); ¹³C NMR (75

MHz, CDCl₃): δ 21.0, 52.0, 52.1, 55.6, 66.7, 68.0, 113.5, 113.7, 114.3, 127.6, 128.3, 138.6, 149.5, 154.4, 165.6, 169.3, 169.4; Ms *m/z* 352 (40, [M⁺]), 309 (20), 278 (39), 277 (100), 261 (29), 249 (28), 233 (41), 223 (68), 219 (40), 205 (44), 195 (51), 177 (54), 161 (89); Anal. Calcd for C₁₇H₂₀O₈: C, 57.95; H, 5.72. Found: C, 57.76; H, 5.59.

Methyl 3-Acetoxy-3-[2-(carbomethoxymethyloxy)-3-ethoxy]phenyl-2-methylenepropanoate (4d): Reaction time: 1 h; yield: 92%; white solid; mp 75-76 °C (Et₂O-PE); IR (KBr): 1743, 1726, 1635, 1473, 1438, 1227 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 1.41 (t, J = 7.0 Hz, 3H, CH₃), 2.09 (s, 3H, CH₃), 3.73 (s, 3H, OCH₃), 3.81 (s, 3H, OCH₃), 4.04 (q, J = 7.0 Hz, 2H, CH₂), 4.68 and 4.74 (AB quartet, 15.8 Hz, 2H, CH₂), 5.70 (s, 1H, CH), 6.40 (s, 1H, CH), 6.85-6.88 (m, 2H, aromatic), 7.00-7.06 (m, 1H, aromatic), 7.12 (s, 1H, CH); ¹³C NMR (75 MHz, CDCl₃): δ 14.7, 21.0, 51.9, 52.0, 64.2, 68.3, 69.7, 113.2, 119.5, 124.2, 127.1, 131.2, 139.0, 145.4, 151.1, 165.5, 169.4, 170.0; Ms m/z 366 (7, [M⁺]), 323 (12), 292 (36), 291 (100), 231 (35), 191 (20); Anal. Calcd for C₁₈H₂₂O₈: C, 59.01; H, 6.05. Found: C, 58.78: H, 6.26.

Methyl 3-Acetoxy-3-[5-bromo-2-(carbomethoxymethyloxy)]phenyl-2-methylenepropanoate (4e): Reaction time: 30 min; yield: 97%; white solid; mp 64-65 °C (Et₂O-PE); IR (KBr): 1745, 1726, 1635, 1487, 1437, 1220 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 2.13 (s, 3H, CH₃), 3.75 (s, 3H, OCH₃), 3.78 (s, 3H, OCH₃), 4.66 (s, 2H, CH₂), 5.74 (s, 1H, CH), 6.45 (s, 1H, CH), 6.65 (d, J = 8.5 Hz, 1H, aromatic), 7.04 (s, 1H, CH), 7.35-7.40 (m, 2H, aromatic); ¹³C NMR (75 MHz, CDCl₃): δ 21.0, 52.0, 52.2, 65.7, 67.4, 113.6, 114.1, 127.9, 129.1, 130.9, 132.2, 138.1, 154.2, 165.4, 168.6, 169.3; Ms m/z 402 (1), 400 (1, [M⁺]), 384 (9), 382 (9), 359 (43), 357 (43), 327 (100), 326 (87); Anal. Calcd for C₁₆H₁₇BrO₇: C, 47.90; H, 4.27. Found: C, 47.72; H, 4.15.

Methyl 3-Acetoxy-3-[2-(carbomethoxymethyloxy)-5-chloro]phenyl-2-methylenepropanoate (4f): Reaction time: 1 h; yield: 95%; white solid; mp 68-70 °C (Et₂O-PE); IR (KBr): 1745, 1727, 1634, 1488, 1438, 1220 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 2.13 (s, 3H, CH₃), 3.75 (s, 3H, OCH₃), 3.78 (s, 3H, OCH₃), 4.66 (s, 2H, CH₂), 5.74 (s, 1H, CH), 6.45 (s, 1H, CH), 6.70 (d, J = 8.5 Hz, 1H, aromatic), 7.04 (s, 1H, CH), 7.22 (dd, J = 8.8 and 2.6 Hz, 1H, aromatic), 7.27 (d, J = 2.6 Hz, 1H, aromatic); ¹³C NMR (75 MHz, CDCl₃): δ 21.0, 52.0, 52.2, 65.8, 67.6, 113.2, 126.8, 127.9, 128.1, 128.8, 129.2, 138.2, 153.8, 165.4, 168.7, 169.3; Ms m/z 358 (2), 356 (6, [M⁺]), 315 (19), 313 (53), 283 (38), 281 (100), 267 (10), 265 (20), 255 (8), 253 (20), 227 (71); Anal. Calcd for C₁₆H₁₇ClO₇: C, 53.87; H, 4.80. Found: C, 53.59; H, 4.62.

Methyl 3-Acetoxy-3-[2-(carbomethoxymethyloxy)-3,5-dichloro]phenyl-2-methylenepropanoate (4g): Reaction time: 30 min; yield: 96%; white solid; mp 71-72 °C (Et₂O-PE); IR (KBr): 1748, 1727, 1635, 1460, 1436, 1216 cm⁻¹; 1 H NMR (300 MHz, CDCl₃): δ 2.11 (s, 3H, CH₃), 3.74 (s, 3H, OCH₃), 3.83 (s, 3H, OCH₃), 4.75 (s, 2H, CH₂), 5.78 (s, 1H, CH), 6.46 (s, 1H, CH), 7.03 (s, 1H, CH), 7.19 (d, J = 2.6 Hz, 1H, aromatic), 7.36 (d, J = 2.6 Hz, 1H, aromatic); 13 C NMR

(75 MHz, CDCl₃): δ 20.9, 52.1, 52.2, 67.7, 69.7, 127.0, 127.6, 128.5, 130.2, 130.4, 134.5, 138.0, 151.0, 165.0, 168.6, 169.2; Ms m/z 392 (2), 390 (3, [M⁺]), 349 (34), 347 (49), 317 (71), 315 (100), 287 (42), 263 (19), 261 (44), 259 (49), 245 (28), 243 (45), 235 (35), 233 (54), 201 (50), 199 (69); Anal. Calcd for $C_{16}H_{16}Cl_2O_7$: C, 49.12; H, 4.12. Found: C, 48.95; H, 4.39.

Dimethyl 2,3-Dihydrobenzo[*b*]**oxepine-2,4-dicarboxylates 5; General Procedure.** A mixture of MBH acetate **4** (1 mmol) and Cs₂CO₃ (0.72 g, 2.2 mmol) in THF (5 mL) was stirred at reflux temperature for 22-96 h. After cooling to rt, the precipitate was filtered, and washed with THF (20 mL). The organic phases were concentrated under reduced pressure. The resulting mixture was chromatographed on silica gel eluting with hexane-EtOAc (4:1) to give **5** as a solid. The physical and spectral data of **5** prepared by this general method are listed as follows.

Dimethyl 2,3-Dihydrobenzo[*b*]**oxepine-2,4-dicarboxylate (5a):** Reaction time: 26 h; yield: 50%; white solid; mp 66-67 °C (Et₂O-PE); IR (KBr): 1757, 1708, 1635, 1487, 1436, 1261 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 3.08 (ddd, J = 18.5, 9.7, and 2.1 Hz, 1H, CH₂), 3.34 (dd, J = 18.5 and 1.8 Hz, 1H, CH₂), 3.83 (s, 3H, OCH₃), 3.84 (s, 3H, OCH₃), 4.59 (dd, J = 9.7 and 1.8 Hz, 1H, CH), 7.05-7.13 (m, 2H, aromatic), 7.27-7.34 (m, 2H, aromatic), 7.63 (d, J = 2.1 Hz, 1H, CH); ¹³C NMR (75 MHz, CDCl₃): δ 35.2, 52.3, 52.6, 77.1, 120.6, 123.2, 123.5, 128.1, 131.1, 134.9, 138.1, 158.2, 167.6, 170.4; Ms m/z 262 (8, [M⁺]), 230 (34), 203 (36), 173 (42), 171 (100), 143 (14), 115 (26); Anal. Calcd for C₁₄H₁₄O₅: C, 64.12; H, 5.38. Found: C, 64.01; H, 5.20.

Dimethyl 7-Methyl-2,3-dihydrobenzo[*b*]**oxepine-2,4-dicarboxylate (5b):** Reaction time: 24 h; yield: 58%; white solid; mp 89-91 °C (Et₂O-PE); IR (KBr): 1758, 1707, 1633, 1496, 1436, 1263, 1220 cm⁻¹; 1 H NMR (300 MHz, CDCl₃): 82.30 (s, 3H, CH₃), 3.06 (ddd, J=18.5, 9.7 and 2.1 Hz, 1H, CH₂), 3.32 (dd, J=18.5 and 1.5 Hz, 1H, CH₂), 3.82 (s, 6H, 2 × OCH₃), 4.55 (dd, J=9.7 and 1.5 Hz, 1H, CH), 6.83-7.11 (m, 3H, aromatic), 7.59 (d, J=2.1 Hz, 1H, CH); 13 C NMR (75 MHz, CDCl₃): 820.3, 35.2, 52.3, 52.5, 77.1, 120.4, 123.2, 128.0, 131.8, 132.5, 135.1, 138.2, 156.1, 167.6, 170.5; Ms m/z 276 (8, [M $^{+}$]), 244 (14), 217 (15), 185 (100), 157 (12), 129 (12); Anal. Calcd for $C_{15}H_{16}O_{5}$: C, 65.21; H, 5.84. Found: C, 65.07; H, 5.72.

Dimethyl 7-Methoxy-2,3-dihydrobenzo[*b*]**oxepine-2,4-dicarboxylate (5c):** Reaction time: 23 h; yield: 54%; white solid; mp 96-97 °C (Et₂O-PE); IR (KBr): 1758, 1708, 1634, 1498, 1436, 1258, 1240 cm⁻¹; 1 H NMR (300 MHz, CDCl₃): δ 3.05 (ddd, J = 18.5, 10.0 and 2.1 Hz, 1H, CH₂), 3.31 (dd, J = 18.5 and 1.8 Hz, 1H, CH₂), 3.79 (s, 3H, OCH₃), 3.82 (s, 3H, OCH₃), 3.83 (s, 3H, OCH₃), 4.53 (dd, J = 10.0 and 1.8 Hz, 1H, CH), 6.80-6.87 (m, 2H, aromatic), 7.05 (d, J = 8.5 Hz, 1H, aromatic), 7.57 (d, J = 2.1 Hz, 1H, CH); 13 C NMR (75 MHz, CDCl₃): δ 35.2, 52.3, 52.5, 55.7, 77.5, 117.2, 118.1, 121.6, 124.3, 128.7, 137.8, 152.3, 155.1, 167.6, 170.5; Ms m/z 292 (42, [M $^{+}$]), 260 (6), 233 (25), 219 (18), 201 (100), 174 (36); Anal. Calcd for C₁₅H₁₆O₆: C, 61.64; H, 5.52. Found: C, 61.35; H, 5.39.

Dimethyl 9-Ethoxy-2,3-dihydrobenzo[b]oxepine-2,4-dicarboxylate (5d): Reaction time: 96 h; yield: 20%; white solid; mp 103-106 °C (Et₂O-PE); IR (KBr): 1736, 1708, 1635, 1463, 1436, 1255, 1205 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 1.46 (t, J = 7.0 Hz, 3H, CH₃), 3.15 (ddd, J = 18.8, 9.7 and 2.1 Hz, 1H, CH₂), 3.35 (dd, J = 18.8 and 1.8 Hz, 1H, CH₂), 3.83 (s, 6H, 2 × OCH₃), 4.10 (q, J = 7.0 Hz, 2H, CH₂), 4.63 (dd, J = 9.7 and 2.1 Hz, 1H, CH), 6.90-7.02 (m, 3H, aromatic), 7.61 (d, J = 1.8 Hz, 1H, CH); ¹³C NMR (75 MHz, CDCl₃): δ 14.9, 35.2, 52.3, 52.6, 65.3, 77.7, 115.8, 122.9, 124.7, 126.6, 128.4, 138.1, 148.5, 150.1, 167.6, 170.2; Ms m/z 307 (10), 306 (54, [M⁺]), 247 (56), 246 (42), 217 (38), 215 (80), 205 (14), 203 (22), 201 (54), 189 (29), 188 (22), 187 (100); Anal. Calcd for C₁₆H₁₈O₆: C, 62.74; H, 5.92. Found: C, 62.57; H, 5.76.

Dimethyl 7-Bromo-2,3-dihydrobenzo[*b*]**oxepine-2,4-dicarboxylate (5e):** Reaction time: 24 h; yield: 43%; white solid; mp 110-111 °C (Et₂O-PE); IR (KBr): 1759, 1709, 1637, 1479, 1436, 1259, 1192 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 3.10 (ddd, J = 18.5, 9.4, and 2.1 Hz, 1H, CH₂), 3.32 (dd, J = 18.5 and 1.8 Hz, 1H, CH₂), 3.82 (s, 3H, OCH₃), 4.58 (dd, J = 9.4 and 1.8 Hz, 1H, CH), 7.01 (d, J = 8.8 Hz, 1H, aromatic), 7.37 (dd, J = 8.8 and 2.6 Hz, 1H, aromatic), 7.45 (d, J = 2.6 Hz, 1H, aromatic), 7.52 (d, J = 1.8 Hz, 1H, CH); ¹³C NMR (75 MHz, CDCl₃): δ 35.0, 52.4, 52.7, 77.1, 115.4, 122.5, 125.4, 129.7, 133.6, 136.5, 136.7, 157.1, 167.1, 170.0; Ms m/z 342 (20), 340 (20, [M⁺]), 310 (37), 308 (37), 283 (24), 281 (15), 253 (55), 251 (100), 250 (54), 249 (72), 224 (17), 222 (18); Anal. Calcd for C₁₄H₁₃BrO₅: C, 49.29; H, 3.84. Found: C, 49.37; H, 3.95.

Dimethyl 7-Chloro-2,3-dihydrobenzo[*b*]**oxepine-2,4-dicarboxylate** (**5f**): Reaction time: 22 h; yield: 53%; white solid; mp 104-106 °C (Et₂O-PE); IR (KBr): 1757, 1711, 1637, 1482, 1436, 1258, 1225, 1196 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 3.09 (ddd, J = 18.5, 9.7, and 2.1 Hz, 1H, CH₂), 3.32 (dd, J = 18.5 and 1.8 Hz, 1H, CH₂), 3.82 (s, 3H, OCH₃), 3.83 (s, 3H, OCH₃), 4.58 (dd, J = 9.7 and 2.1 Hz, 1H, CH), 7.06 (d, J = 8.5 Hz, 1H, aromatic), 7.23 (dd, J = 8.8 and 2.6 Hz, 1H, aromatic), 7.30 (d, J = 2.6 Hz, 1H, aromatic), 7.52 (d, J = 1.8 Hz, 1H, CH); ¹³C NMR (75 MHz, CDCl₃): δ 35.0, 52.4, 52.6, 77.2, 122.1, 124.9, 128.1, 129.7, 130.7, 133.7, 136.6, 156.6, 167.2, 170.0; Ms m/z 298 (5), 296 (14, [M⁺]), 266 (2), 264 (6), 239 (8), 237 (25), 207 (33), 205 (100), 180 (7), 178 (19), 149 (18), 115 (37); Anal. Calcd for C₁₄H₁₃ClO₅: C, 56.67; H, 4.42. Found: C, 56.42; H, 4.27.

Methyl 2-[(2-Carbomethoxybenzo[b]furan)-3-yl]propanoates 9; General Procedure. A mixture of MBH acetate 4 (1 mmol) and Cs₂CO₃ (0.72 g, 2.2 mmol) in DMF (5 mL) was stirred at 67-70 °C for 1-48 h. After cooling to rt, the mixture was diluted with H₂O (10 mL) and extracted with Et₂O (3 × 20 mL). The combined organic layers were dried (MgSO₄) and solvent was evaporated under reduced pressure. The resulting mixture was chromatographed on silica gel eluting with hexane-EtOAc (4:1) to give 9 as a solid. The physical and spectral data of 9 prepared by this general method are listed as follows.

Methyl 2-[(2-Carbomethoxybenzo[b]furan)-3-yl]pro-

panoate (9a): Reaction time: 3 h; yield: 31%; white solid; mp 41-43 °C (Et₂O-PE); IR (KBr): 1737, 1718, 1595, 1436, 1303 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 1.62 (d, J = 7.3 Hz, 3H, CH₃), 3.67 (s, 3H, OCH₃), 4.00 (s, 3H, OCH₃), 4.92 (q, J = 7.3 Hz, 1H, CH), 7.27-7.32 (m, 1H, aromatic), 7.43-7.49 (m, 1H, aromatic), 7.57 (dd, J = 7.6 and 0.9 Hz, 1H, aromatic), 7.69 (dd, J = 7.9 and 0.9 Hz, 1H, aromatic); ¹³C NMR (75 MHz, CDCl₃): δ 16.8, 35.7, 52.2, 52.3, 112.4, 122.0, 123.5, 126.4, 127.9, 128.4, 140.2, 154.5, 160.4, 173.5; Ms m/z 262 (5, [M⁺]), 230 (100), 204 (38), 202 (34), 187 (14), 172 (34), 171 (70); Anal. Calcd for C₁₄H₁₄O₅: C, 64.12; H, 5.38. Found: C, 63.81; H, 5.12.

Methyl 2-[(2-Carbomethoxy-5-methylbenzo[*b***]furan)-3-yl]propanoate (9b):** Reaction time: 1 h; yield: 25%; white solid; mp 106-108 °C (Et₂O-PE); IR (KBr): 1737, 1716, 1587, 1436, 1304 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 1.61 (d, J = 7.3 Hz, 3H, CH₃), 2.45 (s, 3H, CH₃), 3.67 (s, 3H, OCH₃), 3.99 (s, 3H, OCH₃), 4.89 (q, J = 7.3 Hz, 1H, CH), 7.25-7.28 (m, 1H, aromatic), 7.43-7.46 (m, 2H, aromatic); ¹³C NMR (75 MHz, CDCl₃): δ 16.8, 21.4, 35.6, 52.2, 52.3, 111.9, 121.4, 126.5, 128.2, 129.5, 133.2, 140.3, 153.0, 160.5, 173.6; Ms m/z 276 (8, [M⁺]), 245 (14), 244 (93), 218 (42), 216 (50), 201 (19), 187 (42), 185 (100); Anal. Calcd for C₁₅H₁₆O₅: C, 65.21; H, 5.84. Found: C, 64.98; H, 5.96.

Methyl 2-[(2-Carbomethoxy-5-methoxybenzo[*b*]-furan)-3-yl]propanoate (9c): Reaction time: 10 h; yield: 21%; white solid; mp 83-84 °C (Et₂O-PE); IR (KBr): 1736, 1716, 1584, 1480, 1435, 1237 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 1.61 (d, J = 7.3 Hz, 3H, CH₃), 3.68 (s, 3H, OCH₃), 3.84 (s, 3H, OCH₃), 3.99 (s, 3H, OCH₃), 4.89 (q, J = 7.3 Hz, 1H, CH), 7.05-7.09 (m, 2H, aromatic), 7.44-7.47 (m, 2H, aromatic); ¹³C NMR (75 MHz, CDCl₃): δ 16.6, 35.6, 52.1, 52.2, 55.8, 102.9, 113.0, 117.8, 126.8, 128.2, 140.9, 149.6, 156.2, 160.3, 173.5; Ms m/z 292 (25, [M⁺]), 260 (39), 233 (25), 232 (27), 217 (14), 201 (100); Anal. Calcd for C₁₅H₁₆O₆: C, 61.64; H, 5.52. Found: C, 61.43; H, 5.36.

Methyl 2-[(2-Carbomethoxy-7-ethoxybenzo[*b*]furan)-3-yl]propanoate (9d): Reaction time: 48 h; yield: 11%; colorless oil; IR (KBr): 1737, 1718, 1589, 1438, 1316 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 1.51 (t, J = 7.0 Hz, 3H, CH₃), 1.60 (d, J = 7.3 Hz, 3H, CH₃), 3.66 (s, 3H, OCH₃), 3.98 (s, 3H, OCH₃), 4.27 (q, J = 7.0 Hz, 2H, CH₂), 4.90 (q, J = 7.3 Hz, 1H, CH), 6.90-6.93 (m, 1H, aromatic), 7.15-7.24 (m, 2H, aromatic); ¹³C NMR (75 MHz, CDCl₃): δ 14.7, 16.7, 35.7, 52.1, 52.2, 64.5, 110.0, 113.4, 124.2, 128.1, 128.7, 140.4, 144.4, 145.3, 160.4, 173.6; Ms m/z 306 (49, [M⁺]), 274 (100), 248 (48), 246 (58), 218 (70), 215 (66), 187 (67); Anal. Calcd for C₁₆H₁₈O₆: C, 62.74; H, 5.92. Found: C, 62.57; H, 5.79.

Methyl 2-[(5-Bromo-2-carbomethoxybenzo[*b*]furan)-3-yl]propanoate (9e): Reaction time: 2 h; yield: 21%; yellow solid; mp 141-142 °C (Et₂O-PE); IR (KBr): 1736, 1718, 1595, 1435, 1301 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 1.61 (d, J= 7.3 Hz, 3H, CH₃), 3.70 (s, 3H, OCH₃), 4.00 (s, 3H, OCH₃), 4.87 (q, J= 7.3 Hz, 1H, CH), 7.44 (d, J= 8.8 Hz, 1H, aromatic), 7.55 (dd, J= 8.8 and 2.1 Hz, 1H, aromatic), 7.84 (d, J= 2.1 Hz, 1H, aromatic); ¹³C NMR (75

MHz, CDCl₃): δ 16.9, 35.6, 52.4, 52.5, 113.9, 116.7, 124.7, 127.6, 128.3, 131.0, 141.3, 153.2, 160.0, 173.2; Ms m/z 342 (4), 340 (4, [M $^+$]), 310 (92), 308 (91), 282 (29), 280 (28), 251 (100), 249 (74); Anal. Calcd for C₁₄H₁₃BrO₅: C, 49.29; H, 3.84. Found: C, 49.12; H, 3.70.

Methyl 2-[(2-Carbomethoxy-5-chlorobenzo[*b*]furan)-3-yl]propanoate (9f): Reaction time: 5 h; yield: 23%; white solid; mp 123-125 °C (Et₂O-PE); IR (KBr): 1737, 1720, 1595, 1435, 1302 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 1.61 (d, J = 7.3 Hz, 3H, CH₃), 3.69 (s, 3H, OCH₃), 4.00 (s, 3H, OCH₃), 4.88 (q, J = 7.3 Hz, 1H, CH), 7.41 (d, J = 8.8 Hz, 1H, aromatic), 7.49 (d, J = 8.8 Hz, 1H, aromatic), 7.68 (s, 1H, aromatic); ¹³C NMR (75 MHz, CDCl₃): δ 16.9, 35.6, 52.3, 52.4, 113.5, 121.6, 127.7, 128.3 (2 peaks), 129.3, 141.5, 152.8, 160.1, 173.2; Ms m/z 298 (4), 296 (12, [M⁺]), 266 (15), 264 (46), 237 (28), 236 (18), 205 (100), 149 (17), 115 (26); Anal. Calcd for C₁₄H₁₃ClO₅: C, 56.67; H, 4.42. Found: C, 56.49; H, 4.28.

Methyl 2-[(2-Carbomethoxy-5,7-dichlorobenzo[b]-furan)-3-yl]propanoate (9g).

Method A: A mixture of 4g (0.39 g, 1 mmol) and Cs₂CO₃ (0.72 g, 2.2 mmol) in THF (5 mL) was stirred at reflux temperature for 19 h. After cooling to rt, the precipitate was filtered, and washed with THF (20 mL). The organic phases were concentrated under reduced pressure. The resulting mixture was chromatographed on silica gel eluting with hexane-EtOAc (4:1) to give 9g (70 mg, 21%) as a white solid, which was recrystallized from Et₂O-PE; mp 147-149 °C; IR (KBr): 1740, 1711, 1597, 1435, 1308 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 1.60 (d, J = 7.3 Hz, 3H, CH₃), 3.69 (s, 3H, OCH₃), 4.01 (s, 3H, OCH₃), 4.89 (q, J = 7.3 Hz, 1H, CH), 7.47 (d, J = 1.9 Hz, 1H, aromatic), 7.59 (d, J = 1.9 Hz, 1H, aromatic); ¹³C NMR (75 MHz, CDCl₃): δ 16.8, 35.7, 52.5, 52.6, 118.6, 120.3, 127.9, 128.2, 128.7, 129.5, 142.3, 149.0, 159.8, 173.0; Ms m/z 332 (2), 330 (4, [M⁺]), 300 (53), 298 (81), 273 (21), 272 (33), 271 (35), 270 (38), 243 (23), 242 (19), 241 (100), 240 (25), 239 (84); Anal. Calcd for C₁₄H₁₂Cl₂O₅: C, 50.78; H, 3.65. Found: C, 50.62; H, 3.57.

Method B: A mixture of MBH acetate **4g** (0.39 g, 1 mmol) and Cs_2CO_3 (0.72 g, 2.2 mmol) in DMF (5 mL) was stirred at 67-70 °C for 3 h. After cooling to rt, the mixture was diluted with H_2O (10 mL) and extracted with Et_2O (3 × 20 mL). The combined organic layers were dried (MgSO₄) and solvent was evaporated under reduced pressure. The resulting mixture was chromatographed on silica gel eluting with hexane-EtOAc (4:1) to give **9g** (79 mg, 24%) as a solid. The physical and spectral data of **9g** was the same as described above.

(*E*)-Methyl 2-Acetoxymethyl-3-[2-(carbomethoxymethyloxy)phenyl]propenoate (6a): A mixture of 4a (0.32 g, 1 mmol) and DABCO (0.12 g, 1.1 mmol) in THF (5 mL) was stirred at reflux temperature for 24 h. The mixture was diluted with H_2O (10 mL) and extracted with CH_2Cl_2 (3 × 20 mL). The combined organic layers were dried (MgSO₄) and solvent was evaporated under reduced pressure. The resulting mixture was chromatographed on silica gel eluting with hexane-EtOAc (3:1) to give 6a (0.20 g, 63%) as a yellow oil;

IR (CH₂Cl₂): 1759, 1739, 1717, 1634, 1487, 1436, 1225 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 2.09 (s, 3H, CH₃), 3.80 (s, 3H, OCH₃), 3.85 (s, 3H, OCH₃), 4.71 (s, 2H, CH₂), 4.91 (s, 2H, CH₂), 6.77-6.79 (m, 1H, aromatic), 7.01-7.04 (m, 1H, aromatic), 7.26-7.36 (m, 2H, aromatic), 8.21 (s, 1H, CH); ¹³C NMR (75 MHz, CDCl₃): δ 21.0, 52.2, 52.3, 59.8, 65.5, 111.6, 121.7, 124.0, 127.0, 130.3, 131.0, 141.1, 156.0, 167.2, 169.0, 170.7; Ms m/z 322 (1, [M⁺]), 290 (19), 279 (16), 248 (63), 247 (100), 231 (19), 219 (28), 189 (82), 175 (55); Anal. Calcd for C₁₆H₁₈O₇: C, 59.62; H, 5.63. Found: C, 59.45; H, 5.41.

Synthesis of Benzo[b] furan 9a from 6a.

Method A: A mixture of **6a** (0.32 g, 1 mmol) and Cs₂CO₃ (0.72 g, 2.2 mmol) in THF (5 mL) was stirred at reflux temperature for 48 h. After cooling to rt, the precipitate was filtered, and washed with THF (20 mL). The organic phases were concentrated under reduced pressure. The resulting mixture was chromatographed on silica gel eluting with hexane-EtOAc (4:1) to give **9a** (31 mg, 12%) as a solid. The physical and spectral data of **9a** was the same as described in the preparation of **9a** from MBH acetate **4a**.

Method B: A mixture of **6a** (0.32 g, 1 mmol) and Cs_2CO_3 (0.72 g, 2.2 mmol) in DMF (5 mL) was stirred at 67-70 °C for 3 h. After cooling to rt, the mixture was diluted with H_2O (10 mL) and extracted with Et_2O (3 × 20 mL). The combined organic layers were dried (MgSO₄) and solvent was evaporated under reduced pressure. The resulting mixture was chromatographed on silica gel eluting with hexane-EtOAc (4:1) to give **9a** (16 mg, 6%) as a solid. The physical and spectral data of **9a** was the same as described in the preparation of **9a** from MBH acetate **4a**.

(Z)-Methyl 2-Bromomethyl-3-[2-(carbomethoxymethyloxy)phenyl|propenoate (10): A mixture of NBS (0.98 g, 5.5 mmol) and DMS (0.56 mL, 7.5 mmol) in CH₂Cl₂ (10 mL) was added MBH adduct 3a (1.40 g, 5 mmol) at 0 °C. After stirring for 1 h at rt, the reaction mixture was diluted with H_2O (20 mL) and extracted with CH_2Cl_2 (3 × 30 mL). The combined organic layers were dried (MgSO₄) and solvent was evaporated under reduced pressure. The resulting mixture was chromatographed on silica gel eluting with hexane-EtOAc (3:1) to afford 10 (1.31 g, 76%) as a white solid, which was recrystallized from Et₂O-PE; mp 69-72 °C; IR (KBr): 1759, 1715, 1624, 1599, 1485, 1278, 1207 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 3.80 (s, 3H, OCH₃), 3.88 (s, 3H, OCH₃), 4.37 (s, 2H, CH₂), 4.71 (s, 2H, CH₂), 6.77-6.80 (m, 1H, aromatic), 7.09-7.14 (m, 1H, aromatic), 7.34-7.39 (m, 1H, aromatic), 7.70-7.78 (m, 1H, aromatic), 8.08 (s, 1H, CH); ¹³C NMR (75 MHz, CDCl₃): δ 27.3, 52.3, 52.4, 65.5, 111.7, 121.9, 124.0, 129.0, 129.8, 131.0, 138.5, 156.2, 166.6, 169.0; Ms m/z 344 (1), 342 (1, [M⁺]), 313 (1), 311 (1), 263 (35), 231 (72), 203 (100), 173 (65), 143 (46), 131 (73), 115 (84); Anal. Calcd for C₁₄H₁₅BrO₅: C, 49.00; H, 4.41. Found: C, 48.74; H, 4.36.

Synthesis of Dihydrobenzo[*b*]**oxepine 5a from 10:** A mixture of MBH allyl bromide **10** (0.34 g, 1 mmol) and Cs₂CO₃ (0.72 g, 2.2 mmol) in THF (5 mL) was stirred at reflux temperature for 29 h. After cooling to rt, the preci-

pitate was filtered, and washed with THF (20 mL). The organic phases were concentrated under reduced pressure. The resulting mixture was chromatographed on silica gel eluting with hexane-EtOAc (4:1) to give **5a** (60 mg, 23%) as a solid. The physical and spectral data of **5a** was the same as described in the preparation of **5a** from MBH acetate **4a**.

Methyl (2-Formylthiophenoxy)acetate (16): To a suspension of PCC (1.62 g, 7.5 mmol) and celite (2.00 g) in CH₂Cl₂ (20 mL) was added dropwise a solution of alcohol 15 (1.06 g, 5 mmol) in CH₂Cl₂ (10 mL) at rt After stirring for 2 h, the precipitate was filtered, and washed with THF (20 mL). The organic phases were concentrated under reduced pressure. The resulting mixture was chromatographed on silica gel eluting with hexane-EtOAc (3:1) to give 16 (0.89 g, 85%) as a white solid, which was recrystallized from Et₂O-PE; mp 56-58 °C; IR (KBr): 2864, 2743, 1736, 1692, 1676, 1588, 1560, 1462, 1435 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 3.72 (s, 2H, CH₂), 3.73 (s, 3H, OCH₃), 7.35-7.41 (m, 1H, aromatic), 7.48-7.59 (m, 2H, aromatic), 7.85-7.88 (m, 1H, aromatic), 10.4 (s, 1H, CH); ¹³C NMR (75 MHz, CDCl₃): δ 35.5, 52.7, 126.3, 128.7, 132.7, 134.1, 134.2, 139.7, 169.6, 191.6; Ms m/z 210 (100, [M⁺]), 192 (60), 178 (64), 161 (75), 150 (73), 137 (51); Anal. Calcd for C₁₀H₁₀O₃S: C, 57.13; H, 4.79. Found: C, 57.08; H, 4.65.

Methyl Benzo[b]thiophene-2-carboxylate (17): A mixture of **16** (0.63 g, 3 mmol), methyl acrylate (0.81 mL, 9 mmol), DABCO (0.34 g, 3 mmol), and triethanolamine (0.36 g, 2.4 mmol) was stirred at rt for 6 d. The reaction mixture was diluted with H₂O (10 mL) and extracted with CH_2Cl_2 (3 × 20 mL). The combined organic layers were dried (MgSO₄) and the solvent was evaporated under reduced pressure. The resulting mixture was chromatographed on silica gel eluting with hexane-EtOAc (3:1) to produce 17 (0.50 g, 87%) as a white solid, which was recrystallized from Et₂O-PE; mp 69-71 °C (Lit.²⁷ mp 72-73 °C); IR (KBr): 1725, 1521, 1290, 1248 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 3.95 (s, 3H, OCH₃), 7.38-7.48 (m, 2H, aromatic), 7.85-7.89 (m, 2H, aromatic), 8.06 (s, 1H, CH); ¹³C NMR (75 MHz, CDCl₃): δ 52.5, 122.7, 124.9, 125.5, 126.9, 130.6, 133.3, 138.6, 142.2, 163.2; Ms m/z 192 (98, [M⁺]), 161 (100), 133 (21), 89 (24).

References and Notes

- For reviews, see: (a) Hoberg, J. O. *Tetrahedron* **1998**, *54*, 12631.
 (b) Snyder, N. L.; Haines, H. M.; Peczuh, M. W. *Tetrahedron* **2006**, *62*, 9301.
- (a) Basavaiah, D.; Sharada, D. S.; Veerendhar, A. *Tetrahedron Lett.* 2004, 45, 3081.
 (b) Das, B.; Majhi, A.; Banerjee, J.; Chowdhury, N.; Holla, H.; Harakishore, K.; Murty, U. S. *Chem. Pharm. Bull.* 2006, 54, 403.
- For some examples, see: (a) Macias, F. A.; Varela, R. M.; Torres, A.; Molinillo, J. M. G. Tetrahedron Lett. 1999, 40, 4725. (b) Wijnberg, J. B. P. A.; van Veldhuizen, A.; Swarts, H. J.; Frankland, J. C.; Field, J. A. Tetrahedron Lett. 1999, 40, 5767. (c) Asakawa, Y.; Hashimoto, T.; Takikawa, K.; Tori, M.; Ogawa, S. Phytochemistry 1991, 30, 235. (d) Asakawa, Y.; Takeda, R.; Toyota, M.; Takemoto, T. Phytochemistry 1981, 20, 858.
- 4. For recent example, see: (a) Baba, M.; Nishimura, O.; Kanzaki,

 Sabui, S. K.; Venkateswaran, R. V. Tetrahedron Lett. 2004, 45, 2047.

Y.; Kanzaki, N.; Iizawa, Y.; Baba, M.; Shiraishi, M. Bioorg. Med.

- 6. Fujiwara, T.; Koto, Y.; Takeda, T. Tetrahedron 2000, 56, 4859.
- 7. Fuchs, P. L. J. Am. Chem. Soc. 1974, 96, 1607.

Chem. 2005, 13, 363.

- 8. Kahnberg, P.; Lee, C. W.; Grubbs, R. H.; Sterner, O. *Tetrahedron* **2002**, *58*, 5203.
- 9. Nishiguchi, A.; Ikemoto, T.; Ito, T.; Miura, S.; Tomimatsu, K. *Heterocycles* **2007**, *71*, 445.
- 10. Liu, G.; Lu, X. Adv. Synth. Catal. 2007, 349, 2247.
- Li, Y.; Jardine, K. J.; Tan, R.; Song, D.; Dong, V. M. Angew. Chem. Int. Ed. 2009, 48, 9690.
- (a) Aboraia, A. S.; Yee, S. W.; Gomaa, M. S.; Shah, N.; Robotham, A. C.; Makowski, B.; Prosser, D.; Brancale, A.; Jones, G.; Simons, C. *Bioorg. Med. Chem.* 2010, 18, 4939. (b) Negoro, N.; Sasaki, S.; Mikami, S.; Ito, M.; Suzuki, M.; Tsujihata, Y.; Ito, R.; Harada, A.; Takeuchi, K.; Suzuki, N.; Miyazaki, J.; Santou, T.; Odani, T.; Kanzaki, N.; Funami, M.; Tanaka, T.; Kogame, A.; Matsunaga, S.; Yasuma, T.; Momose, Y. *ACS Med. Chem. Lett.* 2010, 1, 290. (c) Li, J.; Rush, T. S., III.; Li, W.; DeVincentis, D.; Du, X.; Hu, Y.; Thomason, J. R.; Xiang, J. S.; Skotnicki, J. S.; Tam, S.; Cunningham, K. M.; Chockalingam, P. S.; Morris, E. A.; Levin, J. I. *Bioorg. Med. Chem. Lett.* 2005, 15, 4961.
- Davis, M. C.; Groshens, T. J.; Parrish, D. A. Synth. Commun. 2010, 40, 3008.
- (a) Suzuki, T.; Horaguchi, T.; Shimizu, T.; Abe, T. Bull. Chem. Soc. Jpn. 1993, 56, 2762, (b) Teague, S. J.; Barber, S. Tetrahedron Lett. 2010, 51, 4720.
- Fall, Y.; Santana, L.; Teijeira, M.; Uriarte, E. Heterocycles 1995, 41, 647.
- For reviews of the Morita-Baylis-Hillman reaction, see: (a) Drews, S. E.; Roos, G. H. P. *Tetrahedron* 1988, 44, 4653. (b) Basavaiah, D.; Rao, P. D.; Hyma, R. S. *Tetrahedron* 1996, 52, 8001. (c) Ciganek, E. *Org. React.* 1997, 51, 201. (d) Langer, P. *Angew. Chem. Int. Ed.* 2000, 39, 3049. (e) Kim, J. N.; Lee, K. Y. *Curr. Org. Chem.* 2002, 6, 627. (f) Basavaiah, D.; Rao, A. J.; Satyanarayana, T. *Chem. Rev.* 2003, 103, 811. (g) Kataoka, T.; Kinoshita, H. *Eur. J. Org. Chem.* 2005, 45. (h) Lee, K. Y.; Gowrisankar, S.; Kim, J. N. *Bull. Korean Chem. Soc.* 2005, 26, 1481. (i) Basavaiah, D.; Rao, K. V.; Reddy, R. J. *Chem. Soc. Rev.*

- 2007, 36, 1581. (j) Singh, V.; Batra, S. Tetrahedron 2008, 64, 4511. (k) Declerck, V.; Martinez, J.; Lamaty, F. Chem. Rev. 2009, 109, 1. (l) Ma, G.-N.; Jiang, J.-J.; Shi, M.; Wei, Y. Chem. Commun. 2009, 5496. (m) Basavaiah, D.; Reddy, B. S.; Badsara, S. S. Chem. Rev. 2010, 110, 5447. (n) Zhong, W.; Liu, Y.; Wang, G.; Hong, L.; Chen, Y.; Chen, X.; Zheng, Y.; Zhang, W.; Ma, W.; Shen, Y.; Yao, Y. Org. Prep. Proced. Int. 2011, 43, 1.
- (a) Kaye, P. T.; Musa, M. A.; Nocanda, X. W. Synthesis 2003, 531.
 (b) Kaye, P. T.; Musa, M. A. Synth. Commun. 2003, 33, 1755.
- (a) Lee, K. Y.; Kim, J. M.; Kim, J. N. Bull. Korean Chem. Soc. 2003, 24, 17. (b) Lesch, B.; Brase, S. Angew. Chem. Int. Ed. 2004, 43, 115.
- Ahn, S.-H.; Lim, H. N.; Lee, K.-J. J. Heterocycl. Chem. 2008, 45, 1701.
- For our recent examples, see: (a) Hong, W. P.; Lim, H. N.; Park, H. W.; Lee, K.-J. Bull. Korean Chem. Soc. 2005, 26, 655. (b) Hong, W. P.; Lee, K.-J. Synthesis 2005, 33. (c) Hong, W. P.; Lee, K.-J. Synthesis 2006, 963. (d) Song, Y. S.; Lee, K.-J. J. Heterocycl. Chem. 2006, 43, 1721. (e) Ji, S.-H.; Hong, W. P.; Ko, S. H.; Lee, K.-J. J. Heterocycl. Chem. 2006, 43, 799. (f) Lim, H. N.; Ji, S.-H.; Lee, K.-J. Synthesis 2007, 2454. (g) Song, Y. S.; Lee, K.-J. Synthesis 2007, 3037. (h) Lim, H. N.; Song, Y. S.; Lee, K.-J. Synthesis 2007, 3376. (i) Jeon, K. J.; Lee, K.-J. J. Heterocycl. Chem. 2008, 45, 615. (j) Park, S. P.; Song, Y. S.; Lee, K.-J. Tetrahedron 2009, 65, 4703. (k) Han, E.-G.; Kim, H. J.; Lee, K.-J. Tetrahedron 2010, 66, 3490. (m) Ahn, S.-H.; Jang, S. S.; Han, E.-G.; Lee, K.-J. Synthesis 2011, 377.
- (a) Clive, D. L. J.; Li, Z.; Yu, M. J. Org. Chem. 2007, 72, 5608. (b)
 Prabhudas, B.; Clive, D. L. J. Angew. Chem. Int. Ed. 2007, 46, 9295. (c)
 Wang, L.; Prabhudas, B.; Clive, D. L. J. J. Am. Chem. Soc. 2009, 131, 6003.
- (a) Buckle, D. R.; Fenwick, A. E.; Outred, D. J.; Rockell, C. J. M. J. Chem. Research (S) 1987, 394. (b) Desai, N. C.; Dave, D.; Shah, M. D.; Vyas, G. D. Indian J. Chem. Sect. B: Org. Chem. Incl. Med. Chem. 2000, 39B, 277; Chem. Abstr. 134, 100808.
- Kim reported similar transformation, see: Lee, K. Y.; Kim, J. M.; Kim, J. N. Synlett 2003, 357.
- 24. Mason, P. H.; Emslie, N. D. Tetrahedron 1994, 50, 12001.
- (a) Das, B.; Venkateswarlu, K.; Krishnaiah, M.; Holla, H.; Majhi,
 A. Helv. Chim. Acta 2006, 89, 1417. (b) Sá, M. M.; Fernandes, L.;
 Ferreira, M.; Bortoluzzi, A. J. Tetrahedron Lett. 2008, 49, 1228.
- Karpenko, A. S.; Dorovskykh, I. V.; Shibinskaya, M. O.; Maltsev, G. V.; Lyakhova, H. A.; Gusyeva, Ju. O.; Zholobak, N. M.; Spivak, N. Ya.; Lyakhov, S. A.; Andronati, S. A. *Ukrainica Bioorganica Acta* 2008, 6, 65; *Chem. Abstr.* 151, 550047.
- (a) Qian, W.; Bao, W.; Zhang, Y. Synlett 1997, 393. (b) Movassagh,
 B.; Shamsipoor, M. Synlett 2005, 121.
- Hsiao, C.-N.; Bhagavatula, L.; Pariza, R. J. Synth. Commun. 1990, 20, 1678.
- 29. Beck, J. R. J. Org. Chem. 1972, 37, 3224.
- 30. Carreras, I.; Scherkenbeck, J.; Paulitz, C. Comb. Chem. High T. Scr. 2005, 8, 643.
- Baettig, U.; Cox, B.; Janus, D.; Leblanc, C.; Sandham, D. A.; Turner, K. L.; Watson, S. J. PCT Int. Appl. WO 2006125593, 2006; Chem. Abstr. 2006, 146, 7693.