Synthesis of Some Novel N⁷-tetrazolo[5,1-*f*]-1,2,4-triazin-8-(7*H*)-one Compounds as Potential Antimicrobial Agents

Susan M. El-Badry and Mamdouh A. M. Taha^{†,*}

Physical and Chemistry Department, Faculty of Education, Alexandria University, Alexandria, Egypt. [†]Chemistry Department, Faculty of Science, Fayoum University, Fayoum 63514, Egypt. ^{*}E-mail: mamdouhamtaha@yahoo.com (Received February 26, 2014; Accepted June 12, 2014)

ABSTRACT. A series of new N⁷-tetrazolo[5,1-f]-1,2,4-triazin-8-(7H)-one derivatives **4–16** were designed and synthesized from **3** with different reagents. The newly prepared compounds were characterized by spectral data and screened for their anti-microbial activities against various bacteria and fungi strains.

Key words: Tetrazolo[5,1-f]-1,2,4-triazin-8-(7H)-one, Synthesis, Structure elucidation, Biological assay

INTRODUCTION

The tetrazole nucleus of several compounds has been published of different heterocyclic rings.^{1–8} Numerous biological properties have been reported for tetrazolo-heterocycles, such as being useful due to their antibacterial,^{3,5,6,9} antiproliferation,¹⁰ anticancer,¹⁰ and anticonvulsant¹¹ activities. Various reviews.^{12–14} dealing with the synthesis of condensed 1,2.4-triazine moiety plays a vital role in many biological activities including antiviral,¹⁵ antihypertensive,^{15,16} blood-platelet aggregation inhibitory,^{16,17} analogesic,¹⁸ and antibacterial properties,^{19,20} as well as some of new anti-HIV and anticancer agents.²¹ In report here high yield synthetic procedures for the preparation of the title compounds and their full characterization data.

EXPERIMENTAL

Melting points were determined in open glass capillaries on a Buchi-530 melting point apparatus an are uncorrected. Spectroscopic data were recorded on the following instruments: Infrared (IR) spectra (KBr, γ cm⁻¹) Perkin Elimer 1240 spectrophtotomer, nuclear magnetic resonance (¹H NMR) spectra (Chemical shift, δ ppm) varion Mercury (300 MHz) spectrometer using TMS as internal standard and electron impact. Mass spectra (El-Ms) GC-MS (QP/000EX) Shimadzu spectrometer at an ionizing voltage of 70 ev. Reaction progress and purity of the compounds wer checked by TLC, making use of Silica Gel plates F254 on Aluminum sheets. Elemental analyses were performed by the Microanalytical center of Cairo University.

N⁷-Potassium salt tetrazolo[5,1-*f*]-1,2,4-triazin-8-(7*H*)one 3

Compound **2**, 2 mmol was disolvesed in 20 cm³ of abs. ethanol and treated with alcoholic potassium hydroxide solution (0.12 g, 2 mmol). The mixture was stirred for 2 h and the potassium salt **3** was filtered washed with abs. ethanol and dried, yield: 0.28 g (73%); m.p. >300 °C.

N⁷-[(2-Oxopropyl)]tetrazolo[5,1-*f*]-1,2,4-triazin-8-(7*H*)one 4

A mixture of (**3**, 2 mmol) and cholroacetone (2 mmol) in dimethylformanide (15 cm³) was refluxed for 3 h. The reaction mixture was filtered under hot, concentrated and the obtained solid was crystallized from abs. ethanol, and dried, yield: 0.22 g (66%); m.p. 180–182 °C; IR (KBr): $\gamma = 1750$ (CO), 1680 (CO, aromatic), 1620 (CN) cm⁻¹; ¹H NMR (DMSO-d_6): $\delta = 8.22$ (s, 1H, CH triazine ring), 5.21 (s, 2H, CH₂), 1.62 (s, 3H, CH₃) ppm; MS: *m/z* (%) = 194 (M⁺, 20). Anal. Calcd. for C₆H₆N₆O₂ (194): C: 37.11; H, 3.09; N, 43.29. Found: C, 37.15; H, 3.21; N, 42.99.

N⁷-[(Ethoxycarbonylmethyl)]tetrazolo[5,1-*f*]-1,2,4-triazin-8-(7*H*)-one 5

Compound **5** was prepared from (**3**, 2 mmol) and ethylchloroacetate (2 mmol) as just described for the preparation of 4. It was crystallized from abs. ethanol, and dried, yield: 0.29 g (76%); m.p. 151–152 °C IR (KBr): $\gamma = 1765$ (CO), 1685 (CO, aromatic), 1620 (CN) cm⁻¹; ¹H NMR (DMSOd₆): $\delta = 8.44$ (s, 1H, CH triazine ring), 5.20 (s, 2H, CH₂), 4.21 (q, 2H, *CH*₂CH₃), 1.31 (t, 3H, CH₂*CH*₃) ppm; MS: *m/z* (%) = 225 (M⁺ + 1, 30). Anal. Calcd. for C₇H₈N₆O₃ (224): C, 37.50; H, 3.57; N, 37.50. Found: C, 37.32; H, 4.11; N, 37.92.

N⁷-[(Hydrazinocarbonylmethyl)]tetrazolo[5,1-*f*]-1,2,4triazin-8-(7*H*)-one 6

To a solution of (5, 1 mmol) in abs. ethanol (30 cm³), hydrazine hydrate (1 mmol) was added and the reaction mixture was heated under stirring for 2 h. After cooling the separated product was filtered and crystallized from abs. ethanol, and dried, yield: 0.21 g (75%); m.p. 170–172 °C; IR (KBr): $\gamma = 3310$, 3270 (NH₂), 3210 (NH), 1680 (CO, aromatic), 1660 (CO, amidic), 1620 (C=N) cm⁻¹; ¹H NMR (DMSO-d₆): $\delta = 10.20$ (s, 1H, NH, exchangeable with D₂O), $\delta = 8.40$ (s, 1H, CH triazine ring), 5.01 (s, 2H, CH₂) 4.40 (s, 2H, NH₂, exchangeable with D₂O) ppm; MS: *m/z* (%) = 212 (M⁺ + 2, 28). Anal. Calcd. for C₅H₆N₈O₂(210): C: 28.57; H, 2.85; N, 53.33. Found: C, 28.61; H, 3.02; N, 53.22.

N⁷-[(3,5-Dimethylpyrazol-1-ylcarbonylmethyl)]tetrazolo[5,1-*f*]-1,2,4-triazin-8-(7*H*)-one 7

A mixture of (6, 1 mmol) and acetylacetone (1 mmol) in abs. ethanol (15 cm³) was heated at 90 °C on water-bath for 3 h. The reaction mixture was cooled, poured onto water and the formed precipitate was filtered and crystallized from ethanol, and dried, yield: 0.28 g (71%); m.p. 210–211 °C; ¹H NMR (DMSO-d₆): δ = 8.25 (s, 1H, CH triazine ring), 6.40 (s, 1H, CH pyrazole ring) 5.31 (s, 2H, CH₂), 2.50, 1.72 (s, 3H each, 2 CH₃) ppm. Anal. Calcd. for C₁₀H₁₀N₈O₂ (274): C: 43.79; H, 3.64; N, 40.87. Found: C, 44.21; H, 4.02; N, 40.44.

N⁷-[(3-Methyl-5-oxopyrazolidin-1-ylcarbonylmethyl)] tetrazolo[5,1-*f*]-1,2,4-triazin-8-(7*H*)-one 8

The title compound **8** was prepared from hydrazide (**6**, 1 mmol) and ethyl acetoacetate (1 mmol) as previously described for the preparation of **7**. It was crystallized from abs. ethanol, and dried yield: 0.22 g (56%); m.p. 260–262 °C; IR (KBr): $\gamma = 1720$ (CO, pyrazole ring), 1680 (CO aromatic), 1675 (CO amidic) cm⁻¹; MS: *m/z* (%) = 277 (M⁺ + 1, 2.23). Anal. Calcd. for C₉H₈N₈O₃ (276): C: 39.13; H, 2.89; N, 40.57. Found: C, 38.66; H, 3.02; N, 40.31.

N⁷-[(1-Phenylthiocarbamoylhydrazinocarbonylmethyl)] tetrazolo[5,1-*f*]-1,2,4-triazin-8-(7*H*)-one 9

Compound (6, 1 mmol) and phenylisothiocyanate (1 mmol) in abs. ethanol was heated under reflux for 1 h. After cooling the precipitated was collected and crystallized from abs. ethanol, and dried, yield: 0.35 g (72%), m.p. 200–202 °C; IR (KBr): γ = 3280, 3210, 3108 (3 NH), 1690 (CO, triazine ring), 1670 (CO, side chain) and 1180 (CS) cm⁻¹; ¹H NMR

(DMSO-d₆): δ = 11.94, 11.82, 11.64 (3s, 3NH, exchangeable with D₂O), 8.30 (s, 1H, CH triazine ring), 7.60–7.20 (m, 5H, *Ar*H), 5.20 (s, 2H, CH₂) ppm. MS: *m/z* (%) = 346 (M⁺ + 1, 11), 345 (M⁺, 7). Anal. Calcd. for C₁₂H₁₁N₉O₂S (345): C: 41.73; H, 3.18; N, 36.52. Found: C, 42.22; H, 3.52; N, 36.41.

N⁷-[(-5-Thioxo-4-phenyl-4*H*-1,2,4-triazol-3-ylmethyl)] tetrazolo[5,1-*f*]-1,2,4-triazin-8-(7*H*)-one 10

The thiocarbamate (**9**, 1 mmol) was refluxed in (15 cm³) of sodium hydroxide (0.34 g) solution for 2 h, filtered after coohing and acidified with 2 N hydrochloric acid. The precipitated of the crude product was filtered, washed several times with cold water and crystallized from abs. ethanol, and dried, yield: 0.20 g (71%); m.p. 215–217 °C; IR (KBr): $\gamma = 3230$ (NH), 1685 (CO, aromatic), 1200 (CS) cm⁻¹, ¹H NMR (DMSO-d₆): $\delta = 10.82$ (s, 1H, NH, exchangeable with D₂O), 8.25 (s, 1H, CH triazine ring), 7.50–7.20 (m, 5H, *Ar*H), 5.10 (s, 2H, CH₂) ppm; MS: *m/z* (%) = 327 (M⁺, 23). Anal. Calcd. for C₁₂H₉N₉OS (327): C: 44.03; H, 2.75; N, 38.53. Found: C, 43.98; H, 3.21; N, 38.67.

N⁷-[(5-Phenylamino-1,3,4-thiadiazol-2-ylmethyl)]tetrazolo[5,1-*f*]-1,2,4-triazin-8-(7*H*)-one 11

To an ice-cold stirred solution of thiocarbamate (**9** 1 mmol) in abs. ethanol (10 cm³), concentrated sulphuric acid (10 cm³), was added over a period of 1 h, the strirring was maintained at ambient temperature for an additional for hours. Then, the reaction mixture was poured onto ice-water. The solid was filtered off and recrystallized from abs. ethanol dried, yield: 0.19 g (67%); m.p. 190–192 °C. IR (KBr): $\gamma = 3187$ (NH) cm⁻¹; ¹H NMR (DMSO-d₆): $\delta = 10.22$ (s, 1H, NH, exchangeable with D₂O), 8.62 (s, 1H, CH triazine ring), 5.31 (s, 2H, CH₂) ppm; MS: m/z(%) = 327 (M⁺, 10). Anal. Calcd. for C₁₂H₉N₉OS (327): C: 44.03; H, 2.75; N, 38.53. Found: C, 44.07; H, 2.66; N, 38.22.

General Procedure for the Synthesis of Arylbenzylidenes 12a-c

A mixture of the acetic acid hydrazide (6, 1 mmol) and an appropriate aromatic aldehyde (1 mmol) namely, benzaldehyde, p-methylbenzaldehyde or p-nitrobenzaldehyde was refluxed in 20 cm³ abs. ethanol for 3 h. After cooling the reparated solid was collected by filtration, dried and crystallized from absolute ethanol. The physico-chemical and spectral data of 12a-c the following:

N^{7} -[(2-Benzylideneacetohydrazide)]tetrazolo[5,1-*f*]-1,2,4-triazin-8-(7*H*)-one 12a:

Yield: 0.33 g (78%); m.p.: 175–177 °C; IR (KBr): γ =

3188 (NH), 1680 (CO, aromatic), 1665 (CO, amidic), 1615 (C=N) cm⁻¹; ¹H NMR (*DMSO*-d₆): δ = 9.30 (s, 1H, NH, exchangeable with D₂O), 9.22 (s, 1H, CH=N), 8.11–7.20 (m, 6H, CH triazine ring + 5 *ArH*), 5.10 (s, 2H, CH₂) ppm; MS: *m/z* (%) = 299 (M⁺ + 1, 17). Anal. Calcd. for C₁₂H₁₀N₈O₂ (298): C: 48.32; H, 3.35; N, 37.58. Found: C, 47.81; H, 3.41; N, 37.88.

N⁷-[(2-p-Methylbenzylideneacetohydrazide)]tetrazolo [5,1-*f*]-1,2,4-triazin-8-(7H)-one 12b:

Yield: 0.22 g (52%); m.p. 153–155 °C; ¹H NMR (DMSOd₆): δ = 9.35 (s, 1H, NH, exchangeable with D₂O), 9.20 (s, 1H, CH=N), 8.20–7.97 (m, 5H, CH + 4 *ArH*), 7.12 (s, 1H, CH triazine ring) 5.20 (s, 2H, CH₂), 2.33 (s, 3H, CH₃) ppm; MS: *m/z* (%) = 312 (M⁺, 12). Anal. Calcd. for C₁₃H₁₂N₈O₂ (312): C: 50.00; H, 3.84; N, 35.89. Found: C, 49.81; H, 4.22; N, 36.32.

N^{7} -[(2-p-Nitrobenzylideneacetohydrazide)]tetrazolo [5,1-*f*]-1,2,4-triazin-8-(7*H*)-one 12c:

Yield: 0.35 g (83%); m.p. 178–180 °C; IR (KBr): $\gamma = 3180$ (NH), 1690 (CO, aromatic), 1660 (CO, amidic), 1620 (C=N) cm⁻¹; MS: *m/z* (%) = 344 (M⁺+1, 10). Anal. Calcd. for C₁₂H₉N₉O₄ (343): C: 41.98; H, 2.62; N, 36.73. Found: C, 42.22; H, 2.55; N, 36.51.

General Procedure for the Synthesis of 5-Aryl-1,3,4oxadiazoles 13a-c

A mixture of the appropriate (12a-c, 1 mmol) and acetic anhydride (15 cm^3) were heated under reflux for 4 h. The reaction mixture was cooled, poured onto ice-water and allowed to stand at ambient temperature. The solid product formed was collected and recrystallized from abs. ethanol to afford the product. The physico-chemical and spectral data of 13a-c the following:

N^{7} -[(4-Acetyl-5-phenyl-4,5-dihydro-1,3,4-oxadiazol-2-ylmethyl)]tetrazolo[5,1-*f*]-1,2,4-triazin-8-(7*H*)-one 13a:

Yield: 0.27 g (79%); m.p. 167–169 °C; IR (KBr): $\gamma = 1760$ (CO), 1680 (CO, aromatic), 1620 (C=N) cm⁻¹; ¹H NMR (DMSO-d₆): $\delta = 8.16-7.15$ (m, 6 H, CH triazine ring + 5 *ArH*), 5.12 (s, 2H, CH₂), 2.08 (s, 1H, CH oxadiazole ring), 2.29 (s, 3H, CH₃) ppm; MS: *m/z* (%) = 340 (M⁺, 7). Anal. Calcd. for C₁₄H₁₂N₈O₃ (340): C: 49.41; H, 3.52; N, 32.94. Found: C, 49.24; H, 3.11; N, 33.21.

N^{7} -[(4-Acetyl-5-p-methylphenyl-4,5-dihydro-1,3,4-oxadiazol-2-yl methyl)]tetrazolo [5,1-*f*]-1,2,4-triazin-8-(7*H*)one 13b:

Yield: 0.19 g (55%); m.p.: 140–142 °C; IR (KBr): $\gamma =$ 1760 (CO), 1680 (CO, aromatic), 1625 (C=N) cm⁻¹; ¹H NMR (DMSO-d₆): $\delta = 8.12-7.30$ (m, 5H, CH triazine ring + 4*ArH*), 5.14 (2, 2H, CH₂), 2.10 (s, 1H, CH oxadiazole ring), 2.30, 2.12, (2s, 3H each, 2CH₃) ppm; MS: *m/z* (%) = 355 (M⁺+1, 10). Anal. Calcd. for C₁₅H₁₄N₈O₃ (354): C: 50.84; H, 3.95; N, 31.63. Found: C, 51.22; H, 4.24; N, 31.42.

N⁷-[(4-Acetyl-5-p-nitrophenyl-4,5-dihydro-1,3,4-oxadiazol-2-yl-methyl)]tetrazolo [5,1-*f*]-1,2,4-triazin-8-7*H*)one 13c:

Yield: 0.32 g (86%); m.p.: 162–164 °C; IR (KBr): $\gamma =$ 1740 (CO), 1690 (CO, aromatic), 1620 (C=N) cm⁻¹; MS: *m/z* (%) = 385 M⁺, 22). Anal. Calcd. for C₁₄H₁₁N₉O₅ (385): C: 43.63; H, 2.85; N, 32.72. Found: C, 44.01; H, 3.11; N, 32.42.

N⁷-[(2-Thioxo-2,3-dihydro-1,3,4-oxadiazol-5-ylmethyl)] tetrazolo[5,1-*f*]-1,2,4-triazin-8-(7*H*)-one 14

To a mixture of (**6**, 1 mmol) in abs. ethanol (15 cm³) was added of potassium hydroxide (0.10 g) in abs. ethanol (20 cm³). Then, carbon disulfide (10 cm³) was added portion wise and the reaction mixture was refluxed till no color of hydrogen sulphide evolved. The solid prouduct obtained after cooling and acidification with dilute hydrochloric acid was collected by filteration and recrystallized from abs. ethanol (yellow color), dried, yield: 0.27 g (75%); m.p. 152–154 °C, IR (KBr): γ = 3295 (NH), 1690 (CO, aromatic), 1220 (CS) cm⁻¹; ¹H NMR (DMSO-d₆): δ = 10.86 (s, 1H, NH, exchangeable with D₂O), 8.50 (s, 1H, CH triazine ring), 5.29 (s, 2H, CH₂) ppm; MS: *m/z* (%) = 252 (M⁺, 12). Anal. Calcd. for C₆H₄N₈OS (252): C: 28.57; H, 1.58; N, 44.44. Found: C, 28.34; H, 2.01; N, 44.49.

N⁷-[(4-Amino-5-thioxo-4,5-dihydro-1*H*-1,2,4-triazol-3-ylmethyl)]tetrazolo[5,1-*f*]-1,2,4-triazin-8-(7*H*)-one 15

A mixture of (14, 1 mmol) and hydrazine hydrate (1 mmol) in 15 cm³ abs. ethanol was refluxed for 2 h. The solvents were removed under reduced pressure, the residue was washed with ether then recrystallized from abs. ethanol, and dried, yield: 0.18 g (58%), m.p. 144–146 °C; IR (KBr): γ = 3300, 3290 (NH₂), 3230 (NH), 1680 (CO, aromatic) cm⁻¹; ¹H NMR (DMSO-d₆): *d* = 11.20 (s, 1H, NH exchangeable with D₂O), 8.30 (s, 1H, CH triazine ring), 5.29 (2, 2H, NH₂, exchangeable with D₂O), 4.95 (s, 2H, CH₂) ppm; MS: *m/z* (%) = 266 (M⁺, 10). Anal. Calcd. for C₆H₆N₁₀OS (266): C: 27.06; H, 2.25; N, 52.63. Found: C, 27.22; H, 2.44; N, 52.41.

General Procedure for the Synthesis of Schiff's Bases 16a-c

A mixture of (15, 1 mmol) and an appropriate aromatic aldchyde (1 mmol) namely, benzenaldehyde, p-methylbenzenaldehy, or p-nitrobenzenaldehyde (1 mmol) was refluxed in 20 cm³ abs. ethanol for 3 h. After cooling the separated solid was collected by filteration, dried and crystallized from abs. ethanol. The physico-chemical and spectra data of **16a–c** the following:

N^{7} -[(4-Benzylideneamino)-5-thioxo)-4,5-dihydro-1*H*-1,2,4-triazol-3-yl-methyl)]-tetrazolo-[5,1-*f*]-1,2,4-triazin-8-(7*H*)-one 16a:

Yield: 0.29 g (74%); m.p. 160–162 °C; IR (KBr): $\gamma = 3195$ (NH), 1675 (CO, aromatic) cm⁻¹; ¹H NMR (DMSO-d₆): $\delta = 10.95$ (s, 1H, NH, exchangeable with D₂O), 9.51 (s, 1H, CH=N), 8.52–8.30 (m, 5H, *ArH*), 8.20 (s, 1H, CH triazine ring), 4.85 (s, 2H, CH₂) ppm; Ms: *m/z* (%) = 355 (M⁺ + 1, 13). Anal. Calcd. for C₁₃H₁₀N₁₀OS (354): C: 44.06; H, 2.82; N, 39.54. Found: C, 45.21; H, 3.01; N, 39.42.

N^7 -[(4-p-Methylbenzylidenoamino)-5-thioxo-4,5-dihydro-1*H*-1,2,4-triazol-3-yl-methyl)]-tetrazolo[5,1-*f*]-1,2,4triazin-8-(7*H*)-one 16b:

Yield: 0.24 g (58%); m.p. 135–137 °C; IR (KBr): γ = 3220 (NH), 1680 (CO, aromatic) cm⁻¹; ¹H NMR (DMSO-d₆): δ = 10.90 (s, 1H, NH, exchangeable with D₂O), 8.90 (s, 1H, CH=N), 8.50–8.15 (m, 4H, *ArH*), 7.30 (s, 1H, CH

triazine ring), 4.79 (s, 2H, CH₂), 2015 (s, 3H, CH₃); MS: m/z (%) = 368 (M⁺, 7). Anal. Calcd. for C₁₄H₁₂N₁₀OS (368): C: 45.65; H, 3.26; N, 38.04. Found: C, 45.71; H, 3.42; N, 38.21.

N^{7} -[(4-p-Nitrobenzylideneamino)-5-thioxo-4,5-dihydro-1*H*-1,2,4-triazol-3-yl-methyl)]-tetrazolo[5,1-*f*]-1,2,4-triazin-8-(7*H*)-one 16c:

Yield: 0.39 g (86%); m.p. 177–178 °C; IR (KBr): $\gamma = 3240$ (NH), 1690 (CO, aromatic) cm⁻¹; MS: *m/z* (%) = 400 (M⁺ + 1, 12). Anal. Calcd. for C₁₃H₉N₁₁O₃S (399): C, 39.09; H, 2.25; N, 38.59. Found: C, 39.22; H, 2.44; N, 38.24.

RESULTS AND DISCUSSION

The structure of the prepared compounds was elucidated using IR, ¹H NMR, and mass spectroscopic methods besides elemental analyses. The pathways leading to the products obtained have been depicted in *Schemes* 1, 2, and 3.

Ethyl 1-aminotetrazole-5-carboxylate [2] (1) was reacted with formamide yielded tetrazolo[5,1-*f*]-1,2,4-triazin-8-(7*H*)-one [2] (2) which upon treatment with potassium hydroxide afforded the potassium salt **3**. Reaction of **3** with chloroacetone and ethyl chloroacetate in dimethylformamide afforded N⁷-oxopropyl and N⁷-ethyloxycarbonyl-methyl derivatives **4** and **5**, respectively. N⁷-Hydrazinocarbonylmethyl **6** was synthesized from the ester **5** with hydrazine hydrate in ethanol. The hydrazide **6** revealed



Scheme 1.

Journal of the Korean Chemical Society



Reagents: i, (CH₃CO)₂CH₂; ii, CH₃COCH₂COCH₂CH₃; iii, PhNCS; iv, NaOH/HCl; v, H₂SO₄





 $\textbf{Reagents:} i, ArCHO; ii, (CH_3CO)_2O; iii, CS_2/KOH; iv, N_2H_4.H_2O$

Ar = $a = C_6H_5$; $b = C_6H_4$ -CH₃(p); $c = C_6H_4$ -NO₂(p)

Scheme 3.

2014, Vol. 58, No. 4

absorption bands at $\gamma = 3310, 3270, 3210, 1680$ and 1660 cm⁻¹ corresponding to NHNH₂ and two CO groups. The ¹H NMR spectrum of compound **6** showed $\delta = 10.20$ (s, 1H, NH exchangeable with D₂O), 8.40 (s, 1H, CH triazine ring), 5.10 (s, 2H, CH₂CO) and 4.40 (s, 2H, NH₂ exchangeable with D₂O) ppm.

Reaction of the acid hydrazide **6** with acetylacetone and ethyl acetoacetate gave pyrazolyl and pyrazolidinyl structures **7** and **8**, respectively. The ¹H NMR spectrum of **7** exhibited, CH pyrazolyl ring and two methyl signals. The IR spectrum of **8** showed characteristic absorption bands at $\gamma = 1720$, 1680 and 1675 cm⁻¹ corresponding to three carbonyl amides of pyrazole, triazine and side chain, respectively.

Compound 6 was refluxed with phenylisothiocyanate, the expected N⁷-phenyl-thiocarbamoylhydrazinocarbonylmethyl derivative 9 was resulted, which ¹H NMR spectrum contained characteristic signals to three NH exchangeable with D₂O and aromatic protons. Alkaline cyclization of the thiocarbamate 9 using sodium hydroxide solution furnished the corresponding 1,2,4-triazolyl nucleus 10, which IR region displayed the disappearance of the two NH and CO absorptions in side chain. The mass spectrum of 10 revealed the molecular ion peak at m/z = 327. Also, the thiosemicarbazide 9 was cyclized with cold concentrated sulphuric acid caused in the formation of the corresponding 1,3,4-thiadiazolyl derivative 11. The ¹H NMR spectrum of compound 11 showed signals for assigned NH (exchangeable with D₂O), CH (triazine ring), aromatic protons, and CH₂ group.

Additionally, the condensation of the acetic acid hydrazide **6** with variety of aromatic aldehydes namely, benzaldehyde, p-methylbenzaldehyde and p-nitrobenzaldehyde in absolute ethanol leading to the benzylidenes 12a-c. The ¹H NMR spectra of the latter products showed the methylidene proton signals and the correct mass spectra. Condensation cyclization of benzylidene structures 12a-c with acetic anhydride afforded the 4-acetyl-5-aryl-1,3,4-oxadiazole congeners 13a-c. The IR spectra of the latter compounds showed the presence of the absorption bands for carbonyl groups at 1760 or 1740 cm⁻¹ and devoid any bands corresponding NH groups and lacked any NH signals in ¹H NMR spectra.

Cyclocondensation reaction of **6** with carbon disulfide in alcoholic potassium hydroxide, by heating under reflux yielded 1,3,4-oxadiazolyl derivative **14**, which showed in IR spectrum NH, CO (aromatic) and CS absorptions and lacked NHNH₂ and CO (amidic) absorptions characteristic of the parent compound **6**. The mass spectrum of **14** revealed a peak corresponding to its molecular ion peak at m/z = 252. The structure of **14** was inferred chemically its reaction with hydrazine hydrate afforded the amino-1,2,4triazole derivative **15**. The IR spectrum of **15** showed a characteristic absorption bands corresponding to NH₂, NH and CO groups, whereas its mass spectrum showed a peak exhibited to its molecular ion at m/z = 266.

Furthermore, condensation of **15** with various of aromatic aldehydes namely, benzaldehyde, p-methylbenzaldelye and p-nitrobenzaldehyde in absolute ethanol afforded the corresponding the expected Schiff's bases **16a–c** which possessed IR absorption characteristic of NH group and lacked NH₂ group. The ¹H NMR spectra of **16a–c** revealed signals for assigned NH (exchangeable with D₂O), methine (CH=N), aromatic protons, CH (triazine ring) and CH₂ group.

The antimicrobial activity of the newly synthesized compounds 4–6, 8–10, 12a,c, 13a,c and 14–16a–c were evaluated

Table 1. Minimal inhibitory concentration (MIC/µg cm⁻³) of compounds 4–6, 8–10, 12a,c, 13a,c and 14–16a,c

Compound	B. subtitis	S. aureus	K. pneumoniae	E. coli	A. niger	C. albican
4	100	>200	100	>200	100	100
5	50	100	100	100	100	100
6	100	50	100	50	100	25
8	100	50	100	50	100	25
9	>200	50	100	100	100	100
10	100	100	>200	100	100	100
12a	100	100	100	100	100	100
12c	100	>200	100	>200	100	100
13a	50	50	100	50	100	25
13c	100	100	50	100	25	100
14	100	50	50	50	25	100
16a	100	100	100	100	100	100
16c	50	100	100	100	100	100
Ampicillin	12.5	12.5	25	25	_	_
Clotrimazole	_	_	-	_	12.5	12.5

Journal of the Korean Chemical Society

against Gram-positive (*Bacillus subtitis* and *Staphylococcus aureus*) and Gram-negative (*Klebsiella pneumoniae* and *Escherichia coli*) bacteria strains and (*Aspergillums niger* and *Condida albican*) fungi strains. The minimal inhibitory concentration (*MIC*/mg) cm^{-3 22} is displayed in *Table* 1, showing that **5**, **13a** and **16c** exhibit an antimicrobial activity against *B. substitis* (25%); **6**, **8**, **9**, **13a** and **14** against *S. aureus* (25%); **13c** and **14** against *K. pneumoniae* (50%); **6**, **8**, **13a** and **14** against *E. coil* (50%) comparable to that of ampicillin. Furthermore, **13c**, **14**, possessed an antimycotic activity against *A. niger* (50%) and **6**, **8** and **13a** against *C. albican* (50%) comparable to that of clotrimazole.

ANTIMICROBIAL SCREENING

Solutions of the test compounds, ampicillin trihydrate and clotrimazole were prepared in *DMSO* at a concentration of 100 μ g/cm³. Twofold dilution of the compounds were prepared (800, 400,, 6.25 g/cm³). The microorganism suspensions at 10⁶ Colony Formin Unit 1 cm³ (*CFU*/cm³) concentration were inoculate to the corresponding wells. Plates were incubated at 36 °C for 24 to 28 h the incubation chamber. The minimal inhibitory concentration (*MIC*) were determined. Controls with *DMSO* and infected media were also investigated.

CONCLUSION

In this paper, the synthesis of some novel N^7 -tetrazolo [5,1-*f*]-1,2,4-triazin-8-(7*H*)-one compounds **4–16** through potassium salt **3** are described. The antimicrobial activities against various bacteria and fungi strains are screened.

Acknowledgments. Publication cost of this paper was supported by the Korean Chemical Society.

REFERENCES

1. El-Badry, S. M.; Taha, M. A. M. J. Korean. Chem. Soc.

2011, 55, 974.

- Taha, M. A. M.; El-Badry, S. M. J. Korean. Chem. Soc. 2010, 54, 414.
- Taha, M. A. M.; El-Badry, S. M. Monatsh. Chem. 2008, 139, 1261.
- Taha, M. A. M. Phosphorus, Sulphur, Silicon Relat. Elem. 2008, 183, 2525.
- 5. Taha, M. A. M. Monatsh. Chem. 2007, 138, 505.
- Taha, M. A. M.; El-Badry, S. M. Phosphorus, Sulfur, Silicon Relat Elem 2007, 182, 1011.
- Taha, M. A. M.; El-Badry, S. M. J. Chin. Chem. Soc. 2006, 53, 1181.
- 8. Taha, M. A. M. J. Chin. Chem. Soc. 2005, 52, 137.
- Karnik, A. V.; Malviya, N. J.; Kulkarni, A. M.; Jadhav, B. L. *Eur. J. Med. Chem.* **2006**, *41*, 891.
- Jantova, S.; Ruzekova, I.; Stantovsky, S.; Spirkova, K. Neoplasma 1997, 44, 240.
- Rubat, C.; Coudert, P.; Couquelet, J. M.; Tronche, P.; Bastide, J.; Bastide, P. Formaco 1990, 45, 331.
- El-Ashry, E. S. H.; Rashed, N.; Taha, M.; Ramadan, E. Adv. Heterocyct Chem. 1994, 59, 39.
- El-Ashry, E. S. H.; Rashed, N.; Mousaad, A.; Ramadan, E. *Adv. Heterocycl Chem.* **1994**, *61*, 207.
- Neunhoeffer, H.; Wiley, P. F. Chem. Heterocycl. Compd. 1978, 33, 749.
- Kaminsky, D. 3-Hydrazino-1,2,4-triazino[5,6-b] indoles.
 U.S. Patent 3: 752891; *Chem. Abstr.* 1973, 79, 149328.
- Monge, A.; Palop, J. A.; Ramierez, C.; Fernandez-Alvarez, E. Acta. Farm. Bonaerense 1987, 6, 157; Chem. Abstr. 109, 121991.
- Monge, A.; Palop J.; Ramirez, C.; Font, M.; Fernandez-Alvarez, E. *Eur. J. Med. Chem.* **1991**, *26*, 179.
- Tamchin, A. B.; Ignat'eva, M. A.; Masyuta, G. F. *Khim-Farm Zh.* **1972**, *6*, 23; *Chem. Abstr.* 77, 43170.
- Dave, A. M.; Bhatt, K. N.; Undavia, N. K.; Trivedi, P. B. J. Indian. Chem. Soc. 1989, 66, 246.
- Shaban, M. A. E.; Taha, M. A. M.; Morgaan, A. E. A. Monatsh. Chem. 2000, 131, 487 and literature cited therein.
- Abdel-Rahman, R. M.; Morsy, J. M.; El Edfawy, S.; Amine H. A. *Pharmazie* 1999, 54, 667.
- 22. Murray, P. R.; Baron, E. J.; Pfaller, M. A.; Tenover, F. C.; Yolken, R. T. Manual of Clinical Microbiology. In *Antimicrobial Agents and Susceptibility Testing*, Woods, G. L., Washington, J. A., Eds.; ASM Press: Washington DC, 1995; p 1327.