

An overview of new oxidation methods for polyacrylonitrile-based carbon fibers

Hye Kyoung Shin¹, Mira Park², Hak-Yong Kim² and Soo-Jin Park^{1,*}

¹Department of Chemistry, Inha University, Incheon 402-751, Korea

²Department of Organic Materials and Fiber Engineering, Chonbuk National University, Jeonju 561-756, Korea

Article Info

Received 4 December 2014

Accepted 2 January 2015

*Corresponding Author

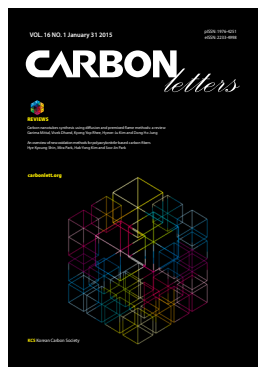
E-mail: sjpark@inha.ac.kr

Tel: +82-32-860-7234

Open Access

DOI: <http://dx.doi.org/10.5714/CL.2015.16.1.011>

This is an Open Access article distributed under the terms of the Creative Commons Attribution Non-Commercial License (<http://creativecommons.org/licenses/by-nc/3.0/>) which permits unrestricted non-commercial use, distribution, and reproduction in any medium, provided the original work is properly cited.



<http://carbonlett.org>

pISSN: 1976-4251

eISSN: 2233-4998

Copyright © Korean Carbon Society

Abstract

The process of oxidizing polyacrylonitrile (PAN)-based carbon fibers converts them into an infusible and non-flammable state prior to carbonization. This represents one of the most important stages in determining the mechanical properties of the final carbon fibers, but the most commonly used methods, such as thermal treatment (200°C to 300°C), tend to waste a great deal of process time, money, and energy. There is therefore a need to develop more advanced oxidation methods for PAN precursor fibers. In this review, we assess the viability of electron beam, gamma-ray, ultra-violet, and plasma treatments with a view to advancing these areas of research and their industrial application.

Key words: carbon fibers, polyacrylonitrile, electron beam, gamma-ray, ultra-violet, plasma

1. Introduction

Carbon fibers are thin filament materials that contain over 92 wt% carbon [1-3], which gives them high strength, stiffness, temperature resistance, thermal conductivity, and good chemical resistance in addition to light weight and low density when compared to most metals and ceramics [4-6]. As a result, carbon reinforced materials have been widely used in a variety of aerospace, automotive and marine applications, as well as various machine parts, wind turbine blades, high-grade sporting goods, etc. [7-9]. Carbon fibers can be classified based on how the fibers are derived, from polyacrylonitrile (PAN), pitch, rayon, or gaseous precursors [10-15]. PAN has tended to be the most widely used precursor for high-performance carbon fibers [16-22]. Most useful carbon materials have also been modified by several surface treatments, such as anodic oxidation [23], plasma [24,25], oxyfluorination [26], ozonization [27], or fluorination [28], Ar ion beam [29-31], several metal coatings [32-34], chemical treatments [35,36]. These treatments enable or enhance function and performance in real industrial applications, including adsorption [37], toxic removal [38,39], catalysis [40-42], adhesion or composites [43,44], electrochemistry [45-47], and so on.

The precursor PAN fiber is an atactic and linear polymer with C≡N bonding, as shown in Fig. 1. This C≡N bond means that the PAN precursor has a glass transition temperature (T_g) of around 120°C, and therefore typically decomposes before melting, meaning that it is essential that it first be stabilized prior to carbonization. Typically, this is accomplished by oxidation of PAN by thermal treatment in air at 200°C to 300°C (Fig. 2), which changes it into infusible and non-flammable fibers with a predominance of cyclic or ladder structures (Fig. 3) that prevent melting during subsequent carbonization [48-56].

Oxidation is therefore one of the most important processes in determining the mechanical properties of the carbon fiber; in this regard, most past reviews have focused on the thermal oxidation of PAN. For example, Rahaman *et al.* [57] briefly reviewed heat treatments for the conversion of PAN precursor fibers into carbon fibers, while Liu and Kumar [58] provided a more detailed report pertaining to recent developments in carbon fiber technology and demon-

strated the relationship between the processing conditions and the chemical/physical structure and tensile properties. However, thermal stabilization not only requires long periods of time (2-3 h), but also has high cost and energy consumption [59]. This has led to the development of advanced methods for oxidizing PAN precursor fibers based on various forms of radiation, which induce a change in the polymer structure, but these still need further improvement to make them more convenient and eco-friendly.

In this review, we summarize the progress that has been made in the various radiation oxidation processes for PAN precursor fibers, and provide some details as to the preparation and characterization of PAN fibers prepared by these methods.

2. Radiation Induced Stabilization of PAN Fibers

2.1. Oxidation by electron beam irradiation

Electron beam irradiation (EBI) is ionizing radiation process performed by a linear accelerator, and has been used industrially ever since it was first applied to the crosslinking of polyethyl-

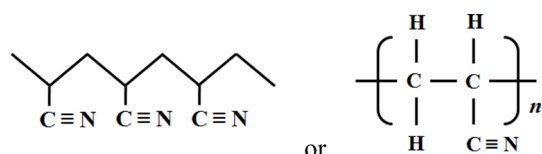


Fig. 1. Molecular structure of polyacrylonitrile precursor.

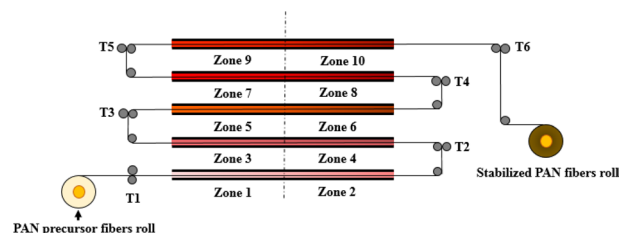


Fig. 2. Schematics of thermal oxidation process line; (Zone 1-Zone 10): temperature zoned, (T1-T6): stretching rollers. PAN: polyacrylonitrile.

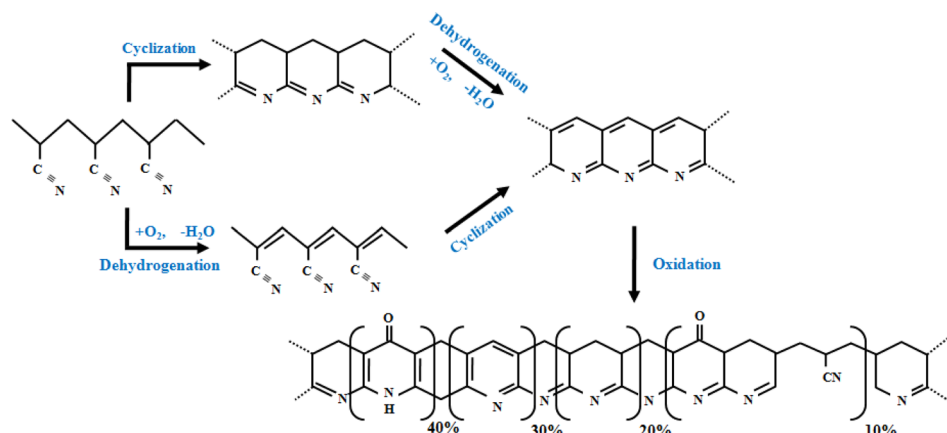


Fig. 3. Schematic depicting the thermal chemistry during the oxidation of polyacrylonitrile fiber.

ene wire insulation in the late 1950s. In more recent years, the number of electron accelerators has grown to exceed 1500, as they are now widely used for a range of scientific and industrial applications [60]. Generally speaking, electron beam (EB) accelerators can be divided into three broad categories, based on their energy output [61]. Low-energy EB accelerators (120-300 keV) are most commonly used for adhesives, coatings for paper, multilayer packaging, and the surface grafting of membranes [62]. Mid-energy EB accelerators (300 keV to 5 MeV) are utilized for the polymerization of monomers, the grafting of monomers onto polymers, the cross-linking of polymers, the degradation of polymers and fiber modification [63-76]. Finally, high-energy (5-10 MeV) EB accelerators are used for the sterilization of medical devices, pharmaceutical and biological products, bio-ethanol products, and the treating of industrial effluent [60,65,77-101]. Through careful selection and control over the EBI conditions used for modifying polymer materials, physical, chemical, and biological properties can be improved without the use of a solvent. Furthermore, since only normal temperatures and pressures are required, it provides a simple and eco-friendly means of producing carbon fibers. Fig. 4 is a schematic illustration of EBI stabilization and the carbonization of PAN fibers.

Shin *et al.* [102] investigated the possibility of using an EB accelerator at an acceleration of 1.14 MeV and beam current of 8

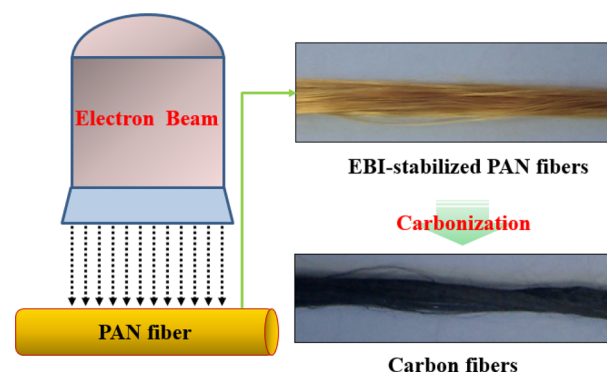


Fig. 4. Schematic illustration of electron beam generation by a linear accelerator and photos of electron beam irradiation (EBI)-stabilized polyacrylonitrile (PAN) fibers and carbon fibers.

mA to oxidize PAN fibers, testing the effects of 500, 1000, 2000, 3000, and 5000 kGy doses of absorbed radiation. In that work, both the PAN precursor fibers and EBI-oxidized fibers were analyzed by Fourier-transform infrared spectroscopy (FT-IR), gel fraction, density, differential scanning calorimetry (DSC), X-ray diffraction (XRD) and mechanical measurements. The structural change in PAN fibers resulting from oxidation by EBI was evidenced by a decrease in the intensity of the peaks at 1451 cm^{-1} and 2244 cm^{-1} , which were attributed to the stretching vibration of C-H and $\text{C}\equiv\text{N}$. The gel fraction and density test results revealed an increase in the degree of cyclization with increasing EBI dose, with a resulting increase in amorphization, causing a reduction in the tensile strength of the fibers. In the typical DSC curves, however, the EBI-oxidized PAN fibers showed a decrease in the activation energy required for exothermic reaction with increasing EBI dose.

Shin *et al.* [103] also looked into producing carbon fiber from PAN fibers through a combination of short thermal treatment and EBI, wherein PAN fibers, irradiated by 1000 kGy of EB energy, were thermally oxidized at 200°C or 250°C for 20 or 40 min, respectively. The results of FT-IR analysis showed that the $\text{C}\equiv\text{N}$ peak at 2244 cm^{-1} almost disappeared after 40 min at 250°C , while the intensity of the $\text{C}=\text{N}$ peak at 1628 cm^{-1} was indicative of an increase in cyclization. Through DSC analysis, thermal treatment at either 200°C or 250°C was found to cause a sharp decrease in the majority of exothermic peaks. The stabilization index of PAN fibers treated at 250°C for 40 min was 99.21%. Finally, the tensile strength of the resulting carbon fiber was around 2.3 GPa, with the cross-sectional image of a single fiber clearly demonstrating the brittle nature of its morphology. This therefore demonstrates that a combination of EBI and thermal treatment can reduce the time normally required for oxidation (2-3 h) by over an hour, yet still produce carbon fibers with a high tensile strength.

Kim *et al.* [104] applied an EB accelerating voltage of 1.14 MeV and a beam current of 1 mA to stabilize PAN nanofiber (NF) mats using dosages ranging from 500 to 5000 kGy. The stabilized mats were subsequently used to form carbon NF mats, which exhibited a similar morphology to the pristine PAN NF mats and contained no significant defects.

Choi *et al.* [105] used the EBI- H_2O_2 method to oxidize PAN NFs, in which 1% H_2O_2 solution-sprayed PAN NFs were irradiated by an EB. In the FT-IR spectrum obtained, EBI- H_2O_2 stabilization was shown to induce the transformation of $\text{C}\equiv\text{N}$ to $\text{C}=\text{N}$ in PAN NFs with significantly less defects than if EBI treatment alone were used. This was particularly evident in the DSC analysis, which revealed that 50 kGy EBI- H_2O_2 treatment reduced the activation energy of exothermic reactions in PAN NFs to a level comparable to that obtained with 500 kGy EBI treatment. Thus, in the EBI- H_2O_2 treatment, electrons are capable of oxidizing PAN NFs at a lower radiation dose than EBI alone.

2.2. Oxidation by gamma-ray irradiation

Gamma-ray radiation, emitted in all directions from radionuclides such as ^{137}Cs or ^{60}Co (Fig. 5) or high energy EBI, which is ionizing radiation generated by electricity and magnetism accelerating electrons to a high energy level, are both known to similarly influence materials [106-108]. However, the gamma

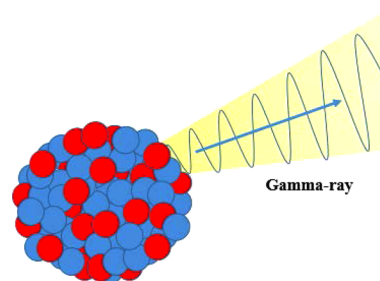


Fig. 5. Schematics of gamma-ray emission from atomic nucleus.

irradiation process is different than that for EBI. Gamma radiation can completely penetrate materials and can be applied to various kinds of materials, but has some drawbacks. The dose is a few orders of magnitude lower and therefore requires a longer residence time compared to EBI.

Tan and Wan [109] used gamma-ray irradiation to investigate structural changes in irradiated PAN precursor fibers. PAN fibers were wound onto a frame of about 20 cm in a glass container and exposed to gamma radiation for 25 or 50 h with an irradiation dose rate of 2.0 kGy/h. Structural changes of the irradiated PAN fibers were characterized by FT-IR, DSC, thermogravimetric analysis (TGA), and XRD. In the FT-IR spectra, gamma-ray irradiated PAN fibers exhibited reduced peak intensity of $\text{C}\equiv\text{N}$ groups but increased peak intensities of $\text{C}=\text{O}$, $\text{C}=\text{N}$ and $\text{C}=\text{C}$ groups. These results suggest that gamma-ray irradiation induces chemical conversions of the linear structure of PAN precursor fibers into ladder structures of stabilized PAN fibers, similar to thermal stabilization. In the DSC analysis, the PAN precursor fibers showed a sharp exothermic peak at 280.9°C while the gamma-ray irradiated PAN fibers showed a broad and reduced peak intensity due to reduction of activation energy by conversion of their chemical structure. In addition, the irradiated PAN fibers showed smaller weight loss in TGA.

As these results show, gamma-ray irradiation exhibits the potential to improve the stabilization of PAN precursor fibers as compared with thermal methods. To increase the yield of carbon fibers, Liu *et al.* [110] preoxidized gamma-ray irradiated PAN fibers and researched the variation in density of the gamma-ray irradiated PAN fibers with and without preoxidation. Density in the stabilized fibers increased due to cyclization and oxygen uptake during stabilization. Density measurement was particularly important in evaluating the influence of radiation processes on the preoxidized PAN fibers. Finally, they found that the gamma-ray irradiated PAN fibers with various types of peroxidation increased in density with an increase in the peroxidation time and temperature.

In addition, Liu *et al.* [111] studied the radiation oxidation of PAN fibers in ways unlike previous researchers. Generally, PAN fibers have mainly been irradiated without oxygen to avoid oxidation degradation. However, the introduction of oxygen in PAN fibers can facilitate thermal oxidation curing, for producing carbon fibers. In the results of the gel fraction, they observed that oxidation reactions were limited to the surface of the PAN fibers because oxygen has difficulty in penetrating the fibers. But, the radiation-oxidized regions have been completely changed to gel by inducing the thermal composition. The above gamma-ray ir-

radiation technologies illustrate the method's potential advantages, and suggest gamma-ray irradiation is one of the advanced oxidation methods that can be applied for oxidation of PAN fibers, except for the very long residence time during oxidation.

2.3. Ultra-violet oxidation

Electromagnetic radiation in the ultra-violet (UV) spectrum has the ability to excite electrons to higher energy orbital levels, which directly affects the nature of atomic bonds and changes the physical and mechanical properties of materials [112,113]. Several researchers have therefore investigated the effect that UV treatment has on the mechanical and chemical properties of various fibers [114,115]. Among them, Paiva *et al.* [116] explored the possibility of using UV oxidation to produce carbon fibers from melt-processible PAN-based copolymers (acrylonitrile [AN]:methyl acrylate [MA] = 88:12), wherein UV-irradiated melt-spun fibers were thermally oxidized prior to being carbonized at 1500°C. However, the carbon fibers obtained by this method exhibited a low mechanical strength of around 350 MPa due to the number and size of defects contained within them.

A later study by Morales and Ogale [117] achieved the UV-assisted stabilization of wet-spun, photoinitiator-modified PAN precursor fibers by adding 4,40-bis(diethylamino)-benzophenone (BDP) as a photoinitiator to reduce the thermal oxidation processing time, and thereby enhance the physical properties of the resulting carbon fibers. The precursor fibers were spun from a solution consisting of PAN powder dissolved in BDP to a mass ratio of 99:1, with the resulting fibers then being irradiated using a UV curing lamp with two different UV sources (mercury or iron halide bulb). They found that those fibers irradiated by a halide bulb exhibited a higher tensile modulus than PAN fibers irradiated by a mercury bulb, which was attributed to the difference in the spectral output of each bulb type. However, in both instances, the tensile modulus was still higher than that of pure PAN fibers, thermally treated PAN fibers with 1 wt% BDP, and UV-irradiated and pure PAN fibers. In addition, the application of 5 min of UV treatment to PAN precursor fibers containing 1 wt% BDP was found to be roughly comparable to the first 83 min of thermal oxidation. These results illustrate the potential savings that UV irradiation can offer in the stabilization of PAN precursor fibers, while still retaining the mechanical properties of the final carbon fiber.

2.4. Plasma oxidation

Plasma can be described as a strongly ionized gas that contains ions, radicals, excited molecules, and free electrons. Owing to its free electrical charge, plasma is electrically conductive and its magnetic field can strongly influence a gas. By applying these principals, polymer materials treated with plasma can be drastically altered in terms of their chemical structure and surface properties. In addition, plasma treatment (Fig. 6) is a simple, effective, and versatile technique for stabilization or surface treatment in the plasma zone [15,118-121].

Lee *et al.* [121] researched the effect of plasma assisted oxidation on the tensile properties of PAN based carbon fibers at different plasma exposure times. PAN fibers were thermally oxidized at 220°C for 30 min with an added plasma treatment for 5 min more. In the FT-IR spectra, the peak intensity at 1595 cm⁻¹ assigned to

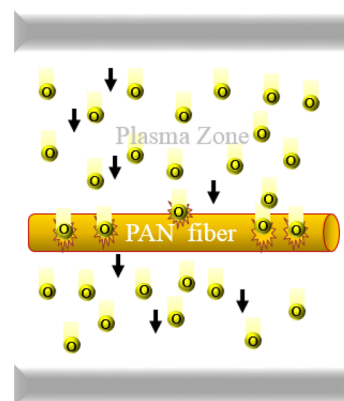


Fig. 6. Schematic diagram of plasma treatment. PAN: polyacrylonitrile.

C≡N bonds increased more than that of PAN fibers which had been thermally oxidized at 220°C for 30 min. This peak was stronger with plasma treatment. However, the peak intensity at 2243 cm⁻¹ assigned to C≡N decreased with more exposure to plasma treatment. In the tensile test results of the obtained carbon fibers, carbon fibers obtained from PAN fibers oxidized by heat and plasma treatment did not exhibit outstanding improvements in tensile properties, compared to PAN fibers oxidized by conventional thermal treatment. However, the results of plasma treatment showed that it could reduce time and temperature in the oxidation process for manufacturing carbon fibers.

The mechanism and efficiency of the plasma oxidation of PAN fibers, and the mechanical properties of the resulting carbon fiber, were further investigated by Lee *et al.* [122]. By comparing the degree of cyclization in plasma- and thermal-oxidized PAN fibers using FT-IR, they found that the extent of cyclization reaction is equal to the intensity of the 1595 cm⁻¹ peak for C=N divided by the intensities of the 1595 cm⁻¹ and 2243 cm⁻¹ peaks. Their results showed that the extent of cyclization in plasma-oxidized PAN fibers is much higher than in thermally oxidized PAN, to the extent that 15 min of plasma treatment is comparable to 120 min of thermal treatment. This means that PAN fibers are more easily and more rapidly oxidized by the oxygen species induced by plasma treatment, and the resulting carbon fiber has a tensile strength (2.8 ± 0.3 GPa) that is 30% greater than carbon fiber produced from thermally oxidized PAN fibers.

3. Conclusion

This review summarizes the feasibility of the three major categories of radiation induced polymer stabilization processes: EB, γ-radiation, UV, and plasma treatment. Of the various precursors available for carbon fiber, PAN precursor fibers were selected as the subject of focus. All three forms of treatment have been proven capable of efficiently transforming the C≡N bonds of PAN into C=N bonds in a manner that is both simpler and faster than conventional thermal treatment. Furthermore, the carbon fibers produced from EB and plasma oxidized PAN fibers exhibit an enhanced tensile strength over carbon fiber obtained by thermal stabilization. It is therefore concluded that further development of these radiation-based oxidation processes can not

only significantly improve the speed of carbon fiber production, but also reduce its environmental impact.

Acknowledgements

This work was supported by Carbon Valley Project of the Ministry of Trade, Industry and Energy, Korea.

References

- [1] Donnet JB, Bansal RC. Carbon Fibers, M. Dekker, New York, NY (1984).
- [2] Fitzer E. Carbon fibres: present state and future expectations. In: Figueiredo JL, Bernardo CA, Baker RTK, Hüttinger KJ, eds. Carbon Fibers Filaments and Composites. NATO ASI Series Vol. 177, Springer, Netherlands, 3 (1990). http://dx.doi.org/10.1007/978-94-015-6847-0_1.
- [3] Park SJ. Carbon Fibers, Springer, New York, NY (2015).
- [4] Minus M, Kumar S. The processing, properties, and structure of carbon fibers. JOM, **57**, 52 (2005). <http://dx.doi.org/10.1007/s11837-005-0217-8>.
- [5] Chand S. Review carbon fibers for composites. J Mater Sci, **35**, 1303 (2000). <http://dx.doi.org/10.1023/A:1004780301489>.
- [6] Beltz LA, Gustafson RR. Cyclization kinetics of poly(acrylonitrile). Carbon, **34**, 561 (1996). [http://dx.doi.org/10.1016/0008-6223\(96\)00005-X](http://dx.doi.org/10.1016/0008-6223(96)00005-X).
- [7] Fitzer E. Pan-based carbon fibers: present state and trend of the technology from the viewpoint of possibilities and limits to influence and to control the fiber properties by the process parameters. Carbon, **27**, 621 (1989). [http://dx.doi.org/10.1016/0008-6223\(89\)90197-8](http://dx.doi.org/10.1016/0008-6223(89)90197-8).
- [8] Serkov AT, Budnitskii GA, Radishevskii MB, Medvedev VA, Zlatoustova LA. Improving carbon fibre production technology. Fibre Chem, **35**, 117 (2003). <http://dx.doi.org/10.1023/A:1024838312261>.
- [9] Perepelkin KE. Oxidized (cyclized) polyacrylonitrile fibres: oxypan. A review. Fibre Chem, **35**, 409 (2003). <http://dx.doi.org/10.1023/B:FICH.0000020769.42823.31>.
- [10] Park SJ, Jang YS, Shim JW, Ryu SK. Studies on pore structures and surface functional groups of pitch-based activated carbon fibers. J Colloid Interface Sci, **260**, 259 (2003). [http://dx.doi.org/10.1016/S0021-9797\(02\)00081-4](http://dx.doi.org/10.1016/S0021-9797(02)00081-4).
- [11] Ma X, Yuan C, Liu X. Mechanical, microstructure and surface characterizations of carbon fibers prepared from cellulose after liquefying and curing. Materials, **7**, 75 (2014). <http://dx.doi.org/10.3390/ma7010075>.
- [12] Wu Q, Pan D. A new cellulose based carbon fiber from a lyocell precursor. Text Res J, **72**, 405 (2002). <http://dx.doi.org/10.1177/004051750207200506>.
- [13] Bronikowski MJ, Willis PA, Colbert DT, Smith KA, Smalley RE. Gas-phase production of carbon single-walled nanotubes from carbon monoxide via the HiPco process: a parametric study. J Vac Sci Technol A, **19**, 1800 (2001). <http://dx.doi.org/10.1116/1.1380721>.
- [14] Shim JW, Park SJ, Ryu SK. Effect of modification with HNO₃ and NaOH on metal adsorption by pitch-based activated carbon fibers. Carbon, **39**, 1635 (2001). [http://dx.doi.org/10.1016/S0008-6223\(00\)00290-6](http://dx.doi.org/10.1016/S0008-6223(00)00290-6).
- [15] Donnet JB, Park SJ. Surface characteristics of pitch-based carbon fibers by inverse gas chromatography method. Carbon, **29**, 955 (1991). [http://dx.doi.org/10.1016/0008-6223\(91\)90174-H](http://dx.doi.org/10.1016/0008-6223(91)90174-H).
- [16] Fennessey SF, Farris RJ. Fabrication of aligned and molecularly oriented electrospun polyacrylonitrile nanofibers and the mechanical behavior of their twisted yarns. Polymer, **45**, 4217 (2004). <http://dx.doi.org/10.1016/j.polymer.2004.04.001>.
- [17] He D, Wang C, Bai Y, Zhu B. Comparison of structure and properties among various PAN fibers for carbon fibers. J Mater Sci Technol, **21**, 376 (2005).
- [18] Morgan P. Carbon Fibers and Their Composites, Taylor & Francis, Boca Raton, FL (2005).
- [19] Shen X, Ji Y, Wang J. Preparation and pH-sensitivity of polyacrylonitrile (PAN) based porous hollow gel fibers. J Appl Polym Sci, **110**, 313 (2008). <http://dx.doi.org/10.1002/app.28176>.
- [20] Bashir Z. A critical review of the stabilisation of polyacrylonitrile. Carbon, **29**, 1081 (1991). [http://dx.doi.org/10.1016/0008-6223\(91\)90024-D](http://dx.doi.org/10.1016/0008-6223(91)90024-D).
- [21] Frank E, Hermanutz F, Buchmeiser MR. Carbon fibers: precursors, manufacturing, and properties. Macromol Mater Eng, **297**, 493 (2012). <http://dx.doi.org/10.1002/mame.201100406>.
- [22] Ouyang Q, Cheng L, Wang H, Li K. Mechanism and kinetics of the stabilization reactions of itaconic acid-modified polyacrylonitrile. Polym Degradation Stab, **93**, 1415 (2008). <http://dx.doi.org/10.1016/j.polymdegradstab.2008.05.021>.
- [23] Ryu SK, Park BJ, Park SJ. XPS analysis of carbon fiber surfaces: anodized and interfacial effects in fiber-epoxy composites. J Colloid Interface Sci, **215**, 167 (1999). <http://dx.doi.org/10.1006/jcis.1999.6240>.
- [24] Park SJ, Kim JS. Influence of plasma treatment on microstructures and acid-base surface energetics of nanostructured carbon blacks: N₂ plasma environment. J Colloid Interface Sci, **244**, 336 (2001). <http://dx.doi.org/10.1006/jcis.2001.7920>.
- [25] Sherwood PMA. Surface analysis of carbon and carbon fibers for composites. J Electron Spectrosc Relat Phenom, **81**, 319 (1996). [http://dx.doi.org/10.1016/0368-2048\(95\)02529-4](http://dx.doi.org/10.1016/0368-2048(95)02529-4).
- [26] Park SJ, Kim BJ. Ammonia removal of activated carbon fibers produced by oxyfluorination. J Colloid Interface Sci, **291**, 597 (2005). <http://dx.doi.org/10.1016/j.jcis.2005.05.012>.
- [27] Park SJ, Kim BJ. Influence of ozone treatment on Cr(VI) adsorption of activated carbon. Korean Chem Eng Res, **44**, 644 (2006).
- [28] Wang YQ, Sherwood PMA. Studies of carbon nanotubes and fluorinated nanotubes by X-ray and ultraviolet photoelectron spectroscopy. Chem Mater, **16**, 5427 (2004). <http://dx.doi.org/10.1021/cm040050t>.
- [29] Dubkova VI, Rodtsevich SP, Komarevich VG, Kotov DA. Influence of ion-beam carbon-fiber surface treatment on the angle of wetting by epoxy oligomers. J Eng Phys Thermophys, **78**, 519 (2005). <http://dx.doi.org/10.1007/s10891-005-0089-3>.
- [30] Park SJ, Seo MK, Kim HY, Lee DR. Studies on PAN-based carbon fibers irradiated by Ar⁺ ion beams. J Colloid Interface Sci, **261**, 393 (2003). [http://dx.doi.org/10.1016/S0021-9797\(03\)00091-2](http://dx.doi.org/10.1016/S0021-9797(03)00091-2).
- [31] Park SJ, Seo MK, Rhee KY. Effect of Ar⁺ ion beam irradiation on the physicochemical characteristics of carbon fibers. Carbon, **41**, 592 (2003). [http://dx.doi.org/10.1016/S0008-6223\(02\)00395-0](http://dx.doi.org/10.1016/S0008-6223(02)00395-0).
- [32] Park SJ, Kim KD. Influence of anodic surface treatment of activated carbon on adsorption and ion exchange properties. J Colloid Interface Sci, **218**, 331 (1999). <http://dx.doi.org/10.1006/jcis.1999.6387>.
- [33] Park SJ, Jang YS. Preparation and characterization of activated

- carbon fibers supported with silver metal for antibacterial behavior. *J Colloid Interface Sci*, **261**, 238 (2003). [http://dx.doi.org/10.1016/s0021-9797\(03\)00083-3](http://dx.doi.org/10.1016/s0021-9797(03)00083-3).
- [34] Heo GY, Hong YT, Park SJ. Preparation and characterization of nickel-coated carbon nanofibers produced from the electrospinning of polyamideimide precursor. *Macromol Res*, **20**, 503 (2012). <http://dx.doi.org/10.1007/s13233-012-0075-5>.
- [35] Kim S, Park SJ. Effect of acid/base treatment to carbon blacks on preparation of carbon-supported platinum nanoclusters. *Electrochim Acta*, **52**, 3013 (2007). <http://dx.doi.org/10.1016/j.electacta.2006.09.060>.
- [36] Park SJ, Seo MK, Nah C. Influence of surface characteristics of carbon blacks on cure and mechanical behaviors of rubber matrix compoundings. *J Colloid Interface Sci*, **291**, 229 (2005). <http://dx.doi.org/10.1016/j.jcis.2005.04.103>.
- [37] Park SJ, Kim KD. Adsorption behaviors of CO₂ and NH₃ on chemically surface-treated activated carbons. *J Colloid Interface Sci*, **212**, 186 (1999). <http://dx.doi.org/10.1006/jcis.1998.6058>.
- [38] Kim BJ, Park H, Park SJ. Toxic gas removal behaviors of porous carbons in the presence of Ag/Ni bimetallic clusters. *Bull Korean Chem Soc*, **29**, 782 (2008). <http://dx.doi.org/10.5012/bkcs.2008.29.4.782>.
- [39] Long RQ, Yang RT. Carbon nanotubes as superior sorbent for dioxin removal. *J Am Chem Soc*, **123**, 2058 (2001). <http://dx.doi.org/10.1021/ja003830l>.
- [40] Chinthaginjala JK, Seshan K, Lefferts L. Preparation and application of carbon-nanofiber based microstructured materials as catalyst supports. *Ind Eng Chem Res*, **46**, 3968 (2007). <http://dx.doi.org/10.1021/ie061394r>.
- [41] Park SJ, Seo MK, Lee JR, Lee DR. Studies on epoxy resins cured by cationic latent thermal catalysts: the effect of the catalysts on the thermal, rheological, and mechanical properties. *J Polym Sci A*, **39**, 187 (2001). [http://dx.doi.org/10.1002/1099-0518\(20010101\)39:1<187::AID-POLA210>3.0.CO;2-H](http://dx.doi.org/10.1002/1099-0518(20010101)39:1<187::AID-POLA210>3.0.CO;2-H).
- [42] Im JS, Kwon O, Kim YH, Park SJ, Lee YS. The effect of embedded vanadium catalyst on activated electrospun CFs for hydrogen storage. *Microporous Mesoporous Mater*, **115**, 514 (2008). <http://dx.doi.org/10.1016/j.micromeso.2008.02.027>.
- [43] Park SJ, Kim BJ. Roles of acidic functional groups of carbon fiber surfaces in enhancing interfacial adhesion behavior. *Mater Sci Eng A*, **408**, 269 (2005). <http://dx.doi.org/10.1016/j.msea.2005.08.129>.
- [44] Park SJ, Lee EJ, Kwon SH. Influence of surface treatment of polyimide film on adhesion enhancement between polyimide and metal films. *Bull Korean Chem Soc*, **28**, 188 (2007). <http://dx.doi.org/10.5012/bkcs.2007.28.2.188>.
- [45] Park SJ, Park BJ, Ryu SK. Electrochemical treatment on activated carbon fibers for increasing the amount and rate of Cr(VI) adsorption. *Carbon*, **37**, 1223 (1999). [http://dx.doi.org/10.1016/S0008-6223\(98\)00318-2](http://dx.doi.org/10.1016/S0008-6223(98)00318-2).
- [46] Wang YQ, Zhang FQ, Sherwood PMA. X-ray photoelectron spectroscopic study of carbon fiber surfaces. 23. Interfacial interactions between polyvinyl alcohol and carbon fibers electrochemically oxidized in nitric acid solution. *Chem Mater*, **11**, 2573 (1999). <http://dx.doi.org/10.1021/cm9902772>.
- [47] Park SJ, Park BJ. Electrochemically modified PAN carbon fibers and interfacial adhesion in epoxy-resin composites. *J Mater Sci Lett*, **18**, 47 (1999). <http://dx.doi.org/10.1023/A:1006673309571>.
- [48] Ko TH. Characterization of PAN-based nonburning (nonflammable) fibers. *J Appl Polym Sci*, **47**, 707 (1993). <http://dx.doi.org/10.1002/app.1993.070470414>.
- [49] Grassie N, McGuchan R. Pyrolysis of polyacrylonitrile and related polymers: I. Thermal analysis of polyacrylonitrile. *Eur Polym J*, **6**, 1277 (1970). [http://dx.doi.org/10.1016/0014-3057\(70\)90046-7](http://dx.doi.org/10.1016/0014-3057(70)90046-7).
- [50] Grassie N, McGuchan R. Pyrolysis of polyacrylonitrile and related polymers: II. The effect of sample preparation on the thermal behaviour of polyacrylonitrile. *Eur Polym J*, **7**, 1091 (1971). [http://dx.doi.org/10.1016/0014-3057\(71\)90141-8](http://dx.doi.org/10.1016/0014-3057(71)90141-8).
- [51] Grassie N, Hay JN. Thermal coloration and insolubilization in polyacrylonitrile. *J Polym Sci*, **56**, 189 (1962). <http://dx.doi.org/10.1002/pol.1962.1205616316>.
- [52] Yun JH, Kim BH, Yang KS, Bang YH, Kim SR, Woo HG. Process optimization for preparing high performance PAN-based carbon fibers. *Bull Korean Chem Soc*, **30**, 2253 (2009). <http://dx.doi.org/10.5012/bkcs.2009.30.10.2253>.
- [53] Yu M, Wang C, Bai Y, Wang Y, Xu Y. Influence of precursor properties on the thermal stabilization of polyacrylonitrile fibers. *Polym Bull*, **57**, 757 (2006). <http://dx.doi.org/10.1007/s00289-006-0629-9>.
- [54] Bansal RC, Donnet JB, Stoeckli F. *Active Carbon*. M. Dekker, New York, NY (1988).
- [55] Cho CW, Cho D, Ko YG, Kwon OH, Kang IK. Stabilization, carbonization, and characterization of PAN precursor webs processed by electrospinning technique. *Carbon Lett*, **8**, 313 (2007).
- [56] Bajaj P, Sreekumar TV, Sen K. Thermal behaviour of acrylonitrile copolymers having methacrylic and itaconic acid comonomers. *Polymer*, **42**, 1707 (2001). [http://dx.doi.org/10.1016/S0032-3861\(00\)00583-8](http://dx.doi.org/10.1016/S0032-3861(00)00583-8).
- [57] Rahaman MSA, Ismail AF, Mustafa A. A review of heat treatment on polyacrylonitrile fiber. *Polym Degradation Stab*, **92**, 1421 (2007). <http://dx.doi.org/10.1016/j.polymdegradstab.2007.03.023>.
- [58] Liu Y, Kumar S. Recent progress in fabrication, structure, and properties of carbon fibers. *Polym Rev*, **52**, 234 (2012). <http://dx.doi.org/10.1080/15583724.2012.705410>.
- [59] Schnabel W. *Polymer Degradation: Principles and Practical Applications*. Hanser Publishers, Munich, Germany (1981).
- [60] Parejo Calvo WA, Duarte CL, Machado LDB, Manzoli JE, Geraldo ABC, Kodama Y, Silva LGA, Pino ES, Somessari ESR, Silveira CG, Rela PR. Electron beam accelerators: trends in radiation processing technology for industrial and environmental applications in Latin America and the Caribbean. *Radiat Phys Chem*, **81**, 1276 (2012). <http://dx.doi.org/10.1016/j.radphyschem.2012.02.013>.
- [61] International Atomic Energy Agency. Industrial electron beam processing. Consultants' Meeting on the "Preparation of the status report on low energy, self-shielded electron accelerators and of industrial scale electron/X-ray irradiators", Vienna, Austria (2008).
- [62] Nishi Y, Sato H, Iwata K, Nishi Y, Iwata K. Effects of homogeneous irradiation of electron beam with low potential on adhesive strength of polymethyl methacrylate composite sheet covered with nylon-6 film. *J Mater Res*, **24**, 3503 (2009). <http://dx.doi.org/doi:10.1557/jmr.2009.0429>.
- [63] Shin BS, Seo DK, Kim HB, Jeun JP, Kang PH. A study of the thermal and mechanical properties of electron beam irradiated HDPE/EPDM blends in the presence of triallyl cyanurate. *J Ind Eng Chem*, **18**, 526 (2012). <http://dx.doi.org/10.1016/j.jiec.2011.11.025>.
- [64] Senna MM, Mohamed RM, Shehab-Eldin AN, El-Hamouly S.

- Characterization of electron beam irradiated natural rubber/modified starch composites. *J Ind Eng Chem*, **18**, 1654 (2012). <http://dx.doi.org/10.1016/j.jiec.2012.03.004>.
- [65] Yoon HJ, Kim SE, Kwon YK, Kim EJ, Lee JC, Lee YS. Synthesis of silver nanostructures on polytetrafluoroethylene (PTFE) using electron beam irradiation for antimicrobial effect. *J Ind Eng Chem*, **18**, 586 (2012). <http://dx.doi.org/10.1016/j.jiec.2011.10.007>.
- [66] Dahal P, Kim YC. Preparation and characterization of modified polypropylene by using electron beam irradiation. *J Ind Eng Chem*, **19**, 1879 (2013). <http://dx.doi.org/10.1016/j.jiec.2013.02.027>.
- [67] Shukushima S, Hayami H, Ito T, Nishimoto S. Modification of radiation cross-linked polypropylene. *Radiat Phys Chem*, **60**, 489 (2001). [http://dx.doi.org/10.1016/S0969-806X\(00\)00395-9](http://dx.doi.org/10.1016/S0969-806X(00)00395-9).
- [68] Chmielewski AG, Al-Sheikhly M, Berejka AJ, Cleland MR, Antoniak M. Recent developments in the application of electron accelerators for polymer processing. *Radiat Phys Chem*, **94**, 147 (2014). <http://dx.doi.org/10.1016/j.radphyschem.2013.06.024>.
- [69] Kang PH, Jeon YK, Jeun JP, Shin JW, Nho YC. Effect of electron beam irradiation on polyimide film. *J Ind Eng Chem*, **14**, 672 (2008). <http://dx.doi.org/10.1016/j.jiec.2008.03.004>.
- [70] An JC. Synthesis of the combined inter- and intra-crosslinked nanohydrogels by e-beam ionizing radiation. *J Ind Eng Chem*, **16**, 657 (2010). <http://dx.doi.org/10.1016/j.jiec.2010.05.013>.
- [71] Miao P, Wu D, Zeng K, Xu G, Zhao Ce, Yang G. Influence of electron beam pre-irradiation on the thermal behaviors of polyacrylonitrile. *Polym Degradation Stab*, **95**, 1665 (2010). <http://dx.doi.org/10.1016/j.polyimdegradstab.2010.05.028>.
- [72] Park M, Pant B, Choi J, Park YW, Lee C, Shin HK, Park SJ, Kim HY. Facile preparation of self-assembled wool-based graphene hydrogels by electron beam irradiation. *Carbon Lett*, **15**, 136 (2014). <http://dx.doi.org/10.5714/CL.2014.15.2.136>.
- [73] Shin HK, Pyo Jeun J, Bin Kim H, Hyun Kang P. Isolation of cellulose fibers from kenaf using electron beam. *Radiat Phys Chem*, **81**, 936 (2012). <http://dx.doi.org/10.1016/j.radphyschem.2011.10.010>.
- [74] Park M, Shin HK, Kim BS, Pant B, Barakat NAM, Kim HY. Facile preparation of graphene induced from electron-beam irradiated graphite. *Mater Lett*, **105**, 236 (2013). <http://dx.doi.org/10.1016/j.matlet.2013.04.027>.
- [75] Liu Y, Park M, Shin HK, Pant B, Park SJ, Kim HY. Preparation and characterization of chitosan-based nanofibers by ecofriendly electrospinning. *Mater Lett*, **132**, 23 (2014). <http://dx.doi.org/10.1016/j.matlet.2014.06.041>.
- [76] Siraj K, Khaleeq-ur-Rahman M, Rafique MS, Nawaz T. Effect of 4MeV electron beam irradiation on carbon films. *Nucl Instr Method Phys Res B*, **269**, 53 (2011). <http://dx.doi.org/10.1016/j.nimb.2010.09.022>.
- [77] Shin H-S, Kim YR, Han B, Makarov IE, Ponomarev AV, Pikaev AK. Application of electron beam to treatment of wastewater from papermill. *Radiat Phys Chem*, **65**, 539 (2002). [http://dx.doi.org/10.1016/S0969-806X\(02\)00348-1](http://dx.doi.org/10.1016/S0969-806X(02)00348-1).
- [78] Supriya P, Sridhar KR, Ganesh S. Fungal decontamination and enhancement of shelf life of edible split beans of wild legume *Canavalia maritima* by the electron beam irradiation. *Radiat Phys Chem*, **96**, 5 (2014). <http://dx.doi.org/10.1016/j.radphyschem.2013.08.007>.
- [79] Dänmark S, Finne-Wistrand A, Schander K, Hakkarainen M, Arvidson K, Mustafa K, Albertsson AC. In vitro and in vivo degradation profile of aliphatic polyesters subjected to electron beam sterilization. *Acta Biomater*, **7**, 2035 (2011). <http://dx.doi.org/10.1016/j.actbio.2011.02.011>.
- [80] Allen JT, Calhoun R, Helm J, Kruger S, Lee C, Mendonsa R, Meyer S, Pageau G, Shaffer H, Whitham K, Williams CB, Farrell JP. A fully integrated 10 MeV electron beam sterilization system. *Radiat Phys Chem*, **46**, 457 (1995). [http://dx.doi.org/10.1016/0969-806X\(95\)00193-2](http://dx.doi.org/10.1016/0969-806X(95)00193-2).
- [81] Odelius K, Pliik P, Albertsson AC. The influence of composition of porous copolyester scaffolds on reactions induced by irradiation sterilization. *Biomaterials*, **29**, 129 (2008). <http://dx.doi.org/10.1016/j.biomaterials.2007.08.046>.
- [82] Luan S, Shi H, Yao Z, Wang J, Song Y, Yin J. Effect of electron beam irradiation sterilization on the biomedical poly (octene-co-ethylene)/polypropylene films. *Nucl Instr Method Phys Res B*, **268**, 1474 (2010). <http://dx.doi.org/10.1016/j.nimb.2010.01.014>.
- [83] Auslender VL, Bryazgin AA, Voronin LA, Polyakov VA, Grodetskiy VP, Izhboldin IK, Mirsaetov OM, Petrov AM, Obidin YT, Ponomaryov VN. Automated technological radiation installation for sterilization of medical goods. *Radiat Phys Chem*, **52**, 459 (1998). [http://dx.doi.org/10.1016/S0969-806X\(98\)00051-6](http://dx.doi.org/10.1016/S0969-806X(98)00051-6).
- [84] Zhang M, Zhu R, Zhang M, Gao B, Sun D, Wang S. High-energy pulse-electron-beam-induced molecular and cellular damage in *Saccharomyces cerevisiae*. *Res Microbiol*, **164**, 100 (2013). <http://dx.doi.org/10.1016/j.resmic.2012.10.023>.
- [85] Park W, Hwang MH, Kim TH, Lee MJ, Kim IS. Enhancement in characteristics of sewage sludge and anaerobic treatability by electron beam pre-treatment. *Radiat Phys Chem*, **78**, 124 (2009). <http://dx.doi.org/10.1016/j.radphyschem.2008.09.010>.
- [86] Ribeiro MA, Sato IM, Duarte CL, Sampa MHO, Salvador VLR, Scapin MA. Application of the electron-beam treatment for Ca, Si, P, Al, Fe, Cr, Zn, Co, As, Se, Cd and Hg removal in the simulated and actual industrial effluents. *Radiat Phys Chem*, **71**, 425 (2004). <http://dx.doi.org/10.1016/j.radphyschem.2004.03.017>.
- [87] Sampa MHO, Relá PR, Casas AL, Mori MN, Duarte CL. Treatment of industrial effluents using electron beam accelerator and adsorption with activated carbon: a comparative study. *Radiat Phys Chem*, **71**, 459 (2004). <http://dx.doi.org/10.1016/j.radphyschem.2004.03.023>.
- [88] Duarte CL, Sampa MHO, Relá PR, Oikawa H, Cherbakian EH, Sena HC, Abe H, Sciani V. Application of electron beam irradiation combined to conventional treatment to treat industrial effluents. *Radiat Phys Chem*, **57**, 513 (2000). [http://dx.doi.org/10.1016/S0969-806X\(99\)00453-3](http://dx.doi.org/10.1016/S0969-806X(99)00453-3).
- [89] Sampa MHO, Duarte CL, Relá PR, Somessari ESR, Silveira CG, Azevedo AL. Remotion of organic compounds of actual industrial effluents by electron beam irradiation. *Radiat Phys Chem*, **52**, 365 (1998). [http://dx.doi.org/10.1016/S0969-806X\(98\)00035-8](http://dx.doi.org/10.1016/S0969-806X(98)00035-8).
- [90] Duarte CL, Geraldo LL, Junior OdAP, Borrelly SI, Sato IM, Sampa MHO. Treatment of effluents from petroleum production by electron beam irradiation. *Radiat Phys Chem*, **71**, 445 (2004). <http://dx.doi.org/10.1016/j.radphyschem.2004.03.021>.
- [91] Moraes MCF, Romanelli MF, Sena HC, Pasqualini da Silva G, Sampa MHO, Borrelly SI. Whole acute toxicity removal from industrial and domestic effluents treated by electron beam radiation: emphasis on anionic surfactants. *Radiat Phys Chem*, **71**, 463 (2004). <http://dx.doi.org/10.1016/j.radphyschem.2004.03.022>.
- [92] Borrelly SI, Gonçalves AA, Oikawa H, Duarte CL, Rocha FR. Electron beam accelerator for detoxification of effluents. When

- radiation processing can enhance the acute toxicity? *Radiat Phys Chem*, **71**, 455 (2004). <http://dx.doi.org/10.1016/j.radphyschem.2004.03.087>.
- [93] Han B, Kyu Kim J, Kim Y, Seung Choi J, Young Jeong K. Operation of industrial-scale electron beam wastewater treatment plant. *Radiat Phys Chem*, **81**, 1475 (2012). <http://dx.doi.org/10.1016/j.radphyschem.2012.01.030>.
- [94] Kim YH, Choi SJ, Park HJ, Lee JH. Electron beam-induced mutants of microalgae *Arthrospira platensis* increased antioxidant activity. *J Ind Eng Chem*, **20**, 1834 (2014). <http://dx.doi.org/10.1016/j.jiec.2013.08.039>.
- [95] Katial RK, Grier TJ, Hazelhurst DM, Hershey J, Engler RJ. Deteriorous effects of electron beam radiation on allergen extracts. *J Allergy Clin Immunol*, **110**, 215 (2002). <http://dx.doi.org/10.1067/mai.2002.126377>.
- [96] Lim SJ, Kim TH, Lee SH, Kim JY, Kim SK. Effects of electron beam irradiation and temperature on the treatment of swine wastewater using an ion exchange biological reactor. *Bioreour Technol*, **137**, 233 (2013). <http://dx.doi.org/10.1016/j.biortech.2013.03.083>.
- [97] Lozada-Castro JJ, Gil-Díaz M, Santos-Delgado MJ, Rubio-Barroso S, Polo-Díez LM. Effect of electron-beam irradiation on cholesterol oxide formation in different ready-to-eat foods. *Innov Food Sci Emerg Technol*, **12**, 519 (2011). <http://dx.doi.org/10.1016/j.ifset.2011.07.005>.
- [98] Ramathilaga A, Murugesan AG. Effect of electron beam irradiation on proximate, microbiological and sensory characteristics of chyavanaprash: ayurvedic poly herbal formulation. *Innov Food Sci Emerg Technol*, **12**, 515 (2011). <http://dx.doi.org/10.1016/j.ifset.2011.06.004>.
- [99] Rivadeneira R, Moreira R, Kim J, Castell-Perez ME. Dose mapping of complex-shaped foods using electron-beam accelerators. *Food Control*, **18**, 1223 (2007). <http://dx.doi.org/10.1016/j.foodcont.2006.07.023>.
- [100] Al-Farisi M, Abuagla A, Mohamed E, Gohs U. The effect of electron beam on dates infestation. *Food Control*, **33**, 157 (2013). <http://dx.doi.org/10.1016/j.foodcont.2013.02.029>.
- [101] Martin DI, Margaritescu I, Cirstea E, Togoe I, Ighigeanu D, Nemtanu MR, Oproiu C, Iacob N. Application of accelerated electron beam and microwave irradiation to biological waste treatment. *Vacuum*, **77**, 501 (2005). <http://dx.doi.org/10.1016/j.vacuum.2004.09.019>.
- [102] Shin H, Jeun J, Kang P. The characterization of polyacrylonitrile fibers stabilized by electron beam irradiation. *Fiber Polym*, **13**, 724 (2012). <http://dx.doi.org/10.1007/s12221-012-0724-5>.
- [103] Shin HK, Park M, Kang PH, Choi HS, Park SJ. Preparation and characterization of polyacrylonitrile-based carbon fibers produced by electron beam irradiation pretreatment. *J Ind Eng Chem*, **20**, 3789 (2014). <http://dx.doi.org/10.1016/j.jiec.2013.12.080>.
- [104] Kim DY, Shin HK, Jeun JP, Kim HB, Oh SH, Kang PH. Characterization of polyacrylonitrile based carbon nanofiber mats via electron beam processing. *J Nanosci Nanotechnol*, **12**, 6120 (2012). <http://dx.doi.org/10.1166/jnn.2012.6346>.
- [105] Choi Y, Park M, Shin HK, Liu Y, Choi JW, Nirmala R, Park SJ, Kim HY. Facile stabilization process of polyacrylonitrile-based electrospun nanofibers by spraying 1% hydrogen peroxide and electron beam irradiation. *Mater Lett*, **123**, 59 (2014). <http://dx.doi.org/10.1016/j.matlet.2014.03.020>.
- [106] Ražem D, Katušin-Ražem B. The effects of irradiation on controlled drug delivery/controlled drug release systems. *Radiat Phys Chem*, **77**, 288 (2008). <http://dx.doi.org/10.1016/j.radphyschem.2007.06.006>.
- [107] Williams HE, Huxley J, Claybourn M, Booth J, Hobbs M, Meehan E, Clark B. The effect of γ -irradiation and polymer composition on the stability of PLG polymer and microspheres. *Polym Degradation Stab*, **91**, 2171 (2006). <http://dx.doi.org/10.1016/j.polymdegradstab.2006.01.006>.
- [108] Igartua M, Hernández RM, Rosas JE, Patarroyo ME, Pedraz JL. Gamma-irradiation effects on biopharmaceutical properties of PLGA microspheres loaded with SPf66 synthetic vaccine. *Eur J Pharm Biopharm*, **69**, 519 (2008). <http://dx.doi.org/10.1016/j.ejpb.2007.12.014>.
- [109] Tan L, Wan A. Structural changes of polyacrylonitrile precursor fiber induced by γ -ray irradiation. *Mater Lett*, **65**, 3109 (2011). <http://dx.doi.org/10.1016/j.matlet.2011.06.090>.
- [110] Liu W, Wang M, Xing Z, Qi Y, Wu G. Radiation-induced cross-linking of polyacrylonitrile fibers and the subsequent regulative effect on the preoxidation process. *Radiat Phys Chem*, **81**, 622 (2012). <http://dx.doi.org/10.1016/j.radphyschem.2012.02.029>.
- [111] Liu W, Wang M, Xing Z, Wu G. Radiation oxidation and subsequent thermal curing of polyacrylonitrile fiber. *Radiat Phys Chem*, **94**, 9 (2014). <http://dx.doi.org/10.1016/j.radphyschem.2013.06.015>.
- [112] Cowd MA. *Polymer Chemistry*, Murray, London (1982).
- [113] Kuleznev VN, Shershnev VA. *The Chemistry and Physics of Polymers*, Mir Publishers, Moscow (1990).
- [114] Fouda IM, Shabana HM, El-Sharkawy FM. Changes in orientation caused by UV irradiation of nylon 6 fibers. *J Appl Polym Sci*, **88**, 3202 (2003). <http://dx.doi.org/10.1002/app.12063>.
- [115] Nough SA. The effect of UV radiation on the optical properties of cellulose triacetate. *Radiat Measur*, **27**, 499 (1997). [http://dx.doi.org/10.1016/S1350-4487\(97\)00006-1](http://dx.doi.org/10.1016/S1350-4487(97)00006-1).
- [116] Paiva MC, Kotasthane P, Edie DD, Ogale AA. UV stabilization route for melt-processible PAN-based carbon fibers. *Carbon*, **41**, 1399 (2003). [http://dx.doi.org/10.1016/S0008-6223\(03\)00041-1](http://dx.doi.org/10.1016/S0008-6223(03)00041-1).
- [117] Morales MS, Ogale AA. Wet-spun, photoinitiator-modified polyacrylonitrile precursor fibers: UV-assisted stabilization. *J Appl Polym Sci*, **130**, 2494 (2013). <http://dx.doi.org/10.1002/app.39442>.
- [118] Řezničková A, Kolská Z, Hnatowicz V, Stopka P, Švorčík V. Comparison of glow argon plasma-induced surface changes of thermoplastic polymers. *Nucl Instr Method Phys Res B*, **269**, 83 (2011). <http://dx.doi.org/10.1016/j.nimb.2010.11.018>.
- [119] Švorčík V, Kolářová K, Slepíčka P, Macková A, Novotná M, Hnatowicz V. Modification of surface properties of high and low density polyethylene by Ar plasma discharge. *Polym Degradation Stab*, **91**, 1219 (2006). <http://dx.doi.org/10.1016/j.polymdegradstab.2005.09.007>.
- [120] Park SJ, Kim BJ. Influence of oxygen plasma treatment on hydrogen chloride removal of activated carbon fibers. *J Colloid Interface Sci*, **275**, 590 (2004). <http://dx.doi.org/10.1016/j.jcis.2004.03.011>.
- [121] Lee SW, Lee HY, Jang SY, Jo SM, Lee HS, Lee S. Tensile properties and morphology of carbon fibers stabilized by plasma treatment. *Carbon Lett*, **12**, 16 (2011).
- [122] Lee SW, Lee HY, Jang SY, Jo S, Lee HS, Choe WH, Lee S. Efficient preparation of carbon fibers using plasma assisted stabilization. *Carbon*, **55**, 361 (2013). <http://dx.doi.org/10.1016/j.carbon.2012.10.062>.