



Research Article

Effects of chlorine dioxide gas fumigation on the storage quality of astringent persimmon (*Diospyros kaki* T.) *Cheongdobansi*

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Abstract Because of their short harvest season, large quantities of persimmons must often be processed within a limited time. Therefore, new methods to extend their storage life are required. This study examined the effects of chlorine dioxide (ClO₂) gas fumigation for various treatment periods on the storage quality of astringent persimmons *Cheongdobansi* under low-temperature conditions. The conditions consisted of continuous treatment with ClO₂, treatment for 2 weeks with ClO₂, and no treatment, all of which are stored at low temperatures. Control samples (storage 0 days) without any treatment were prepared and all experiments were conducted for 10 weeks at two-week intervals. The ClO₂ gas treatment maintained the moisture content, color value, hardness, soluble tannin content, and sensory characteristics. However, ClO₂ gas treatment did not affect the soluble solids, pH, and total sugar content. In particular, continuous treatment with ClO₂ maintained the storage quality after 6–8 weeks of storage, particularly the hardness and weakness (sensory evaluation). The results suggest the potential of continuous treatment with ClO₂ as a highly effective method for maintaining the freshness of *Cheongdobansi*.

Keywords *Cheongdobansi*, astringent persimmon, chlorine dioxide gas, fumigation, storage



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1. Introduction

Persimmons (*Diospyros kaki* Thunb.) are a globally cultivated fruit, with significant production in its native region of East Asia (Nam et al., 1998). Various horticultural varieties exist and they are generally classified as astringent or non-astringent, which determines their post-harvest consumption method (Kim et al., 2004). Among the astringent varieties, *Cheongdobansi* is a variety produced in Cheongdo, North Gyeongsang Province, at a rate of approximately 40,000 tons per year. *Cheongdobansi* becomes edible only after astringency removal or softening owing to its high water-soluble tannin content. Furthermore, it has generally been processed and distributed as a soft persimmon owing to its tender flesh and lack of seeds (Cheongdo Agricultural Technology Center, 2023). However, it is increasingly processed and distributed in a slice-dried form for

extended shelf life and added value (Kim et al., 2009). *Cheongdobansi* are harvested in late September and kept refrigerated until the peeled fruits are processed into dried and slice-dried persimmons for mass distribution. Therefore, one major drawback is that a large quantity of persimmons must be processed within a limited time frame (Kim et al., 2018; Praeger et al., 2018). For this reason, it is important to develop post-harvest storage technologies that ensure that the fruit remains safe for human consumption, minimize external and internal physical damage to the fruit, and preserve its quality characteristics to maintain consistency in the quality of dried persimmon products.

Various processing methods are utilized to preserve the quality of fruits that are harvested en masse at a single time point, such as low-temperature storage (Kim et al., 2017), high CO₂ treatment (Kim et al., 2012), UV irradiation (Jang et al., 2012), hot water treatment (Park and Kim, 2002), controlled atmosphere (CA) storage (Chung et al., 2005), modified atmosphere storage (Kim et al., 2017), and 1-methylcyclopropene treatment (Win et al., 2017). Among these, ClO₂ treatment is utilized to enhance the storage quality of agricultural produce (Kim et al., 2016). Some key benefits of ClO₂ are that it is readily degraded in the air, leaving no residue on the food after treatment, and it can be applied over a broad pH range (pH 3-8) (Malka and Park, 2022). Furthermore, owing to its high water solubility and stability as dissolved gas in aqueous solutions, it is primarily used for fumigation in gaseous and aqueous forms (Moon et al., 2017). In contrast to chlorine, which is generally used to extend the storage of various fresh agricultural produce, ClO₂ does not generate carcinogens such as trihalomethane, or alter the nutritional and sensory qualities of the produce

because it does not react with organic substances (Praeger et al., 2018). Moreover, ClO₂ has been approved as a disinfectant for fresh agricultural produce and food processing industries owing to its efficacy and safety (FDA, 1998). Maintaining the storage quality of agricultural products requires technologies for regulating their respiration rate and ethylene biosynthesis. ClO₂ controls water transpiration by regulating the opening and closing of plant stomata (Chen and Zhu, 2011), and prevents weight loss by reducing the respiration rate (Guo et al., 2014). Furthermore, it delays ripening by regulating ethylene biosynthesis-related genes, namely ACS2, ACO1, and ACO3 (Guo et al., 2013). It can also prevent spoilage by inhibiting microbial growth, thereby preserving the sensory characteristics of fruits and vegetables, such as color and flavor (Chen et al., 2010).

ClO₂ gas has been used to enhance the storage quality of fruits such as tomatoes (Yang et al., 2020), strawberries (Kim and Hwang, 2019), paprika (Kang et al., 2015), and broccoli (Park and Lee, 2018). Studies that have used ClO₂ have primarily directly treated fruits and vegetables with aqueous ClO₂ (Park et al., 2008) or investigated the optimal concentration of ClO₂ gas (Kim et al., 2005). Various studies have been conducted to improve the storage quality of persimmons, including CO₂ gas adjustment (Cia et al., 2006), combined treatment with potassium lactate and heat (Naser et al., 2018), and edible coating based on gum arabic (Saleem et al., 2020). However, there is still a paucity of research on extending the storage quality of *Cheongdobansi* using ClO₂ gas. Therefore, this study analyzed the quality characteristics of *Cheongdobansi* over the storage period based on the duration of ClO₂ gas fumigation, and evaluated the possibility of extending the shelf life using this treatment. Therefore, this

study analyzed the quality characteristics of *Cheongdobansi* over the storage period based on the duration of ClO₂ gas fumigation. The chemical, physical, and sensory properties of the *Cheongdobansi* were evaluated and compared after various storage periods, with untreated samples were used as a control. From these results, we evaluated the possibility of extending the shelf life using this treatment.

2. Materials and methods

2.1. Experimental materials and storage conditions

The persimmons used in this study were a variety of astringent persimmon known as *Cheongdobansi* (*Diospyros kaki* T.). The persimmons were harvested in October 2022 and were purchased from a farm in Cheongdo, North Gyeongsang Province. Following harvest, the control group (CONT) was analyzed immediately (week 0). Subsequently, all other samples were stored at 0±1°C and 30% relative humidity. Persimmons stored under a low temperature conditions only (PNT), persimmons stored under low temperature conditions after two weeks of ClO₂ fumigation (P2T), and persimmons stored under continuous chlorine dioxide fumigation (PAT) were analyzed over a 10 week period at two-week intervals. ClO₂ gas was generated using a ClO₂ gas generator (CA-300, Purgofarm Inc., Yongin, Korea) fabricated according to the electrochemical method described by Gates (1998). The high-purity ClO₂ gas generated from a NaClO₂ aqueous solution passed through a porous membrane electrode assembly, underwent a series of electrochemical reactions, and flowed into the collection chamber via a vent. The gas stream was controlled with a regulator valve to maintain a ClO₂ gas concentration of 60 mg/L in the storage facility (Analytical Technology, Inc., Collegeville, PA, USA).

2.2. Determination of appearance

A smartphone (iPhone 11 Pro, Apple Inc., CA, USA) was placed 60 cm above a non-glossy white background to capture images of the appearance of the persimmons. To minimize shadows, a studio photo box (Puluz Photo Studio Light Box, Shenzhen Puluz Technology Co., Ltd., Guangdong, China) was used with 1,690 LM LED lights installed on two sides. Then, the *Cheongdobansi* was halved and placed on the white background to photograph its appearance. All pictures were taken in a dark room at 20°C.

2.3. Determination of moisture content

Changes in the moisture content were determined using the air-oven method of the Codex Alimentarius. A 3 g sample was weighed and spread on a aluminum dish. The sample was then dried in an oven at 105°C for 5 h and cooled in a desiccator for 30 min before measurement. The results were expressed as a percentage (%) using the following equation:

$$\text{Moisture (\%)} = \frac{W_1 - W_0}{W_0} \times 100 \quad (1)$$

W₀: the initial weight of sample.

W₁: the weight of sample after drying.

2.4. Determination of soluble solids and pH

A quarter of the *Cheongdobansi* flesh was minced using laboratory scissors and ground at 8,000 rpm for 30 s. The ground sample (10 g) was mixed with distilled water (400 g) in a media bottle (50 times dilution). An ultrasonicator (DH WUC, D22H, Daihan Scientific Co., Ltd., Wonju, Korea) was used to extract the mixture for 30 min at 200 W (50% of the maximum ultrasound power) at 40 kHz. Then, the extract was filtered through a Whatman® No. 2 filter paper to prepare the final *Cheongdobansi* extract. The soluble solid content and pH were measured

three times using a refractometer (0–32 °Brix) (Maste- α , Atsgo Co., Tokyo, Japan) and pH meter (Orion 3-Star, Thermo Scientific Co., Waltham, MA, USA), respectively.

2.5. Measurement of chromaticity and color difference

A random sample was selected and cut at a point 1.5 cm away from the stem, and the chromaticity of the cut surface was measured. A colorimeter (CR0400, Minolta, Tokyo, Japan), calibrated with a white plate ($L^*=61.04$, $a^*=35.70$, $b^*=58.02$), was used to take 10 measurements of the L^* (lightness), a^* (redness), and b^* (yellowness) values, and the average values were calculated. The color difference was calculated using the following equation based on the measured L^* , a^* , and b^* values:

$$\Delta E = \sqrt{\Delta L^{*2} + \Delta a^{*2} + \Delta b^{*2}} \quad (2)$$

ΔL^* : Difference lightness value between the sample stored for 0 d

Δa^* : Difference redness value between the sample stored for 0 d

Δb^* : Difference yellowness value between the sample stored for 0 d

2.6. Measurement of hardness

Hardness was measured using a texture profile analyzer (Compac-100II, SunScientific Co., Tokyo, Japan). A No.5 (\varnothing 5 mm) probe was used and the test conditions were set as follows: test mode, 20; probe penetration distance (%), 5 mm; table speed, 180 mm/min; and maximum load cell stress, 2 kg. The sample was cut at a point 1.5 cm from the stem, and 10 cut surfaces were measured to calculate the average.

2.7. Determination of total sugar content

The total sugar content was quantified using the

phenol-sulfuric acid method (Dubois et al., 1956). The *Cheongdobansi* extract prepared in Section 2.5 was diluted 800 times. A mixture of 1 mL of the diluted solution and 0.5 mL of 5% Folin-Ciocalteu's reagent (Sigma Chemicals, Co., St. Louis, MO, USA) was prepared. This mixture was incubated with 2.5 mL of sulfuric acid (Duksan Pure Chemicals Co., Ltd., Ansan, Korea) in a water bath at 85°C for 15 min and then left to stand at room temperature for 30 min. The absorbance was measured at 480 nm using a UV-Visible spectrophotometer (Evolution 201, Thermo Fisher Scientific Inc., Madison, WI, USA). Glucose (Duksan Pure Chemicals Co., Ltd., Ansan, Korea) was used as the standard material for calculating the sugar content, and a standard curve was generated to express the total sugar content as glucose equivalents.

2.8. Determination of soluble tannin content

The soluble tannin content was measured using the Folin-Denis method (Folin and Denis, 1912). The *Cheongdobansi* extract prepared in Section 2.5 was diluted eight times. One milliliter of the diluted solution was mixed with 1 mL of Folin-Ciocalteu's reagent (Sigma Chemicals, Co., St. Louis, MO, USA) at a 1:1 ratio and allowed to stand in a dark room for 15 min. After adding 10% Na_2CO_3 (Duksan Pure Chemicals Co., Ltd., Ansan, Korea), the mixture was vigorously shaken and left to stand in the dark for 1 h. The absorbance was measured at 760 nm using a UV-Visible spectrophotometer (Evolution 201, Thermo Fisher Scientific Inc., Madison, WI, USA). The soluble tannin content was calculated from a standard curve prepared with tannic acid (Sigma Chemicals, Co.).

2.9. Sensory evaluation

A sensory evaluation was conducted with 15

students from Kyungpook National University. The evaluation parameters were: degree of clearness, weakness, sweet flavor, and off-flavor, using a seven-point scale (1: very opaque or weak to 7: very clear or strong). Based on this, the overall acceptability was evaluated on a seven-point scale (1: very poor to 7: very good). The highest score for marketability was set to 2. Five *Cheongdobansi* samples were selected and each *Cheongdobansi* was cut into quarters before providing them to the participants. This sensory evaluation was conducted safely and with final exemption from the Kyungpook National University Institutional Review Board (IRB) (Approval number: KNU-2022-0488).

2.10. Statistical analysis

The results were analyzed with one-way analysis

of variance (ANOVA) using the SPSS (Statistical Package for Social Sciences, SPSS Inc., Chicago, IL, USA) software package. Duncan's multiple range test was performed to analyze the significance of the differences across storage periods and samples. The significance of the differences among the samples at week 2 were analyzed with the t-test ($p < 0.05$).

3. Results and discussion

3.1. Moisture content

Table 1 shows the changes in the weight loss rate of *Cheongdobansi* under different ClO_2 storage conditions. There were no significant differences in the moisture content based on the storage period. However, after week 8 of storage, the PAT and P2T samples had slightly higher moisture contents. Lee

Table 1. Moisture content, soluble solid content, and pH of *Cheongdobansi* persimmon with different storage treatment

Storage week	Sample	Moisture content (%)	Soluble solid content (°Brix)	pH
0	CONT ¹⁾	81.37±0.19 ²⁾	19.73±0.64	5.79±0.07
2	PNT	81.40±0.34 ^{A3),NS4)}	19.00±0.20 ^{A,NS}	5.52±0.010 ^{A,NS}
	PAT	81.12±0.50 ^{B,NS}	19.03±0.21 ^{C,NS}	5.65±0.02 ^{A,NS}
4	PNT	80.90±0.60 ^{Ab5)}	19.03±0.21 ^{Bb}	5.58±0.01 ^{Ba}
	P2T	79.50±0.10 ^{Aa}	18.87±0.12 ^{Ca}	5.63±0.01 ^{Ab}
	PAT	79.81±0.51 ^{Aa}	19.00±0.00 ^{Cb}	5.63±0.01 ^{Ab}
6	PNT	80.90±0.60 ^{Ab}	18.87±0.12 ^{Ab}	5.84±0.02 ^{Da}
	P2T	79.50±0.10 ^{Aa}	18.53±0.12 ^{Ba}	5.87±0.03 ^{Ba}
	PAT	79.81±0.51 ^{Aa}	18.67±0.12 ^{Bab}	5.87±0.03 ^{Ca}
8	PNT	80.62±0.24 ^{Aa}	18.47±0.12 ^{Ba}	5.87±0.03 ^{Eb}
	P2T	81.34±0.07 ^{Cb}	18.33±0.12 ^{ABa}	5.95±0.02 ^{Cb}
	PAT	81.27±0.19 ^{Bb}	18.33±0.12 ^{Aa}	5.88±0.03 ^{Ca}
10	PNT	80.41±0.73 ^{Aa}	18.43±0.12 ^{Bb}	5.81±0.02 ^{Ca}
	P2T	80.35±0.57 ^{Ba}	18.13±0.12 ^{Aa}	5.94±0.03 ^{Cb}
	PAT	80.96±0.04 ^{Ba}	18.97±0.06 ^{Cc}	5.94±0.03 ^{Bb}

¹⁾CONT, no treatment; PNT, only stored at 0±1°C and RH 30%; P2T, chlorine dioxide was treated at concentration 60 mg/L for 2 weeks and after stored at 0±1°C and RH 30%; PAT, all was stored by chlorine dioxide treatment at concentration 60 mg/L, and stored at 0±1°C and RH 30%.

²⁾Values are mean±standard deviation (n=3).

³⁾A-E Means significantly different between groups of samples by Duncan's multi-range test (A<B<C<D<E) ($p < 0.05$).

⁴⁾NS Means not significantly different between groups of samples on 2 weeks by t-test.

⁵⁾a-c Means significantly different between groups of storage period by Duncan's multi-range test (a<b<c) ($p < 0.05$).

et al. (2011) reported that the moisture content of *Cheongdobansi* immediately after harvest ranges from 80.40–84.70% depending on the harvesting time. Hence, all samples showed a high moisture content overall until week 10 of storage. In general, the moisture content of fresh produce decreases with increasing storage time due to increased transpiration (Park et al., 2004). The resulting skin shrinkage causes the appearance of the fruit to change, which directly affects the sensory quality characteristics of the fruit (Choi et al., 2012). Kim and Hwang (2019) reported that ClO₂ treatment suppresses moisture loss in fruits by reducing respiration rates. They confirmed that continuous ClO₂ treatment throughout storage is more effective than a one-time treatment with ClO₂ gas early in the storage period to maintain the quality of strawberries. Moreover, Ku et al. (2006) demonstrated that a combination of low-temperature and ClO₂ treatment lowered the weight loss of *Agaricus bisporus* during storage by regulating their transpiration and respiration. In addition, Wang et al. (2014) found that extended ClO₂ treatment delays transpiration in plums. Therefore, ClO₂ treatment has a positive effect on maintaining the initial moisture content throughout storage.

3.2. Soluble solids and pH

Table 1 shows the changes in the soluble solids content and pH of the *Cheongdobansi* under various ClO₂ storage conditions. The soluble solids content of fruits generally indicates sweetness and is used as an objective indicator of palatability. The pH not only affects microbial growth but also directly affects the quality and safety of food (Kim et al., 2005). The soluble solids content of the *Cheongdobansi* exhibited no significant differences among the samples or storage periods. Besada et al. (2009) reported that

the soluble solids content of persimmons did not differ significantly between ripening stages. Similarly, Chang et al. (2007) found that the soluble solids content of grapes treated with ClO₂ packaging remained similar from the beginning of storage until week 10 of storage. The pH of the *Cheongdobansi* did not exhibit significant differences overall over the days of storage, although a slight increase was observed with increasing storage period. No trends were observed between the results and the storage treatment method. Ultimately, ClO₂ gas treatment did not affect the pH of *Cheongdobansi*. Similarly, Du et al. (2007) reported that the pH of bell peppers was not affected by the ClO₂ gas concentration during storage.

3.3. Chromaticity and color difference

Table 2 shows the changes in the internal chromaticity (CIE L*, a*, b*) and color difference (ΔE) of the *Cheongdobansi* during storage under different ClO₂ treatment conditions. In terms of lightness (L* value), all samples exhibited a decreasing trend during storage. In particular, PNT exhibited a greater variation in L* value compared to those of the P2T and PAT samples over the storage period. In terms of redness (a* value), there was an overall increasing trend followed by a decreasing trend during storage; however, no significant differences were observed among the treatment groups. In terms of yellowness (b* value), all samples exhibited a steady decreasing trend during storage. The decrease was most pronounced in PNT, which were stored at a low temperature. At week 10, the color difference (ΔE) was 9.97 ± 1.39 for PNT, 7.42 ± 3.05 for P2T, and 5.74 ± 2.12 for PAT, indicating that the greatest color change occurred in the low-temperature-stored sample. This was visually confirmed by the overall darkening of color, as shown in Fig. 1.

Table 2. Color value (CIE L*, a*, b*) and color different (ΔE) of *Cheongdobansi* persimmon with different storage treatments

Storage week	Sample	Color value			ΔE
		L*	a*	b*	
0	CONT ¹⁾	70.67±1.43 ²⁾	13.49±0.05	58.31±2.26	0.00±0.00
2	PNT	66.90±1.33 ^{B3),NS4)}	15.63±1.09 ^{B,NS}	57.74±0.07 ^{C,NS}	3.55±0.80 ^{A,NS}
	PAT	65.37±1.16 ^{A,NS}	15.29±0.58 ^{B,NS}	58.18±0.76 ^{B,NS}	3.58±1.36 ^{A,NS}
4	PNT	69.33±1.40 ^{Cb}	15.43±0.13 ^{ABa}	55.52±2.50 ^{Ca}	4.15±1.31 ^{Aa}
	P2T	66.42±0.71 ^{Ba}	16.85±0.11 ^{Aa}	55.82±1.14 ^{Ba}	4.93±1.98 ^{Aa}
	PAT	66.23±1.59 ^{ABa}	15.11±1.05 ^{Ba}	56.54±0.35 ^{ABa}	4.53±2.51 ^{ABa}
6	PNT	66.13±0.56 ^{Bb}	13.91±3.06 ^{Aa}	51.60±0.37 ^{Ba}	4.98±1.42 ^{Aa}
	P2T	66.37±1.03 ^{Ba}	14.21±0.09 ^{Aa}	55.18±1.10 ^{ABa}	5.18±2.59 ^{Aa}
	PAT	67.61±0.85 ^{Ba}	16.24±0.98 ^{Ba}	54.64±1.15 ^{ABa}	4.82±0.59 ^{ABa}
8	PNT	67.01±1.11 ^{Ba}	14.06±0.87 ^{ABb}	53.64±1.51 ^{Bb}	7.33±2.86 ^{Ba}
	P2T	68.69±0.14 ^{Cb}	11.60±1.15 ^{Aa}	50.29±2.05 ^{Aa}	8.72±1.91 ^{Bb}
	PAT	66.37±0.99 ^{ABa}	11.51±1.17 ^{Aa}	52.57±2.13 ^{Aab}	6.28±1.91 ^{Ba}
10	PNT	61.17±1.60 ^{Aa}	13.88±1.18 ^{Ab}	49.82±1.20 ^{Aa}	9.97±1.39 ^{Cb}
	P2T	64.55±1.36 ^{Ab}	12.24±1.17 ^{Aa}	51.28±2.85 ^{ABa}	7.42±3.05 ^{ABab}
	PAT	66.84±1.02 ^{ABc}	12.59±0.64 ^{Aab}	52.75±1.98 ^{Aa}	5.74±2.12 ^{ABa}

¹⁾CONT, no treatment; PNT, only stored at 0±1°C and RH 30%; P2T, chlorine dioxide was treated at concentration 60 mg/L for 2 weeks and after stored at 0±1°C and RH 30%; PAT, all was stored by chlorine dioxide treatment at concentration 60 mg/L, and stored at 0±1°C and RH 30%.

²⁾Values are mean±standard deviation (n=5).

^{3)A-C}Means significantly different between groups of samples by Duncan’s multi-range test (A<B<C) (p<0.05).

^{4)NS}Means not significantly different between groups of samples on 2 weeks by t-test.

^{5)a-c}Means significantly different between groups of storage period by Duncan’s multi-range test (a<b<c) (p<0.05).

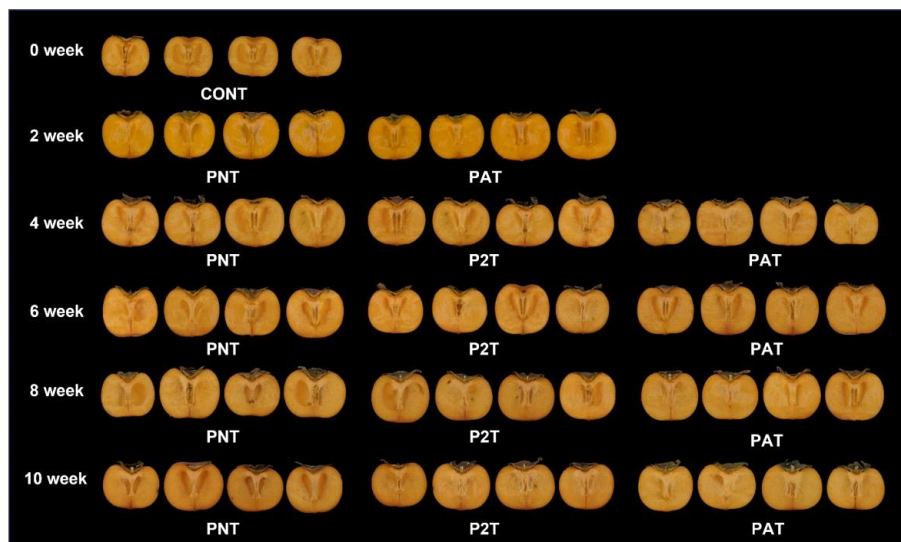


Fig. 1. Photographs of *Cheongdobansi* persimmon with different storage treatments. In the codes listed: CONT, no treatment; PNT, only stored at 0±1°C and RH 30%; P2T, chlorine dioxide was treated at concentration 60 mg/L for 2 weeks and after stored at 0±1°C and RH 30%; PAT, all was stored by chlorine dioxide treatment at concentration 60 mg/L, and stored at 0±1°C and RH 30%.

Moreover, a significant difference was observed between P2T and PAT from week 8 of storage. The internal color of a fruit is an important factor for determining fruit quality throughout ripening (Mendoza and Aguilera, 2004). Indicators of deteriorating appearance during the storage of persimmons include color change and injury. A decrease in lightness and yellowness and an increase in redness are the main causes of color change (Jeong, 2001). Overall, PAT exhibited small color differences throughout storage, and the internal color changes in terms of lightness, redness, and yellowness were smaller compared to those of the other samples. Therefore, it can be inferred that continuous ClO₂ treatment during storage helps maintain the internal color of *Cheongdobansi*. Similar trends were observed in studies by Mahmoud et al. (2007) and Choi et al. (2013).

3.4. Hardness

Fig. 2 illustrates the changes in the hardness of the *Cheongdobansi* under different ClO₂ storage conditions. Hardness, measured using a texture profile analyzer, refers to the force required for the material to yield or deform during the first compression or shear state (Peleg, 2019). Therefore, hardness is considered an index of the mechanical properties of fruits (King and O'Donoghue, 1995). The hardness of the PNT samples decreased significantly with increasing storage period. P2T had a greater hardness than that of PAT until week 6. However, the hardness declined overall after week 8. In other words, hardness decreased in all samples with increasing storage period, although P2T and PAT showed significantly higher hardness values than PNT. Moreover, all samples treated with ClO₂ gas maintained similar hardness values to those of week 4 during storage. During the ripening

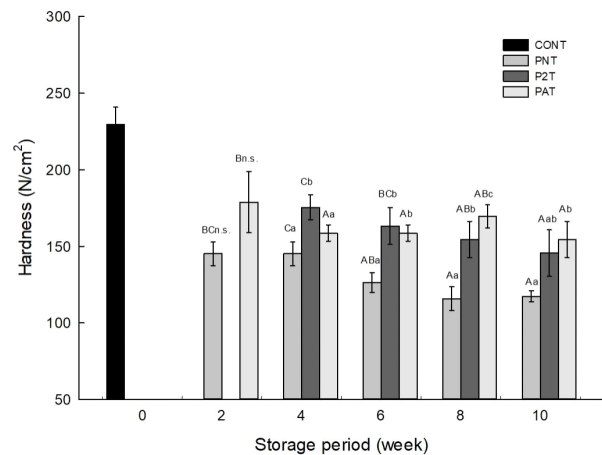


Fig. 2. Hardness of *Cheongdobansi* persimmon with different storage treatments. In the codes listed; CONT, no treatment; PNT, only stored at 0±1°C and RH 30%; P2T, chlorine dioxide was treated at concentration 60 mg/L for 2 weeks and after stored at 0±1°C and RH 30%; PAT, all was stored by chlorine dioxide treatment at concentration 60 mg/L, and stored at 0±1°C and RH 30%; Values are mean±SD (n=5); ^{A-C}Means significantly different between groups of samples by Duncan's multi-range test (A(B)(C) (p<0.05); ^{NS}Means not significantly different between groups of samples on 2 weeks by t-test; ^{a-c}Means significantly different between groups of storage period by Duncan's multi-range test (a(b)(c) (p<0.05).

process, fruits soften owing to the enzymatic breakdown of the cell walls and pectin. Therefore, the ripening process affects the mechanical characteristics of fresh fruits (Chauhan et al., 2006). ClO₂ fumigation delays ripening by inhibiting the activity of factors involved in ethylene biosynthesis, which is the basis for regulating the quality of fresh produce (Guo et al., 2014). Consequently, respiration decreases and softening is delayed, which suppresses alterations in the mechanical properties of *Cheongdobansi* during storage. Therefore, ClO₂ fumigation is expected to have a positive effect on the storage quality of *Cheongdobansi* by preserving its hardness, a major factor in evaluating the storage quality of fresh persimmons (Kim, 1999). Similar results were found in the studies of Choi et al. (2013) and Kang et al. (2015), in which ClO₂ gas treatment maintained the hardness of cherry

tomatoes and paprika.

3.5. Total sugar content

The primary sugar components in astringent persimmons are sucrose and fructose. The glucose content increases throughout ripening as the sucrose is broken down into glucose and fructose by invertase (Jeong et al., 2010). Table 3 shows the changes in the total sugar content during the storage of *Cheongdobansi* under different ClO₂ treatment conditions. Overall, there were no significant differences in the total sugar content

Table 3. Total sugar contents and soluble tannin contents of *Cheongdobansi* persimmon with different storage treatments

Storage week	Sample	Total sugar content (mg/g)	Soluble tannin content (mg/g)
0	CONT ¹⁾	178.86±10.27 ²⁾	5.02±0.05
2	PNT	180.04±8.61 ^{AB3)NS4)}	4.70±0.30 ^{ANS}
	PAT	200.33±15.71 ^{ANS}	4.02±0.13 ^{ANS}
4	PNT	176.97±9.43 ^{Aa5)}	2.96±0.09 ^{Ba}
	P2T	194.67±14.25 ^{Aa}	4.60±0.37 ^{Abc}
	PAT	187.35±18.08 ^{Aa}	4.16±0.69 ^{Ab}
6	PNT	194.43±4.27 ^{BCa}	1.30±0.07 ^{Ca}
	P2T	198.44±5.35 ^{Aa}	3.59±1.23 ^{Bb}
	PAT	195.14±11.33 ^{Aa}	4.57±0.54 ^{Ac}
8	PNT	197.97±4.33 ^{Ca}	1.80±0.18 ^{BCa}
	P2T	189.95±15.36 ^{Aa}	2.84±0.38 ^{Cb}
	PAT	189.71±22.49 ^{Aa}	2.93±0.41 ^{Bb}
10	PNT	194.67±10.81 ^{BCa}	1.69±0.61 ^{Ca}
	P2T	189.00±18.48 ^{Aa}	2.45±0.52 ^{Cb}
	PAT	190.18±24.26 ^{Aa}	3.05±0.36 ^{Bc}

¹⁾CONT, no treatment; PNT, only stored at 0±1°C and RH 30%; P2T, chlorine dioxide was treated at concentration 60 mg/L for 2 weeks and after stored at 0±1°C and RH 30%; PAT, all was stored by chlorine dioxide treatment at concentration 60 mg/L, and stored at 0±1°C and RH 30%.

²⁾Values are mean±standard deviation (n=3).

^{3)A-C}Means significantly different between groups of samples by Duncan's multi-range test (p<0.05) (A<B<C).

^{4)NS}Means not significantly different between groups of samples on 2 weeks by t-test.

^{5)a-c}Means significantly different between groups of storage period by Duncan's multi-range test (a<b<c) (p<0.05).

based on the storage treatment. However, PNT showed a slight increase in the total sugar content with increasing storage time from week 6 of storage. The trends in the soluble solids and total sugar contents were generally similar. ClO₂ treatment has been reported to affect the enzymatic breakdown of cell wall components and suppress fruit softening (Kim et al., 2018). This may be the reason for the increasing trend in total sugar content in the PNT samples, which were stored under low-temperature conditions only. Jiang et al. (2022) reported that ClO₂ treatment suppressed the increase in the total sugar content of grapes during storage.

3.6. Soluble tannin content

Plant-derived tannins are water soluble and coagulate the mucosal proteins on the tongue, resulting in a strong astringent taste that affects palatability (Seo et al., 2000). However, the tannin content gradually decreases as they react with the acetaldehyde produced and accumulated in fruits during the late ripening process. In other words, the astringency of the fruit decreases during storage (Shahkoomahally and Ramezani, 2013). Table 3 shows the changes in the soluble tannin content of *Cheongdobansi* during storage under different ClO₂ treatment conditions. Generally, the samples treated with ClO₂ showed a higher soluble tannin content than those stored at low temperatures only over the days of storage. The PNT sample showed a sharp decrease in tannin content starting from week 6 of storage. The PAT and P2T samples showed a slow decreasing trend during storage; however, no trends between the tannin content and the treatment conditions were observed. ClO₂ treatment can delay the polymerization of tannin molecules by suppressing the ripening and softening of persimmons. However, the PNT samples ripened rapidly, resulting in the

transformation of the tannins from soluble to insoluble, which led to a relatively low soluble tannin content. Sim et al. (2016) reported that the tannin content of a water extract of unripened *Cheongdobansi* was 2.88 mg/g. Therefore, the reason for the high tannin content may be due to differences in the harvesting time (Bian et al., 2015).

3.7. Sensory evaluation

In general, persimmons are harvested in a single harvest for distribution. During this process, they can ripen and become soft owing to respiration and transpiration (Jeong et al., 2010). As persimmons ripen, their internal transparency and sweetness increase, while their mechanical properties deteriorate.

Consequently, decay can occur, which can critically impact the sensory quality characteristics. Therefore, we evaluated the sensory qualities of *Cheongdobansi* stored under different ClO₂ conditions in terms of clearness, weakness, sweet flavor, off-flavor, and overall preference (Table 4). Overall, the clearness increased with storage time, and P2T and PAT treated with ClO₂ exhibited low levels from week 6. This can be observed visually in Fig. 1, in which the transparent area is wider. Kim et al. (2009) confirmed the correlation between tissue texture and clearness in persimmons. A similar trend was observed between the transparency and weakness. Compared to the CONT samples, the weakness mostly increased with increasing storage time,

Table 4. Sensory characteristics of *Cheongdobansi* persimmon with different storage treatments (score: 7-point scale)

Storage week	Sample	Sensory characteristics				
		Degree of clearness	Weakness	Sweet flavor	Off flavor	Overall acceptability
0	CONT ¹⁾	1.93±0.59	1.87±0.83	3.00±0.93	1.20±0.56	5.47±1.06
2	PNT	2.33±0.62 ^{A3),NS4)}	2.07±0.70 ^{A,NS}	4.00±1.31 ^{A,NS}	1.27±0.46 ^{A,NS}	5.00±0.85 ^{BSD5)}
	PAT	2.40±0.74 ^{A,NS}	1.73±0.70 ^{A,NS}	3.53±1.19 ^{A,NS}	1.27±0.46 ^{A,NS}	5.13±1.06 ^{ABSD}
4	PNT	3.20±1.08 ^{Ba6)}	3.67±1.29 ^{ABb}	3.87±1.36 ^{Aa}	1.13±0.35 ^{Aa}	5.00±1.13 ^{Ba}
	P2T	3.20±1.15 ^{Aa}	2.87±1.46 ^{Aab}	3.53±1.64 ^{ABa}	1.27±0.46 ^{Aa}	5.00±1.00 ^{Aa}
	PAT	2.93±1.33 ^{ABa}	2.67±0.90 ^{Ba}	3.60±1.30 ^{Ba}	1.40±0.51 ^{Aa}	5.00±0.85 ^{ABa}
6	PNT	5.20±1.37 ^{Cb}	5.60±5.57 ^{Cb}	4.27±1.16 ^{Aa}	1.20±0.41 ^{Aa}	4.33±0.72 ^{ABa}
	P2T	3.20±1.21 ^{Aa}	3.87±1.60 ^{Aab}	4.47±1.64 ^{Ba}	1.33±0.49 ^{Aa}	4.80±1.08 ^{Aa}
	PAT	3.13±1.19 ^{ABa}	2.60±1.64 ^{ABa}	4.40±1.06 ^{Ba}	1.20±0.41 ^{Aa}	4.73±1.03 ^{Aa}
8	PNT	4.60±1.35 ^{Ca}	4.13±1.68 ^{ABa}	4.53±1.19 ^{Aa}	1.33±0.62 ^{Aa}	3.80±1.37 ^{Aa}
	P2T	3.67±1.76 ^{Aa}	3.47±1.19 ^{Aa}	4.20±1.52 ^{ABa}	1.27±0.59 ^{Aa}	4.27±1.39 ^{Aa}
	PAT	3.53±1.30 ^{Ba}	3.67±1.35 ^{Ca}	4.40±1.18 ^{Ba}	1.13±0.35 ^{Aa}	5.53±0.74 ^{Bb}
10	PNT	5.00±0.85 ^{Cb}	4.93±1.28 ^{Bc}	4.60±1.64 ^{Ab}	1.13±0.35 ^{Aa}	3.80±1.08 ^{Aa}
	P2T	3.67±0.72 ^{Aa}	3.60±1.40 ^{Ab}	3.20±1.15 ^{Aa}	1.13±0.35 ^{Aa}	4.20±1.37 ^{Aa}
	PAT	3.20±1.42 ^{ABa}	1.93±0.88 ^{ABa}	2.73±0.80 ^{Aa}	1.20±0.56 ^{Aa}	4.60±1.18 ^{Aa}

¹⁾CONT, no treatment; PNT, only stored at 0±1°C and RH 30%; P2T, chlorine dioxide was treated at concentration 60 mg/L for 2 weeks and after stored at 0±1°C and RH 30%; PAT, all was stored by chlorine dioxide treatment at concentration 60 mg/L, and stored at 0±1°C and RH 30%.

²⁾Values are mean±standard deviation (n=15).

^{3)A-C}Means significantly different between groups of samples by Duncan's multi-range test (A(B(C) (p<0.05).

^{4)NS}Means not significantly different between groups of samples on 2 weeks by t-test.

^{5)SD}Means significantly different between groups of samples on 2 weeks by t-test.

^{6)a-c}Means significantly different between groups of storage period by Duncan's multi-range test (a(b(c) (p<0.05).

which was inversely proportional to the hardness trend. The samples treated with ClO₂ exhibited relatively low weakness compared to PNT. In particular, the value was significantly lower overall in PAT than in P2T. This indicates that ClO₂ treatment can delay changes in the mechanical properties of *Cheongdobansi* over the storage period. According to Cha et al. (2019), the sweetness of a fruit increases as the fruit ripens, and the sweet aroma also increases overall throughout the storage period. In particular, a significant difference was observed between the samples stored under different conditions at week 10. The common causes of off-flavor are environmental contaminants, microbial growth, lipid oxidation, or enzymatic degradation. Among these causes, spoilage from microbial growth has the greatest impact on the flavor of the sample (Wikes et al., 2000). However, the off-flavor values in all samples were similar to that of CONT, and no significant differences were observed based on the storage period or condition. This suggests that no microbial decay or accumulation of foreign substances occurred due to ClO₂ treatment, and the flavor characteristics were not affected. The overall preference score for all the stored samples was lower than that for CONT, and it tended to decline as the storage period increased. However, the preference was higher for the ClO₂-treated samples than for PNT over the storage period. In particular, the participants showed a high preference for PAT until week 10 of storage. In conclusion, ClO₂ treatment improved the sensory qualities of *Cheongdobansi* during storage, with PAT treatment in particular maintaining good storage quality.

4. Summary

In this study, we comparatively analyzed changes

in the quality of *Cheongdobansi* after varying durations of ClO₂ gas fumigation treatment under low-temperature storage conditions. Treating *Cheongdobansi* with ClO₂ gas helped maintain the moisture content, color, hardness, soluble tannin content, and sensory characteristics. However, ClO₂ gas treatment did not affect the soluble solids content, pH, and total sugar content. Regarding the duration of ClO₂ gas treatment, the quality of the PAT samples, which were continuously treated with ClO₂ gas, was better preserved after 6-8 weeks of storage. In particular, their mechanical properties, namely hardness and weakness in sensory evaluation, were the most influenced by ClO₂ gas treatment. These findings suggest that a packaging system that continuously releases ClO₂ gas during distribution could contribute to preserving the freshness of *Cheongdobansi*. Additional research is needed to shed light on the correlation between stoma opening and closing and ripening based on the duration of ClO₂ gas treatment. Based on these findings, identifying the optimal ClO₂ gas treatment duration for each type of fruit could enhance the distribution and storability of fruits and vegetables.

Conflict of interests

The authors declare no potential conflicts of interest.

Author contributions

Conceptualization: Kim J, Moon KD. Methodology: Kim J, Kim JS. Formal analysis: Kim JS, Kim M, Kim JH, Kim I, Nam I. Validation: Kim J, Kim JS. Writing - original draft: Kim J. Writing - review & editing: Kim J, Kim JS, Kim M, Kim JK, Moon KD.

Ethics approval

The sensory evaluation of this research was safely

carried out with the approval of exemption (No. KNU-2022-0488) from the IRB of Kyungpook National University.

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